

AN ABSTRACT OF THE THESIS OF

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Title: A Microcomputer Software Package for Simulation of Non-Ideal  
Aqueous Electrolyte Systems at Equilibrium

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The non-ideal aqueous electrolyte simulator (NAELS) is composed of three major parts: a Newton-Raphson non-linear optimization program written by Weare, et al (1987); an activity coefficient subroutine for non-ideal electrolyte systems based on Pitzer's model; and an extensive, user expandable database. It is robust, stable, and requires neither thermodynamic data nor initial guesses as input. NAELS provides very good estimates of equilibrium speciation and solubility in concentrated electrolyte systems. NAELS was assembled as a technical utility package for use on IBM-compatable microcomputers.

**A Microcomputer Software Package for Simulation of Non-Ideal  
Aqueous Electrolyte Systems at Equilibrium**

**by**

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## A Microcomputer Software Package for Simulation of Non-Ideal Aqueous Electrolyte Systems at Equilibrium

### INTRODUCTION

If one is faced with a simple (single salt) aqueous solubility problem, one could simply look up a value in a reference book or measure the solubility experimentally. When there are a number of different salts present in the system, the problem of predicting the phase assemblage becomes disproportionately more complex due to the interdependent nature of the chemical species involved. Each specie affects the behavior of all of the other species present resulting in a highly non-linear problem. This interdependence is weak enough to be ignored at low concentrations (i.e. when the total ionic strength is less than about 0.1 moles per kilogram). However, when concentrations are sufficiently high such that solid phases precipitate, then the interdependence of ion activities must be taken into consideration if any reasonable predictions are to be made. The problem is further compounded by the fact that the equilibrium concentrations of the various species in any system will usually vary by several orders of magnitude. This type of problem is ideally suited for a computer.

Solving highly non-ideal equilibrium problems with a computer requires three things: 1) A mathematical formulation of the equilibrium problem and a suitable algorithm to solve it. 2) A method of estimating the activities of the solution species which takes into account their interdependent nature. 3) The necessary physical, chemical, and thermodynamic data of the species in the system as well as any empirical data required by the activity model.

The overall objective of this thesis is to provide a microcomputer software package for simulating non-ideal aqueous electrolyte systems at equilibrium. This project has been guided by the following specific objectives:

- 1) To assemble a reliable yet robust aqueous equilibrium simulator that can handle concentrated (highly non-ideal) multicomponent electrolyte systems.
- 2) It should provide good estimates of equilibrium concentrations and solid phase speciation at temperatures up to 100°C.

- 3) All necessary data should be read from a database that is easily expanded to meet the user's individual needs.
- 4) The program should run on a IBM or compatable microcomputer as this is what is typically readily available to the average engineer.

A number of computer programs are available for calculating chemical equilibria in electrolyte systems. These are reviewed in the following section of this thesis. All of these programs failed to meet the requirements set forth in at least one aspect.

The result of this thesis is NAEELS (Nonideal Aqueous Electrolyte Systems). NAEELS is a microcomputer software package that calculates the equilibrium composition of aqueous electrolyte systems. In the most general case it can solve aqueous systems with a gas phase and several solid phases present. The soluble gaseous species, however, must be at constant partial pressure (i.e. an open system). It is currently dimensioned to handle a maximum of 18 cations, 18 anions, 8 neutral aqueous species, and 96 total species in all phases combined. It uses a modified Newton-Raphson algorithm to find the phase assemblage at which the total free energy of the system is at a minimum. This algorithm, developed by Weare, et al. (1987), is combined with a user expandable data base and various activity coefficient estimation methods to form the program NAEELS. The development of NAEELS and the content of its components are described in this thesis, and examples of its application to solubility problems is demonstrated.

## REVIEW OF EXISTING ELECTROLYTE EQUILIBRIUM PROGRAMS

There are a number of existing computer programs for solving equilibrium problems. None could be found that meet all of the objectives of this project. Some of the more widely used ones are summarized here.

### WATEQ

The WATEQ program was developed by the U.S. Geological Survey in order to predict both trace and major species concentrations and solid formation. It is limited to 25°C and low to moderate ionic strength (0-3 molal). It uses an extended Debye-Huckel activity coefficient model. The reader is referred to Zemaitis (1986) for the specifics of the Debye-Huckel model. It is limited to low ionic strengths. It is written in FORTRAN and will run on a microcomputer.

### REDEQL

This program was developed by the Environmental Protection Agency. It is based on the Davies equation and is intended for water quality use. The program includes an extensive databank for metal-ligand complexes. Based upon the Newton-Raphson technique, it allows for solid formation. It is limited to moderate ionic strengths (0-1 molal) and 25°C. It is also microcomputer compatible.

### ECES

This is an extensive general purpose vapor-liquid-solid prediction program for aqueous electrolyte systems developed by OLI Systems Inc., 52 South Street, Morristown, NJ 07960. It uses an optional Bromley or Pitzer model to estimate activity coefficients. It can model systems from 0-200 atm., 0-250°C, and 0-30 molal ionic strength. It requires a VAX, UNIVAC, or CDC/NOS computer to run. It has an extensive private database.

### MICROQL

Written in the BASIC computer language, MICROQL has no database and no model for activity coefficients. The user inputs constant activity coefficients. It will simulate systems at any temperature as the user inputs all thermodynamic data. It was written for microcomputer use.

### SOLGASMIX

Created at the University of Umea, Sweden (1974), SOLGASMIX is a general gas-liquid-solid equilibrium program. There is no thermodynamic database and the user must also supply the activity coefficient model for the type of problem under consideration. The program is proprietary.

### GIBBS

Created by R. Gautem for his PhD thesis at the Dept. of Chemical Engineering of the University of Pennsylvania (1979). Written in the form of a FORTRAN subroutine, GIBBS computes physical and chemical equilibrium. The user must supply the thermodynamic data as well as the activity coefficient equations. This program is public domain and can be run on a microcomputer.

### ISIS

Created by B. Kelly for his M.S. Thesis at Oregon State University (1988), ISIS uses Pitzer's model for estimating activity coefficients. It also includes a database. ISIS was written for use on a CDC Cyber computer and had limited success when used on a microcomputer.

There are several considerations when evaluating an equilibrium program. Computers solve problems iteratively. The stability and robustness of the algorithm determine whether or not the program will converge to a solution and how long it will take. In all of the above programs (except possibly ECES) the activity coefficients are treated as constants and then revised at each step of the iterative process. In highly non-ideal electrolyte systems the activity coefficients can change faster than the mole numbers as the iterations proceed. Since the activity is the product of the activity coefficient and the mole number, the algorithm can cycle or diverge rather than converge.

Another consideration is the activity model itself. In dilute solutions, the simpler models (i.e. extended Debye-Huckel, Davies, etc) work fine. However, when electrolyte concentrations are high enough such that solids precipitate, then the total ionic strength will be moderate to high in most systems. This is especially true in multicomponent systems. Pitzer's model provides good estimates at high ionic strengths and at temperatures other than

25°C. and the empirical parameters for common species can be found in the literature. Weare, (1987) has developed an extended Pitzer model which takes into account the interactions between charged and neutral species as well as unlike neutrals. Unfortunately, only a very few of the empirical parameters have been measured at this time. Each unique doublet or triplet of species has associated with it an empirical parameter. Pitzer considered interactions between only charged species. Weare choose to include neutral species. The reader can estimate the number of combinations possible.

The presence of a user expandable database is of enormous convenience when using an equilibrium simulator if one is dealing with many different systems. If a database is not present, then all of the necessary thermodynamic and mass balance information, along with the empirical parameters for the activity coefficient model must be read from an input file in matrix form. If a species is added or removed, then the entire matrix must be restructured, redimensioned, and reentered.

## THE SOLUBILITY PROBLEM

We know from thermodynamics that any system will seek equilibrium at its lowest energy state. In a chemical system at equilibrium, the Gibb's free energy is at a minimum. The general expression for the free energy of species  $i$  is written as:

$$G_i = \mu_i n_i$$

where  $\mu_i$  is the chemical potential and is simply the Gibb's free energy for species  $i$  on a molar basis.  $n_i$  is the number of moles of species  $i$ . The total free energy of the system is just the summation of the free energies of all of the species in all of the phases. At equilibrium this sum will be at a minimum:

$$\text{Minimize } G = \sum_i \mu_i n_i \text{ for all species in all phases.}$$

The problem is subject to the constraints of bulk composition, mass balance, electroneutrality, and non-negativity (ie  $n_i \geq 0$ ). As an optimization problem it would be written as follows:

$$\text{minimize } G = \sum_{j=1}^{n^t} \mu_j n_j$$

subject to 
$$\sum_{j=1}^{n^t} A_{ji} n_j = b_i \quad i=1, m_c \quad (\text{mass balance})$$

$$\sum_{j \text{ in } s} z_j n_j = 0 \quad s=1, e_c \quad (\text{charge balance})$$

$$n_j \geq 0 \quad \text{for all } j \quad (\text{non-negativity})$$

where  $G$  is the Gibbs free energy

$\mu_j$  is the chemical potential of species  $j$

$n_j$  is the number of moles of species  $j$

$n^t$  is the total number of species in the system

$m_c$  is the number of system elements

$A_{ji}$  is the number of moles of element i in one mole of species j

$z_j$  is the charge of the  $j^{th}$  species in electrolyte solution phase s

$e_c$  is the number of electrolyte solution phases

$b_i$  is the number of moles of element i

To solve the problem the method of Lagrangian multipliers is used to incorporate the constraints into an unconstrained objective function. The reader is referred to Weare, et al., (1987) for a complete description of the algorithm used to find the solution phase assemblage. It is outlined here. The unconstrained Lagrangian function is written as

$$L(\vec{n}, \vec{t}, \vec{\kappa}, \vec{\eta}, \vec{\omega}) = \sum_{j=1}^{n^t} \mu_j n_j - \sum_{i=1}^c \kappa_i \sum_{j=1}^{n^t} (A_{ji} n_j - b_i) - \sum_{i=1}^e \eta_i \sum_{j \text{ in } i} z_j n_j \\ - \sum_{j=1}^{n^t} \omega_j (n_j - t_j^2)$$

where  $\kappa_i$  is the Lagrangian multiplier for mass balance constraint i

$\eta_i$  is the Lagrangian multiplier for charge balance constraint i

$\omega_j$  is the Lagrangian multiplier for the  $j^{th}$  non-negativity constraint

$t_j^2$  is the slack variable for the inequality constraint

The necessary conditions for a local minimum are obtained by differentiating L with respect to all variables and setting the derivatives equal to zero:

$$\frac{\partial L}{\partial n_k} = 0 = \mu_k - \sum_{i=1}^c A_{ki} \kappa_i - \omega_k \text{ for species k in non-electrolyte phase}$$

$$\frac{\partial L}{\partial n_k} = 0 = \mu_k - \sum_{i=1}^c A_{ki} \kappa_i - \eta_s z_k - \omega_k \text{ for species k in electrolyte phase s}$$

$$\frac{\partial L}{\partial \kappa_k} = 0 = \sum_{j=1}^{n^t} (A_{jk} n_j - b_k) \quad k=1, c$$

$$\frac{\partial L}{\partial \eta_k} = 0 = \sum_{j \text{ in } k} z_j n_j \quad k=1, e$$

$$\frac{\partial L}{\partial t_k} = 0 = 2\omega_k t_k \quad k=1, n^t$$

$$\frac{\partial L}{\partial \omega_k} = 0 = n_k - t_k^2 \quad k=1, n^t$$

The algorithm begins with a feasible point and tests for a local minimum. If a minimum is not present, then a Newton-Raphson step is taken and tests are performed to determine if the addition or removal of a phase will further decrease the system free energy. When a local minimum has been reached and addition or removal of a phase results in an increase in free energy, then a global minimum has been reached.

## ACTIVITY COEFFICIENT ESTIMATION

In order to carry out the minimization procedure we must have an expression for the chemical potential. The expression for the chemical potential of species  $i$  is written as

$$\mu_i = \mu_i^{\circ} + RT\ln(a_i) \quad (1)$$

where  $\mu_i^{\circ}$  is the standard chemical potential for species  $i$ .  $R$  is the ideal gas constant and  $T$  is absolute temperature.  $a_i$  is the activity of species  $i$ . For solids (and gases at constant partial pressure) the activity is taken as unity. The chemical potential for solids reduces to  $\mu_i = \mu_i^{\circ}$ . For solution species, activity is expressed as the product of the activity coefficient,  $\gamma$ , and the molality,  $m$ :

$$a_i = \gamma_i m_i \quad (2)$$

Combining equations 1 and 2 and rearranging yields

$$\frac{\mu_i}{RT} = \frac{\mu_i^{\circ}}{RT} + \ln(\gamma_i) + \ln(m_i) \quad (3)$$

In an aqueous ideal solution phase the activity coefficients are taken as unity. The free energy function then becomes

$$\frac{\mu_i}{RT} = \frac{\mu_i^{\circ}}{RT} + \ln(m_i) \quad (4)$$

A relatively simple model for activity coefficients is the Davies (1962) equation. It is valid for  $I < 0.5$  mol/liter. Equation 5 is used to calculate activity coefficients for charged species only; neutral species are assigned unit activity.

$$\ln \gamma_k = -3A^{\phi} Z_k^2 \left[ \frac{\sqrt{I}}{1+\sqrt{I}} - 0.3I \right] \quad (5)$$

$$I = \frac{1}{2} \sum_j m_j z_j^2 \quad \text{ionic strength} \quad (6)$$

$$A^{\phi} = \frac{1}{3} (2\pi N_0 d_w / 1000)^{1/2} (e^2 / DkT)^{3/2} \quad (7)$$

$Z_i$  is the charge of species  $k$ ,  $I$  is the ionic strength, and  $A^\phi$  is the Debye-Hückel constant for osmotic coefficients. In spite of its appearance,  $A^\phi$  is a function of temperature only. Beyer and Staples (1986) provide a thorough discussion of  $A^\phi$  as well as a tabulation of  $A^\phi$  versus  $T$  from 0 to 350°C. This data was used to fit the polynomial  $A^\phi(T)$  in the database.

Pitzer's model, though rather complex, provides the best estimates of solubility in multicomponent systems. A brief summary of the equations used is presented here for convenience. This model contains many measured values referred to as 'ion interaction parameters'. A familiarity with the equations below simplifies the task of searching the literature for ion interaction parameters not already contained in the database. For a complete derivation of the model refer to Pitzer and coworkers (1973-1975). In the equations that follow the subscripts M, c, and c' refer to cations while X, a, and a' refer to anions. The interaction parameters will be labeled as such.

$$\ln \gamma_M = z_M^2 F + \sum_a m_a (2B_{Ma} + ZC_{Ma}) + \sum_c m_c (2\Phi_{Mc} + \sum_a m_a \psi_{Mca}) + \sum_{a \neq a'} m_a m_{a'} \psi_{aa'M} + |z_M| \sum_c \sum_a m_c m_a C_{ca} \quad (8a)$$

$$\ln \gamma_X = z_X^2 F + \sum_c m_c (2B_{cX} + ZC_{cX}) + \sum_a m_a (2\Phi_{xa} + \sum_c m_c \psi_{xac}) + \sum_{c \neq c'} m_c m_{c'} \psi_{cc'X} + |z_X| \sum_c \sum_a m_c m_a C_{ca} \quad (8b)$$

$$F = -A^\phi \left( \frac{\sqrt{I}}{1+b\sqrt{I}} + \frac{2}{b} \ln (1+b\sqrt{I}) \right) + \sum_c \sum_a m_c m_a B'_{ca} + \sum_{c \neq c'} m_c m_{c'} \Phi'_{cc'} + \sum_{a \neq a'} m_a m_{a'} \Phi'_{aa'} \quad (9)$$

$b$  is an empirical parameter equal to 1.2.  $\psi_{cc'a}$  and  $\psi_{caa'}$  are ternary mixing parameters and exist for unlike cation-cation-anion and cation-anion-anion triplets.

$$Z = \sum_i m_i |z_i| \quad (10)$$

$$C_{MX} = C_{MX}^\phi / 2 \sqrt{|z_M z_X|} \quad (11)$$

$$B_{MX} = \beta_{MX}^0 + \beta_{MX}^1 g(\alpha_1 \sqrt{I}) + \{\beta_{MX}^2 g(\alpha_2 \sqrt{I})\} \quad (12a)$$

$$B'_{MX} = \beta_{MX}^1 g'(\alpha_1 \sqrt{I})/I + \{\beta_{MX}^2 g'(\alpha_2 \sqrt{I})/I\} \quad (12b)$$

$\beta_{MX}^0$ ,  $\beta_{MX}^1$ ,  $\beta_{MX}^2$ , and  $C_{MX}^*$  are single salt mixing parameters. They exist for each cation-anion combination. However  $\beta_{MX}^2$  is only defined for salts in which both ions are divalent (ie  $\text{CaSO}_4$ ).  $\alpha_1=1.4$  and  $\alpha_2=12.0$  for 2-2 electrolytes in equations 12. For all other charge combinations,  $\alpha_1=2.0$  and the bracketed terms do not exist. The functions  $g$  and  $g'$  are defined below with  $x = \alpha \sqrt{I}$ .

$$g(x) = 2[1 - (1+x)e^{-x}]/x^2 \quad (13a)$$

$$g'(x) = -2[1 - (1+x+\frac{1}{2}x^2)e^{-x}]/x^2 \quad (13b)$$

$$\Phi_{ij} = \theta_{ij} + {}^e\theta_{ij}(I) \quad (14a)$$

$$\Phi'_{ij} = {}^e\theta'_{ij}(I) \quad (14b)$$

$${}^e\theta_{ij}(I) = \frac{z_i z_j}{4I} \left[ J0(X_{ij}) - \frac{1}{2} J0(X_{ii}) - \frac{1}{2} J0(X_{jj}) \right] \quad (15a)$$

$${}^e\theta'_{ij}(I) = \frac{z_i z_j}{8I^2} \left[ J1(X_{ij}) - \frac{1}{2} J1(X_{ii}) - \frac{1}{2} J1(X_{jj}) \right] - \frac{{}^e\theta_{ij}}{I} \quad (15b)$$

$$X_{ij} = z_i z_j A^* \sqrt{I}$$

$\theta_{ij}$  is a parameter that accounts for interactions between ions of like sign. It exists for unlike cation-cation and anion-anion pairs.  ${}^e\theta_{ij}(I)$  and  ${}^e\theta'_{ij}(I)$  are zero when  $z_i=z_j$  and are functions of ionic strength, ion pair type, and temperature only. Equations 15 are for cations. Similar equations exist for unlike anion pairs. The expressions for  $J0(X)$  and  $J1(X)$  are given by Pitzer as

$$J0(X) = \frac{X}{4} - 1 + \frac{1}{X} \int_0^\infty \left[ 1 - e^{(-\frac{X}{Y} e^{-Y})} \right] Y^2 dY \quad (16a)$$

$$J1(X) = \frac{X}{4} - \frac{1}{X} \int_0^\infty \left[ 1 - (1 + \frac{X}{Y} e^{-Y}) e^{(-\frac{X}{Y} e^{-Y})} \right] Y^2 dY \quad (16b)$$

NAELS uses a Chebychev expansion to evaluate the above integrals numerically.

In most equilibrium simulators the activity coefficients are treated as constants from one step to the next and then recalculated based on the new

concentration values. This works as long as the change in activity coefficients from one step to the next is small compared to the change in concentrations. As discussed previously, in highly non-ideal systems the activity coefficients can change faster than the concentrations from step to step. It makes little sense, then, to go to the trouble of using sophisticated algorithms to update concentrations from step to step only to multiply them by ever changing 'constants'. NAEIS treats activity coefficients as variable functions of mole numbers and therefore includes the derivatives of activity coefficients with respect to mole numbers in the algorithm.

## DATABASE

The database contains all of the necessary thermodynamic data and molecular formulas to describe each chemical species. It is divided in three sections: Mass balance and chemical potential, Pitzer betas, and Pitzer theta-psis. Because of its size, the entire database is not listed in this thesis. Portions of each section are shown in Figures 1, 2, and 3. All of the data found in the database was taken from the periodic table, Wagman, et.al. (1982), Woods and Garrels, (1987), Frederick and Kim (1988), Pitzer (1979), and Weare (1987). The dimensionless free energy values for the species indicated by an asterick in Appendix B were calculated from solubility data found in Linke (1965), and Seidell (1935).

The first column in the mass balance section contains the species identification number. Water is assigned number 99. Cations, begining with  $H^+$ , are assigned numbers 100 through 199. Anions, begining with  $OH^-$ , are assigned numbers 200 through 299. Neutral solution species are numbered 300 through 399, and pure species (solids and gasses) are numbered from 400 onward. The next column is a character field 24 spaces wide containing the name or chemical formula of the species. Next is a 2 space integer field for the charge of the species. This is left blank for solid species. Following this is 4 space real number field followed by a 3 space integer field. There are 7 pairs of these fields. The integer fields contain atomic numbers of elements as found on the periodic table. The real numbers are the molar amounts found in a given species. For example, specie 215 is  $H_2PO_4^-$ . It carries a charge of -1. It is made up of 1 atom of element 15 (phosphorous), 4 atoms of element 8 (oxygen), and 2 atoms of element 1 (hydrogen). The second line of each species entry contains a 24 space character field followed by the coefficients of a fourth order polynomial. The character variable contains the temperature range (in  $^{\circ}C$ ) over which the polynomial is valid as well as a flag indicating the source of the data used to generate the polynomial. The polynomial relates the dimensionless standard chemical potential (from Equation 3) to temperature and has the following form:

$$\frac{\mu^{\circ}}{RT} = C_0 + C_1(T - T_{ref}) + C_2(T - T_{ref})^2 + C_3(T - T_{ref})^3 + C_4(T - T_{ref})^4 \quad (17)$$

where R is the gas constant, T is in kelvin, and  $T_{ref}=298.15$  K. This form was chosen for convenience. If  $\mu$  is known only at 25°C then  $C_0$  will be the only non-zero coefficient. The chemical potential is defined conceptually as follows:

$$\mu = H - TS \quad (18)$$

The difference in chemical potential between two temperatures at constant pressure would then be

$$\mu_2 - \mu_1 = H_2 - H_1 - (T_2 S_2 - T_1 S_1) \quad (19)$$

but  $H_2 - H_1 = \int_1^2 C_p dT$  and  $S_2 = S_1 + \int_1^2 \frac{C_p}{T} dT$  (20)

If  $\mu_1$  is arbitrarily set equal to the standard free energy of formation at 25°C, and  $S_1$  is the standard entropy at 25°C; then equations 19 and 20 can be integrated, substituted, and rearranged to yield:

$$\mu_2^\circ = \Delta G_f^\circ + \overline{C_p}_{t_0}^{t_2} \Delta T - t_2 \overline{C_p}_{t_0}^{t_2} \ln(t_2/t_0) - S_{25}^\circ \Delta T \quad (21)$$

where  $t_0=25$ °C,  $\Delta T=(t_2-t_0)$ , and  $\overline{C_p}_{t_0}^{t_2}$  is the average heat capacity between  $t_0$  and  $t_2$ . If  $C_p(T)$  data are available, then the integrals in equation 20 can be carried out numerically. Heat capacity equations are published in the literature for many compounds. If the heat capacity of a solid is known only at 25°C, then as an approximation it can be assumed constant over a small temperature range. For solution species, heat capacities change with temperature considerably but are typically available only at 25°C if at all. Appendix C outlines the method of estimating  $C_p(T)$  for solution species that was used in this work. Equation 21 should be used only for solids and only in the absence of experimental solubility data. The method of using solubility data to generate chemical potentials for solid (or gaseous) species is discussed in the section on sources of error. This is the preferred method for these species.

It is important to note that  $\mu_2^\circ$  is not, in general, equivalent to the standard free energy of formation at  $t_2$ . The difference between the two is the free energy required to raise the temperature of the elements in the formation reaction from  $t_0$  to  $t_2$ . Equilibrium calculations are based on

differences in chemical potential between reactants and products in a balanced equilibrium reaction. The aforementioned difference between  $\mu_2^\circ$  and  $\Delta G_f$  at  $t_2$  is the same on both sides of a balanced equilibrium reaction and drops out when  $\Delta G_{rxn}$  is calculated. Therefore,  $\Delta G_{rxn}$  is identical to  $\Delta\mu_{2,rxn}^\circ$ . The extra work of calculating  $\Delta G_f$  as a function of temperature is unnecessary.

Figure 2 shows the second section of the database which contains the Pitzer beta values. This section is preceded by polynomial coefficients for the function  $A^\phi(T)$ , and values for the Pitzer alphas. Each line corresponds to a cation-anion combination referenced by species identification numbers in the left hand columns. The column for  $\beta^2$  is blank for all except 2-2 electrolytes.

The last section, shown in Figure 3, is for Pitzer theta and psi values. In the first half of this section a theta will exist for each unlike cation-cation pair and a psi will exist for each unlike cation-cation-anion triple. The theta will appear on the same line as the two species I.D. numbers. Starting on the following line, the psis will be listed, one for each anion in the order they appear in the database. For example theta for  $H^+ - Mg^{+2}$  is 0.0891 and psi for  $H^+ - Mg^{+2} - HSO_4^-$  is -0.0178. The other half of this section is a completely analogous list of anion-anion theta's and cation-anion-anion psi's. Adding data to the Pitzer section involves more than simply adding lines to the end of a list. The arrangement should be studied carefully before any attempt is made to build onto it.

FIGURE 1

Selected portions of mass balance section of database  
 FORMAT(I4, A24, I2, 7(F4.3, I3) /14X, 5E13.5) aqueous  
 FORMAT(I4, A24, 2X, 7(F4.3, I3) /14X, 5E13.5) solids

99 H2O	0	2.	1	1.	8			
	-9.56611E+1	2.91704E-1	-9.82177E-4	2.57158E-6	-3.53354E-9			
100 H+	+1	1.	1					
25-200	0.00000E+0	4.35345E-3	-1.27810E-4	2.85164E-7	-3.67000E-10			
101 Na+	+1	1.	11					
25-200	-1.05656E+2	3.22011E-1	-1.13238E-3	2.89251E-6	-3.93920E-9			
102 K+	+1	1.	19					
25-200	-1.14275E+2	3.24604E-1	-1.13209E-3	2.90867E-6	-3.97770E-9			
103 Mg++	+2	1.	12					
25-200	-1.83472E+2	7.05555E-1	-2.40987E-3	6.13515E-6	-8.33210E-9			
104 MgOH+	+1	1.	12	1.	8	1.	1	
25	-2.52822E+2							
:								
112 Mn++	+2	1.	25					
25-200	-9.20184E+1	3.60767E-1	-1.29107E-3	3.26602E-6	-4.42840E-9			
113 MnOH+	+1	1.	25	1.	8	1.	1	
25-200	-1.63382E+2	5.60973E-1	-1.91448E-3	4.88671E-6	-6.63760E-9			
200 OH-	-1	1.	8	1.	1			
25-200	-6.34344E+1	2.14183E-1	-5.44900E-4	1.45959E-6	-2.02110E-9			
201 Cl-	-1	1.	17					
25-200	-5.29394E+1	1.38462E-1	-3.06790E-4	8.53212E-7	-1.19710E-9			
202 CO3--	-2	1.	6	3.	8			
25-200	-2.12926E+2	7.38806E-1	-2.04005E-3	5.37846E-6	-7.40120E-9			
:								
215 H2PO4-	-1	1.	15	4.	8	2.	1	
25-200	-4.55971E+2	1.46582E+0	-4.62432E-3	1.19862E-5	-1.64150E-8			
216 VO3-	-1	1.	23	3.	8			
25-200	-3.16115E+2	1.02330E+0	-3.05737E-3	7.99006E-6	-1.09820E-8			
300 H2CO3(aq)		0.3.	8	1.	6	2.	1	
25	-2.51410E+2							
301 NaHCO3(aq)	0	1.	11	3.	8	1.	6	1.
25	-3.42461E+2							
302 MgCO3(aq)	0	1.	12	3.	8	1.	6	
25	-4.04426E+2							
:								
307 H3PO4(aq)	0	1.	15	4.	8	3.	1	
25	-4.60945E+2							
400 SiO2		1.	14	2.	8			
25-200*	-3.43184E+2	1.12871E+0	-3.64467E-3	9.43049E-6	-1.29007E-8			
401 Al2O3		2.	13	3.	8			
25-200*	-6.38322E+2	2.11400E+0	-6.82365E-3	1.76541E-5	-2.41494E-8			
402 Al(OH)3		1.	13	3.	8	3.	1	
25	-4.65659E+2							
403 AlCl3		1.	13	3.	17			
25-200*	-2.53667E+2	8.03667E-1	-2.63307E-3	6.84154E-6	-9.37387E-9			
404 AlCl3*6H2O		1.	13	3.	17	6.	8	12.
25-200*	-9.12159E+2	2.92210E+0	-9.54994E-3	2.47961E-5	-3.39650E-8			
405 Al2(SO4)3		2.	13	3.	16	12.	8	
25-200*	-1.25056E+3	4.08540E+0	-1.32543E-2	3.43418E-5	-4.70031E-8			

FIGURE 2  
Pitzer betas, sample section of database  
FORMAT( I3, 1X, I3, 3X, 4F14.8)

```

:
486 KA13Si3O10(OH)2      1. 19  3. 14  3. 13 12.  8  2.  1
   25-200*    -2.26251E+3  7.44237E+0  -2.40404E-2  6.22104E-5  -2.51059E-8
487 KH2PO4      1. 19  1. 15  4.  8  2.  1
   25-200*    -5.71174E+2  1.85567E+0  -6.01954E-3  1.55961E-5  -2.13458E-8
488 2Na2SO4*Na2CO3      2. 16  6. 11 11.  8  1.  6
   25*     -1.44996E+3

900 APHI
  3.76978E-1  4.44771E-4  4.92829E-6  -1.41090E-8  4.31240E-11
  1.2          2.0          1.4          12.0
               BETA0        BETA1        BETA2      C-PHI
100 200      .00000      .00000      .00000
100 201      .20332     -.01668      -.00372
100 202      .00000      .00000      .00000
100 203      .00000      .00000      .00000
100 204      .00000      .00000      .00000
100 205      .02980      .00000      .04380
100 206      .20650     .55560      .00000
100 207      .00000      .00000      .00000
100 208      .00000      .00000      .00000
100 209      .00000      .00000      .00000
100 210      .00000      .00000      .00000
100 211      .00000      .00000      .00000
100 212      .00000      .00000      .00000
100 213      .00000      .00000      .00000
100 214      .00000      .00000      .00000
100 215      .00000      .00000      .00000
100 216      .00000      .00000      .00000
101 200      .08640     .25300      .00440
101 201      .07722     .25183      .00106
101 202      .03990     1.38900      .00440
:
:
```

FIGURE 3  
 Pitzer theta-psis, sample section of database  
 FORMAT( I3, 1X, I3, 8X, F9.5, 2( / 7X, 9F8.4))

```

    :
113 214      .00000      .00000      .00000
113 215      .00000      .00000      .00000
113 216      .00000      .00000      .00000
901
100 101      .03600
              .0000  -.0033  .0000  .0000  .0000  .0000  -.0129  .0000  .0000
              .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000
100 102      .00670
              .0000  -.0081  .0000  .0000  .0000  .1970  -.0265  .0000  .0000
              .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000
100 103      .08910
              .0000  -.0006  .0000  .0000  .0000  .0000  -.0178  .0000  .0000
              .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000
100 104      .00000
              .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000
              .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000
100 105      .00000
              .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000
              .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000
100 106      .06820
              .0000  .0043  .0000  .0000  .0000  .0000  .0000  .0000  .0000
              .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000
100 107      .00000
              .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000
              .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000
100 108      .00000
              .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000
              .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000  .0000
    :

```

## THE INPUT FILE

An example input file is shown in Figure 4. Since the input file is formatted, it is best to modify an existing input file to suit a user's individual needs rather than to type one from scratch. It must be a standard ASCII file with the left margin set to zero.

The first line is a label for the file and is ignored by the program. The next four lines are print options for the output file. They should be assigned either 0 or 1 depending on the options desired. When IPRINT=1, diagnostics of the main Newton minimization procedure are written to the output file at each Newton iteration. Diagnostics include mole numbers, free energies, step directions, and the type of step (feasibility, phase addition or deletion, refinement, or normal step). Unless a simulation is crashing for no apparent reason it is not recommended that IPRINT be set to 1 as the resulting output file will be quite large.

When IPR=1, the solution to the problem is written to the output file. When IPR2=1, the diagnostics of the subminimization procedure (called the "dual" problem) are written to the output file. When IPR3=1, the phase (Lagrangian) multipliers at the solution point will be written to the output file.

All of these input variables have integer values. The next input, the system temperature, must be in the form of a real number. Fortran distinguishes between real and integer numbers with the use of a decimal point. Input the system temperature in degrees Celcius.

At this point it is easiest to skip a few lines and return to them later. The line labeled "NAME" is for a user specified identifier that will be written to the output file. The next two lines are column headings and must be left as is. The symbols indicate available column width for convenience. In this section the species that are to be considered as possibly present in the phase are specified along with the type of phase under consideration. The aqueous phase must be entered first. The first line of any phase will consist of a integer in the first column indicating the type of phase:

- 1—Pure phase (i.e. solid or gas)
- 2—Aqueous ideal solution phase

3—Aqueous phase with Davies activity coefficient model

4—Aqueous phase with Pitzer activity coefficient model

Following the phase type indicator, all species that are to be considered possibly present in the phase should be listed by their database I.D. numbers. Executing the program "DBLIST.EXE" will produce a list of all the species, both aqueous and solid, contained in the database (appendix B). The aqueous phase must be entered first and the first specie must be water (specie #99). The remaining aqueous species must be in ascending order as shown in the example. A zero marks the end of a phase. A second zero would mark the end of the input file. A phase type identifier marks the presence of another phase (usually a solid or gas phase). Although each solid is technically a separate phase, all solids and gases can be listed together as one phase. A gas phase is distinguished from a solid phase by specifying its partial pressure. Aside from that they are treated identically.

Once the identities of the component species have been specified, the total elemental concentrations for the whole system must be specified. The variable TAU is used for this purpose. Its units are either moles or molality depending on the value of TAU chosen for specie #99. The number of moles in a kilogram of water is 55.50807. If one wishes to work in molal units, then TAU for specie #99 should be 55.550807. If there are several solutes dissolved in 10.0 moles of water then TAU would be 10.0 for specie #99. TAU can be specified for aqueous species as well as pure phases but not both within a given reaction set. For example if TAU were specified for  $K^+$  and for  $Cl^-$ , then TAU for  $KCl(s)$  must be left blank; and vise-versa. It is usually simplest to input all salts in aqueous form in their most ionized state (i.e. aluminum in the form of  $Al^{3+}$ , phosphate in the form of  $PO_4^{3-}$ , etc.) and let NAEELS deal with the speciation.

When predicting the solubility of a particular salt it is tempting to specify an absurdly large amount of its component ions on the assumption that the excess will simply precipitate. While this approach will work for a anhydrate salt, it will cause numerical difficulties with a hydrated salt. In effect, the water is used up. Since most precipitates in aqueous systems are hydrates, it is best to calculate solubilities by starting with a subsaturated system and progressively concentrate it until the precipitate appears. This

means making several runs gradually increasing the bulk concentration until a saturation is reached. If TAU is specified for any aqueous species, electroneutrality must be observed. An alternative is to specify TAU for one or more solids and let them dissolve.

The next input to consider is ETA. The natural logarithm of ETA (if non-zero) is added to the standard chemical potential of the species. This is how the fugacity of a gas is fixed in a open system. Input ETA in units of atmospheres. Specifying a value for ETA has no effect on TAU, which must be dealt with separately. By definition, a gas in an open system at constant partial pressure (ie O<sub>2</sub>, N<sub>2</sub>, CO<sub>2</sub>, etc in the atmosphere) can never be used up. However a finite bulk concentration must be specified. TAU for a gaseous species should be large enough such that not all of it is dissolved into the system.

It is possible to perform a series of calculations with NAEIS, making changes in the total system concentration at each step. For example, titration, evaporation, dilution, precipitation followed by filtration, etc., all fall in this category. The variables NTR, FRAX, ALPHA, ALPHA2, SIG, and ZETA are the parameters used to control the multi-step calculations. NTR is the desired number of steps in the series (NTR ≥ 1). FRAX is the fraction of solids formed that is to be removed at each step (0 ≤ FRAX ≤ 1). When FRAX=0 no solid removal takes place. When FRAX=1 all solids formed are removed at each step. This is useful to model a process involving stepwise filtration. ALPHA and SIG are used to add or remove an absolute amount of a specie at each step (0 ≤ ALPHA ≤ 1, SIG < 0 for removal, SIG > 0 for addition). The number of moles added or removed is the product ALPHA\*SIG. The stoichiometry must be chosen to ensure that electroneutrality is preserved after the step. ALPHA and SIG are typically used for step-wise additions of a solute. ALPHA2 and ZETA are used to add or remove a specific fraction of a specie at each step (0 ≤ ALPHA2 ≤ 1, ZETA < 0 for removal, ZETA > 0 for addition). The number of moles added or removed is the product ALPHA2\*ZETA multiplied by the equilibrium molality of the specie. The most common usage of ALPHA2 and ZETA is to add or remove water. It is important to note that a step-wise change, whether fractional or absolute, is relative to the equilibrium concentration of a given specie, which may be quite different than the value

of TAU for that specie.

Examples of input files are shown on the following pages. The corresponding output file for example 1 is shown in figure 6 in the section 'OUTPUT FILE'.

FIGURE 4  
Example input file for  $\text{CaCl}_2\text{-MgCl}_2$  system

INPUT FILE FOR EQUILIBRIUM SIMULATION PROGRAM

IPRINT=0

IPR=1

IPR2=0

IPR3=0

SYSTEM TEMPERATURE (C)=25.0

NTR=5

ALPHA=1.0

FRAX=0.8

ALPHA2=0.0

NAME:  $\text{CaCl}_2\text{-MgCl}_2$  system

#	TAU	SIG	ETA	ZETA
4				
99	55.50807		—3.0	
100				
103	4.500			
104				
106	0.5000		0.1	
107				
200				
201	10.00		0.2	
0				
1				
427				
435				
436				
450				
0				
0				

Figure 4 shows an example input file for a 5 step evaporation sequence. 4.5 moles of magnesium chloride and 0.5 moles of calcium chloride are dissolved in 1 kilogram of water. At each step, 3 moles of water are removed, 0.1 moles of calcium chloride are added, and 80% of the solids (if any) are removed. The solid species to be considered are  $\text{MgCl}_2\cdot 6\text{H}_2\text{O}$ ,  $\text{Ca}(\text{OH})_2$ ,  $\text{CaCl}_2\cdot 6\text{H}_2\text{O}$ , and  $\text{CaCl}_2\cdot 2\text{MgCl}_2\cdot 12\text{H}_2\text{O}$ . Pitzer's activity model is to be used for the aqueous phase.

FIGURE 5  
Example input file for  $\text{CaCO}_3\text{-Ca(OH)}_2$  system

INPUT FILE FOR EQUILIBRIUM SIMULATION PROGRAM

IPRINT=0  
IPR=1  
IPR2=0  
IPR3=0  
SYSTEM TEMPERATURE (C)=25.0  
NTR=10  
ALPHA=0.1  
FRAX=0.0  
ALPHA2=0.05  
NAME:  $\text{CaCO}_3\text{-Ca(OH)}_2$  system

#	TAU	SIG	ETA	ZETA
99	55.50807			-1.0
100				
106		2.0		
107				
108				
200		4.0		
202				
203				
204				
300				
301				
304	5.00			
0				
1				
435				
445				
489	10.00		0.005	
0				
0				

Figure 5 shows an example input file for a 10 step sequence. A 5.0 molal calcium carbonate solution is placed in contact with a carbon dioxide atmosphere at 0.005 atm. partial pressure. At each step 0.2 moles of calcium hydroxide are added while 5% of the water is removed. The 10 moles specified for initial  $\text{CO}_2$  is somewhat arbitrary. In the case of a gas in an open system, an amount needs to be specified such that not all of it is used up.  $\text{CaOH}_2$ ,  $\text{CaCO}_3$ , and  $\text{CO}_2(\text{gas})$  are the pure phases to be considered.

### THE OUTPUT FILE

The output file for Example 1 is shown in Figure 6. It begins by printing all of the relevant data retrieved from the database: dimensionless standard chemical potential, Pitzer parameters, atomic composition, bulk composition, etc. The solution itself lists total moles, the logarithm of the activity, and the molar concentration, of each species. The molality is printed in both scientific and standard notation for convenience. The activity of water and the ionic strength are also provided. Only the solids that actually formed are listed along with their respective mole numbers. A solution will be printed for each step in a multistep simulation.

Some other information of lesser utility is included in the output file. ITER is the number of full Newton steps taken in the convergence process. NAEELS will take only a partial step if a full step crosses a constraint boundary. IFUN is the number of functional evaluations at a full Newton step. If ITER=IFUN, then all Newton steps were full steps. IREFIN is the number of refinement steps taken. When a solution species is at or below a boundary value, a small refinement step is taken. The MAX RESIDUAL is the largest absolute value of the residual vector that is formed by subtracting the dot product of the solution vector and the mass balance matrix from the vector of total mole numbers. It is written in equation form as

$$\text{MAX } \|A_{ij} \cdot n_j - b_i\| \quad j=1, n^t \text{ and } i=1, m_c$$

This number should be significantly smaller than the smallest value in the MOLES column of the solution. NAEELS calculates a SCALE FACTOR that is unique to each minimization based on the characteristics of the unconstrained Lagrangian objective function. The entire problem is scaled up to improve the precision of the solution. For detailed information on the algorithm itself the reader is referred to Weare, et al. (1987).

FIGURE 6  
Example output file for  $\text{CaCl}_2\text{-MgCl}_2$  system

SOLUTION PHASE 1		PHASE TYPE 4	
SPECIES		CHARGE	U0/RT
1 H <sub>2</sub> O		0	-95.66110000
2 H <sup>+</sup>		1	.00000000
3 Mg <sup>++</sup>		2	-183.47200000
4 MgOH <sup>+</sup>		1	-252.82200000
5 Ca <sup>++</sup>		2	-223.32200000
6 CaOH <sup>+</sup>		1	-289.81500000
7 OH <sup>-</sup>		-1	-63.43440000
8 Cl <sup>-</sup>		-1	-52.93940000

NON-IDEAL AQUEOUS ELECTROLYTE SOLUTION DATA

APHI= .39097      BETA= 1.200      ALPHA= 2.000 1.400 12.000

		B0	B1	B2	Cphi
H <sup>+</sup>	OH <sup>-</sup>	.00000	.00000	.00000	.00000
H <sup>+</sup>	Cl <sup>-</sup>	.20332	-.01668	.00000	-.00372
Mg <sup>++</sup>	OH <sup>-</sup>	.00000	.00000	.00000	.00000
Mg <sup>++</sup>	Cl <sup>-</sup>	.35573	1.61738	.00000	.00474
MgOH <sup>+</sup>	OH <sup>-</sup>	.00000	.00000	.00000	.00000
MgOH <sup>+</sup>	Cl <sup>-</sup>	-.10000	1.65800	.00000	.00000
Ca <sup>++</sup>	OH <sup>-</sup>	-.17470	-.23030	.00000	.00000
Ca <sup>++</sup>	Cl <sup>-</sup>	.32579	1.38412	.00000	-.00174
CaOH <sup>+</sup>	OH <sup>-</sup>	.00000	.00000	.00000	.00000
CaOH <sup>+</sup>	Cl <sup>-</sup>	.00000	.00000	.00000	.00000

THETA-PSI	THETA	OH <sup>-</sup>	Cl <sup>-</sup>
H <sup>+</sup> Mg <sup>++</sup>	.0891	.0000	-.0006
H <sup>+</sup> MgOH <sup>+</sup>	.0000	.0000	.0000
H <sup>+</sup> Ca <sup>++</sup>	.0682	.0000	.0043
H <sup>+</sup> CaOH <sup>+</sup>	.0000	.0000	.0000
Mg <sup>++</sup> MgOH <sup>+</sup>	.0000	.0000	.0280
Mg <sup>++</sup> Ca <sup>++</sup>	.0070	.0000	-.0120
Mg <sup>++</sup> CaOH <sup>+</sup>	.0000	.0000	.0000
MgOH <sup>+</sup> Ca <sup>++</sup>	.0000	.0000	.0000
MgOH <sup>+</sup> CaOH <sup>+</sup>	.0000	.0000	.0000
Ca <sup>++</sup> CaOH <sup>+</sup>	.0000	.0000	.0000

THETA-PSI	THETA	H <sup>+</sup>	Mg <sup>++</sup>	MgOH <sup>+</sup>	Ca <sup>++</sup>	CaOH <sup>+</sup>
OH <sup>-</sup> Cl <sup>-</sup>	-.0500	.0000	.0000	.0000	-.0250	.0000

PURE PHASES

SPECIES	U0/RT
9 $\text{MgCl}_2\cdot 6\text{H}_2\text{O}$	-852.97000000
10 $\text{Ca}(\text{OH})_2$	-362.46300000
11 $\text{CaCl}_2\cdot 6\text{H}_2\text{O}$	-893.80500000
12 $\text{CaCl}_2\cdot 2\text{MgCl}_2\cdot 12\text{H}_2\text{O}$	-2015.79000000

FIGURE 6 (continued)

## CONSTRAINT EQUATIONS

SPECIES	0	1	8	12	17	20
1 H <sub>2</sub> O	.0	2.0	1.0	.0	.0	.0
2 H <sup>+</sup>	1.0	.0	.0	.0	.0	.0
3 Mg <sup>++</sup>	2.0	.0	.0	1.0	.0	.0
4 MgOH <sup>+</sup>	1.0	1.0	1.0	1.0	.0	.0
5 Ca <sup>++</sup>	2.0	.0	.0	.0	.0	1.0
6 CaOH <sup>+</sup>	1.0	1.0	1.0	.0	.0	1.0
7 OH <sup>-</sup>	-1.0	1.0	1.0	.0	.0	.0
8 Cl <sup>-</sup>	-1.0	.0	.0	.0	1.0	.0
9 MgCl <sub>2</sub> *6H <sub>2</sub> O	.0	12.0	6.0	1.0	2.0	.0
10 Ca(OH) <sub>2</sub>	.0	2.0	2.0	.0	.0	1.0
11 CaCl <sub>2</sub> *6H <sub>2</sub> O	.0	12.0	6.0	.0	2.0	1.0
12 CaCl <sub>2</sub> *2MgCl <sub>2</sub> *12H	.0	24.0	12.0	2.0	6.0	1.0

RANK= 5 MAXIMUM PHASES= 4

INPUT DATA CaCl<sub>2</sub>-MgCl<sub>2</sub> solubility

NTR,ALPHA,FRAZ,ALPHA2= 5 1.00000 .80000 .00000

SPECIES	TAU	SIG	ETA	ZETA
1 1 H <sub>2</sub> O	5.55081D+01	-3.00000D+00	0.00000D+00	0.00000D+00
3 1 Mg <sup>++</sup>	4.50000D+00	0.00000D+00	0.00000D+00	0.00000D+00
5 1 Ca <sup>++</sup>	5.00000D-01	1.00000D-01	0.00000D+00	0.00000D+00
8 1 Cl <sup>-</sup>	1.00000D+01	2.00000D-01	0.00000D+00	0.00000D+00

STEP 1 SCALE FACTOR= 9.0077D-03

SPECIES	MOLES	LN(A)	CONC	CONC
1 H <sub>2</sub> O	5.55081D+01	-8.05002D-01	5.55084D+01	55.50837
2 H <sup>+</sup>	4.95595D-06	1.22278D+01	4.95598D-06	.00000
3 Mg <sup>++</sup>	4.50000D+00	1.52827D+00	4.50002D+00	4.50002
4 MgOH <sup>+</sup>	4.94609D-06	-1.64166D+01	4.94612D-06	.00000
5 Ca <sup>++</sup>	5.00000D-01	-5.36143D+00	5.00003D-01	.50000
6 CaOH <sup>+</sup>	8.82285D-09	-2.31067D+01	8.82290D-09	.00000
7 OH <sup>-</sup>	1.04237D-09	-2.08039D+01	1.04238D-09	.00000
8 Cl <sup>-</sup>	1.00000D+01	7.66219D+00	1.00001D+01	10.00006

AH2O= .44708708 IONIC STRENGTH= 15.00007838 mol/kg

CONVERGENCE CRITERIA 1.077D-17 1.556D-08 0.000D+00

ITER= 10 IFUN= 18 IREFIN= 1

STEP 2 SCALE FACTOR= 9.5223D-03

SPECIES	MOLES	LN(A)	CONC	CONC
1 H <sub>2</sub> O	5.25081D+01	-9.23134D-01	5.55084D+01	55.50837
2 H <sup>+</sup>	5.09816D-06	-1.20758D+01	5.38947D-06	.00001
3 Mg <sup>++</sup>	4.49999D+00	-1.41153D+00	4.75712D+00	4.75712
4 MgOH <sup>+</sup>	5.08809D-06	-1.65700D+01	5.37883D-06	.00001
5 Ca <sup>++</sup>	6.00000D-01	-5.23566D+00	6.34284D-01	.63428
6 CaOH <sup>+</sup>	9.34971D-09	-2.32511D+01	9.88395D-09	.00000
7 OH <sup>-</sup>	7.19375D-10	-2.10740D+01	7.60480D-10	.00000
8 Cl <sup>-</sup>	1.02000D+01	8.23443D+00	1.07828D+01	10.78283

AH2O= .39727187 IONIC STRENGTH= 16.17423594 mol/kg

CONVERGENCE CRITERIA 1.210D-17 3.351D-09 0.000D+00

ITER= 4 IFUN= 4 IREFIN= 1

FIGURE 6 (continued)

STEP 3 SCALE FACTOR= 1.0099D-02

SPECIES	MOLES	LN(A)	CONC	CONC
1 H2O	4.95081D+01	-1.06542D+00	5.55084D+01	55.50837
2 H+	5.23386D-06	-1.19171D+01	5.86820D-06	.00001
3 Mg++	4.49999D+00	-1.27630D+00	5.04539D+00	5.04539
4 MgOH+	5.22391D-06	-1.67358D+01	5.85704D-06	.00001
5 Ca++	7.00000D-01	-5.14997D+00	7.84839D-01	.78484
6 CaOH+	9.47302D-09	-2.34664D+01	1.06211D-08	.00000
7 OH-	4.78277D-10	-2.13751D+01	5.36244D-10	.00000
8 Cl-	1.04000D+01	8.87384D+00	1.16605D+01	11.66047
AH2O=	.34458235	IONIC STRENGTH=	17.49069246	mol/kg
CONVERGENCE CRITERIA	1.104D-17	7.136D-09	0.000D+00	
ITER=	4	IFUN=	4	IREFIN= 1

STEP 4 SCALE FACTOR= 1.0751D-02

SPECIES	MOLES	LN(A)	CONC	CONC
1 H2O	4.33541D+01	-1.15415D+00	5.55084D+01	55.50837
2 H+	4.84053D-06	-1.19161D+01	6.19758D-06	.00001
3 Mg++	3.97433D+00	-1.43630D+00	5.08853D+00	5.08853
4 MgOH+	4.83007D-06	-1.69855D+01	6.18419D-06	.00001
5 Ca++	8.00000D-01	-5.15081D+00	1.02428D+00	1.02428
6 CaOH+	1.00864D-08	-2.35570D+01	1.29141D-08	.00000
7 OH-	3.71690D-10	-2.14648D+01	4.75893D-10	.00000
8 Cl-	9.54866D+00	9.35431D+00	1.22256D+01	12.22563
AH2O=	.31532387	IONIC STRENGTH=	18.33844247	mol/kg
9 MgCl2*6H2O	5.25669D-01	0.00000D+00	1.00000D+00	1.00000
CONVERGENCE CRITERIA	2.810D-17	4.443D-10	0.000D+00	
ITER=	6	IFUN=	12	IREFIN= 1

STEP 5 SCALE FACTOR= 1.2200D-02

SPECIES	MOLES	LN(A)	CONC	CONC
1 H2O	3.43453D+01	-1.18360D+00	5.55084D+01	55.50837
2 H+	3.97916D-06	-1.21246D+01	6.43106D-06	.00001
3 Mg++	2.97287D+00	-2.06687D+00	4.80471D+00	4.80471
4 MgOH+	3.96736D-06	-1.74369D+01	6.41198D-06	.00001
5 Ca++	9.00000D-01	-5.34106D+00	1.45457D+00	1.45457
6 CaOH+	1.14569D-08	-2.35681D+01	1.85165D-08	.00000
7 OH-	3.46996D-10	-2.12857D+01	5.60809D-10	.00000
8 Cl-	7.74575D+00	9.75794D+00	1.25186D+01	12.51856
AH2O=	.30617424	IONIC STRENGTH=	18.77783878	mol/kg
9 MgCl2*6H2O	1.10659D+00	0.00000D+00	1.00000D+00	1.00000
CONVERGENCE CRITERIA	4.291D-17	7.398D-08	0.000D+00	
ITER=	3	IFUN=	3	IREFIN= 1
NORMAL TERMINATION OF EQUIL				

## SOME SOLUBILITY PREDICTIONS

Figures 7 through 11 compare various solubility curves as predicted by NAEELS with experimental data from the literature. Qualitatively, NAEELS has predicted the shape of the curves and the approximate position of the invariant points. Invariant points (labeled 'ip' on the graphs) are points where all of the phases that can coexist do coexist at equilibrium (i.e. the system has zero degrees of freedom). Quantitatively, only the invariant points will be compared. If the solubility estimate is perfect, then the experimental invariant points will be identical to those predicted by NAEELS. The noninvariant points generated by NAEELS were not intended to duplicate the experimental points; but rather, that the resulting curves should coincide. For the invariant points the error will be defined for each solid as follows:

$$\% \text{ error} = \frac{\text{NAEELS solubility} - \text{experimental solubility}}{\text{experimental solubility}} \times 100\%$$

In Figure 7 points 2 and 3 are the invariant points as measured experimentally and as predicted by NAEELS, respectively. The error is 47% with respect to KCl and 3% with respect to  $\text{CaCl}_2 \cdot 6\text{H}_2\text{O}$ . In the region between point 1 and the invariant points, only KCl forms as a precipitate. In the region between point 4 and the invariant points, only  $\text{CaCl}_2 \cdot 6\text{H}_2\text{O}$  forms. Qualitatively, the agreement is good away from the invariant point. In this type of situation the predicted invariant point could be moved closer to the experimental point by adjusting the value of the Pitzer parameter  $\theta_{\text{K}^+ \cdot \text{Ca}^+}$  until the most satisfactory fit is achieved. This approach is commonly practiced when attempting to model a particular electrolyte system. It is not done here because the intent is for NAEELS to be a generally applicable simulator.

In the region between points 5 and 6 in Figure 8, the solid formed is  $\text{MgCl}_2 \cdot 6\text{H}_2\text{O}$ . Point 6 represents an invariant point both experimentally and as predicted by NAEELS. The error is 0.7% on the Mg axis and 2% on the Ca axis. In the region between points 6 and 7 (or 8), only the compound salt  $\text{CaCl}_2 \cdot 2\text{MgCl}_2 \cdot 12\text{H}_2\text{O}$  forms. Point 7 is the invariant point for the formation of the complex salt together with  $\text{CaCl}_2 \cdot 6\text{H}_2\text{O}$  as predicted by NAEELS. Point 8 is the corresponding experimental point. The error is 12% on the Mg axis and 7% on the Ca axis. Below point 8, only  $\text{CaCl}_2 \cdot 6\text{H}_2\text{O}$  forms.

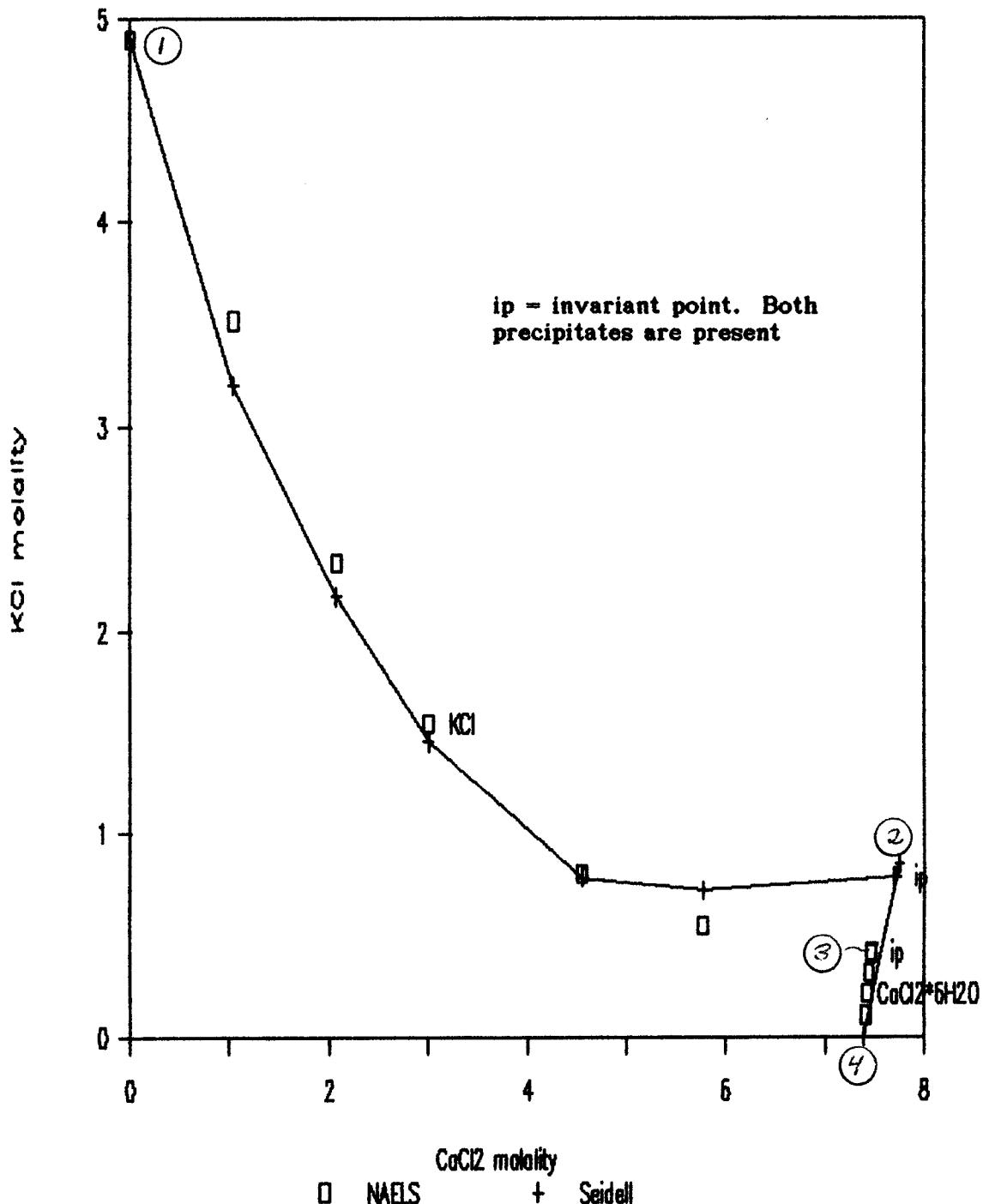
In Figure 9 the regions to the left of the invariant points are where the hydrated salt  $\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$  forms. The invariant points are where both the anhydrate and the hydrated salt coexist. To the right of the invariant points the anhydrate salt exists alone until point 9 is reached, where the solid  $\text{NaOH} \cdot \text{H}_2\text{O}$  precipitates. The invariant points are off by 30 % for  $\text{Na}_2\text{SO}_4$  and 18% for  $\text{NaOH}$ . The interesting point to notice about this graph is the similarity in the shapes of the curves. The curious vertical deflection of the curves appears to be due, at least in part to a discrepancy in the pure salt solubility of  $\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$ . The curves should start from the same point on the  $\text{Na}_2\text{SO}_4$  axis. The fact that they do not indicates that the experimental data would lead to a different value for the standard free energy of formation for  $\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$  than the one used to generate the NAEELS curve in Figure 9. This discrepancy could have been corrected to yield a better fit (in practice the single salt solubilities should be consistent with the experimental data). However, the procedure used in this work was to use a single salt solubility that was measured independently of the system of comparison. Adjusting the standard free energy for  $\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$  would yield a much better fit in this case.

In Figure 10 the  $\text{Na}_2\text{CO}_3\text{-Na}_2\text{SO}_4$  system predictions are shown. The region to the left of the invariant points is the solubility curve for  $\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$  while the region below is for  $\text{Na}_2\text{CO}_3 \cdot 10\text{H}_2\text{O}$ . The error in the invariant points is 7% on the  $\text{Na}_2\text{CO}_3$  axis and 8% on the  $\text{Na}_2\text{SO}_4$  axis.

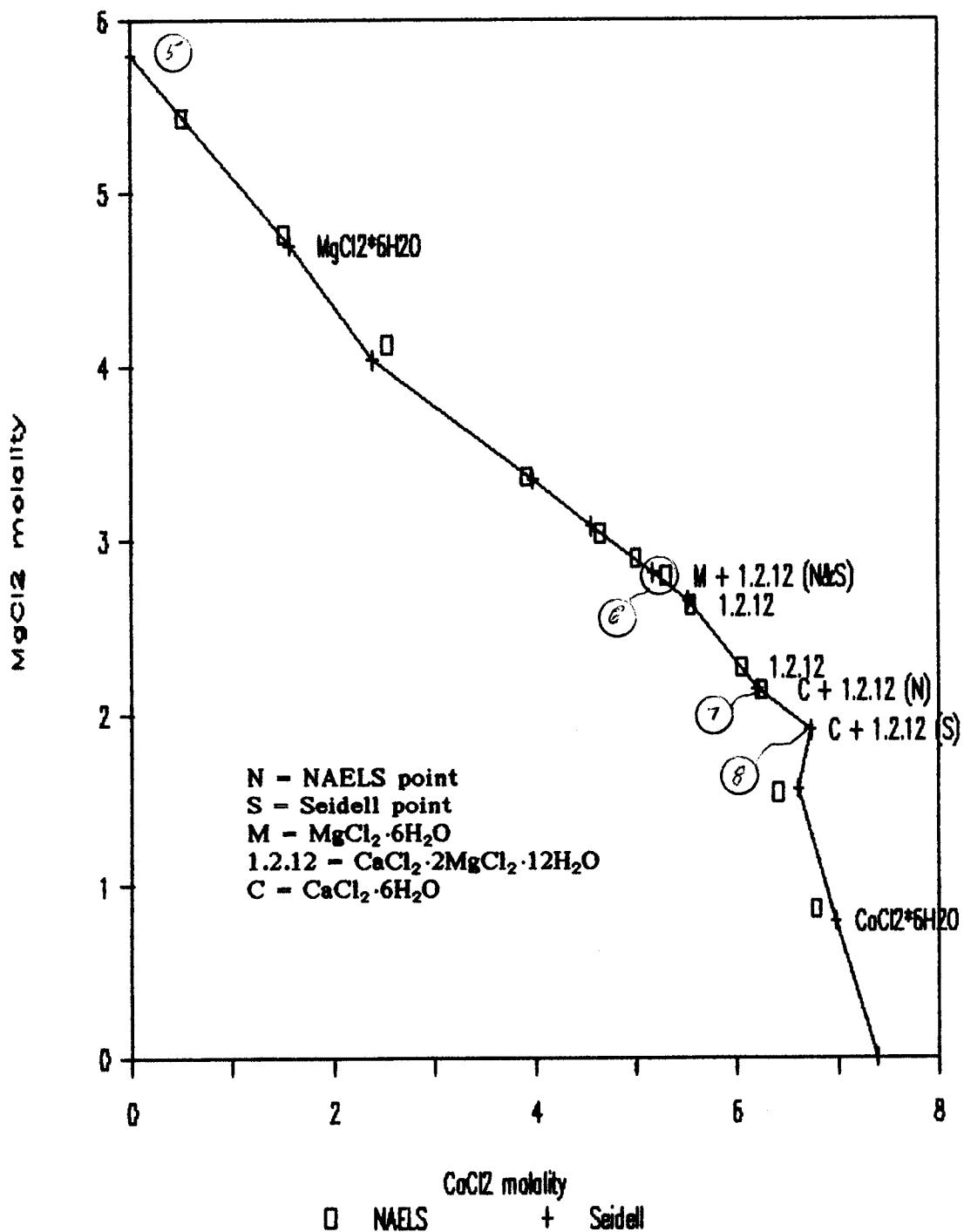
In Figure 11  $\text{NaCl}$  is the solid formed above the invariant points while  $\text{Na}_2\text{CO}_3 \cdot \text{H}_2\text{O}$  is formed below the invariant points. The invariant point error is 0.2% on the  $\text{Na}_2\text{CO}_3$  axis and 14% on the  $\text{NaCl}$  axis.

Qualitatively, these are good estimates of solubility. They are representative of the type of predictions that can be made with NAEELS. They would doubtless be better if more of the Pitzer ion interaction parameters were known for the many possible combinations of ions found in these systems. Whether or not the predictions are sufficiently accurate is a question that must be answered by the individual user.

Figure 7  
 $\text{CaCl}_2\text{--KCl}$  solubility at 25°C



**Figure 8**  
**CaCl<sub>2</sub>—MgCl<sub>2</sub> solubility at 25°C**



NaOH— $\text{Na}_2\text{SO}_4$  solubility at 25°C

Figure 9

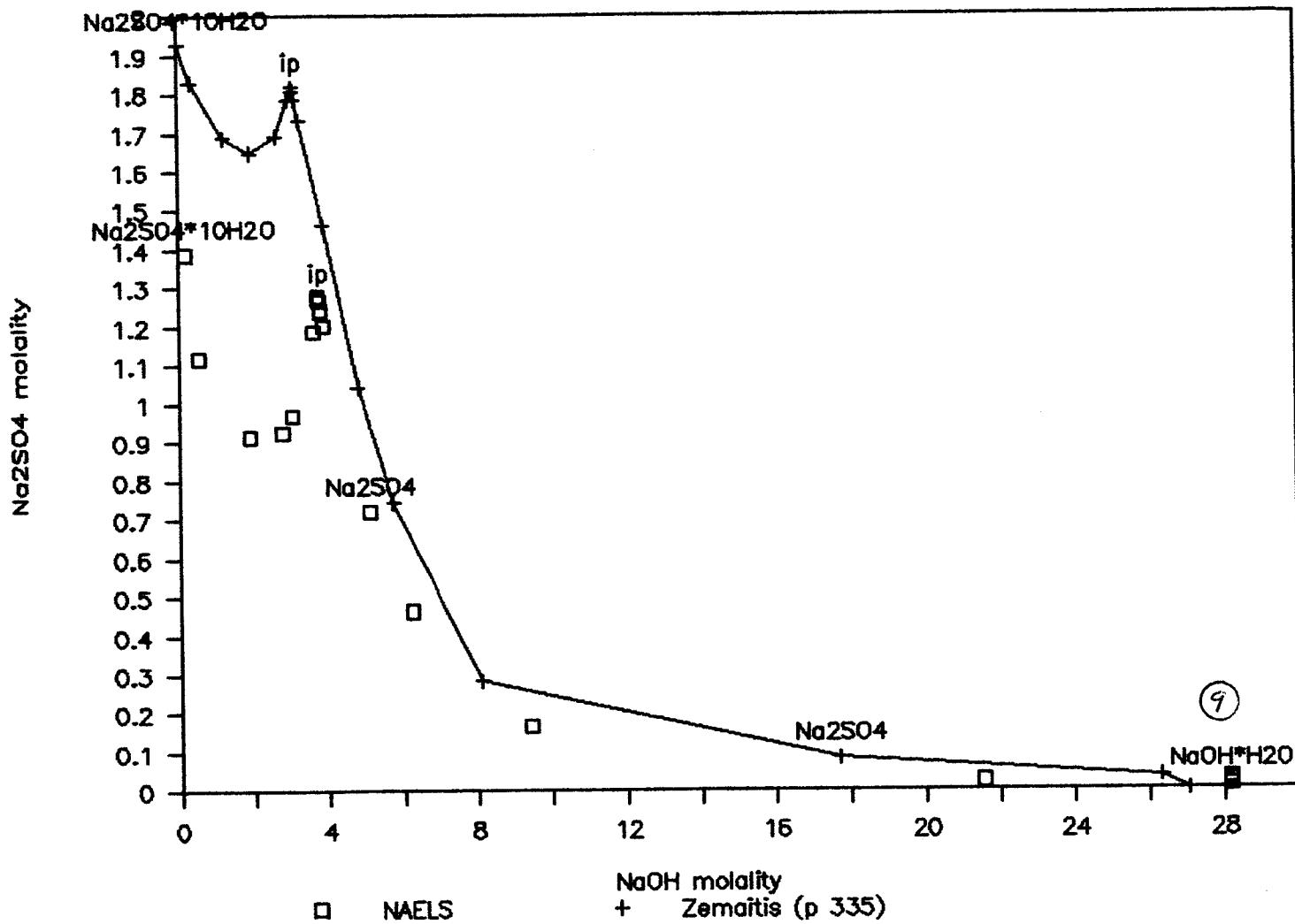


Figure 10  
 $\text{Na}_2\text{CO}_3$ — $\text{Na}_2\text{SO}_4$  solubility at 25°C

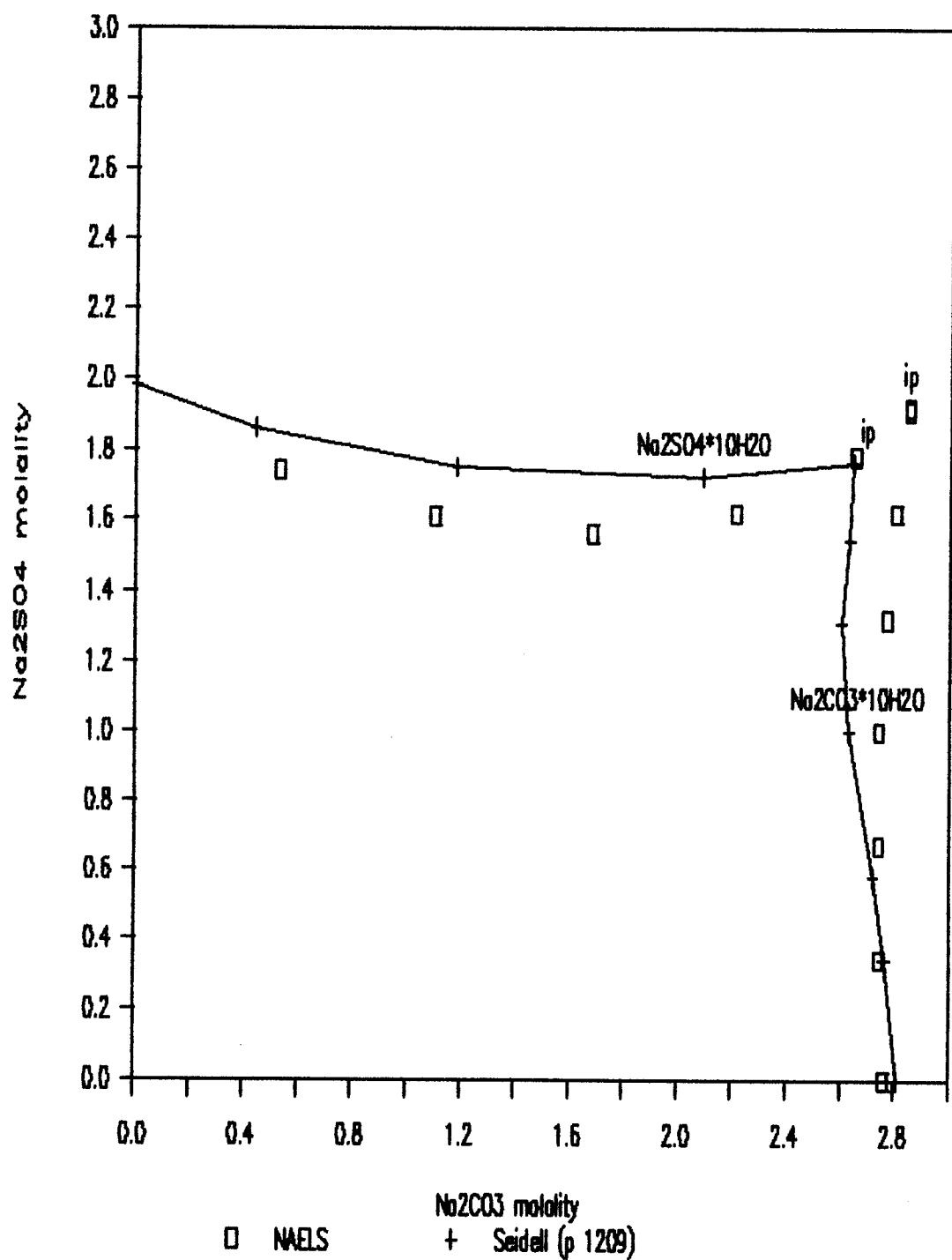
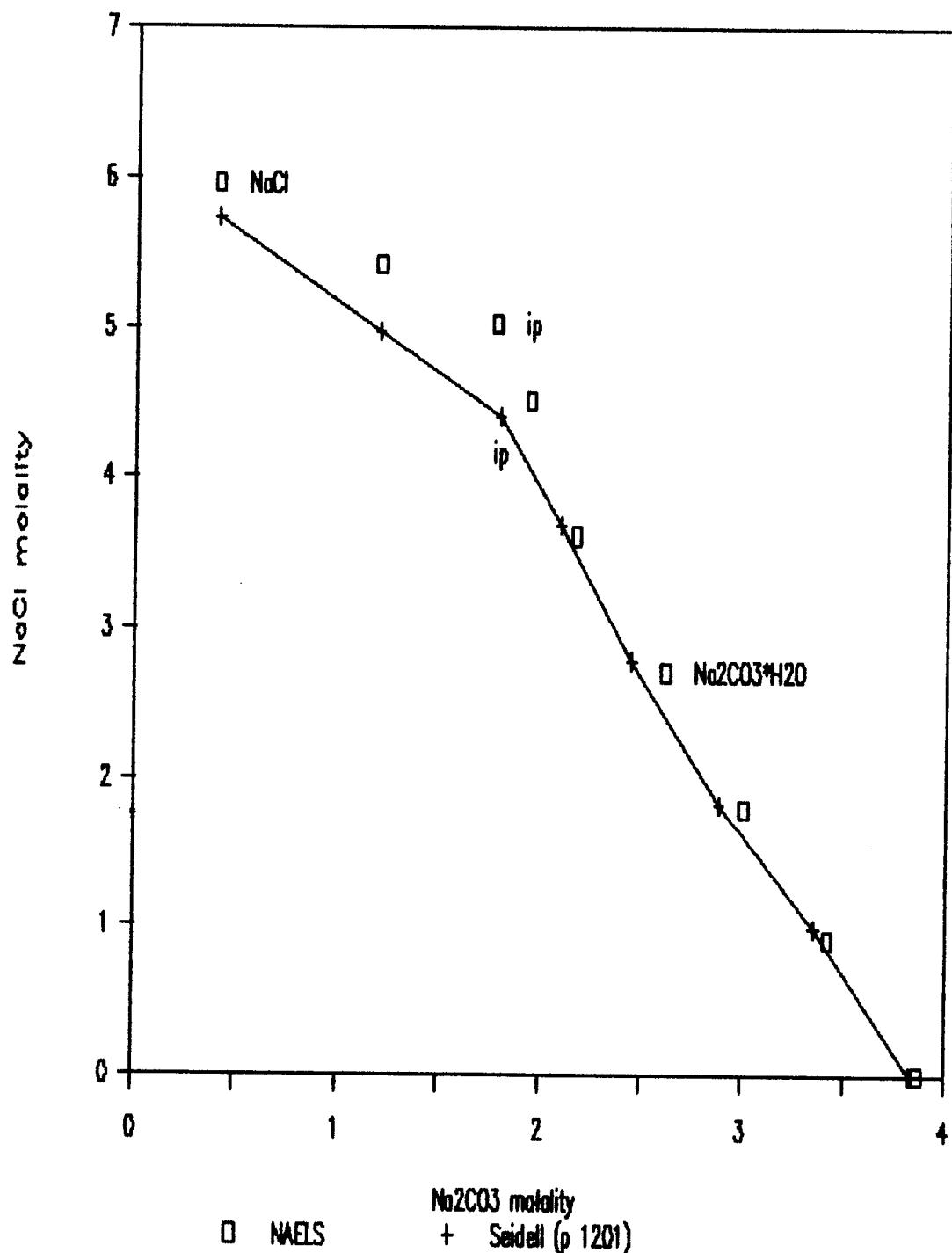


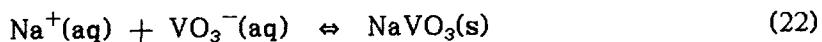
Figure 11  
NaCl— $\text{Na}_2\text{CO}_3$  solubility at 40°C



## LIMITATIONS AND SOURCES OF ERROR

The major drawback associated with using a free energy minimization approach to solving equilibrium problems is the need for very accurate data. Wagman, et al., lists only one value of  $\Delta G_f^\circ$  for each species (i.e. the value that the authors feel is from the most reliable source). Woods and Garrels make no such judgements and present thermodynamic data from as few as one to as many as a dozen or more sources for the more common species. It has been said that a man with one watch always knows what time it is while a man with two is never sure. Such is the case in solubility predictions. Discrepancies of a few percent in standard free energy values can translate into much larger discrepancies in solubility predictions. The three solubility curves in Figure 12 were generated using three of the standard free energy values for KCl at 25°C found in Woods and Garrels. The same standard free energy for NaCl was used throughout. Note how the small variations in  $\Delta G_f^\circ$  lead to relatively large variations in predicted solubility. The predicted locations of the invariant points (circled) are also affected.

Figure 12 illustrates why it is better to start with experimental solubility data and back-calculate what the chemical potential of the precipitate should be to produce a good fit. NAEELS can easily do this. Linke (1965) is an excellent source of solubility data. Whatever the source, solubility data is usually in the form of mass of dissolved solute per mass of solvent (or per mass of solution). This must be converted to units of molality. The procedure is best illustrated with an example. Assume that the vanadate ion was already in the database but the solid was a new addition. Consider the formation of sodium vanadate:



At equilibrium the sum of the chemical potentials on each side of Equation 22 must be equal. If Equation 1 is rearranged and written for each species, the result will be

$$\frac{\mu_{\text{Na}^+}^\circ}{RT} + \ln(a_{\text{Na}^+}) + \frac{\mu_{\text{VO}_3^-}^\circ}{RT} + \ln(a_{\text{VO}_3^-}) = \frac{\mu_{\text{NaVO}_3}^\circ}{RT} + \ln(a_{\text{NaVO}_3}) \quad (23)$$

The activity of the solid is taken as unity and all four terms on the left hand side of Equation 23 can be found in the output file of NAEELS. The dimensionless standard chemical potential of the solid is the only unknown. Seidell reports the solubility of sodium vanadate at 25°C to be 21.10 grams per 100 grams H<sub>2</sub>O. Its molecular weight is 121.93. Therefore the saturation concentration of both Na<sup>+</sup> and VO<sub>3</sub><sup>-</sup> is 1.7305 mol/kg. This concentration is then used in the input file for both Na<sup>+</sup> and VO<sub>3</sub><sup>-</sup>. H<sup>+</sup>, OH<sup>-</sup> and H<sub>2</sub>O are included in the input list and no solids are listed. Equation 23 must hold at saturation equilibrium. NAEELS will calculate the equilibrium activities on the left hand side of Equation 23 (the standard chemical potentials are already known). Executing NAEELS produces the following values, respectively, for the terms in Equation 23:

$$(-105.656) + (-0.0375964) + (-316.115) + (-0.498978) = \frac{\mu_{\text{NaVO}_3}^{\circ}}{RT} + 0.0$$

$$\frac{\mu_{\text{NaVO}_3}^{\circ}}{RT} = -422.308 \text{ dimensionless}$$

This number can now be added to the database as the dimensionless standard chemical potential for sodium vanadate (C<sub>0</sub> in Equation 17). If data is available at several temperatures, then the procedure can be repeated at each temperature to generate values which can then be fit to a polynomial of the form of Equation 17.

A word of caution is in order with respect to fitting chemical potential data to a polynomial. Whether the chemical potential data to be fitted comes from Equation 21 or from solubility data, some data sets turn out smoother than others. A utility called 'TK Solver' was used for all polynomial fitting in this project. The regression coefficient, R<sup>2</sup>, exceeded 0.9998 for all of the polynomials in the database (0.99999 was typical). Once the polynomial has been fit it is wise to check the discrepancy between the data points and the corresponding functional values. If the fit is unacceptable, then a new polynomial should be fit over a smaller 'temperature range of interest'.

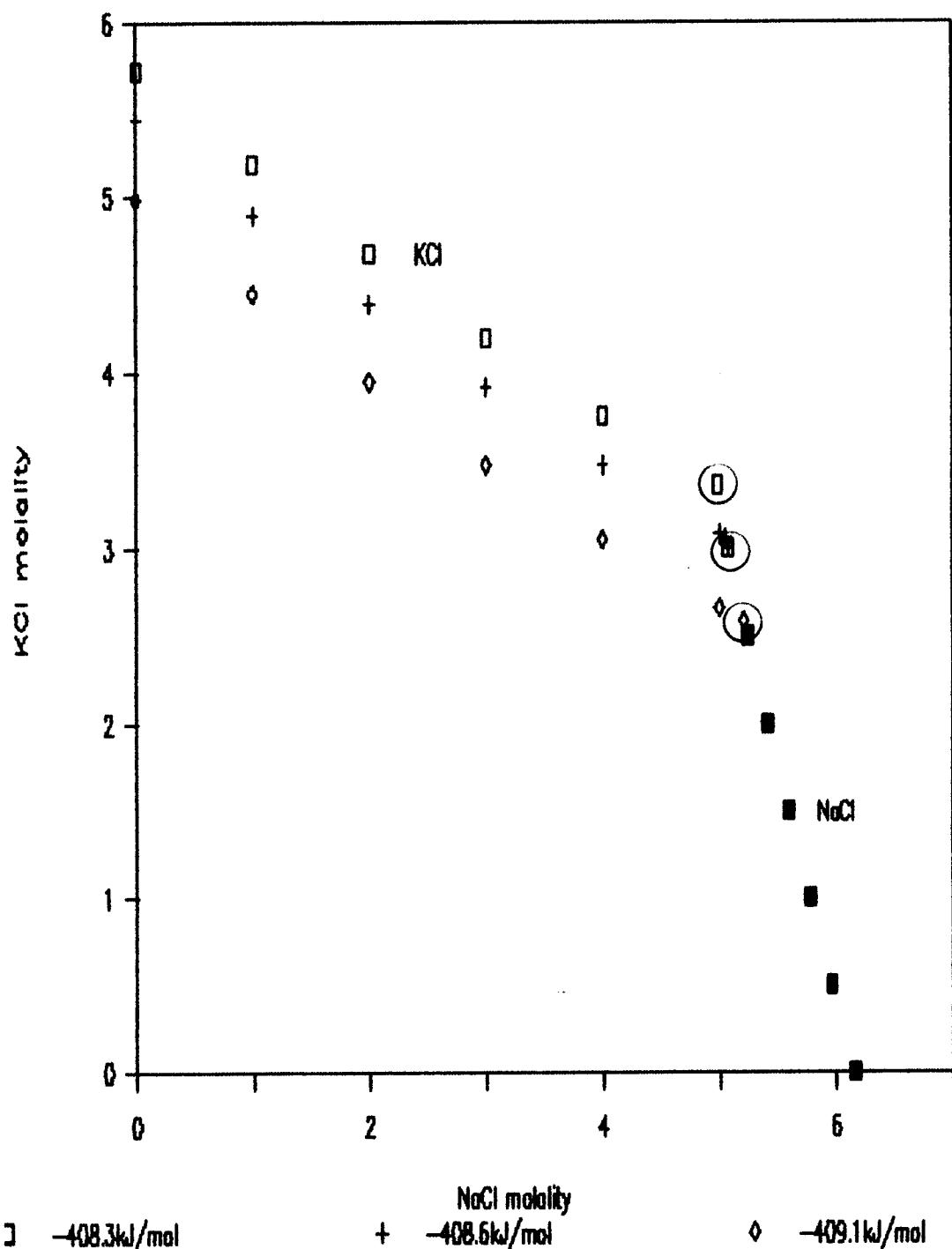
The lack of Pitzer parameters is another potential source of error in solubility predictions. It is obvious from a quick look at the Pitzer section of the database that the majority of the parameters are zero. The importance of Pitzer parameters varies with the concentration of the species in question.

For a dilute species the Pitzer parameters will never be missed if they are not available. The resulting product of the Pitzer parameters and the low molality is negligible compared to the corresponding terms for the major species in Equations 8. For the major species in a solution it is good to have them if they have been measured. The work of Pitzer and coworkers has been going on for decades but when one looks at the number of ions that exist, and the number of pairs and triplets that result, then the lack of data is not surprising. The amount of Pitzer data in the database does not represent an exhaustive search of the literature. Frederick and Kim (1988), Zemaitis, et al. (1986), and Weare, et al. (1987) were sources used in this work. A thorough search of the literature with emphasis on Pitzer would doubtless provide a good deal of parameters to add to the database.

With regard to systems at temperatures other than 25°C, the ion interaction parameters are assumed constant while  $A^\phi$  is a function of temperature. In reality Pitzer ion interaction parameters do vary with temperature. Pitzer (1979) gives derivatives with respect to temperature at 25°C for a few species. The slopes are very small at 25°C. While it would be safe to use these slopes to extrapolate parameters near 25°C, it is not sound to use a derivative at 25°C to extrapolate to a value at 100°C. Due to the small number of species for which temperature derivatives are available, and the fact even those are only known at 25°C, the Pitzer parameters are assumed constant over the range 25-100°C.

Figure 12

KCl—NaCl solubility  
dependence on KCl standard free energy



## CONCLUSIONS

In the course of this thesis an aqueous electrolyte simulator was assembled that meets all of the criteria set forth in the introduction.

It has never failed to converge to a solution in the course of this work even though the ionic strength frequently exceeded 30 molal. The time required to converge varied from under one minute to four minutes on an IBM compatable AT using an INTEL 80286 processer operating at 12 MHz. At the time of this writing this is not considered to be a very powerful computer.

Non-ideality, in any aqueous system, is greatest at saturation. At saturation the ionic strength is high and precipitates are in equilibrium with the aqueous phase. Therefore, solubility predictions provide the most challenging test for a program such as NAEELS. Qualitatively, the solubility diagrams predicted by NAEELS agreed well with the experimental data from the literature. NAEELS predicted the overall shapes of the experimental solubility diagrams and, in most cases, closely predicted the locations of the invariant points.

A large, user expandable database was assembled so as to minimize the amount of input required on the part of the user. Unfortunately, at the time of this writing, Pitzer ion interaction parameters have only been measured for a few dozen of the more common ionic species. Even fewer of the ternary (ion triplets) mixing parameters have been measured. As time goes on, hopefully the volume of Pitzer data will increase.

## FUTURE WORK

NAELS could be improved by expanding the database. Solubility data, preferably over a range of temperatures, could be used to add solid species to the database (the procedure is outlined in this work). Solution species can be added to the database with a trial and error procedure using solubility data. Solution species can be added directly if the standard free energy of formation is known.

A literature search of Pitzer ion interaction parameters would doubtless yield a number of them not found during the course of this work. Following a literature search, the experimental measurement of Pitzer ion interaction parameters for the user's system(s) of interest would be fruitful.

The capability of modeling a closed gas-liquid-solid system would make NAELS a more powerful utility. A closed system would require a model for predicting the activity coefficients of the gaseous species. As gaseous species enter and leave the solution phase, their partial pressures change. The total pressure is the sum of the partial pressures. If the total system pressure is to remain constant, then the system volume must be allowed to vary. If the volume is fixed, then the total pressure will vary. Since solubility is largely pressure dependent, a closed system is more complex to model. Constant pressure and constant volume closed systems could be two additional modeling capabilities for NAELS.

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## **APPENDICES**

## APPENDIX A

### Fortran language and hardware considerations

NAELS is written in the FORTRAN 77 computer language and compiled with the MICROSOFT FORTRAN compiler (version 4.1). There are 5 files associated with NAELS: The executable file NAELS.EXE, the database THERMO.DAT, the input file SALT, the output file SOLUTION, and the database listing program DBLIST.EXE. NAELS requires approximately 580 K of free RAM to run.

The user should consult the Microsoft manuals for technical support with regard to the compiler or linker. The options chosen for compiling this work are as follows:

/G2-selects the 80286 processor instruction set.  
/Os-optimizing procedures favor smallest code size.  
/Ge-enables stack probes to check available stack space.  
/FPi-generates in-line instructions and selects the emulator math package

Once the individual subroutines were compiled they were linked using the MS overlay linker. When a program is overlaid, specified parts of the program share the same space in RAM memory and are only loaded if and when they are needed. This reduces the required amount of free RAM necessary to run the program. NAELS was overlaid in the fashion shown below. The subroutines in parentheses are overlays while the remaining subroutines constitute the resident part of the program.

PRIMAL+DUAL+(MAIN+DATA+DATAP)+(CONADD+CONDEL)  
+(REFINE)+(GRAD+HESS+ELECT)+(DADD+DDEL)+(PHSPRT)

**APPENDIX B**  
**List of species in database**

THE FOLLOWING IS A LIST OF THE CHEMICAL SPECIES FOUND IN THE DATABASE "THERMO.DAT", ALONG WITH THEIR RESPECTIVE I.D. NUMBERS, AND THE TEMPERATURE RANGES OVER WHICH THE FREE ENERGY POLYNOMIAL IS VALID. AN ASTERISK INDICATES THAT THE FREE ENERGY DATA WERE CALCULATED USING STANDARD STATE DATA FOUND IN THE LITERATURE. THE ABSENCE OF AN ASTERISK INDICATES THAT THE FREE ENERGY DATA WERE CALCULATED FROM PUBLISHED SOLUBILITY DATA. CONSULT THE USERS MANUAL FOR MORE INFORMATION.

99 H2O		100 H+	25-200
101 Na+	25-200	102 K+	25-200
103 Mg++	25-200	104 MgOH+	25
105 MgHCO3+	25	106 Ca++	25-200
107 CaOH+	25	108 CaHCO3+	25
109 Al+++	25-200	110 AlOH++	25
111 Al(OH)2+	25	112 Mn++	25-200
113 MnOH+	25-200	200 OH-	25-200
201 Cl-	25-200	202 CO3--	25-200
203 HCO3-	25-200	204 NaCO3-	25
205 SO4--	25-200	206 HSO4-	25-200
207 NaSO4-	25	208 KSO4-	25
209 Al(OH)4-	25-200	210 S--	25-200
211 HS-	25-200	212 H3SiO4-	25
213 PO4---	25-200	214 HPO4--	25-200
215 H2PO4-	25-200	216 VO3-	25-200
300 H2CO3(aq)	25	301 NaHCO3(aq)	25
302 MgCO3(aq)	25	303 MgSO4(aq)	25
304 CaCO3(aq)	25	305 CaSO4(aq)	25
306 H4SiO4(aq)	25	307 H3PO4(aq)	25
400 SiO2	25-200*	401 Al2O3	25-200*
402 Al(OH)3	N/A	403 AlCl3	25-200*
404 AlCl3*6H2O	25-200*	405 Al2(SO4)3	25-200*
406 Al2(SO4)3*6H2O	25-200*	407 Al2SiO5	25-200*
408 Al2Si2O7*2H2O	25-200*	409 Al6Si2O13	25-200*
410 Al2Si4O10(OH)2	25-200*	411 Mn(OH)2	25-200*
412 MnCl2	25-200*	413 MnCl2*H2O	25-200*
414 MnCl2*2H2O	25-200*	415 MnCl2*4H2O	25-200*
416 MnS	25-200*	417 MnSO4	25-200*
418 MnHPO4	25*	419 MnCO3	25-200*
420 MnSiO3	25-200*	421 Mn2SiO4	25-200*
422 Mg(OH)2	25-200*	423 MgCl2	25-200*
424 MgCl2*H2O	25-200*	425 MgCl2*2H2O	25-200*
426 MgCl2*4H2O	25-200*	427 MgCl2*6H2O	25-200*
428 MgS	25-200*	429 MgSO4	25-200*
430 MgSO4*H2O	25-200*	431 MgSO4*6H2O	25-200*
432 Mg3(PO4)2	25-200*	433 MgCO3	25-200*
434 (MgCO3)3*Mg(OH)2*3H2O	25*	435 Ca(OH)2	25-200*
436 CaCl2	25-200*	437 CaS	25-200*
438 CaSO4	25-200*	439 CaSO4*2H2O	25-200*

440	Ca3(PO4)2	25-200*	441	CaHPO4	25-200*
442	CaHPO4*2H2O	25-200*	443	Ca(H2PO4)2*H2O	25-200*
444	Ca10(PO4)6(OH)2	25-200*	445	CaCO3	25-200*
446	CaO*SiO2	25-200*	447	CaO*Al2O3	25-200*
448	CaO*Al2O3*2SiO2*H2O	25-200*	449	CaCO3*MgCO3	25-200*
450	(CaO)2*MgO*2SiO2	25-200*	451	NaOH	25-200*
452	NaOH*H2O	25-64	453	NaCl	25-100
454	Na2S	25-200*	455	Na2SO4	25-200*
456	NaHSO4	25-200*	457	NaHSO4*H2O	25-200*
458	Na2SO4*10H2O	25	459	Na3PO4	25-200*
460	NaH2PO4	25-200*	461	Na2HPO4	25-200*
462	Na2HPO4*2H2O	25-200*	463	Na2HPO4*7H2O	25-200*
464	Na2HPO4*12H2O	25-200*	465	Na2CO3	25-200*
466	Na2CO3*H2O	25-200*	467	Na2CO3*7H2O	25-200*
468	Na2CO3*10H2O	20-32	469	NaHCO3	25
470	NaHCO3*Na2CO3*2H2O	25	471	Na2SiO3	25-200*
472	NaAlSiO4	25-200*	473	NaVO3	25-75
474	KOH	25-200*	475	KOH*2H2O	25-200*
476	KCl	25-200*	477	K2S	25-200*
478	K2SO4	25-200*	479	KHSO4	25-200*
480	K2CO3	25-200*	481	KHCO3	25-200*
482	KA1(SO4)2	25-200*	483	KA1(SO4)2*3H2O	25-200*
484	KA1(SO4)2*12H2O	25-200*	485	KA1SiO4	25-200*
486	KA13Si3O10(OH)2	25-200*	487	KH2PO4	25-200*
488	2Na2SO4*Na2CO3	25*			

## APPENDIX C

### Ionic heat capacity estimation

Partial molar heat capacity data for ions in solution is usually available only at 25°C, if at all. In general, heat capacities for solution species vary considerably with temperature. It is therefore expedient to make use of a model for the estimation of ionic heat capacities as a function of temperature. The model begins with the relationship between entropy and heat capacity:

$$S_t^\circ = S_{t_0}^\circ + \int_{t_0}^t C_p \, d(\ln T) \quad (C-1)$$

$C_p$  can be replaced with an average value over the range  $t_0$  to  $t$ :

$$S_t^\circ = S_{t_0}^\circ + \overline{C_p}|_{t_0}^t \int_{t_0}^t d(\ln T) \quad (C-2)$$

Integrating and rearranging yields,

$$\overline{C_p}|_{t_0}^t = \frac{S_t^\circ - S_{t_0}^\circ}{\ln(t/t_0)} \quad (C-3)$$

$t_0$  and  $t$  can be any two temperatures. For convenience  $t_0$  will be the standard reference temperature 298.15 K. Standard state entropies of aqueous species are usually available in the literature. Wagman (1982), and Woods and Garrels (1987) are excellent references. The only remaining unknown in Equation C-3 is  $S_t^\circ$ . The correspondence principle developed by Criss and Cobble (1964) states, "...If the standard state is chosen properly by fixing the entropy of H<sup>+</sup>(aq) at each temperature, then the ionic entropies at one temperature are linearly related to their corresponding entropies at 25°C". In equation form the model looks like this:

$$S_t^\circ = a(t) + b(t)S_{25abs}^\circ \quad (C-4)$$

$$\text{where } S_{25abs}^\circ = S_{25}^\circ - 5.0Z \quad (C-5)$$

$S_{25}^\circ$  in Equation C-5 is the conventional standard state entropy and  $Z$  is the charge on the ion.  $a(t)$  and  $b(t)$  in Equation C-4 are functions of temperature and ion type. Criss and Cobble's values for  $a(t)$  and  $b(t)$  are presented in Table 1.  $S_t^\circ$  is then used in Equation C-3 to obtain a partial molar heat

capacity over a particular temperature range.

TABLE 1  
Criss and Cobble's entropy parameters  
(in CAL/MOLE\*°C)

t, °C	Simple cations		Simple anions and OH <sup>-</sup>		Oxy anions XO <sub>n</sub> <sup>-m</sup>		Acid oxy anions XO <sub>n</sub> (OH) <sup>-m</sup>		S <sub>abs</sub> <sup>o</sup> (H <sup>+</sup> )
	a(t)	b(t)	a(t)	b(t)	a(t)	b(t)	a(t)	b(t)	
25	0	1.000	0	1.000	0	1.000	0	1.000	-5.0
60	3.9	0.955	-5.1	0.969	-14.0	1.217	-13.5	1.38	-2.5
100	10.3	0.876	-13.0	1.000	-31.4	1.476	-30.3	1.894	2.0
150	16.2	0.792	-21.3	0.989	-46.4	1.687	-50.	2.381	6.5
200	23.3	0.711	-30.2	0.981	-67.	2.02	-70.	2.960	11.1

Criss and Cobble's model, in its original form, is of limited utility. In order to calculate the chemical potential as a continuous function of temperature, one must be able to calculate the heat capacity in Equation C-3 as a continuous function of temperature. It turns out that the entropy parameters in Table 1 are themselves linear functions of temperature. Performing linear regression on the parameters in Table 1 produces the equations in Table 2, where T is in degrees celcius.

TABLE 2  
Entropy parameters  
as a function of temperature

	Regression coeff. R <sup>2</sup>
Simple cations	
a = 0.133886 T - 3.58581	0.991
b = -0.00168537 T + 1.04717	0.997
Simple anions and OH <sup>-</sup>	
a = -1.174220 T + 4.72151	0.998
b = -0.00018599 T + 1.0077	0.952
Oxy anions	
a = -0.377372 T + 8.69877	0.997
b = 0.00569148 T + 0.871012	0.994
Acid oxy anions	
a = -0.401098 T + 10.1575	0.999
b = 0.0111676 T + 0.728066	0.998
S <sub>25</sub> <sup>o</sup> for H <sup>+</sup>	
a = 0.09416006 T - 7.635905	0.995

Combining the equations in Table 2 with Equations C-4 and C-5 produces a partial molar heat capacity function which is used to calculate the chemical potential as a function of temperature for charged aqueous species. Since the reference state for H<sup>+</sup>(aq) in this model is a function of temperature, the chemical potential function for any charged species added to the database must be generated using the procedure outlined above. This is necessary to maintain interal consistency in the database. However, if an aqueous species can be formed by combining species already found in the database (ie. neutral salt species), then it is best to use equilibrium data (if available) to generate the chemical potential function for the new species.

Criss and Cobbles method requires that the standard state entropy, S<sub>25</sub><sup>o</sup>, be known. For many ions this quantity is not available in the literature. One method of estimating S<sub>25</sub><sup>o</sup> is that of Connick and Powell (1953). The entropy of an oxy-anion of the form XO<sub>n</sub><sup>-z</sup> can be estimated fairly well from the empirical relation

$$S_{25}^o = 43.5 - 46.5(Z - 0.28n) \quad \frac{\text{cal}}{\text{mol} \cdot ^\circ\text{C}}$$

Once again, a literature search would be in order as there are other methods for estimating entropies.

**APPENDIX D**  
**Nomenclature and glossary**

- $\alpha_1$  Pitzer parameter equal to 1.4 for 2-2 electrolytes and 2.0 for all others.
- $\alpha_2$  Pitzer parameter equal to 12.0 for 2-2 electrolytes and 0.0 for all others.
- $\beta_{ca}^0$  Pitzer empirical parameter for the salt c-a.
- $\beta_{ca}^1$  Pitzer empirical parameter for the salt c-a.
- $\beta_{ca}^2$  Pitzer empirical parameter for the 2-2 salt c-a.
- $\gamma_i$  Activity coefficient for ion i (dimensionless).
- $\eta_i$  Lagrangian multiplier for charge balance constraint i.
- $\theta_{cc}$  Pitzer ion interaction parameter for unlike cation pairs (empirical).
- $\theta_{aa}$  Pitzer ion interaction parameter for unlike anion pairs (empirical).
- $K_i$  Lagrangian multiplier for mass balance constraint i.
- $\mu_i$  Chemical potential of (or Gibb's energy per mole of) species i.
- $\mu_i^\circ$  Standard state chemical potential of species i.
- $\pi$  Pi
- $\psi_{caa}$  Pitzer ternary interaction parameter for unlike ion triplets (empirical).
- $\psi_{acc}$  Pitzer ternary interaction parameter for unlike ion triplets (empirical).
- $\omega_i$  Lagrangian multiplier for species constraint i.
- $A_{ij}$  Number of moles of element i in species j.
- $a_i$  Activity of species i.
- $A^\phi$  Debye-Hückel constant for osmotic coefficients on a natural log basis.
- $b_i$  Number of moles of element i.
- $b$  Pitzer parameter equal to 1.2
- $C_{ca}$  Pitzer empirical parameter for the salt c-a.
- $C_p$  Heat capacity at constant pressure.
- $D$  Dielectric constant of water (temperature dependent).
- $e_c$  Number of electrolyte phases.
- $e$  Electronic charge.  $1.60 \times 10^{-19}$  coulombs
- $G$  Total system (Gibb's) free energy.
- $\Delta G_f^\circ$  Standard state free energy of formation
- $\Delta G_{rxn}$  Free energy change in a chemical reaction.
- $H$  Enthalpy
- $I$  Ionic strength (mole/kg).
- $k$  Boltzman's constant ( $1.381 \times 10^{-23}$  joule/kelvin)
- $L$  Lagrangian objective function to be minimized

$m_i$  molality of species i (moles/kg).  
 $m_c$  number of mass balance constraints  
 $n_i$  mole number of species i (moles).  
 $n^t$  total number of chemical species in the system.  
 $N_o$  Avagadro's number.  
 $R$  Gas constant (8.314 joules/mole\*kelvin).  
 $S$  Entropy  
 $S_{25}^\circ$  Standard state entropy (joules/mole\*kelvin).  
 $T$  Absolute temperature (kelvin).  
 $t$  Temperature (celcius).  
 $T_{ref}$  Reference temperature (298.15 kelvin).  
 $t_i^2$  Slack variable for the non-zero mole number constraint for species i.  
 $z_i$  Charge of species i.

**Species:** A compound made up of elements and formed by one or more equilibrium reactions. It may be charged or neutral and it may exist in any phase. For example, CO<sub>2</sub>(gas) and CO<sub>2</sub>(aq) are two different species.

**Phase assemblage** The number of phases present in a system at equilibrium as well as the composition of each phase. The solution to an equilibrium problem.

**Element:** A chemical element as found in the periodic table.

**Empirical:** Measured experimentally

## APPENDIX F

## NAELS source code

```

C-----  

C THE MAIN PROGRAM. THIS PROGRAM CALCULATES CHEMICAL  

C EQUILIBRIUM BY MINIMIZING THE GIBBS FREE ENERGY.  

C INDEPENDENT REACTIONS DIRECTIONS ARE CALCULATED  

C THROUGH A QR FACTORIZATION OF THE CONSTRAINT MATRIX. A  

C SET OF VARIABLES ARE THUS GENERATED THAT CORRESPOND  

C TO THESE REACTION DIRECTIONS. SINCE THE PROGRESS TO  

C EQUILIBRIUM THAT IS DEFINED BY THESE VARIABLES DO NOT  

C VIOLATE ANY CONSTRAINTS, THE PROBLEM HAS BEEN  

C TRANSFORMED INTO AN UNCONSTRAINED MINIZATION PROBLEM.  

C NEWTONS METHOD IS USED TO CONVERGE TO A SOLUTION. THE  

C LAGRANGIAN MULTIPLIERS WHICH CORRESPOND TO THE  

C NEGATIVE NATURAL LOGARITHM OF THE SATURATION RATIO  

C ARE THEN EVALUATED IN ORDER TO DETERMINE WHETHER OR  

C NOT OTHER PHASES SHOULD COME INTO THE SYSTEM. THE  

C SEQUENCE OF NEWTON ITERATIONS AND MULTIPLIER  

C EVALUATIONS IS CONTINUED UNTIL GLOBAL CONVERGENCE IS  

C ACHIEVED.  

C-----  


```

```

C CURRENT DIMENSIONS:  

C A MAXIMUM TOTAL OF 96 SPECIES IN ALL INPUT PHASES,  

C A MAXIMUM OF 60 SOLUTION SPECIES IN ALL SOLUTION  

C PHASES, A MAXIMUM TOTAL OF 46 AQUEOUS SPECIES: 18  

C MAXIMUM CATIONS, 18 MAXIMUM ANIONS, AND 8 MAXIMUM  

C NEUTRALS. (NOTE THAT ONE SPECIES MUST BE H2O).  

C-----  


```

```

IMPLICIT DOUBLE PRECISION(A-H,O-Z)  

IMPLICIT INTEGER*2 (I-N)  

INTEGER*2 Z  

LOGICAL ERROR,CHARGE,BLANK,OUTFIL  

DIMENSION DEL(96),BB(96),U0(96),JH(96),PF(96),JFRAX(96)  

CHARACTER*24 A1  

CHARACTER*50 WATLAB  

C-----  


```

```

C ERROR: IF PRIMAL RETURNS ERROR = .TRUE., THE PROBLEM HAS  

C NOT CONVERGED OR HAS CONVERGED TO A NON-CONVEX  

C SOLUTION.  

C  

C CHARGE: = .TRUE. IF THE PROGRAM HAS TO MAKE A CHARGE  

C BALANCE CORRECTION  

C-----  


```

```

COMMON/CONST/Q(96,96),MACT(96),JA(96),JI(96),NSC,NCC,NDC,NC0C  

1, N0EC  

C-----  

C-----  


```

C Q(I,I): NON-DEGENERATE MASS, CHARGE AND SPECIES  
C CONSTRAINT MATRIX

C MACT(J)=I : VECTOR THAT MAPS THE SPECIES NUMBER J TO  
C EITHER CONSTRAINT NUMBER I IF THAT SPECIES IS  
C CONSTRAINED OR DEGENERATE, OR TO A NEGATIVE NUMBER  
C IF THE SPECIES IS UNCONSTRAINED (I). THE VALUE OF A  
C POSITIVE I ALSO CORRESPONDS TO AN ELEMENT IN THE JA  
C ARRAY WHEN THE CONSTRAINT CORRESPONDS TO A SPECIES  
C CONSTRAINT. THE ABSOLUTE VALUE OF THE NEGATIVE  
C NUMBER CORRESPONDS TO THE ELEMENT OF THE JI ARRAY  
C THAT IS EQUAL TO THAT SPECIES NUMBER.

C JA(J)=I: VECTOR THAT MAPS CONSTRAINT NUMBER J TO  
C SPECIES NUMBER I. IF J IS LESS THEN OR EQUAL TO NC0C,  
C THE NUMBER OF NON-DEGENERATE MASS AND CHARGE  
C BALANCE CONSTRAINTS, THE SPECIES I IS DEGENERATE. THE  
C NUMBER OF DEGENERATE CONSTRAINTS CANNOT BE MORE  
C THEN NC0C. REASON: THE TOTAL NUMBER OF SPECIES IS  
C NSC. THEREFORE THE MAXIMUM NUMBER OF (DEGENERATE  
C AND NON-DEGENERATE) SPECIES CONSTRAINTS IS NSC.  
C THE MAXIMUM NUMBER OF NON-DEGENERATE SPECIES  
C CONSTRAINTS IS NCC-NC0C (TOTAL LINEARLY INDEPENDENT  
C CONSTRAINTS MINUS THE CHARGE AND MASS BALANCE  
C CONSTRAINTS). THEREFORE THE MAXIMUM NUMBER OF  
C DEGENERATE CONSTRAINTS

C JI(J)=I: VECTOR THAT MAPS THE JTH INACTIVE SPECIES TO  
C UNCONSTRAINED, NON-DEGENERATE SPECIES I

C EXAMPLES:

C AN AQUEOUS SYSTEM WITH 20 AQUEOUS SPECIES AND 20  
C PURE EVAPORITES. SPECIES 24 HAS PRECIPITATED AND  
C ONLY IT AND THE SOLUTION PHASE ARE PRESENT:

C JI(21) = 24  
C MACT(24) = -21

C ASSUME NOW THAT SPECIES 24 COMPLETELY DISSOLVES AND  
C THAT THE SYSTEM HAS 9 MASS AND CHARGE BALANCE  
C CONSTRAINTS. WE THEN HAVE 9 + 20 (SOLID SPECIES HELD  
C OUT) = 29 TOTAL CONSTRAINTS:

C JA(29) = 24  
C MACT(24) = 29

C NSC: TOTAL NUMBER OF SPECIES

C NCC: TOTAL NUMBER OF NON-DEGENERATE CONSTRAINTS

```

C      NDC: TOTAL NUMBER OF DEGENERATE SPECIES
C
C      NC0C: TOTAL NUMBER OF NON-DEGENERATE MASS AND CHARGE
C          BALANCE CONSTRAINTS
C
C      N0EC: TOTAL NUMBER OF CHARGE BALANCE CONSTRAINTS
C
C-----
C      DIMENSIONS AND STRUCTURE OF THE QT AND R MATRICES.
C      (T=TRANSPOSE) (A IS THE MATRIX OF LINEARLY INDEPENDENT
C      CONSTRAINTS).
C
C      \Q1T\    \R11\R12\
C      \---\    ---\---
C      \Q2T\ A = \ 0 \R22\
C      \---\    ---\---
C      \Q3T\    \ 0 \ 0 \
C
C      R11 AND R22 ARE UPPER TRIANGULAR.
C
C      DIMENSIONS: Q1T: NC0C X NSC
C                  Q2T: (NCC-NC0C) X NSC
C                  Q3T: (NSC-NCC) X NSC
C
C                  A: NSC X (NCC-NC0C)
C
C                  R11: NC0C X NC0C
C                  R12: NC0C X (NCC-NC0C)
C                  R22: (NCC-NC0C) X (NCC-NC0C)
C
C-----
C      COMMON/LABEL/Z(96),U0RT(96),A1(96)
C-----
C      Z(J):    CHARGE ON EACH SPECIES J
C
C      U0RT(J): STANDARD CHEMICAL POTENTIAL OF SPECIES J
C
C      A1(J): CHARACTER ARRAY FOR THE NAME OF SPECIES J
C
C-----
C      COMMON/PHASE/R11(96,16),JMIN(96),JMAX(96),KSPEC(96),KTYP(96)
C      1 ,NP0C,N00C
C-----
C      R11(J,J): STORE "R" ARRAY FROM "QR" FACTORIZATION
C
C      JMIN(J)=I : MAP PHASE NUMBER J TO MINIMUM SPECIES NUMBER
C                  I IN PHASE J
C
C      JMAX(J)=I : MAP PHASE NUMBER J TO MAXIMUM SPECIES
C                  NUMBER I IN PHASE J
C

```

```

C-----  

C  

C      EXAMPLES:  

C  

C          SAME PHASES AS IN ABOVE EXAMPLE  

C  

C          JMIN(1) = 1      JMAX(1) = 20  

C          JMIN(2) = 21     JMAX(2) = 21  

C          JMIN(3) = 22     JMAX(3) = 22  

C  

C-----  

C  

C          KSPEC(J)=I: MAPS SPECIES NUMBER J TO PHASE NUMBER I.  

C          IN SAME EXAMPLE,  

C          KSPEC(1) = 1  

C          KSPEC(2) = 1  

C          KPSEC(21) = 2  

C  

C          KTYPE(J)=I: MAPS PHASE NUMBER J TO PHASE TYPE I  

C          I = 1  PURE PHASE  

C          I = 2  IDEAL SOLUTION PHASE  

C          I = 3  NON-IDEAL DAVIES AQUEOUS PHASE  

C          I = 4  NON-IDEAL PITZER AQUEOUS PHASE  

C          IF I IS NEGATIVE, THAT PHASE  

C          IS NOT PRESENT  

C  

C          NP0C: THE TOTAL NUMBER OF PHASES (INCLUDING THOSE THAT  

C          ARE NOT PRESENT)  

C  

C          N00C: THE TOTAL (INCLUDING DEGENERATE) CHARGE AND MASS  

C          BALANCE CONSTRAINTS  

C  

C-----  

C          COMMON/SEARCH/X(96),P(96),G(96),W0(96),H(64,64),B(96),S(32)  

C-----  

C  

C          X(J): MOLE NUMBER OF EACH SPECIES J  

C  

C          P(J): STEP DIRECTION OF SPECIES J AT EACH ITERATION  

C  

C          G(J): CHEMICAL POTENTIAL OF SPECIES J  

C  

C          W0(J): STORED COMPOSITIONS OF SPECIES J IF J IS IN  

C          A SOLUTION PHASE THAT IS NOT PRESENT.  

C          (FOR USE IN SUB-MINIMIZATION PROBLEM)  

C  

C          H(I,J): HESSIAN MATRIX  

C  

C          B(J): PROJECTED NON-DEGENERATE MASS AND CHARGE  

C          BALANCE VECTOR (Q1 (TRANSPOSE) X)  

C  

C          S(J): IMPOSED CONSTRAINT BOUNDARIES ON NON-CONVEX  

C          SOLUTION PHASES (USED IN THE SUB-MINIMIZATION  

C          PROBLEM)

```

```

C-----  

C      COMMON/PRINT/PP,PP2,PD  

C-----  

C  

C      PP: PRINT (MOSTLY) DIAGNOSTICS IN PRIMAL INCLUDING THE  

C          MOLE NUMBERS, FREE ENERGY, STEP DIRECTION, CHEMICAL  

C          POTENTIALS AT EACH NEWTON ITERATION AS WELL AS THE  

C          TYPE OF STEP (FEASIBILITY, PHASE ADDITION, PHASE  

C          DELETION, REFINEMENT, OR NORMAL STEP).  

C  

C      PP2: PRINT SOLUTION TO THE PROBLEM  

C  

C      PD: PRINT (MOSTLY) DIAGNOSTICS IN DUAL. THE DIGANOSTICS  

C          THAT ARE PRINTED ARE THE SAME VARAIBLES AS PRINTED  

C          BY PP EXCEPT HERE THEY APPLY TO THE DUAL SUBROUTINE  

C          (THE SUB-MINIZATION PROCEDURE)  

C  

C-----  

C      COMMON/CONST2/RR(96,16)  

C-----  

C  

C      RR(J,J): UNFACTORED CONSTRAINT MATRIX (INCLUDING  

C          DEGENERATE CONSTRAINTS)  

C  

C-----  

C      COMMON/PH/LACT,NFILE,P0  

C-----  

C      COMMON/MMM/MUL,WWW(96)  

C-----  

C  

C      MUL: PRINT MULTIPLIERS (SEE THE FIRST STATEMENT)  

C  

C      WWW(J): MULTIPLIER OF SPECIES J  

C  

C-----  

C      PARAMETER(BMIN=1.0D-07,DEL9=1.0D-03)  

C      PARAMETER(DELMP=0.5D0**49)  

C      PARAMETER(DEL2=1.0D-12)  

C-----  

C  

C      BMIN: THE MINIMUM AMOUNT (NON-SCALED) OF A PARTICULAR  

C          COMPONENT  

C  

C      DEL9: MAXIMUM CHARGE IMBALANCE (IF GREATER THEN THIS,  

C          EXIT, IF SMALLER, CORRECT IT.  

C  

C      DELMP: APPROXIMATE MACHINE PRECISION SCALED TO ONE  

C          (I.E. .000000000000001 MEANS 16 DIGITS OF PRECISION)  

C  

C      DEL2: CONSTRAINT BOUNDARY VALUE  

C  

C      DEL(J): PROJECTED MASS BALANCE STEP DIRECTION  

C          (Q1 (TRANSPOSE) (DELTA)X) THE PROJECTED CHANGE IN  

C          THE ABSOLUTE NUMBER OF MOLES READ FROM THE

```

C INPUT FOR EACH PROBLEM. (E.G. IF "SIG" FOR SPECIES "1"  
C (SEE THE EXPLANATION OF LINES 14-16 IN THE INPUT  
C EXAMPLE) IS EQUAL TO ".1", AND ALPHA IS EQUAL TO  
C ONE, ".1" MOLES OF SPECIES "1" WILL BE ADDED AFTER  
C EACH ITERATION. DEL(J) STORES THE Q1 PROJECTED  
C VALUES OF THESE MASS BALANCE CHANGES. SEE  
C ALSO "PF".

C BB(J): UNPROJECTED, MASS AND CHARGE BALANCE ARRAY  
C (INCLUDING DEGENERATE CONSTRAINTS)

C U0(J): STANDARD CHEMICAL POTENTIAL FOR P = 1. I.E. IF P=2  
C U0RT(J)=U0(J)+LN(2.)

C JH(J)=I : VECTOR THAT MAPS SPECIES NUMBER J THAT INCLUDE  
C SPECIES THAT ARE WITHHELD TO THE ACTUAL SPECIES C  
C NUMBER I. IF THE SPECIES IS WITHHELD, I IS EQUAL TO -1

C EXAMPLE:  
C SPECIES 5 IS WITHHELD

C INPUT SPECIES NUMBER	C ACTUAL SPECIES NUMBER (JH)
C 1	C 1
C 2	C 2
C 3	C 3
C 4	C 4
C 5	C -1
C 6	C 5
C 7	C 6
C 8	C 7

C PF(J): MASS BALANCE STEP DIRECTION FOR PERCENT CHANGE  
C IN SPECIES J (I.E. AN EVAPORATION)

C I=JFRAX(J): IF I IS EQUAL TO ONE, SPECIES J ( A MINERAL)  
C WILL NOT BE FRACTIONATED. IF I IS EQUAL TO  
C ZERO, SPECIES J CAN BE FRACTIONATED.

C WATLAB: CHARACTER ARRAY THAT STORES A LABEL THAT  
C IDENTIFIES A PARTICULAR COMPOSITION

C-----  
C LOGICAL PP,PP2,PD,PD2,MUL

C SAS

C OPEN(UNIT=4, FILE='SALT', STATUS='OLD')  
C OPEN(UNIT=5, FILE='THERMO.DAT', STATUS='OLD')  
C REWIND(4)  
C REWIND(5)  
C INQUIRE(FILE='SOLUTION', EXIST=OUTFIL)

```

IF(OUTFIL) THEN
  WRITE(*,5058)
5058      FORMAT( 1X,'THE OUTPUT FILE "SOLUTION" ALLREADY
EXISTS'
1 1X,'FROM A PREVIOUS SIMULATION. YOU CAN USE THE'
2 1X,'DOS RENAME COMMAND NOW OR JUST HIT THE RETURN'
3 1X,'KEY TO OVERWRITE THE OLD FILE. PROGRAM EXECUTION'
4 1X,'WILL RUSUME AUTOMATICALLY EITHER WAY.')
PAUSE
OPEN(UNIT=6, FILE='SOLUTION', STATUS='UNKNOWN')
CLOSE(UNIT=6, STATUS='DELETE')
END IF
OPEN(UNIT=6,FILE='SOLUTION',STATUS='NEW')
REWIND(6)

C SAS
C
C       READ(4,5002) IPRINT,IPR,IPR2,IPR3
C-----
C
C       FORMAT(6I5)
C       ALL PRINT OPTION LOGICAL VARIABLES ARE INITIALIZED TO
C       .FALSE.
C       IPRINT = 1:  PP SET EQUAL TO .TRUE.
C       IPR = 1:    PP2 SET EQUAL TO TRUE.
C       IPR2 = 1:   PD  SET EQUAL TO .TRUE.
C       IPR3 = 1:   MUL SET EQUAL TO .TRUE.
C
C       THE LOGICAL VARIABLES ARE EXPLAINED ABOVE
C
C-----
C       WRITE(*,*) 'CALLING DATA'
C       CALL DATA(JFRAX,M00,JH)

IF(M00.LT.NC0C) THEN
C
C       THIS "IF" STATEMENT SHOULD NEVER EXECUTE
C
C       WRITE(6,9988)
9988      FORMAT('ERROR IN MATRIX FACTORIZATION ROUTINE')
STOP
END IF
PP=.FALSE.
PP2=.FALSE.
PD=.FALSE.
PD2=.FALSE.
MUL=.FALSE.
IF(IPRINT.EQ.1) PP=.TRUE.
IF(IPR.EQ.1) PP2=.TRUE.
IF(IPR2.EQ.1) PD=.TRUE.
IF(IPR3.EQ.1) MUL=.TRUE.
NS=NSC
DO 30 J=1,NS
U0(J)=U0RT(J)

```

```

30  CONTINUE
    IWAT=0
    REWIND(4)
500  READ(4,5007) NTR,ALPHA,FRAZ,ALPHA2
C-----
C
C      IWAT : THE NUMBER OF THE CURRENT BRINE SYSTEM BEING
C      CONSIDERED (E.G. IF THE EVAPORATION SEQUENCE OF BRINE NO.
C      1 IS BEING CALCULATED IWAT =1. IF BRINE NO. 2 IS THEN READ
C      IN AND EVAPORATED, IWAT=2). THIS VARIABLE IS USED ONLY FOR
C      THE PURPOSE OF OUTPUT.
C
C      NTR: THE NUMBER OF ITERATIONS (I.E. THE NUMBER OF
C      EVAPORATION STEPS)
C
C      ALPHA: THE STEP LENGTH FOR THE ABSOLUTE NUMBER OF
C      MOLES THAT WILL CHANGE AT EACH STEP
C
C      FRAZ: THE FRACTIONATION FACTOR (BETWEEN 0. AND 1.)
C
C      ALPHA2: THE STEP LENGTH FOR THE PERCENT NUMBER OF
C      MOLES THAT WILL CHANGE AT EACH STEP
C
C      FORMAT(I5,3F10)
C
C      SEE THE EXPLANATION OF THE READ AFTER THE NEXT ONE
C      FOR AN EXAMPLE.
C
C-----
C      READ(4,5001) WATLAB
C-----
C
C      A LABEL THAT IS WRITTEN TO THE OUTPUT FILE TO IDENTIFY
C      THE SOLUTION FORMAT(A30)
C
C-----
10   IWAT=IWAT+1
    WRITE(6,5000) IWAT,IWAT,IWAT
    WRITE(6,5001) WATLAB
    N00=N00C
    N0E=N0EC
    N0EP1=N0E+1
    NC0=NC0C
    DO 10 I=1,M00
    B(I)=0.0D0
    BB(I)=0.0D0
    DEL(I)=0.0D0
10   CONTINUE
    DO 13 K=1,NP0C
    KTYP(K)=-IABS(KTYP(K))
13   CONTINUE
    DO 11 J=1,NS
    X(J)=0.0D0
    P(J)=0.0D0

```

```

PF(J)=0.0D0
IF(MACT(J).GT.NC0) THEN
  WRITE(*,*) 'CALLING CONDEL FROM MAIN'
  CALL CONDEL(J)
END IF

C
C      WHEN A NEW COMPOSITION IS READ IN, RELAX ALL SPECIES
C      CONSTRAINTS
C
C
11  CONTINUE
BBMAX=0.0D0
WRITE(6,5004) NTR,ALPHA,FRAX,ALPHA2
IF(FRAX.LT.0.0D0.OR.FRAX.GT.1.0D0) GO TO 9995
WRITE(6,5010)
PHI=0.0D0
PHI2=0.0D0

C SAS
C   REWIND INPUT FILE AND READ STARTING CONCENTRATIONS, ETC.
J=1
BLANK=.FALSE.
REWIND(4)
READ(4,5056)
5056 FORMAT(//////////)
100 READ(4,5016) NSPEC,TAU,SIG,ETA,ZETA

C-----
C
C      FORMAT(I5,4F15)

C      J: SPECIES NUMBER. SPECIES NUMBERS ARE ASSIGNED TO
C      SPECIES IN ORDER AS THEY ARE READ IN WITHIN DATA.F

C      TAU: INITIAL MOLE NUMBER OF SPECIES J

C      SIG: ABSOLUTE CHANGE AT EACH STEP IN THE MASS
C            REPRESENTED BY SPECIES J (MULTIPLIED BY ALPHA)

C      ETA: THE NATURAL LOG OF THIS VARIABLE (IF NON-ZERO)
C            IS ADDED TO THE CHEMICAL POTENTIAL OF SPECIES J.
C            THE MOST COMMON USE OF THIS VARIABLE IS TO
C            FIX A GASES FUGACITY.

C      ZETA: PERCENT CHANGE AT EACH STEP IN THE MASS
C            REPRESENTED BY SPECIES J (MULTIPLIED BY ALPHA2)

C      EXAMPLE: DO FOUR STEPS EVAPARATING A 2M. SODIUM
C      CHLORIDE 1M. SODIUM SULFATE SOLUTION IN EQUILIBRIUM WITH
C      A CARBON DIOXIDE ATMOSPHERE AT A PRESSURE OF .00034 ATM.
C      WHERE AT EACH STEP THE SOLIDS ARE FRACTIONATED 10%, 30%
C      OF THE WATER IS REMOVED AT EACH STEP, AND .1 MOLES OF
C      SODIUM SULFATE ARE ADDED AT EACH STEP
C

```

```

C      SPECIES   SPECIES NUMBER
C
C      H2O          1
C      NA+          2
C      CL-          3
C      SO4--         4
C      CO2(AQ)      5
C      HALITE       6
C      THENARDITE    7
C      MIRABALITE    8
C      CO2(GAS)      9
C
C      THEN NTR=1, ALPHA= 1., FRAX=.1, ALPHA2=.3
C
C      SPECIES     J      TAU      SIG      ETA      ZETA
C
C      H2O          1      55.508   0.0           -1.
C      NA+          2      4.          .2           0.0
C      CL-          3      2.          0.           0.0
C      SO4-2        4      1.          .1           0.
C      CO2 (GAS)    9      10.         0.0      .00034   0.0
C
C      THIS COMPLETES THE INPUT: THE FOLLOWING IS A SAMPLE INPUT
C      FILE FOR A SIMPLE SODIUM CHLORIDE WATER SOLUTION AT 25
C      DEGREES. (THE NUMBERS ON THE LEFT ARE USED WOULD NOT BE IN
C      THE INPUT FILE. THEY ARE PUT HERE ONLY FOR REFERENCE IN
C      THIS EXAMPLE. THE "\\" SYMBOL REPRESENTS THE LEFT BORDER OF
C      THE FILE AND SHOULD NOT BE IN THE INPUT FILE EITHER.
C
C-----  

IF(NSPEC.EQ.0) THEN  

  IF(BLANK) GO TO 101  

  BLANK=.TRUE.  

ELSE IF(NSPEC.LT.10) THEN  

  BLANK=.FALSE.  

ELSE IF(TAU.EQ.0.0D0) THEN  

  BLANK=.FALSE.  

  J=J+1  

ELSE  

  JJ=JH(J)  

  PF(JJ)=ZETA  

  IF(JJ.EQ.-1) THEN  

    WRITE(6,5555) J  

    STOP  

  END IF  

  IF(ETA.NE.0.0D0) THEN  

    IF(IABS(KTYP(KSPEC(JJ))).NE.1) THEN  

      WRITE(6,9989)  

9989      1      FORMAT('CHANGES IN THE STANDARD CHEMICAL',  

2      'POTENTIALS OF SOLUTION SPECIES ARE NOT'  

      'ALLOWED')  

      STOP  

    END IF  

    P0=ETA

```

```

        U0RT(JJ)=U0(JJ)+DLOG(ETA)
        G(JJ)=U0RT(JJ)
    END IF
    K0=KSPEC(JJ)
    KTYP(K0)=IABS(KTYP(K0))
    WRITE(6,5012) JJ,K0,A1(JJ),TAU,SIG,ETA,ZETA
    X(JJ)=X(JJ)+TAU
    PHI=PHI+SIG*Z(JJ)
    PHI2=PHI2+ZETA*Z(JJ)
    DO 12 I=N0EP1,NC0
    DEL(I)=DEL(I)+SIG*Q(JJ,I)
12     CONTINUE
    J=J+1
    BLANK=.FALSE.
    END IF
    GO TO 100
101   CONTINUE
    IF(DABS(PHI).GT.DELMP.OR.DABS(PHI).GT.DELMP) THEN
    WRITE(6,5050) PHI,PHI2
5050   FORMAT('STEP DIRECTION NOT CHARGE BALANCED ',1P2D12.3)
    STOP
    END IF
    DO 104 K=1,NP0C
    IF(KTYP(K).GT.0) THEN
    CHARGE=.FALSE.
    JM=JMIN(K)
    JX=JMAX(K)
    SIG=0.0D0
    PHI=0.0D0
    ICMAX=0
    IAMAX=0
    ZCMAX=0.0D0
    ZAMAX=0.0D0
    DO 106 J=JM,JX
    DO 116 I=1,M00
    BB(I)=BB(I)+RR(J,I)*X(J)
116   CONTINUE
    IF(Z(J).NE.0) THEN
        PHI=PHI+X(J)*Z(J)
        IF(Z(J).LT.0) THEN
            IF(X(J).GT.ZAMAX) THEN
                IAMAX=J
                ZAMAX=X(J)
            END IF
            ELSE IF(Z(J).GT.0) THEN
                IF(X(J).GT.ZCMAX) THEN
                    ICMAX=J
                    ZCMAX=X(J)
                END IF
            END IF
        END IF
    END IF
106   CONTINUE
    IF(DABS(PHI).GT.DELMP) THEN
        IF(DABS(PHI).GT.DEL9*DMAX1(ZCMAX,ZAMAX)) THEN

```

```

      WRITE(6,5052) PHI
5052    FORMAT('SOLUTION NOT CHARGE BALANCED ',1PD12.3)
      STOP
      ELSE
        CHARGE=.TRUE.
        II=0
        IF(PHI.LT.0.0D0) THEN
          X(ICMAX)=X(ICMAX)-PHI/Z(ICMAX)
          II=ICMAX
        ELSE
          X(IAMAX)=X(IAMAX)-PHI/Z(IAMAX)
          II=IAMAX
        END IF
        IF(II.EQ.0) THEN
C
C   THIS IF STATEMENT SHOULD NEVER BE TRUE
C
          WRITE(6,9990)
9990    FORMAT('NO SPECIES FOUND FOR CORRECTING',
           'CHARGE BALANCE')
          STOP
        ENDIF
        DO 117 I=1,M00
        BB(I)=BB(I)-RR(II,I)*PHI/Z(II)
117      CONTINUE
        END IF
        END IF
        DO 109 I=N0EP1,NC0
        DO 108 J=JM,JX
        B(I)=B(I)+Q(J,I)*X(J)
108      CONTINUE
109      CONTINUE
        IF(CHARGE) THEN
          WRITE(6,5015)
5015    FORMAT(//,'SOLUTION PHASE CHARGE BALANCE CORRECTION')
          WRITE(6,5017)
          DO 118 J=JM,JX
          IF(X(J).GT.0.0D0) WRITE(6,5012) J,K,A1(J),X(J)
118      CONTINUE
          END IF
        END IF
104      CONTINUE
        DO 107 I=N0EP1,M00
        IF(DABS(BB(I)).LT.BMIN) THEN
          WRITE(6,5051) I,BB(I)
5051    FORMAT('MASS BALENCE ERROR ',I5,2X,1PD12.3)
          STOP
        END IF
107      CONTINUE
        DO 600 ITR=1,NTR
        IF(ITR.EQ.1) ERROR=.TRUE.
        WRITE(*,*) 'CALLING PRIMAL'
        CALL PRIMAL(ITR,ERROR)
        IF(ERROR) THEN

```

```

      WRITE(6,5014) ITR,ITR,ITR
      STOP
      ELSE
        IF(FRAX.GT.0.0D0) THEN
          DO 28 K=1,NP0C
            IF(KTYP(K).EQ.1) THEN
              IF(JFRAX(K).EQ.0) THEN
                J0=JMIN(K)
                TAU=X(J0)*FRAX
                X(J0)=X(J0)-TAU
                DO 29 I=N0EP1,NC0
                  B(I)=B(I)-TAU*Q(J0,I)

29       CONTINUE
        IF(X(J0).LE.DEL2) KTYP(K)=-IABS(KTYP(K))
        END IF
      END IF
      CONTINUE
    END IF
    DO 36 J=1,NS
    DO 26 I=N0EP1,NC0
      B(I)=B(I)+ALPHA2*X(J)*PF(J)*Q(J,I)
26     CONTINUE
36     CONTINUE
    DO 27 I=N0EP1,NC0
      B(I)=B(I)+ALPHA*DEL(I)
27     CONTINUE
    END IF
600   CONTINUE
      WRITE(6,2500)
2500   FORMAT('NORMAL TERMINATION OF EQUIL')
      STOP

9995  WRITE(6,5025) FRAX
      STOP
5002  FORMAT( / 7X,I5 / 4X,I5 / 5X,I5 / 5X,I5 / 5X,I5)
5003  FORMAT(4F15.7)
5001  FORMAT( 5X,A50)
5000  FORMAT(3(15('`'),I5),15(``'))
5004  FORMAT('NTR,ALPHA,FRAX,ALPHA2= ',I5,3F10.5)
5006  FORMAT(F10.5)
5007  FORMAT( ////////////// 4X,I5 / 6X,F10.5/ 5X,F10.5/ 7X,F10.5)
5008  FORMAT(/,2X,'NO.',10X,'B(INPUT)',6X,'B(PROJ)',17X,'DEL(INPUT',
1 4X,'DEL(PROJ)')
5009  FORMAT(I5,2(5X,1P2D13.5,0PF6.3))
5010  FORMAT(11X,'SPECIES           X           P           A ')
5017  FORMAT(11X,'SPECIES           X')
5012  FORMAT(2I5,1X,A8,1P4D12.5)
5013  FORMAT(' TEMPERATURE = ',F10.4,' PRESSURE= ',F10.4)
5014  FORMAT('*****',3I5,'*****')
5016  FORMAT(I3,4F15.7)
5025  FORMAT('FRAX = ',1PD12.3)
5555  FORMAT(' SPECIES NO. ',I5,' WITHHELD')
      END

```

```

C-----  

C THIS SUBROUTINE READS IN THE INFORMATION ABOUT EACH  

C SPECIES (NAME, CHARGE, STANDARD CHEMICAL POTENTIAL  

C DIVIDED BY RT, AND THE SPECIFICATION OF COMPONENTS) AND  

C FACTORS THE INITIAL CONSTRAINT MATRIX (I.E. CHARGE AND  

C MASS BALANCE BUT NO SPECIES CONSTRAINTS).  

C  

C-----  

C-----  

C SUBROUTINE DATA(JFRAX,M0,JH)  

C-----  

C  

C M0: TOTAL NUMBER OF MASS AND CHARGE BALANCE  

C CONSTRAINTS INCLUDING DEGENERATE ONES (SET TO M00 IN  

C AMAIN2.F)  

C  

C-----  

C IMPLICIT DOUBLE PRECISION(A-H,O-Z)  

C IMPLICIT INTEGER*2 (I-N)  

C INTEGER*2 Z  

C INTEGER*2 LC(100),JV(9)  

C CHARACTER*24 A1  

C LOGICAL BLANK,ELECT,POS,NEG,SOLN,IDEAL,NODEG,REDUN  

C-----  

C  

C BLANK: SET TO .TRUE. WHEN A BLANK LINE IS READ  

C  

C ELECT: SET TO .TRUE. IF THE INPUT PHASE BEING READ IS AN  

C ELECTROLYTE PHASE  

C  

C POS: SET TO .TRUE. IF SPECIES BEING READ IS POSITIVE  

C  

C NEG: SET TO .TRUE. IF SPECIES BEING READ IS NEGATIVE  

C  

C POS AND NEG ARE USED TO INSURE THAT FOR AN ELECTROLYTE  

C SOLUTION PHASE, BOTH POSITIVE AND NEGATIVE SPECIES ARE  

C DEFINED IN THE INPUT FILE.  

C  

C SOLN: SET TO .TRUE. IF INPUT PHASE BEING READ IS A SOLUTION  

C PHASE  

C  

C IDEAL: SET TO .TRUE. IF PHASE BEING READ IS AN IDEAL  

C SOLUTION  

C PHASE (THE VARIABLE IS SET BUT NOT USED)  

C  

C NODEG: SET TO .TRUE. IF DURING THE QR FACTORIZATION, A  

C CONSTRAINT IS FOUND TO BE NOT DEGENERATE  

C  

C REDUN: SET TO .TRUE. IF A SPECIES IS FOUND TO BE  

C DEGENERATE  

C-----  

C-----  

C COMMON/CONST2/RR(96,16)

```

```

COMMON/CONST/Q(96,96),MACT(96),JA(96),JI(96),NSC,NCC,NDC,NC0C
1,N0EC
COMMON/LABEL/Z(96),U0RT(96),A1(96)
COMMON/PHASE/R11(96,16),JMIN(96),JMAX(96),KSPEC(96),KTYP(96)
1,NP0C,N00C
COMMON/SEARCH/X(96),P(96),G(96),W0(96),H(64,64),B(96)
COMMON/DBASE/NCAT(18),NANI(18),ICAT,IANI,NCD,NAD,NND,TEMP
DIMENSION R(96,96),V(96),JAN(96),JH(96),JFRAX(96)

```

---

C-----  
C  
C R: STORE UNFACTORIED CONSTRAINT MATRIX. AS THE QR ALGORITHM  
C PROCEEDS IT BECOMES THE UPPER TRIANGULAR MATRIX. R IS A  
C LOCAL VARIABLE WHICH SHARES ITS SPACE WITH THE H MATRIX.  
C THE R MATRIX IS STORED GLOBALLY IN R11.

C  
C IAN: AN ARRAY THAT STORES THE CONSTRAINTS ACCORDING TO THE  
C NEW CONSTRAINT NUMBERS GIVEN BY LC. THIS ARRAY IS USED ONLY  
C TO PRINT THE CONSTRAINT TABLE ON THE OUTPUT

C  
C-----  
C

PARAMETER(DEL4=1.0D-13)

---

C-----  
C  
C DEL4: VALUE THAT DETERMINES CONSTRAIN DEGENERACY.  
C I.E. IF IN R WE HAVE:

C X X X  
C 0 X X  
C 0 0 0  
C 0 0 0  
C 0 0 0  
C 0 0 0

C  
C CONSTRANT 3 IS DEGENERATE SINCE THE DIAGONAL  
C ELEMENT IS EQUAL TO DEL4( VERY CLOSE TO ZERO)

---

C-----  
C  
C PARAMETER(NMAX=96,MMAX=16)  
C EQUIVALENCE (H(1,1),R(1,1))  
C  
C INITIALIZE Q MATRIX AND LC VECTOR  
C  
2000 FORMAT(/,1X,'SOLUTION PHASE',I2,3X,'PHASE TYPE',I2,/  
1,5X,'SPECIES',16X,'CHARGE',6X,'U0/RT')  
2001 FORMAT(/,1X,'PURE PHASES',/,5X,'SPECIES',32X,'U0/RT')  
2004 FORMAT(/,1X,'IDEAL SOLUTION PHASE',I2,/  
1,5X,'SPECIES',8X,'CHARGE',10X,'U0/RT')  
DO 1 J=1,100  
LC(J)=0  
1 CONTINUE  
DO 2 I=1,NMAX  
DO 2 J=1,NMAX  
Q(J,I)=0.0D0  
2 CONTINUE

```

C
C -----
C      SYSTEM DATA INPUT LOOP
C
C      (A) READ SOLUTION PHASES
C          (1) IDENT LINE FOR PHASE IDENTIFICATION
C              IDENT = 2 3 AND 4
C          (2) SPECIES LINES
C              A) READ SPECIES ID NUMBER
C                  B) SEARCH DATABASE FOR COMPOSITION AND
C                      CHEMICAL POTENTIAL DATA
C                  C) REPEAT FOR NEXT SPECIES
C          (3) ZERO LINE
C          (4) NON-IDEAL SOLUTION DATA READ IN DATAP
C              IF IDENT > 2 (IDENT = 2 FOR IDEAL SOLUTION)
C                  A) SEARCH DATABASE FOR PITZER DATA
C
C      (B) READ PURE PHASES
C          (1) IDENT LINE IDENT = 1
C          (2) PURE PHASE LINES
C              A) READ SPECIES I.D. NUMBER
C                  B) SEARCH DATABASE ETC.
C                  C) REPEAT FOR NEXT SPECIES
C          (3) ZERO LINE
C
C      NCAT(ICAT)=SPECIES I.D. NUMBER. VECTOR THAT MAPS THE Ith
C      CATION TO ITS DATABASE IDENTIFICATION NUMBER
C
C      NANI(I)=SPECIES I.D.NUMBER. VECTOR THAT MAPS THE Ith ANION
C      TO ITS DATABASE IDENTIFICATION NUMBER
C
C
C      WHEN IDENT IS GREATER THAN 2 THE SUBROUTINE DATAP IS
C      CALLED TO READ IN NON-IDEAL SOLUTION MODEL PARAMETERS.
C      THE PARAMETER IDENT IS USED TO IDENTIFY THE TYPE OF
C      MODEL TO BE USED FOR THE PHASE. FOR NON-IDEAL SOLUTION
C      PHASES GRAD,HESS,AND DATAP USE IDENT.
C
C      NCOMP=0
C      NP0=0
C      M0=0
C      J=1
C      SOLN=.TRUE.
C
C      SAS
C          READ TOTAL NUMBER OF CATIONS, ANIONS, AND NEUTRALS IN
C          DATABASE
C          READ(5,1002) NCD,NAD,NND
C
C          READ SYSTEM TEMPERATURE (CELCIUS)
C          READ(4,1003) TEMP
C
C          ICAT=0

```

```

IANI=0

      READ(4,1008)
100  READ(4,1000) IDENT
C-----
C
C      IDENT: PHASE TYPE:
C      1 PURE PHASE
C      2 IDEAL SOLUTION PHASE
C      3 AQUEOUS SOLUTION PHASE WITH DAVIES ACTIVITY
C          COEFFICIENTS
C      4 AQUEOUS SOLUTION PHASE WITH PITZER ACTIVITY
C          COEFFICIENTS
C      FORMAT(I5)
C SAS
C-----
1000 FORMAT( I3,60X)
IF(IDENT.LT.1) GO TO 103

106  NP0=NP0+1
KTYP(NP0)=-IDENT
C
C      KTYP IS INITIALIZED TO ALL NEGATIVE NUMBERS. THAT IS, ALL
C      PHASES ARE NOT PRESENT. WHEN MOLE NUMBERS FOR SPECIES
C      ARE READ IN IN AMAIN2.F, KTYP FOR THE CORRESPONDING PHASE
C      IS SET TO POSITIVE.
C
C      JMIN(NP0)=J
C      ELECT=.FALSE.
C      NSPEC=0
C      IF(IDENT.GT.1) THEN
C          IF(.NOT.SOLN) THEN
C              WRITE(6,990)
990      FORMAT('SOLUTION PHASE CANNOT FOLLOW A PURE PHASE')
C          STOP
C      ENDIF
C      NSOLNC=NP0
C      IF(IDENT.EQ.2) THEN
C          WRITE(6,2004) NP0
C      ELSE
C          WRITE(6,2000) NP0,IDENT
C      END IF
C      ELSE IF(IDENT.EQ.1.AND.SOLN) THEN
C          SOLN=.FALSE.
C          WRITE(6,2001)
C          NP0=NP0-1
C          JJ=J
C      END IF
C
C SAS
C      SEARCH DATABASE FOR SPECIES INDICATED IN INPUT FILE
99   READ(4,1004) NSPEC1
BLANK=.FALSE.
IF (NSPEC1.EQ.0) THEN

```

```

BLANK=.TRUE.
GO TO 101
ELSE IF (NSPEC1.LT.99) THEN
  WRITE(*,1006) NSPEC1
  STOP
ELSE IF (NSPEC1.EQ.99) THEN
  READ(5,1005) NSPEC2, A1(J), Z(J), (V(K),JV(K),K=1,7),
1    U0RT0, U0RT1, U0RT2, U0RT3, U0RT4
  IF (NSPEC2.NE.99) THEN
    WRITE(*,1006) NSPEC1
    STOP
  END IF
  GO TO 101
ELSE IF (NSPEC1.LT.200) THEN
50   READ(5,1005) NSPEC2, A1(J), Z(J),(V(K),JV(K),K=1,7),
1     U0RT0, U0RT1, U0RT2, U0RT3, U0RT4
  IF (NSPEC2.EQ.NSPEC1) THEN
    ICAT=ICAT+1
    NCAT(ICAT)=NSPEC2
    GO TO 101
  ELSE IF (NSPEC2.GT.NSPEC1) THEN
    WRITE(*,1006) NSPEC1
    STOP
  ELSE
    GO TO 50
  END IF
ELSE IF (NSPEC1.LT.300) THEN
51   READ(5,1005) NSPEC2, A1(J), Z(J), (V(K),JV(K),K=1,7),
1     U0RT0, U0RT1, U0RT2, U0RT3, U0RT4
  IF (NSPEC2.EQ.NSPEC1) THEN
    IANI=IANI+1
    NANI(IANI)=NSPEC2
    GO TO 101
  ELSE IF (NSPEC2.GT.NSPEC1) THEN
    WRITE(*,1006) NSPEC1
    STOP
  ELSE
    GO TO 51
  END IF
ELSE IF (NSPEC1.LT.400) THEN
52   READ(5,1005) NSPEC2, A1(J), Z(J), (V(K),JV(K),K=1,7),
1     U0RT0, U0RT1, U0RT2, U0RT3, U0RT4
  IF (NSPEC2.EQ.NSPEC1) THEN
    GO TO 101
  ELSE IF (NSPEC2.GT.NSPEC1) THEN
    WRITE(*,1006) NSPEC1
    STOP
  ELSE
    GO TO 52
  END IF
ELSE
53   READ(5,1007) NSPEC2, A1(J), (V(K),JV(K),K=1,7),
1     U0RT0, U0RT1, U0RT2, U0RT3, U0RT4
  IF (NSPEC2.EQ.NSPEC1) THEN

```

```

        GO TO 101
ELSE IF (NSPEC2.GT.NSPEC1) THEN
    WRITE(*,1006) NSPEC1
    STOP
ELSE IF (NSPEC2.EQ.0) THEN
    WRITE(*,1006) NSPEC1
    STOP
ELSE
    GO TO 53
END IF
END IF

1002 FORMAT( ///// I4 / I4 / I4 /)
1003 FORMAT( 23X, F10.5)
1004 FORMAT(I4)
1005 FORMAT(I4, A24, I2, 7(F4.3,I3) /14X, 5E13.5)
1006 FORMAT(1X,'SPECIES NUMBER ',I3,'NOT FOUND IN DATABASE OR',
1 ' INPUT NOT IN ASCENDING ORDER')
1007FORMAT(I4,A24,2X,7(F4.3,I3) /14X,5E13.5)
1008FORMAT( /////)
101   CONTINUE

```

C SAS

```

IF(.NOT.SOLN) THEN
    IF(BLANK) GO TO 103
    JFRAX(NP0+1)=0
    IF(Z(J).GT.0) THEN
        IF(Z(J).EQ.1) THEN
            WRITE(6,5999) A1(J)
            JH(JJ)=-1
            JJ=JJ+1
            GO TO 101
        ELSE IF(Z(J).EQ.2) THEN
            JFRAX(NP0+1)=1
        ELSE
            WRITE(6,991) Z(J)
            FORMAT('ILLEGAL PURE PHASE OPTION',I5)
        END IF
    END IF
    NP0=NP0+1
    Ktyp(NP0)=-1
    JMIN(NP0)=J
    JMAX(NP0)=J
    KSPEC(J)=NP0
    WRITE(*,*) 'CALLING DATAP @ LN 341'
    CALL DATAP(NP0)
    X(J)=0.0D0
    P(J)=0.0D0

```

C SAS

```

U0RT(J)=U0RT0 +U0RT1*(TEMP-25.0D0) +U0RT2*(TEMP-25.0D0)**2
1+U0RT3*(TEMP-25.0D0)**3 +U0RT4*(TEMP-25.0D0)**4

```

C SAS

```

G(J)=U0RT(J)

```

```

      W0(J)=1.0D0
      Z(J)=0
      JH(JJ)=J
      JJ=JJ+1
      WRITE(6,2007) J,A1(J),U0RT(J)
2007    FORMAT(1X,I3,1X,A24,7X,F17.8)
      ELSE
      IF(BLANK) GO TO 102
          JH(J)=J
          KSPEC(J)=NP0
          X(J)=0.0D0
          P(J)=0.0D0
C SAS
      U0RT(J)=U0RT0 +U0RT1*(TEMP-25.0D0) +U0RT2*(TEMP-25.0D0)**2
      1+U0RT3*(TEMP-25.0D0)**3 +U0RT4*(TEMP-25.0D0)**4
C SAS
      G(J)=U0RT(J)
      W0(J)=.01D0
      IF(Z(J).NE.0) ELECT=.TRUE.
      NSPEC=NSPEC+1
      WRITE(6,2003) J,A1(J),Z(J),U0RT(J)
2003    FORMAT(1X,I3,1X,A24,I2,F17.8)
      END IF
      DO 4 K=1,9
      IF(JV(K).GT.0) THEN
          IF(LC(JV(K)).LE.0) THEN
              NCOMP=NCOMP+1
              LC(JV(K))=NCOMP
          END IF
          Q(J,LC(JV(K)))=V(K)
      END IF
4     CONTINUE
      J=J+1
      GO TO 99
102    JMAX(NP0)=J-1
      W0(JMIN(NP0))=1.0D0

      IF(NSPEC.LT.2) THEN
          WRITE(6,992)
992        FORMAT('A SOLUTION PHASE MUST HAVE MORE THEN ONE
SPECIES')
          STOP
      ENDIF
      IF(ELECT) THEN
          M0=M0+1
          IAN(M0)=0
          JM=JMIN(NP0)
          JX=JMAX(NP0)
          POS=.TRUE.
          NEG=.TRUE.
          DO 5 I=1,NMAX
              R(I,M0)=0.0D0
              RR(I,M0)=R(I,M0)
5         CONTINUE

```

```

DO 7 I=JM,JX
R(I,M0)=Z(I)
RR(I,M0)=R(I,M0)
IF(Z(I).LT.0) NEG=.FALSE.
IF(Z(I).GT.0) POS=.FALSE.
7 CONTINUE
IF(POS.OR.NEG) THEN
  WRITE(6,993)
  FORMAT('YOU MUST HAVE BOTH POSITIVE AND NEGATIVE
993 IONS')
  STOP
ENDIF
END IF

C
C IDENT=1 PURE PHASE
C IDENT=2 IDEAL SOLUTION
C IDENT=3 AQUEOUS DAVIES EQN
C IDENT=4 AQUEOUS PITZER EQN
C
WRITE(*,*) 'CALLING DATAP @ LN 460'
CALL DATAP(NP0)
GO TO 100
C
C -----
C
C FORM CONSTRAINT MATRIX IN R
C
103 NS=J-1
MX=M0
DO 11 I=1,100
IF(LC(I).EQ.0) GO TO 11
M0=M0+1
IF(M0.GT.MMAX) THEN
  WRITE(6,5500) M0
5500 FORMAT(I5,' TOO MANY CONSTRAINTS,ENLARGE RR AND R11
ARRAY')
  END IF
  IAN(M0)=I
  TAU=1.0D+20
  DO 10 J=1,NS
  R(J,M0)=Q(J,LC(I))
  RR(J,M0)=R(J,M0)
  IF(R(J,M0).GT.1.0D-05.AND.R(J,M0).LT.TAU) TAU=R(J,M0)
10 CONTINUE

C
C STORE SPECIES THAT HAS THE LEAST AMOUNT OF A GIVEN
C COMPONENT (USED TO CALCULATE THE SCALING FACTOR IN
C PRIMAL.F) SINCE WE WISH TO FIND THE MINIMUM VALUE, TAU IS
C INITIALIZED AT A VERY LARGE NUMBER.
C
C EXAMPLE: THE FOLLOWING SPECIES ALL CONTAIN HYDROGEN:
C
C H2O: 2 HYDROGENS
C

```

```

C MIRABALITE : 10 HYDROGENS
C GYPSUM : 4 HYDROGENS
C FOR THE HYDROGEN COMPONENT, TAU WOULD BE EQUAL TO "2"
C BECAUSE H2O HAS THE LEAST NUMBER OF HYDROGENS PER
C MOLE OF SPECIES.
C
C
C
B(M0)=TAU
11 CONTINUE
WRITE(6,2002) (IAN(K),K=1,M0)
2002 FORMAT(///,1X,'CONSTRAINT EQUATIONS',/,5X,'SPECIES',9X,20I5)
DO 12 J=1,NS
WRITE(6,2005) J,A1(J),(R(J,K),K=1,M0)
2005 FORMAT(1X,I3,1X,A16,20F5.1)
12 CONTINUE
C
C -----
C INITIALIZE ORTHOGONAL CONSTRAINT MATRIX
C
C INITIALIZE THE Q MATRIX TO THE IDENTITY MATRIX
C AND ALL VALUES OF THE MACT ARRAY EQUAL TO NEGATIVE
C NUMBERS (I.E. NO SPECIES CONSTRAINTS; SEE AMAIN2.F
C AND THE DICTIONARY). THUS ALL SPECIES ARE PRESENT
C (INACTIVE) AND THEY ALL HAVE CORRESPONDING ELEMENTS IN
C THE JI ARRAY. THEREFORE THERE ARE NO ELEMENTS
C (INITIALLY) IN THE JA ARRAY.
C
C
DO 14 I=1,NS
DO 13 J=1,NS
Q(J,I)=0.0D0
13 CONTINUE
JA(I)=0
JI(I)=I
MACT(I)=-I
Q(I,I)=1.0D0
14 CONTINUE
C
C -----
C QR FACTOR CONSTRAINT MATRIX (SEE STRANG (1980))
C
NC=0
DO 22 MM=1,M0
MP1=NC+1
SIG=0.0D0
NODEG=.FALSE.
DO 15 I=MP1,NS
V(I)=R(I,MM)
R(I,MM)=0.0D0
IF(DABS(V(I)).GT.DEL4) NODEG=.TRUE.
SIG=SIG+V(I)*V(I)

```

```

15  CONTINUE
    IF(NODEG) THEN
        NC=MP1
        SIG=DSIGN(DSQRT(SIG),V(NC))
        IF(NC.EQ.NS) THEN
            R(NC,MM)=V(NC)
        ELSE
            V(NC)=V(NC)+SIG
            PI=1.0D0/(V(NC)*SIG)
            R(NC,MM)--SIG
            R11(NC,MM)--SIG
            DO 18 J=1,NS
            TAU=0.0D0
            DO 16 I=NC,NS
            TAU=TAU+Q(J,I)*V(I)
16      CONTINUE
            TAU=TAU*PI
            DO 17 I=NC,NS
            Q(J,I)=Q(J,I)-TAU*V(I)
            IF(DABS(Q(J,I)).LT.DEL4) Q(J,I)=0.0D0
17      CONTINUE
18      CONTINUE
        IF(MM.LT.M0) THEN
            MMP1=MM+1
            DO 21 J=MMP1,M0
            TAU=0.0D0
            DO 19 I=NC,NS
            TAU=TAU+R(I,J)*V(I)
19      CONTINUE
            TAU=TAU*PI
            DO 20 I=NC,NS
            R(I,J)=R(I,J)-TAU*V(I)
20      CONTINUE
            R11(NC,J)=R(NC,J)
21      CONTINUE
            END IF
        END IF
    END IF
22  CONTINUE
C      NCP1=NC+1
DO 27 I=1,M0
    IF(I.GT.MX) THEN
        R11(NCP1,I) = B(I)
    ELSE
        R11(NCP1,I)=1.0D+20
    END IF
27  CONTINUE
NINACT=NS
ND=0
C      FIND SPECIES WHICH ARE REDUNDANT WITH CHARGE AND MASS
C      BALANCE CONSTRAINTS.
C

```

```

C      DO 28 J=1,NS
C      REDUN=.TRUE.

C      IF THE COLUMNS OF THE Q3 MATRIX (COLUMNS NCP1 THROUGH
C      NS OF THE FULL Q MATRIX) ARE ALL VERY CLOSE TO ZERO FOR
C      A GIVEN SPECIES (I.E. FOR A GIVEN ROW OF THE Q3 MATRIX),
C      THAT SPECIES IS DEGENERATE.

C
C
C      DO 29 I=NCP1,NS
C      IF(DABS(Q(J,I)).GT.DEL4) REDUN=.FALSE.

29  CONTINUE

C      IF THE NUMBER OF CHARGE AND MASS BALANCE CONSTRAINTS
C      EQUALS THE NUMBER OF SPECIES, ALL SPECIES ARE DEGENERATE
C      (THERE IS NO Q3 MATRIX).

C
C      IF(NS.EQ.NC) REDUN=.TRUE.
C      IF(REDUN) THEN
C          ND=ND+1
C          JA(ND)=J
C          IT--MACT(J)
C          JI(IT)=JI(NINACT)
C          MACT(JI(IT))=-IT
C          NINACT=NINACT-1
C          MACT(J)=ND
C      END IF

28  CONTINUE
NSC=NS
NCC=NC
NDC=ND
NC0C=NC
N00C=M0
N0EC=MX
NP0C=NP0
NPMAX=NC-MX
WRITE(6,2008) NC,NPMAX
2008 FORMAT(5X,'RANK=',I3,5X,'MAXIMUM PHASES=',I3)
RETURN

C
5999 FORMAT(80(''),/,'A24,' WITHHELD',/,'80(''))
END

```

```

C-----  

C      THIS SUBROUTINE READS IN THE NONIDEAL SOLUTION  

C      PARAMETERS.  

C  

C      THE INPUT ORDER IS: H2O,CATIONS,ANIONS,NEUTRALS.  

C-----  

SUBROUTINE DATAP(KPP)  

IMPLICIT DOUBLE PRECISION(A-H,O-Z)  

IMPLICIT INTEGER*2 (I-N)  

DOUBLE PRECISION IONIC,MOLAL  

INTEGER*2 Z,ZMAX  

CHARACTER*24 A1  

DOUBLE PRECISION DUMPSI(100)  

COMMON/LABEL/Z(96),U0RT(96),A1(96)  

COMMON/PHASE/R11(96,16),JMIN(96),JMAX(96),KSPEC(96),KTYP(96)  

1,NP0C,N00C  

  COMMON/AQSOL/JP(46),JMC,JXC,JMA,JXA,JMB,JXB,ALPHA(3)  

1,BETA,B0(18,18),B1(18,18),B2(18,18),CCA(18,18),TA(153)  

2,PSIA(153,18),TC(153),PSIC(153,18),MOLAL(46)  

3,F1(3),F2(3),F3(3),ZSUM  

C-----  

C      JP (J)=I: SPECIES J IN THE AQUEOUS SOLUTION IS THE ITH CATION,  

C      ANION, OR NEUTRAL SPECIES. THIS IS USED TO ACCESS THE  

C      SOLUTION PARAMETER ARRAYS AND MATRICES.  

C  

C      EXAMPLE:  

C      SPECIES (J)    JP(J)  

C      H2O        1      0  

C      NA         2      1  

C      K          3      2  

C      CL         4      1  

C      SO4        5      2  

C      HCO3       6      3  

C      CO2(AQ)   7      1  

C  

C      JMC: MINIMUM CATION SPECIES NUMBER  

C  

C      JXC: MAXIMUM CATION SPECIES NUMBER  

C  

C      JMA: MINIMUM ANION SPECIES NUMBER  

C  

C      JXA: MAXIMUM ANION SPECIES NUMBER  

C  

C      JMB: MINIMUM NEUTRAL SPECIES NUMBER  

C  

C      JXB: MAXIMUM NEUTRAL SPECIES NUMBER  

C  

C      ALPHA (J): THE THREE PITZER ALPHAS (ALPHA FOR 1-2 AND LESS,  

C                  ALPHA FOR GREATER THEN 1-2, AND ALPHA FOR THE  

C                  BETA2 TERM.

```

C      BETA: B VALUE IN PITZER LIMITING LAW.

C      B0 (I,J): PITZER'S B0 VALUE FOR CATION I AND ANION J  
 C            (REFERENCED BY JP)

C      B1 (I,J): PITZER'S B1 VALUE FOR CATION I AND ANION J  
 C            (REFERENCED BY JP)

C      B2 (I,J): PITZER'S B2 VALUE FOR CATION I AND ANION J  
 C            (REFERENCED BY JP)

C      CCA (I,J): PITZER'S CPHI VALUE FOR CATION I AND ANION J  
 C            (REFERENCED BY JP)

C      TA (J) : PITZER'S THETA VALUE FOR THE JTH ANION-ANION THETA  
 C            REFERENCED BY JP.

C      EXAMPLE: ANIONS CL,SO4, AND HCO3-:  
 C            TA(1) = THETA CL-SO4  
 C            TA(2) = THETA CL-HCO3  
 C            TA(3) = THETA SO4-HCO3

C      PSIA (J,I): PITZER'S PSI VALUE FOR THE JTH ANION-ANION AND ITH  
 C            CATION PSI REFERENCED BY JP.  
 C            EXAMPLE: CATIONS NA AND K, ANIONS CL, SO4, HCO3

C      PSIA(1,1) = PSI CL-SO4-NA  
 C      PSIA(1,2) = PSI CL-SO4-K  
 C      PSIA(2,1) = PSI CL-HCO3-NA  
 C      PSIA(2,2) = PSI CL-HCO3-K  
 C      PSIA(3,1) = PSI SO4-HCO3-NA  
 C      PSIA(3,2) = PSI SO4-HCO3-K

C      TC (J) : PITZER'S THETA VALUE FOR THE JTH CATION-CATION  
 C            THETA REFERENCED BY JPA (ANALAGOUS TO TA ABOVE).

C      PSIC (J,I) : PITZER'S PSI VALUE FOR THE JTH CATION-CATION AND  
 C            ITH ANION PSI REFERENCED BY JP (ANALAGOUS TO PSIA  
 C            ABOVE).

C      MOLAL (J) : THE MOLALITY OF SPECIES J.

C      F1 (J) : THE EXPONENTIAL TERM IN PITZER'S OSMOTIC  
 C            COEFFICIENT EXPRESSION FOR EACH VALUE OF ALPHA  
 C            (E.G. EXP(-ALPHA\*SQRT(I))).

C      F2 (J) : THE "G(X)" TERM IN PITZER'S LOG OF THE ACTIVITY  
 C            COEFFICIENT EXPERSSSIONS FOR EACH VALUE OF ALPHA.  
 C            (X = -ALPHA\*SQRT(I)).

C      F3 (J) : THE DERIVATIVE OF G(X) WITH RESPECT TO IONIC  
 C            STRENGTH.

C      ZSUM : THE SUM OF THE ABSOLUTE VALUES OF THE CHARGE ON

C EACH SPECIES TIMES ITS MOLALITY. (THIS TERM  
 C MULTIPLIES THE CPHI IN SOME INSTANCES IN THE PITZER  
 C EQUATIONS)

C-----  
 C COMMON/ELCT/APHI,IONIC,ETH(8,8),ETHP(8,8),ETHP2(8,8),ZMAX

C-----  
 C APHI : A PHI.

C-----  
 C IONIC : THE IONIC STRENGTH.

C-----  
 C ETH(I,J) : THE ELECTROSTATIC CONTRIBUTION TO THETA IN THE  
 C PITZER EQUATIONS. (THETA = ETH(J,J) + THETA). I AND J  
 C ARE THE REPRESENT THE ABSOLUTE VALUES OF THE  
 C LIKE CHARGES.

C-----  
 C ETHP (I,J) : THE FIRST DERIVATIVE OF ETH WITH RESPECT TO  
 C IONIC STRENGTH.

C-----  
 C ETHP2 (I,J) : THE SECOND DERIVATIVE OF ETH WITH RESPECT TO  
 C IONIC STRENGTH.

C-----  
 C ZMAX : THE ABSOLUTE VALUE OF THE HIGHEST CHARGE FOUND  
 C ON ANY ONE THE AQUEOUS SPECIES.

C-----  
 C SAS

C-----  
 C COMMON/DBASE/NCAT(18),NANI(18),ICAT,IANI,NCD,NAD,NND,TEMP

C-----  
 C NCAT : MATRIX OF DATABASE SPECIES NUMBERS OF CATIONS IN  
 C THE SYSTEM

C-----  
 C NANI : ANALOGOUS TO NCAT

C-----  
 C ICAT : TOTAL NUMBER OF CATIONS IN THE CURRENT SYSTEM

C-----  
 C IANI : TOTAL NUMBER OF ANIONS IN THE CURRENT SYSTEM

C-----  
 C NCD : TOTAL NUMBER OF CATIONS IN THE DATABASE (READ FROM  
 C DATABASE)

C-----  
 C NAD : TOTAL NUMBER OF ANIONS IN THE DATABASE

C-----  
 C NND : TOTAL NUMBER OF NEUTRALS IN THE DATABASE

C-----  
 C TEMP : SYSTEM TEMPERATURE

C-----  
 C SAS

1000 FORMAT(I3,1X,I3,3X,4F14.8)

1003 FORMAT(I3,1X,I3,8X, F9.5, 2(/ 7X,9F8.4))

```

1002 FORMAT(1X,'DATABASE ERROR IN PITZER SECTION. MISSING
1 LINES',2I4)
1004 FORMAT(2X,5F14.6)
1005 FORMAT(/////////)
1006 FORMAT()
1007 FORMAT(I3)
2001 FORMAT(//,10X,'NON-IDEAL AQUEOUS ELECTROLYTE SOLUTION
DATA')
3000 FORMAT(/,5X,'APHI=',F8.5,5X,'BETA=',F8.3,5X,'ALPHA=',3F8.3)
3001 FORMAT(/,24X,'B0',8X,'B1',8X,'B2',7X,'CPHI')
3002 FORMAT(1X,A8,1X,A8,4F10.5)
3003 FORMAT(/,19X,'THETA',4X,6(A8),(/ 32X,6A8))
3004 FORMAT(2A8,7(F8.4),(/ 28X,6F8.4))
3011 FORMAT(/,5X,'DAVIES CONSTANT A=',F8.6)
JTYPE=IABS(KTYP(KPP))
IF(JTYPE.LE.1.OR.JTYPE.EQ.2.AND.KPP.GT.1) RETURN
C
C      IF THE PHASE IS A PURE MINERAL OR AN IDEAL SOLUTION NON-
C      AQUEOUS PHASE, EXIT.
C
JM=JMIN(KPP)
JX=JMAX(KPP)
JMC=JM+1
JXC=0
JXA=0
JXB=0
ZMAX=0
IF(JX-JM.GT.46) THEN
    WRITE(6,9988) JX-JM+1
    GO TO 600
END IF
9988 FORMAT('TOO MANY AQUEOUS SPECIES ',I5)
DO 5 J=JMC,JX
IF(Z(J)) 2,3,4
2 JXA=J
IF(JXB.GT.0) GO TO 600
IF(ZMAX.LT.-Z(J)) ZMAX--Z(J)
JP(J)=J-JXC
IF(JP(J).GT.18) THEN
    WRITE(6,9989) J,JP(J)
    GO TO 600
END IF
9989 FORMAT('TOO MANY ANIONS ',2I5)
GO TO 5
3 JXB=J
JP(J)=J-JXA
IF(JP(J).GT.8) THEN
    WRITE(6,9990) J,JP(J)
    GO TO 600
END IF
9990 FORMAT('TOO MANY NEUTRALS ',2I5)
GO TO 5
4 JXC=J
IF(JXA.GT.0) GO TO 600

```

```

IF(JXB.GT.0) GO TO 600
IF(ZMAX.LT.Z(J)) ZMAX=Z(J)
JP(J)=J-JM
IF(JP(J).GT.18) THEN
  WRITE(6,9991) J,JP(J)
  GO TO 600
END IF
9991 FORMAT('TOO MANY CATIONS ',2I5)
5 CONTINUE
JMA=JXC+1
JMB=JXA+1
IF(JXA*JXC.EQ.0) GO TO 600
IF(JTYPE.LE.2) RETURN

C
C   IF THE PHASE IS AN AQUEOUS IDEAL SOLUTION PHASE, EXIT
C SAS
9 READ(5,1003) NSPEC
IF(NSPEC.NE.900) GO TO 9
C   ADVANCING DATA FILE TO PITZER SECTION
C
C   READ APHI POLYNOMIAL
C
READ(5,1004) APHI0,APHI1,APHI2,APHI3,APHI4
APHI=APHI0 + APHI1*TEMP + APHI2*TEMP**2 + APHI3*TEMP**3
1 +APHI4*TEMP**4

C   IF DAVIES MODEL IS USED REWIND DATA FILE AND RETURN
IF(JTYPE.EQ.3) THEN
  REWIND(5)
  READ(5,1005)
  WRITE(6,3011) APHI*3.0D0/DLOG(10.0D0)
  RETURN
END IF

C   THIS SECTION READS IN PITZER DATA
C   READ B, ALPHA0, ALPHA1, AND ALPHA2
READ(5,1004) BETA,(ALPHA(I),I=1,3)
READ(5,1006)
WRITE(6,2001)
WRITE(6,3000) APHI,BETA,(ALPHA(I),I=1,3)
WRITE(6,3001)
NC=JP(JXC)
NA=JP(JXA)

C
C   READ BETA0, BETA1, BETA2, AND CPHI
C   AND CONVERT CPHI TO C.
C
II=0
DO 10 I=JMC,JXC
JJ=0
II=II+1

```

```

DO 10 J=JMA,JXA
JJ=JJ+1
11 READ(5,1000) NSPECC,NSPECA,B0(JP(I),JP(J)),B1(JP(I),JP(J))
1,B2(JP(I),JP(J)),CPHI
IF (NSPECC.EQ.NCAT(II)) THEN
  IF (NSPECA.EQ.NANI(JJ)) THEN
    WRITE(6,3002) A1(I),A1(J)
1      ,B0(JP(I),JP(J)),B1(JP(I),JP(J)),B2(JP(I)
2      ,JP(J)),CPHI
    CCA(JP(I),JP(J))=CPHI/SQRT(-4.0E0*Z(I)*Z(J))
    GO TO 10
  ELSE IF (NSPECA.GT.NANI(JJ)) THEN
    WRITE(*,1002)
    STOP
  ELSE
    GO TO 11
  END IF
ELSE IF (NSPECC.GT.NCAT(II)) THEN
  WRITE(*,1002)
ELSE
  GO TO 11
END IF
10 CONTINUE

```

C MOVE POINTER TO BEGINING OF THETA/PSI DATA

```

12 READ(5,1007) NSPECC
IF (NSPECC.NE.901) GO TO 12

```

```

L=0
JXCM1=JXC-1
IF(JXCM1.LT.JMC) GO TO 701
WRITE(6,3003)(A1(K),K=JMA,JXA)

```

C      READ CATION-CATION THETAS AND CATION-CATION-ANION PSI'S  
C

```

DO 30 I=1,ICAT-1
DO 30 J=I+1,ICAT
L=L+1

```

```

31 READ(5,1003) NSPEC1,NSPEC2,TC(L),(DUMPSI(K),K=1,NAD)
IF (NSPEC1.EQ.NCAT(I)) THEN
  IF(NSPEC2.EQ.NCAT(J)) THEN

```

```

    DO 32 K=1,IANI
    PSIC(L,K)=DUMPSI(NANI(K)-199)
32   CONTINUE
    WRITE(6,3004) A1(I+1),A1(J+1)
1      ,TC(L),(PSIC(L,K),K=1,NA)
    GO TO 30

```

```

  ELSE IF (NSPEC2.GT.NCAT(J)) THEN
    WRITE(*,1002)
    STOP
  ELSE
    GO TO 31
END IF

```

```

ELSE IF (NSPEC1.GT.NCAT(I)) THEN
    WRITE(*,1002)
    STOP
ELSE
    GO TO 31
END IF
30  CONTINUE

701  CONTINUE

C  MOVE POINTER TO NEXT SECTION
33  READ(5,1007) NSPEC1
    IF (NSPEC1.NE.902) GO TO 33

    JXAM1=JXA-1
    IF(JXAM1.LT.JMA) GO TO 700
    WRITE(6,3003)(A1(K),K=JMC,JXC)
    L=0
C
C      READ ANION-ANION THETAS AND ANION-ANION-CATION PSI'S
C
    DO 40 I=1,IANI-1
    DO 40 J=I+1,IANI
    L=L+1

41  READ(5,1003) NSPEC1,NSPEC2,TA(L),(DUMPSI(K),K=1,NCD)
    IF (NSPEC1.EQ.NANI(I)) THEN
        IF(NSPEC2.EQ.NANI(J)) THEN
            DO 42 K=1,ICAT
            PSIA(L,K)=DUMPSI(NCAT(K)-99)
42      CONTINUE
            WRITE(6,3004) A1(I+1+ICAT),
1                  ,A1(J+1+ICAT),TA(L)
2                  ,(PSIA(L,K),K=1,NC)
            GO TO 40
        ELSE IF (NSPEC2.GT.NANI(J)) THEN
            WRITE(*,1002)
            STOP
        ELSE
            GO TO 41
        END IF
    ELSE IF (NSPEC1.GT.NANI(I)) THEN
        WRITE(*,1002)
        STOP
    ELSE
        GO TO 41
    END IF
40  CONTINUE
    L=0

700  CONTINUE

C      REWIND DATA FILE AND ADVANCE
    REWIND(5)

```

```
READ(5,1005)

      RETURN
C SAS
600  WRITE(6,1001)
1001 FORMAT(' DATAP READ ERROR  ')
      STOP
      END
```

```

C -----
C
C      THIS SUBROUTINE IS THE MAIN PROJECTED NEWTON MINMIZATION
C      PROCEDURE
C
C -----
SUBROUTINE PRIMAL(IRUN,ERROR)
IMPLICIT DOUBLE PRECISION(A-H,O-Z)
IMPLICIT INTEGER*2 (I-N)
INTEGER*2 Z
CHARACTER*24 A1
COMMON/CONST/Q(96,96),MACT(96),JA(96),JI(96),NSC,NCC,NDC,NC0C
1,N0EC
COMMON/LABEL/Z(96),U0RT(96),A1(96)
COMMON/PHASE/R11(96,16),JMIN(96),JMAX(96),KSPEC(96),KTYP(96)
1,NP0C,N00C
COMMON/SEARCH/X(96),P(96),G(96),W0(96),H(64,64),B(96)
COMMON/PRINT/PP,PP2,PD
DIMENSION YS(96),JDG(96),KPH(96),KREM(96)
LOGICAL SUBMIN,ZOUTEN(96),POSDEF,REMPHS,ADDPHS,SENS
LOGICAL NORMAL,ERROR,REFIN,REMP,PREM,GRADF,GRADP
LOGICAL REMDEG
LOGICAL RR
LOGICAL PP,PP2,PD,FIRST
PARAMETER(DELMP=0.5D0**55)
PARAMETER(DELLB=1.0D-20)
PARAMETER(DEL4=1.0D-13,DEL8=4.0D0*DELMP*(2.0D0**5))
PARAMETER(DEL2=1.0D-12)
PARAMETER(DEL2P=DEL2*(1.0D0+DEL2),DEL2M=DEL2*(1.0D0-DEL2))
PARAMETER(DEL1=1.0D-07,DEL5=0.1D0,DEL6=1.0D-04,DEL7=1.0D-07)
PARAMETER(NFUN=1000)
C
C -----
C      INITIALIZATION FOR MINIMIZATION
C
FIRST=.FALSE.
IF(ERROR) FIRST=.TRUE.
ERROR=.FALSE.
NS=NSC
NC0=NC0C
NSP1=NS+1
NC0P1=NC0+1
NPHAS0=NP0C
N00=N00C
C
C      CALCULATE SCALE FACTOR
C
SCALE=1.0D+20
DO 2 I=1,N00
TAU=0.0D0
DO 1 J=1,NC0
TAU=TAU+B(J)*R11(J,I)
1 CONTINUE
IF(TAU*SCALE.GT.R11(NC0P1,I)) SCALE=R11(NC0P1,I)/TAU

```

```

2 CONTINUE
      WRITE(6,5000) IRUN,SCALE
5000  FORMAT(//,' STEP',I5,' SCALE FACTOR=',1PD12.4)
      IF(SCALE.EQ.1.0D+20) THEN
          ERROR=.TRUE.
          RETURN
      END IF
C
C      FIND INITIAL PHASE ASSEMBLAGE, INITIALIZE MOLE NUMBER AND
C      DIRECTION VECTORS, EVALUATE INITIAL FREE ENERGY, AND
C      DELETE OR ADD SPECIES CONSTRAINTS
C
      ITER=0
      ISENS=0
      IFUN=0
      ISUB=0
      IREFIN=0
      NPHAS=0
      GFE0=0.0D0
      PTGA=0.0D0
      ALPHA=0.0D0
      NDG=0
      DO 5 K=1,NPHAS0
      IF(KTYP(K).GT.1) THEN
          NPHAS=NPHAS+1
          KPH(NPHAS)=K
          JM=JMIN(K)
          JX=JMAX(K)
          DO 3 J=JM,JX
              ZOUTEN(J)=.FALSE.
              X(J)=X(J)*SCALE
2323    FORMAT(I5,1PD25.16)
              P(J)=0.0D0
              Q(J,NSP1)=0.0D0
              IF(X(J).GT.DEL2P) THEN
                  IF(MACT(J).GT.NC0) THEN
                      WRITE(*,*) 'CALLING CONDEL @ LN 91'
                      CALL CONDEL(J)
                  END IF
              ELSE
                  IF(X(J).LT.DELLB) X(J)=DELLB
                  IF(MACT(J).LT.0) THEN
                      WRITE(*,*) 'CALLING CONADD @ LN 97'
                      CALL CONADD(J)
                  ELSE IF(MACT(J).LT.NC0P1) THEN
                      NDG=NDG+1
                      JDG(NDG)=J
                  END IF
              END IF
          CONTINUE
          WRITE(*,*) 'CALLING GRAD FROM PRIMAL @ LN 105'
          CALL GRAD(K,ALPHA,PTGA,GFE0)
      ELSE IF(KTYP(K).EQ.1) THEN
          NPHAS=NPHAS+1

```

```

KPH(NPHAS)=K
J=JMIN(K)
X(J)=X(J)*SCALE
P(J)=0.0D0
Q(J,NSP1)=0.0D0
ZOUTEN(J)=.FALSE.
IF(X(J).LT.DELLB) X(J)=DELLB
IF(MACT(J).GT.NC0) THEN
  WRITE(*,*) 'CALLING CONDEL @ LN 117'
  CALL CONDEL(J)
END IF
GFE0=GFE0+X(J)*G(J)
ELSE
  JM=JMIN(K)
  JX=JMAX(K)
  DO 4 J=JM,JX
  ZOUTEN(J)=.FALSE.
  X(J)=0.0D0
  P(J)=0.0D0
  Q(J,NSP1)=0.0D0
  IF(MACT(J).LT.0) THEN
    WRITE(*,*) 'CALLING CONADD @ LN 130'
    CALL CONADD(J)
  ELSE IF(MACT(J).LT.NC0P1) THEN
    NDG=NDG+1
    JDG(NDG)=J
  END IF
4  CONTINUE
END IF
5  CONTINUE
IF(NDG.GT.0) THEN
  DO 6 JD=1,NDG
  J=JDG(JD)
  IF(MACT(J).LT.0) THEN
    WRITE(*,*) 'CALLING CONADD @ LN 143'
    CALL CONADD(J)
  END IF
6  CONTINUE
END IF
NORMAL=.FALSE.
ADDPHS=.FALSE.
REMPHS=.FALSE.
REFIN=.FALSE.

C
C -----
C      SENSITIVITY STEP CALCULATION - TEST FOR FEASIBILITY
C
1000 SENS=.FALSE.
DO 9 I=1,NC0
  TAU=B(I)*SCALE
  DO 8 K=1,NPHAS
    JM=JMIN(KPH(K))
    JX=JMAX(KPH(K))
    DO 7 J=JM,JX

```

```

    TAU=TAU-X(J)*Q(J,I)
7  CONTINUE
8  CONTINUE
YS(I)=TAU
IF(DABS(TAU).GT.DEL4) SENS=.TRUE.
9  CONTINUE
C
C   IF FEASIBLE COMPUTE NORMAL DESCENT DIRECTION, OTHERWISE
C   COMPUTE A SENSITIVITY DIRECTION TO FEASIBLE DOMAIN
C
IF(SENS) THEN
  ISENS=ISENS+1
  NCMIN=NC0P1
  NJMIN=1
  IF(ISENS.GT.100) THEN
    WRITE(6,9991) ISENS
9991  FORMAT(' PRIMAL 9991 (ISENS)=',I5)
    ERROR=.TRUE.
    RETURN
  END IF
ELSE
  NORMAL=.TRUE.
  GO TO 1002
END IF
C
C -----
C   MODIFIED STEP DIRECTION CALCULATION
C
1001 NC=NCC
ND=NDC
NU=NS-NC
NUP1=NU+1
NUT=NUP1
NUP2=NU+2
NINACT=NU-ND+NC0
PTG=0.0D0
AMAX=1.0D0
J0=0
IF(NC.GE.NCMIN) THEN
  DO 11 I=NCMIN,NC
    JJ=JA(I)
    IF(DABS(P(JJ)).GT.DELLB) THEN
      TAU=P(JJ)
      Q(JJ,NSP1)=TAU
      PTG=PTG+TAU*G(JJ)
    ELSE
      TAU=0.0D0
    END IF
    IF(I.GT.NJMIN) THEN
      IM1=I-1
      DO 10 J=NJMIN,IM1
        TAU=TAU-Q(JJ,J)*YS(J)
      CONTINUE
    END IF
  11
END IF

```

```

      YS(I)=TAU/Q(JJ,I)
11    CONTINUE
      END IF
      IF(ND.GT.0) THEN
        DO 14 JD=1,ND
          J=JA(JD)
          TAU=0.0D0
          DO 12 I=NJMIN,NC
            TAU=TAU+Q(J,I)*YS(I)
12    CONTINUE
      IF(KTYP(KSPEC(J)).GT.1) THEN
        PI=X(J)+AMAX*TAU
        IF(X(J)+AMAX*TAU.LT.DELLB) THEN
          IF(PP) THEN
            WRITE(6,3867) J,X(J),TAU
3867    FORMAT(' DEG SPEC BOUNDARY',I5,2F25.20)
          END IF
          IF(X(J).GT.DELLB) THEN
            IF(REMPHS) GO TO 1012
            AMAX=(DELLB-X(J))/TAU
            J0=J
          ELSE IF(DABS(TAU).LT.DEL4.AND..NOT.REFIN) THEN
            TAU=0.0D0
          ELSE IF(TAU.LT.0.0D0) THEN
            IF(REMPHS) GO TO 1012
              IF(PP) THEN
                WRITE(6,3876) J,TAU,X(J)
                FORMAT(' ND ZERO STEP',I5,1P2D25.16)
3876    END IF
              AMAX=0.0D0
              J0=J
            END IF
          END IF
        ELSE IF(KTYP(KSPEC(J)).EQ.1) THEN
          IF(X(J)+AMAX*TAU.LT.0.0D0) THEN
            AMAX=-X(J)/TAU
            J0=J
          END IF
        ELSE IF(DABS(TAU-P(J)).LT.DEL4) THEN
          TAU=P(J)
        ELSE IF(REMPHS) THEN
          GO TO 1012
        ELSE IF(SENS) THEN
          IF(TAU.LT.0) THEN
            J0=J
          ELSE
            K0=KSPEC(J)
            NPHAS=NPHAS+1
            KTYP(K0)--KTYP(K0)
            KPH(NPHAS)=K0
            IF(KTYP(K0).GT.1) THEN
              JM=JMIN(K0)
              JX=JMAX(K0)
              DO 13 J2=JM,JX

```

```

X(J2)=DELLB
CONTINUE
ALPHA=0.0D0
WRITE(*,*) 'CALLING GRAD FROM PRIMAL @ LN 274'
CALL GRAD(K0,ALPHA,PTG,GFE0)
END IF
END IF
ELSE
  WRITE(6,9393) ADDPHS,REFIN,J
  FORMAT(' ADDPHS MOD STEP ERROR',3I5)
  ERROR=.TRUE.
  RETURN
END IF
Q(J,NSP1)=TAU
P(J)=TAU
PTG=PTG+TAU*G(J)
CONTINUE
14 IF(.NOT.SENS) THEN
  IF(J0.GT.0) THEN
    IF(X(J0)+AMAX*P(J0).LE.X(J0)*DEL2) THEN
      AMAX=X(J0)*(DEL2-1.0D0)/P(J0)
      IF(X(J0)+AMAX*P(J0).GT.DELLB) J0=0
    END IF
  END IF
  ELSE IF(J0.GT.0) THEN
    J1=0
    DO 43 II=NC0P1,NC
    I=(NC0P1+NC)-II
    J2=JA(I)
    TAU=Q(J0,I)
    IF(I.LT.NC) THEN
      IP1=I+1
      DO 63 I2=IP1,NC
      TAU=TAU-YS(I2)*Q(JA(I2),I)
    CONTINUE
    END IF
    YS(I)=TAU/Q(J2,I)
    IF(YS(I).GT.DEL4.AND.J2.GT.J1) THEN
      J1=J2
    END IF
 43 CONTINUE
  IF(J1.EQ.0) THEN
    WRITE(6,9999) J0
    FORMAT(I5,' PRIMAL 9999 (NO FEASIBLE SOLUTION)')
    ERROR=.TRUE.
    RETURN
  END IF
  WRITE(*,*) 'CALLING CONDEL @ LN 318'
  CALL CONDEL(J1)
  WRITE(*,*) 'CALLING CONADD @ LN 320'
  CALL CONADD(J0)
  P(J0)=0.0D0
  Q(J0,NSP1)=0.0D0
  IF(PP) THEN

```

```

      WRITE(6,5006) J0,J1
5006    FORMAT(' PRIMAL SENSITIVITY PIVOT',2I5)
           END IF
           GO TO 1000
       END IF
   END IF
IF(NINACT.EQ.0) GO TO 1006
DO 62 JJ=1,NINACT
J=JI(JJ)
TAU=0.0D0
DO 61 I=NJMIN,NC
TAU=TAU+Q(J,I)*YS(I)
61  CONTINUE
Q(J,NSP1)=TAU
62  CONTINUE
DO 16 I1=2,NUP2
IM1=I1-1
DO 15 I2=1,IM1
H(I2,I1)=0.0D0
15  CONTINUE
16  CONTINUE
H(NUP1,NUP2)=-1.0D0
GO TO 1005

C
C -----
C     NEWTON STEP DIRECTION CALCULATION
C
1002  NC=NCC
ND=NDC
NU=NS-NC
NUT=NU
NUP1=NU+1
NUP2=NU+2
NINACT=NU-ND+NC0
IF(NU.EQ.0) THEN
    Q3MAX=0.0D0
    WMIN--DEL7
    SUBMIN=.TRUE.
    ISUB=ISUB+1
    IF(PP.OR.PD) WRITE(6,6050) ISUB,ITER,IFUN,NU
6050    FORMAT('SUBSPACE MINIMUM ',4I5)
    POSDEF=.TRUE.
    GO TO 1004
  END IF
  POSDEF=.FALSE.

C
C     REDUCED GRADIENT VECTOR CALCULATION
C
1003  SIG=0.0D0
PTPMAX=0.0D0
Q3MAX=0.0D0
PTG=0.0D0
AMAX=0.999D0
J0=0

```

```

SIG=0.0D0
DO 19 L=1,NU
TAU=0.0D0
LQ=L+NC
DO 18 JJ=1,NINACT
TAU=TAU+G(JI(JJ))*Q(JI(JJ),LQ)
18  CONTINUE
YS(L)=TAU
IF(DABS(TAU).GT.Q3MAX) Q3MAX=DABS(TAU)
19  CONTINUE
IF(Q3MAX.LT.DEL1) THEN
  SUBMIN=.TRUE.
  ISUB=ISUB+1
    IF(PP.OR.PD) THEN
      WRITE(6,6050) ISUB,ITER,IFUN,NU
      END IF
    WMIN=-DEL7
  ELSE
    SUBMIN=.FALSE.
    WMIN=-DMAX1(Q3MAX,DEL7)
  END IF
C
C   EVALUATE MULTIPLIERS (IF POSDEF)
C
1004 IF(POSDEF) THEN
NP=NPHAS
WRITE(*,*) 'CALLING DUAL @ LN 405'
CALL DUAL(SUBMIN,ADDPHS,ZOUTEN,ERROR,WMIN,np,NCMIN,KPH)
IF(ERROR) THEN
  RETURN
ELSE IF(ADDPHS) THEN
  NPSAV=NPHAS
  NPHAS=NP
  NJMIN=NCMIN
  NORMAL=.FALSE.
  GO TO 1001
ELSE IF(SUBMIN) THEN
  IREFIN=IREFIN+1
  WRITE(*,*) 'CALLING REFINE @ LN 417'
  CALL REFINE(Q2MAX,NCMIN)
  IF(Q2MAX.GT.DEL1) THEN
    NJMIN=NCMIN
    NORMAL=.FALSE.
    REFIN=.TRUE.
    GO TO 1001
  ELSE
C
C     OPTIMAL SOLUTION FOUND WITHIN TOLERANCE OF DEL1
C
    IF(ITER.EQ.0) GFEA = GFE0
    IF(PP2) THEN
      WRITE(*,*) 'CALLING PHSPRT @ LN 430'
      CALL PHSPRT(SCALE,GFEA,ITER,IFUN,IREFIN,
Q2MAX,Q3MAX)
    1
  
```

```

        END IF
        RETURN
    END IF
    END IF
END IF
DO 59 L=1,NU
LP1=L+1
DO 17 L2=1,L
H(L2,LP1)=0.0D0
17  CONTINUE
H(L,NUP2)=YS(L)
59  CONTINUE
C
C      PROJECTED HESSIAN MATRIX CALCULATION
C
1005 DO 25 KP=1,NPHAS
K=KPH(KP)
IF(KTYP(K).GT.1) THEN
    JM=JMIN(K)
    JX=JMAX(K)
    WRITE(*,*) 'CALLING HESS FROM PRIMAL @ LN 452'
    CALL HESS(K)
    DO 24 L1=1,NU
    LQ1=L1+NC
    DO 21 J1=JM,JX
    TAU=0.0D0
    DO 20 J2=JM,JX
    JP=MAX0(J1,J2)
    TAU=TAU+H(JP,J1+J2-JP)*Q(J2,LQ1)
20   CONTINUE
    YS(J1)=TAU
21   CONTINUE
    DO 23 L2=L1,NUT
    LQ2=L2+NC
    TAU=H(L1,L2+1)
    DO 22 J1=JM,JX
    TAU=TAU+YS(J1)*Q(J1,LQ2)
22   CONTINUE
    H(L1,L2+1)=TAU
23   CONTINUE
24   CONTINUE
END IF
25   CONTINUE
C
C      REDUCED DIRECTION VECTOR CALCULATION
C
POSDEF=.TRUE.
GAMMA=0.0D0
ZETA=0.0D0
THETA=0.0D0
DO 27 I=1,NU
IP1=I+1
IF(DABS(H(I,IP1)).GT.GAMMA) GAMMA=DABS(H(I,IP1))
IF(I.LT.NU) THEN

```

```

      DO 26 J=IP1,NU
      IF(DABS(H(I,J+1)).GT.ZETA) ZETA=DABS(H(I,J+1))
26    CONTINUE
      IF(I.EQ.1) THETA=ZETA
      END IF
27    CONTINUE
      DELTA=DEL8*DMAX1(1.0D0,GAMMA,ZETA)
      BETA2=1.0D0/DMAX1(GAMMA,DELTA,ZETA/NU)
      TAU=DMAX1(DABS(H(1,2)), DELTA, THETA*THETA*BETA2)
      IF(TAU.GT.H(1,2)) POSDEF=.FALSE.
      H(1,2)=1.0D0/TAU
      IF(NU.EQ.1) THEN
          H(1,3)=H(1,3)*H(1,2)
      ELSE
          DO 31 J=2,NU
          JM1=J-1
          JP1=J+1
          JP2=J+2
          PHI=H(J,JP1)
          DO 28 K=1,JM1
          PI=H(K,JP1)*H(K,K+1)
          PHI=PHI-PI*H(K,JP1)
          H(K,JP1)=PI
28      CONTINUE
          THETA=0.0D0
          DO 30 I=JP2,NUP2
          TAU=H(J,I)
          DO 29 K=1,JM1
          TAU=TAU-H(K,JP1)*H(K,I)
29      CONTINUE
          H(J,I)=TAU
          IF(I.LT.NUP2) THEN
              IF(DABS(TAU).GT.THETA) THETA=DABS(TAU)
          END IF
30      CONTINUE
          SIG=DMAX1( DABS(PHI), DELTA, THETA*THETA*BETA2)
          IF(SIG.GT.PHI) POSDEF=.FALSE.
          H(J,JP1)=1.0D0/SIG
31      CONTINUE
          H(NU,NUP2)=H(NU,NUP2)*H(NU,NUP1)
          DO 33 JMN=2,NU
          J=NUP1-JMN
          JP1=J+1
          TAU=H(J,NUP2)*H(J,JP1)
          DO 32 K=JP1,NU
          TAU=TAU-H(J,K+1)*H(K,NUP2)
32      CONTINUE
          H(J,NUP2)=TAU
33      CONTINUE
      END IF
C
C      TEST FOR ACCIDENTAL MAXIMUM
C
      IF(NORMAL.AND.SUBMIN) THEN

```

```

IF(POSDEF) GO TO 1004
C
C          LOCAL MAXIMUM
C
C          WRITE(6,9992)
9992      FORMAT' PRIMAL 9992 (LOCAL MAXIMUM)'
            ERROR=.TRUE.
            RETURN
END IF
C
C          FEASIBLE DIRECTION VECTOR CALCULATION AND MAXIMUM STEP
C          LENGTH
C
DO 35 JJ=1,NINACT
J=JI(JJ)
TAU=0.0D0
DO 34 I=1,NUT
TAU=TAU-Q(J,I+NC)*H(I,NUP2)
34      CONTINUE
P(J)=TAU
PTG=PTG+TAU*G(J)
IF(KTYP(KSPEC(J)).GT.1) THEN
    IF(X(J)+AMAX*TAU.LT.DEL2M) THEN
        AMAX=DMAX1( (DEL2-X(J))/TAU ,0.0D0)
        J0=J
    END IF
ELSE
    IF(X(J)+AMAX*TAU.LT.0.0D0) THEN
        AMAX=DMAX1( -X(J)/TAU ,0.0D0)
        J0=J
    END IF
END IF
35      CONTINUE
C
C          TEST FOR DESCENT DIRECTIONS (IERR=1)
C
1006      SIG=0.0D0
PI=0.0D0
IF(PP) THEN
    WRITE(*,*) 'CALLING QTEST @ LN 578'
    CALL QTEST
DO 890 I=1,NC0
    TAU=B(I)*SCALE
    TAU2=0.0D0
    DO 891 J=1,NS
        TAU=TAU-Q(J,I)*X(J)
        TAU2=TAU2+P(J)*Q(J,I)
891      CONTINUE
    IF(DABS(TAU).GT.SIG) SIG=DABS(TAU)
    IF(DABS(TAU2).GT.PI) PI=DABS(TAU2)
890      CONTINUE
    WRITE(6,894) SIG,PI
894      FORMAT( 'FEASIBILITY TEST ',1P2D27.19)
    PI=-PTG

```

```

TAU=-GFE0
SIG=-WMIN
DO 888 J=1,NS
TAU=TAU+X(J)*G(J)
PI=PI+P(J)*G(J)
SIG=SIG+Q(J,NSP1)*G(J)
C           WRITE(6,8738) J,P(J),G(J)
8738       FORMAT(I5,1P3D16.8)
888        CONTINUE
IF(.NOT.ADDPHS) SIG=0.0D0
WRITE(6,8738) J0,PI,TAU,SIG
SIG=0.0D0
PHI=0.0D0
DO 896 KP=1,NPHAS
K=KPH(KP)
JM=JMIN(K)
JX=JMAX(K)
DO 960 J=JM,JX
IF(NORMAL) THEN
    TAU=G(J)
    IF(DABS(TAU).GT.PHI) PHI=DABS(TAU)
ELSE
    TAU=0.0D0
END IF
IF(KTYP(K).GT.1) THEN
    DO 961 J2=JM,JX
    JP=MAX0(J,J2)
    PI=H(JP,J+J2-JP)*P(J2)
    TAU=TAU+PI
    IF(DABS(PI).GT.PHI) PHI=DABS(PI)
CONTINUE
961        END IF
YS(J)=TAU
CONTINUE
960        CONTINUE
896        SIG=0.0D0
IF(NC.LT.NS) THEN
    NCP1=NC+1
    DO 963 I=NCP1,NS
    TAU=0.0D0
    DO 964 J=1,NS
    TAU=TAU+Q(J,I)*YS(J)
    CONTINUE
    IF(DABS(TAU).GT.SIG) SIG=DABS(TAU)
CONTINUE
964        END IF
    WRITE(6,962) SIG,PHI,POSDEF
963        FORMAT(' P TEST ',1P2D25.17,I5)
962        FORMAT(' P TEST ',1P2D25.17,I5)
END IF
C
C
IF(SENS) GO TO 1013
IF(PTG.GT.0.0D0) THEN
    IF(NORMAL.OR.REFIN) THEN

```

```

      WRITE(6,9994) NORMAL,REFIN,PTG,Q3MAX,PTPMAX
9994    FORMAT(' PRIMAL 9994  PTG > 0',2I5,1P3D12.4)
      ERROR=.TRUE.
      RETURN
      ELSE
        IERR=1
        GO TO 1012
      END IF
    END IF

C
C   CONSTRAINT ENCOUNTERED FOR FULL STEP
C   LIMIT MAXIMUM DECREASE OF SOLUTION PHASE VARIABLE TO
C   FACTOR OF DEL6
C

IF(J0.GT.0.AND.NORMAL) THEN
  IF(KTYP(KSPEC(J0)).GT.1) THEN
    IF(MACT(J0).LT.0) THEN
      DELTA=DEL6
    ELSE
      DELTA=DEL4
    END IF
    IF((DELTA-1.0D0)*X(J0).GT.AMAX*p(J0)) THEN
      AMAX=(DELTA-1.0D0)*X(J0)/P(J0)
      IF(X(J0)+AMAX*p(J0).GT.DEL2P) J0=0
    END IF
  END IF
END IF

C
C   EVALUATE FUNCTION VALUE AT MAXIMUM STEP
C

1013  IFUN=IFUN+1
      IF(IFUN.GT.NFUN) THEN
        WRITE(6,9995) IFUN
9995    FORMAT(' PRIMAL 9995 (NFUN EXCEEDED),I6)
        ERROR=.TRUE.
        RETURN
      END IF
      ALPHA=AMAX
      PTGA=0.0D0
      GFEA=0.0D0
      DO 36 K=1,NPHAS
      KP=KPH(K)
      RR=.FALSE.
      IF(REMPHS) THEN
        DO 89 KK=1,NPREM
        IF(KP.EQ.KREM(KK)) RR=.TRUE.
89     CONTINUE
      END IF
      IF(IABS(KTYP(KP)).GT.1.AND..NOT.RR) THEN
        WRITE(*,*) 'CALLING GRAD FROM PRIMAL @ LN 696'
        CALL GRAD(KP,ALPHA,PTGA,GFEA)
      ELSE IF(KTYP(KP).LT.-1) THEN
        JM=JMIN(KP)
        JX=JMAX(KP)

```

```

DO 83 J=JM,JX
GFEA=GFEA+(X(J)+ALPHA*P(J))*G(J)
PTGA=PTGA+P(J)*G(J)
83    CONTINUE
ELSE
J=JMIN(KP)
GFEA=GFEA+(X(J)+ALPHA*P(J))*G(J)
PTGA=PTGA+P(J)*G(J)
END IF
36    CONTINUE
IF(SENS) GO TO 1009
C
C      IF FREE ENERGY IS REDUCED TAKE MAXIMUM STEP
C
C      IF(GFEA.LT.GFE0) GO TO 1009
C
C      IF FREE ENERGY DECREASING AND MAXIMUM POSSIBLE
C      IMPROVEMENT IS INSIGNIFICANT COMPARED TO GFE0
C      TAKE MAXIMUM STEP
C
C      IF(PTGA.LT.DELMP) THEN
IF(DABS(AMAX*PTG).LT.DEL8*DABS(GFE0)) GO TO 1009
DELG=(GFEA-GFE0)/DABS(GFE0)
WRITE(6,9996) AMAX,PTG,PTGA,GFEA,DELG
9996    FORMAT('PRIMAL 9996 (NON-CONVEXITY)',1P5D12.5)
GO TO 1009
END IF
C
C      LINESEARCH
C
AMIN=0.0D0
ETA=DEL5*DABS(PTG)
IF(PP) THEN
WRITE(6,9638) ALPHA,PTG,PTGA,GFE0,GFEA
END IF
1007  AMAX=ALPHA
PTG2=PTGA
IF(DABS(GFEA-GFE0).LT.DELMP*DABS(GFE0).AND.PTGA.LT.0.0D0)
1    GO TO 1008
IF(DABS(PTGA).LT.ETA) THEN
IF(GFEA.LT.GFE0) GO TO 1008
IF(PTGA.LT.DELMP) THEN
DELG=(GFEA-GFE0)/DABS(GFE0)
IF(DABS(ALPHA*PTG).GT.DEL8*DABS(GFE0)) THEN
WRITE(6,9980) AMAX,PTGA,GFEA,DELG
END IF
FORMAT('PRIMAL 9980 (NON-CONVEXITY)',1P4D12.5)
9980    GO TO 1008
END IF
END IF
ALPHA=AMIN-(AMAX-AMIN)*PTG/(PTG2-PTG)
IFUN=IFUN+1
IF(IFUN.GT.NFUN) THEN
WRITE(6,9995) IFUN

```

```

        ERROR=.TRUE.
        RETURN
    END IF
    PTGA=0.0D0
    GFEA=0.0D0
    DO 37 K=1,NPHAS
    KP=KPH(K)
    RR=.FALSE.
    IF(REMPHS) THEN
        DO 88 KK=1,NPREM
        IF(KP.EQ.KREM(KK)) RR=.TRUE.
88     CONTINUE
    END IF
    IF(IABS(KTYP(KP)).GT.1.AND..NOT.RR) THEN
        WRITE(*,*) 'CALLING GRAD FROM PRIMAL @ LN 769'
        CALL GRAD(KP,ALPHA,PTGA,GFEA)
    ELSE IF(KTYP(KP).LT.-1) THEN
        JM=JMIN(KP)
        JX=JMAX(KP)
        DO 84 J=JM,JX
        GFEA=GFEA+(X(J)+ALPHA*P(J))*G(J)
        PTGA=PTGA+P(J)*G(J)
84     CONTINUE
    ELSE
        J=JMIN(KP)
        GFEA=GFEA+(X(J)+ALPHA*P(J))*G(J)
        PTGA=PTGA+P(J)*G(J)
    END IF
37     CONTINUE
            IF(PP) THEN
                WRITE(6,9638) ALPHA,PTG,PTGA,GFE0,GFEA
9638   FORMAT(' LINESEARCH',1P5D24.16,/)

            END IF
            IF(PTGA*PTG2.LT.0.0D0) THEN
                PTG=PTG2
                AMIN=AMAX
            ELSE
                PTG=PTG*0.5D0
            END IF
            GO TO 1007
1008   IF(J0.GT.0) THEN
            IF(KTYP(KSPEC(J0)).GT.1) THEN
                IF(X(J0)+ALPHA*P(J0).GT.DEL2P) J0=0
            ELSE
                IF(X(J0)+ALPHA*P(J0).GT.DELMP) J0=0
            END IF
        END IF
C
C      UPDATE X = X + ALPHA*P
C
1009   GFE0=GFEA
        ITER=ITER+1
C
C      UPDATE NORMAL STEP

```

```

C
      IF(NORMAL) THEN
        DO 38 JJ=1,NINACT
          J=JI(JJ)
          X(J)=X(J)+ALPHA*P(J)
        CONTINUE
        IF(PP) THEN
          WRITE(6,5001) ITER,IFUN,J0,GFE0,ALPHA
          WRITE(6,5107)
5001      FORMAT(' PRIMAL NORMAL',3I5,1PD25.17,1P2D20.8)
          DO 6000 J=1,NS
            KK=(KSPEC(J))
            WRITE(6,5002) J,KK,KTYP(KK),MACT(J),X(J),G(J),P(J)
6000      CONTINUE
5002      FORMAT(4I5,F25.21,1P3D13.4)
        END IF
        IF(J0.EQ.0) GO TO 1003
        WRITE(*,*) 'CALLING CONADD @ LN 826'
        CALL CONADD(J0)
        IF(NDC.GT.ND) THEN
          NDP1=ND+1
          DO 85 JD=NDP1,NDC
            P(JA(JD))=0.0D0
            Q(JA(JD),NSP1)=0.0D0
85      CONTINUE
        END IF
        K0=KSPEC(J0)
        IF(KTYP(K0).EQ.1) THEN
          X(J0)=0.0D0
          P(J0) = 0.0D0
          Q(J0,NSP1)=0.0D0
          KTYP(K0)=-KTYP(K0)
          NPHAS=NPHAS-1
          DO 46 K=1,NPHAS
            IF(KPH(K).EQ.K0) KPH(K)=KPH(NPHAS+1)
          CONTINUE
          GO TO 1002
46      ELSE
          X(J0)=DEL2
          JM=JMIN(K0)
          JX=JMAX(K0)
          TAU=0.0D0
          PI=0.0D0
          DO 47 J=JM,JX
            TAU=TAU+X(J)
            PI=PI+P(J)
          CONTINUE
          P(J0)=0.0D0
          Q(J0,NSP1)=0.0D0
47      IF(TAU+(1.0D0-ALPHA)*PI.LT.1.0D-05) THEN
            KCON=NCC+1
            NCMIN=KCON
            KTYP(K0)=-KTYP(K0)
            TAU=1.0D0/TAU

```

```

        DO 49 J=JM,JX
        W0(J)=X(J)*TAU
        P(J)--X(J)
        II=MACT(J)
        IF(II.LT.0) THEN
            WRITE(*,*) 'CALLING CONADD @ LN 868'
            CALL CONADD(J)
        ELSE IF(II.GT.NC0) THEN
            IF(II.LT.NCMIN) NCMIN=II
        END IF
        CONTINUE
        IF(NDC.GT.ND) THEN
            NDP1=ND+1
            DO 99 JD=NDP1,NDC
            P(JA(JD))=0.0D0
            Q(JA(JD),NSP1)=0.0D0
            CONTINUE
        99
        END IF
        NPREM=1
        KREM(1)=K0
        NJMIN=NCMIN
        AMAX=1.0D0
        NORMAL=.FALSE.
        REMPHS=.TRUE.
        GO TO 1001
    ELSE
        GO TO 1002
    END IF
    END IF
C
C -----
C     UPDATE MODIFIED STEP
C
    ELSE
        NDG=0
        PREM=.FALSE.
        REMP=.FALSE.
        GRADF=.FALSE.
        IF(REFIN) REMP=.TRUE.
        NPREM=0
        IF(PP) THEN
            DO 1112 J=1,NS
            YS(J)=0.0D0
        1112
        CONTINUE
        END IF
        DO 39 KP=1,NPHAS
        K0=KPH(KP)
        IF(KTYP(K0).LT.0) KTYP(K0)--KTYP(K0)
        REMDEG=.TRUE.
        JPH=0
        IF(KTYP(K0).EQ.1) THEN
            J=JMIN(K0)
            X(J)=X(J)+ALPHA*P(J)
            YS(J)=P(J)

```

```

P(J)=0.0D0
Q(J,NSP1)=0.0D0
IF(X(J).GT.DEL4) THEN
  IF(MACT(J).GT.NC0) THEN
    WRITE(*,*) 'CALLING CONDEL @ LN 921'
    CALL CONDEL(J)
    REMP=.FALSE.
    REMDEG=.FALSE.
  END IF
ELSE
  PREM=.TRUE.
  KTyp(K0)--1
  IF(MACT(J).LT.0) THEN
    WRITE(*,*) 'CALLING CONADD @ LN 930'
    CALL CONADD(J)
    JPH=JPH+1
  ELSE IF(MACT(J).LT.NC0P1) THEN
    NDG=NDG+1
    JDG(NDG)=J
  END IF
END IF
ELSE
  GRADP=.FALSE.
  JM=JMIN(K0)
  JX=JMAX(K0)
  TAU=0.0D0
  DO 40 J=JM,JX
  X(J)=X(J)+ALPHA*P(J)
  TAU=TAU+X(J)
  YS(J)=P(J)
  P(J)=0.0D0
  Q(J,NSP1)=0.0D0
  IF(X(J).GT.DEL2P) THEN
    IF(MACT(J).GT.NC0) THEN
      WRITE(*,*) 'CALLING CONDEL @ LN 951'
      CALL CONDEL(J)
      REMP=.FALSE.
      REMDEG=.FALSE.
    END IF
  ELSE
    IF(X(J).LT.DELLB) THEN
      X(J)=DELLB
      GRADP=.TRUE.
    END IF
    IF(MACT(J).LT.0) THEN
      WRITE(*,*) 'CALLING CONADD @ LN 962'
      CALL CONADD(J)
      JPH=JPH+1
    ELSE IF(MACT(J).LT.NC0P1) THEN
      NDG=NDG+1
      JDG(NDG)=J
    END IF
  END IF
CONTINUE

```

```

        IF(REMPHS.AND.TAU.LT.DEL4*10.0D0) THEN
          DO 44 J=JM,JX
            X(J)=0.0D0
            CONTINUE
            KTyp(K0)--KTyp(K0)
            PREM=.TRUE.
            GRADP=.FALSE.
        ELSE IF(REMP.AND.TAU.LT.100.D0*DEL2.OR.REMDEG
1           .AND.TAU.LT.100.0D0*DEL2.AND.JPH.GT.0) THEN
          TAU=1.0D0/TAU
          DO 45 J=JM,JX
            W0(J)=X(J)*TAU
            CONTINUE
            NPREM=NPREM+1
            KREM(NPREM)=K0
            END IF
            IF(GRADP) GRADF=.TRUE.
        END IF
39        CONTINUE
        IF(NDG.GT.0) THEN
          DO 41 JD=1,NDG
            J=JDG(JD)
            IF(MACT(J).LT.0) THEN
              WRITE(*,*) 'CALLING CONADD @ LN 994'
              CALL CONADD(J)
            END IF
            CONTINUE
        END IF
        IF(J0.GT.0) THEN
          IF(MACT(J0).LT.0) THEN
            WRITE(*,*) 'CALLING CONADD @ LN 1001'
            CALL CONADD(J0)
            IF(PP) THEN
              WRITE(6,7878) J0,X(J0)
              FORMAT(' MOD STEP 7878',I5,1PD12.4)
            END IF
          END IF
        END IF
        IF(PREM) THEN
          NP=NPHAS
          NPHAS=0
          DO 42 K=1,NP
            IF(KTyp(KPH(K)).GT.0) THEN
              NPHAS=NPHAS+1
              KPH(NPHAS)=KPH(K)
            END IF
            CONTINUE
        END IF
        IF(GRADF) THEN
          IF(PP) WRITE(6,3444)
3444        FORMAT(' SOL. VAR. HIT ZERO')
          GFE0=0.0D0
          PTG=0.0D0
          ALP=0.0D0
        END IF
      END IF
    END IF
  END IF
END IF

```

```

DO 110 KP=1,NPHAS
K=KPH(KP)
IF(KTYP(K).GT.1) THEN
  WRITE(*,*) 'CALLING GRAD FROM PRIMAL @ LN 1028'
  CALL GRAD(K,ALP,PTG,GFE0)
ELSE
  J=JMIN(K)
  GFE0=GFE0+X(J)*G(J)
END IF
CONTINUE
END IF
110
IF(PP) THEN
  IF(ADDPHS) THEN
    WRITE(6,5003) ITER,IFUN,J0,GFE0,ALPHA
  ELSE IF(REMPHS) THEN
    WRITE(6,5004) ITER,IFUN,J0,GFE0,ALPHA
  ELSE IF(REFIN) THEN
    WRITE(6,5005) ITER,IFUN,J0,GFE0,ALPHA
  ELSE IF(SENS) THEN
    WRITE(6,5106) ITER,IFUN,J0,GFE0,ALPHA
  END IF
  5003 FORMAT(' PRIMAL ADDPHS',3I5,1PD25.17,1PD20.8)
  5004 FORMAT(' PRIMAL REMPHS',3I5,1PD25.17,1PD20.8)
  5005 FORMAT(' PRIMAL REFIN ',3I5,1PD25.17,1PD20.8)
  5106 FORMAT(' PRIMAL SENS ',3I5,1PD25.17,1PD20.8)
  5107 FORMAT(' J K0 KTYP MACT',24X,'X',12X,'G',12X,'P')
  WRITE(6,5107)
  DO 6001 J=1,NS
  KK=(KSPEC(J))
  WRITE(6,5002) J,KK,KTYP(KK),MACT(J),X(J),G(J),YS(J)
  CONTINUE
  END IF
  6001
  IF(SENS) THEN
    GO TO 1000
  ELSE IF(NPREM.EQ.0) THEN
    ADDPHS=.FALSE.
    REMPHS=.FALSE.
    REFIN=.FALSE.
    NORMAL=.TRUE.
    GO TO 1002
  ELSE
    ADDPHS=.FALSE.
    REMPHS=.TRUE.
    REFIN=.FALSE.
    NORMAL=.FALSE.
    KCON=NCC+1
    NCMIN=NCC+1
    DO 78 KR=1,NPREM
    K0=KREM(KR)
    KTYP(K0)=KTYP(K0)
    JM=JMIN(K0)
    JX=JMAX(K0)
    DO 82 J=JM,JX
    P(J)--X(J)
  END IF
END IF

```

```

        II=MACT(J)
        IF(II.GT.NC0.AND.II.LT.NCMIN) NCMIN=II
82      CONTINUE
78      CONTINUE
        NJMIN=NCMIN
        AMAX=1.0D0
        GO TO 1001
        END IF
    END IF
C -----
C MODIFIED STEP FAILED
C
C     IERR= -J, P(J) NOT SET CORRECTLY
C     IERR= 1, PTG > 0
C     IERR= 2, AMAX NOT FEASIBLE
C
1012 IF(REMPHS) THEN
        REMPHS=.FALSE.
        DO 79 K=1,NPREM
        K0=KREM(K)
        KTYP(K0)--KTYP(K0)
79      CONTINUE
        IF(KCON.LE.NC) THEN
            DO 51 II=KCON,NC
            I=(KCON+NC)-II
            JR=JA(I)
            WRITE(*,*) ' CALLING CONDEL @ LN 1105'
            CALL CONDEL(JR)
51      CONTINUE
        END IF
        IF(PP) THEN
            WRITE(6,5007) IERR,(KREM(I),I=1,NPREM)
5007      FORMAT(' PRIMAL SOLUTION PHASE REMOVAL FAILED ',20I5)
            END IF
        ELSE IF(SUBMIN) THEN
            WRITE(6,9998)
9998      FORMAT(' PRIMAL 9998 (CONSTRAINT DELETION STEP)')
            ERROR=.TRUE.
            RETURN
        ELSE
            NPHAS=NPSAV
            ADDPHS=.FALSE.
            IF(PP) THEN
                WRITE(6,9998)
            END IF
        END IF
        DO 52 J=1,NS
        P(J)=0.0D0
        Q(J,NSP1)=0.0D0
52      CONTINUE
        NORMAL=.TRUE.
        GO TO 1002
    END

```

```

C-----
C
C      THIS SUBROUTINE CALCULATES THE CHEMICAL POTENTIALS FOR
THE
C      SOLUTION SPECIES IN PHASE KPP AS WELL AS THE SOLUTION'S
C      CONTRIBUTION TO THE FREE ENERGY, AND ITS CONTRIBUTION
TO THE
C      DERIVATIVE OF THE FREE ENERGY WITH RESPECT TO THE STEP
C      DIRECTION (P).
C
C-----
C      SUBROUTINE GRAD(KPP,STSZ,ZETA,GFE)
C
C      COMPUTE CHEMICAL POTENTIALS FOR AQUEOUS ELECTROLYTE
C      SOLUTIONS USING PITZER AND KIM (1974) MODEL WITH
C      ELECTROSTATIC MIXING TERMS FROM PITZER(1975).
C
C      IMPLICIT DOUBLE PRECISION(A-H,O-Z)
C      IMPLICIT INTEGER*2 (I-N)
C      DOUBLE PRECISION IONIC,MOLAL
C      INTEGER*2 Z,ZMAX
C      CHARACTER*24 A1
C      COMMON/LABEL/Z(96),U0RT(96),A1(96)
C      COMMON/PHASE/R11(96,16),JMIN(96),JMAX(96),KSPEC(96),KTYP(96)
1,NP0C,N00C
      COMMON/SEARCH/X(96),P(96),G(96),W(96),H(64,64),B(96)
      COMMON/AQSOL/JP(46),JMC,JXC,JMA,JXA,JMB,JXB,ALPHA(3)
1,BETA,B0(18,18),B1(18,18),B2(18,18),CCA(18,18),TA(153)
2,PSIA(153,18),TC(153),PSIC(153,18),MOLAL(46),F1(3),F2(3),F3(3)
3,ZSUM
      COMMON/ELCT/APHI,IONIC,ETH(8,8),ETHP(8,8),ETHP2(8,8),ZMAX
      COMMON/PH/LACT
      DATA WH2O/0.0180153D0/
      STEP=STSZ
      IF(IABS(KTYP(KPP)).EQ.2) GO TO 500
C SAS
      IF(IABS(KTYP(KPP)).EQ.3) GO TO 600
C SAS
      JM=JMIN(KPP)
      JX=JMAX(KPP)
      IONIC=0.0D0
      ZSUM=0.0D0
      PHI=0.0D0
      CONVR=1.0D0/(WH2O*(X(JM)+STEP*P(JM)) )
      DO 2 J=JMC,JX
      S=(X(J)+STEP*P(J))*CONVR
      G(J)=U0RT(J)+DLOG(S)
      PHI=PHI+S
      ZSUM=ZSUM+IABS(Z(J))*S
      IONIC=IONIC+S*Z(J)*Z(J)
      MOLAL(J)=S
2     CONTINUE
      IONIC=IONIC*0.5D0
      S=DSQRT(IONIC)

```

```

SH=S
T=1.0D0+BETA*S
PHI=PHI-2.0D0*APHI*S*IONIC/T
FZ2=-APHI*(S/T+2.0*DLOG(T)/BETA)
FZ=0.0D0
DO 3 I=1,3
T=ALPHA(I)*S
IF(T.LT.1.0D-07) GO TO 4
F1(I)=DEXP(-T)
F2(I)=2.0D0*(1.0D0-(1.0D0+T)*F1(I))/(T*T)
F3(I)=(F1(I)-F2(I))/IONIC
GO TO 3
4 F1(I)=1.0D0-T*(1.0D0-T*(.5D0-T*(.16666666666667D0
1 -T*4.16666666666D-02)))
F2(I)=1.D0-T*(.66666666666667D0-T*(0.25D0-T*(
1 6.66666666666666D-02-T*(1.388888888888D-02-T*2.380952381D-03)))
IF(T.LT.1.0D-21)T=1.0D-21
F3(I)=(-.33333333333333D0/T+.25D0-T*(.1D0
1 -T*(2.7777777777778D-02-T*5.952380952D-03)))*ALPHA(I)*ALPHA(I)
3 CONTINUE
GAMCL=FZ2+IONIC*(.0967D0+.2122D0*(F2(1)+F1(1))-IONIC*.00126D0)
WRITE(*,*) 'CALLING ELECT @ LN 74'
CALL ELECT
DO 13 I=JMC,JXA
KK=MIN0(Z(I),2)
DO 12 J=JMA,JXA
S=MOLAL(I)*MOLAL(J)
K=MIN0(KK,-Z(J))
PHI=PHI+2.0D0*S*(B0(JP(I),JP(J))+B1(JP(I),JP(J))*F1(K)
1 +B2(JP(I),JP(J))*F1(3)+ZSUM*CCA(JP(I),JP(J)) )
T=2.0D0*(B0(JP(I),JP(J))+B1(JP(I),JP(J))*F2(K)
1 +B2(JP(I),JP(J))*F2(3) )+ZSUM*CCA(JP(I),JP(J))
G(I)=G(I)+MOLAL(J)*T
G(J)=G(J)+MOLAL(I)*T
FZ2=FZ2+S*(B1(JP(I),JP(J))*F3(K)+B2(JP(I),JP(J))*F3(3) )
FZ=FZ+S*CCA(JP(I),JP(J))
12 CONTINUE
13 CONTINUE
L=0
JXAM1=JXA-1
IF(JMA.GT.JXAM1) GO TO 100
DO 30 I=JMA,JXAM1
IP1=I+1
DO 36 J=IP1,JXA
L=L+1
THETA=TA(L)+ETH(-Z(I),-Z(J))
T=THETA
S=MOLAL(I)*MOLAL(J)
FZ2=FZ2+S*ETHP(-Z(I),-Z(J))
DO 28 K=JMC,JXC
T=T+MOLAL(K)*PSIA(L,JP(K))
G(K)=G(K)+S*PSIA(L,JP(K))
28 CONTINUE
PHI=PHI+2.0D0*S*(T+IONIC*ETHP(-Z(I),-Z(J)))

```

```

T=T+THETA
G(I)=G(I)+MOLAL(J)*T
G(J)=G(J)+MOLAL(I)*T
36  CONTINUE
30  CONTINUE
100 JXCM1=JXC-1
    IF(JMC.GT.JXCM1) GO TO 102
    L=0
    DO 34 I=JMC,JXCM1
    IP1=I+1
    DO 33 J=IP1,JXC
    L=L+1
    THETA=TC(L)+ETH(Z(I),Z(J))
    T=THETA
    S=MOLAL(I)*MOLAL(J)
    FZ2=FZ2+S*ETHP(Z(I),Z(J))
    DO 32 K=JMA,JXA
    T=T+MOLAL(K)*PSIC(L,JP(K))
    G(K)=G(K)+S*PSIC(L,JP(K))
32  CONTINUE
    PHI=PHI+2.0D0*S*(T+IONIC*ETHP(Z(I),Z(J)))
    T=T+THETA
    G(I)=G(I)+MOLAL(J)*T
    G(J)=G(J)+MOLAL(I)*T
33  CONTINUE
34  CONTINUE
C*****
C*****
102 G(JM)=U0RT(JM)-WH2O*PHI
    T=G(JM)*P(JM)
    DO 9 I=JMC,JX
    G(I)=G(I)+IABS(Z(I))*(FZ+IABS(Z(I))*FZ2)
80  T=T+G(I)*P(I)
9   CONTINUE
    ZETA=ZETA+T
    DO 169 J=JM,JX
    GFE=GFE+(X(J)+STEP*P(J))*G(J)
169 CONTINUE
    IF(LACT.EQ.0) THEN
        AL1=G(JMA)-DLOG(MOLAL(JMA))-U0RT(JMA)
        PHI=AL1-GAMCL
67   DO 68 J=JMC,JXA
        G(J)=G(J)+Z(J)*PHI
68   CONTINUE
    END IF
    RETURN
500 JM=JMIN(KPP)
    JX=JMAX(KPP)
    S=0.0D0
    DO 70 J=JM,JX
70   S=S+X(J)+STEP*P(J)
    S=1.0D0/S
    T=0.0D0
    DO 71 J=JM,JX

```

```

G(J)=U0RT(J)+DLOG(S*(X(J)+STEP*P(J)))
GFE=GFE+(X(J)+STEP*P(J))*G(J)
71   T=T+G(J)*P(J)
      ZETA=ZETA+T
      RETURN
C SAS
600   JM=JMIN(KPP)
      JX=JMAX(KPP)
      IONIC=0.0D0
      CONVR=1.0D0/(WH2O*(X(JM)+STEP*P(JM)))
      DO 72 J=JMC,JX
      S=(X(J)+STEP*P(J))*CONVR
      IONIC=IONIC+S*Z(J)*Z(J)
72   CONTINUE
      IONIC=0.5D0*IONIC
      S=0.0D0
      DO 73 J=JM,JX
      S=S+X(J)+STEP*P(J)
73   CONTINUE
      S=1.0D0/S
      T=0.0D0
      DAVIES=(DSQRT(IONIC)/(DSQRT(IONIC)+1.0D0)-0.3D0*IONIC)
      DO 74 J=JM,JX
      G(J)=U0RT(J)+DLOG(S*(X(J)+STEP*P(J)))-3*APHI*Z(J)*Z(J)*DAVIES
      GFE=GFE+(X(J)+STEP*P(J))*G(J)
      T=T+G(J)*P(J)
74   CONTINUE
      ZETA=ZETA+T
      RETURN
C SAS
      END

```

```

C-----  

C  

C      THIS SUBROUTINE CALCULATES THE IONIC STRENGTH AND  

C      CHARGE DEPENDENT ELECTROSTATIC TERMS IN THE EXPRESSION  

C      FOR THETA AS WELL AS THEIR FIRST AND SECOND DERIVATIVES  

C      USING A CHEBYCHEV EXPANSION.  

C-----  

SUBROUTINE ELECT  

IMPLICIT DOUBLE PRECISION(A-H,O-Z)  

IMPLICIT INTEGER*2 (I-N)  

COMMON/ELCT/APHI,ST,ETH(8,8),ETHP(8,8),ETHP2(8,8),NZ  

DIMENSION A1(21),A2(21)  

DATA A1/-0.000000000010991D0,-.000000000002563D0  

1,0.000000000001943D0,0.000000000046333D0,-.000000000050847D0  

1,-.0000000000821969D0,0.0000000001229405D0,0.000000013522610D0  

1,-.000000025267769D0,-.0000000202099617D0,0.000000396566462D0  

1,0.000002937706971D0,-.000004537895710D0,-.000045036975204D0  

1,0.000036583601823D0,0.000636874599598D0,0.000388260636404D0  

1,-.007299499690937D0,-.029779077456514D0,-.060076477753119D0  

1,1.925154014814667D0/  

     DATA A2/0.000000000237816D0,-.000000002849257D0  

1,-.000000006944757D0,0.000000004558555D0,0.0000000080779570D0  

1,0.000000216991779D0,-.0000000250453880D0,-.000003548684306D0  

1,-.000004583768938D0,0.000034682122751D0,0.000087294451594D0  

1,-.000242107641309D0,-.000887171310131D0,0.001130378079086D0  

1,0.006519840398744D0,-.001668087945272D0,-.036552745910311D0  

1,-.028796057604906D0,0.150044637187895D0,0.462762985338493D0  

1,0.628023320520852D0/  

     NZM1=NZ-1  

IF(NZM1.EQ.0) GO TO 6  

X0=6.0D0*APHI*DSQRT(ST)  

DO 2 II=1,NZ  

DO 1 JJ=II,NZ  

X=II*JJ*X0  

IF(X.GT.1.0D0) GO TO 20  

Y=4.0D0*(X**0.2D0)  

G=0.2D0*Y/X  

G2=-0.8D0*G/X  

Y=Y-2.0D0  

BI=0.0D0  

BIP1=0.0D0  

DBI=0.0D0  

DBIP1=0.0D0  

D2BI=0.0D0  

D2BIP1=0.0D0  

DO 10 I=1,21  

BIP2=BIP1  

BIP1=BI  

BI=Y*BIP1-BIP2+A1(I)  

DBIP2=DBIP1  

DBIP1=DBI  

DBI=BIP1+Y*DBIP1-DBIP2  

D2BIP2=D2BIP1  

D2BIP1=D2BI

```

```

D2BI=2.0D0*DBIP1+Y*D2BIP1-D2BIP2
10 CONTINUE
GO TO 40
20 Y=4.444444444444444D0/(X**0.1D0)
G=-0.1D0*Y/X
G2=-1.1D0*G/X
Y=Y-2.444444444444444D0
BI=0.0D0
BIP1=0.0D0
DBI=0.0D0
DBIP1=0.0D0
D2BI=0.0D0
D2BIP1=0.0D0
DO 25 I=1,21
BIP2=BIP1
BIP1=BI
BI=Y*BIP1-BIP2+A2(I)
DBIP2=DBIP1
DBIP1=DBI
DBI=BIP1+Y*DBIP1-DBIP2
D2BIP2=D2BIP1
D2BIP1=D2BI
D2BI=2.0D0*DBIP1+Y*D2BIP1-D2BIP2
25 CONTINUE
40 ETH(IJ,JJ)=0.25D0*X+0.5D0*(BI-BIP2)-1.0D0
THP=0.5D0*(DBI-DBIP2)
ETHP2(IJ,JJ)=X*X*(G2*THP+0.5D0*G*G*(D2BI-D2BIP2))
ETHP(IJ,JJ)=X*(G*THP+0.25D0)
1 CONTINUE
2 CONTINUE
C=0.5D0/ST
C1=0.5D0*C
C2=C*C1
C3=C*C2
DO 4 I=1,NZM1
IP1=I+1
DO 3 J=IP1,NZ
TH=I*j*C1*(ETH(I,J)-0.5D0*(ETH(I,I)+ETH(J,J)))
THP=-TH/ST+I*j*C2*(ETHP(I,J)-0.5D0*(ETHP(I,I)+ETHP(J,J)))
THP2=C*(-TH/ST-5.0D0*THP)
THP2=THP2+I*j*C3*(ETHP2(I,J)-0.5D0*(ETHP2(I,I)+ETHP2(J,J)))
ETH(I,J)=TH
ETH(J,I)=TH
ETHP(I,J)=THP
ETHP(J,I)=THP
ETHP2(I,J)=THP2
3 ETHP2(J,I)=THP2
4 CONTINUE
6 DO 5 I=1,NZ
ETH(I,I)=0.0D0
ETHP(I,I)=0.0D0
5 ETHP2(I,I)=0.0D0
RETURN
END

```

```

C-----  

C THIS SUBROUTINE CALCULATES THE LOWER TRIANGLE OF THE  

C SECOND DERIVATIVE (SYMMETRIC) MATRIX OF THE FREE  

C ENERGY.  

C-----  

C-----  

SUBROUTINE HESS(KPP)  

IMPLICIT DOUBLE PRECISION(A-H,O-Z)  

IMPLICIT INTEGER*2 (I-N)  

DOUBLE PRECISION IONIC,MOLAL  

INTEGER*2 Z,ZMAX,ZI,ZJ  

CHARACTER*24 A1  

COMMON/LABEL/Z(96),U0RT(96),A1(96)  

COMMON/PHASE/R11(96,16),JMIN(96),JMAX(96),KSPEC(96),KTYP(96)  

1,NP0C,N00C  

COMMON/SEARCH/X(96),P(96),G(96),W(96),H(64,64),B(96)  

COMMON/AQSOL/JP(46),JMC,JXC,JMA,JXA,JMB,JXB,ALPHA(3)  

1,BETA,B0(18,18),B1(18,18),B2(18,18),CCA(18,18),TA(153)  

2,PSIA(153,18),TC(153),PSIC(153,18),MOLAL(46),F1(3),F2(3),F3(3)  

3,ZSUM  

COMMON/ELCT/APHI,IONIC,ETH(8,8),ETHP(8,8),ETHP2(8,8),ZMAX  

COMMON/PRINT/PP,PP2,PD  

LOGICAL PP,PP2,PD  

DIMENSION GT(64),HT(64)  

DATA WH2O/0.0180153D0/  

IF(IABS(KTYP(KPP)).EQ.2) GO TO 500  

IF(IABS(KTYP(KPP)).EQ.3) GO TO 700  

JM=JMIN(KPP)  

JX=JMAX(KPP)  

DO 2 J=JM,JX  

GT(J)=0.0D0  

HT(J)=0.0D0  

DO 1 I=J,JX  

H(I,J)=0.0D0  

1 CONTINUE  

2 CONTINUE  

S=DSQRT(IONIC)  

T=1.0D0+BETA*S  

FZ2=-APHI*(0.5D0+T)/(S*T*T)  

DO 3 I=1,3  

F1(I)=-(2.0D0*F3(I)+0.5D0*ALPHA(I)*F1(I)/S)/IONIC  

3 CONTINUE  

DO 13 I=JMC,JXC  

KK=MIN0(Z(I),2)  

DO 12 J=JMA,JXA  

S=MOLAL(I)*MOLAL(J)  

K=MIN0(KK,-Z(J))  

H(J,I)=2.0D0*(B0(JP(I),JP(J))+B1(JP(I),JP(J))*F2(K)  

1 +B2(JP(I),JP(J))*F2(3))+ZSUM*CCA(JP(I),JP(J))  

T=B1(JP(I),JP(J))*F3(K)+B2(JP(I),JP(J))*F3(3)  

GT(I)=GT(I)+MOLAL(J)*T  

GT(J)=GT(J)+MOLAL(I)*T

```

```

FZ2=FZ2+MOLAL(I)*MOLAL(J)*(B1(JP(I),JP(J))*F1(K)
1      +B2(JP(I),JP(J))*F1(3))
HT(I)=HT(I)+MOLAL(J)*CCA(JP(I),JP(J))
HT(J)=HT(J)+MOLAL(I)*CCA(JP(I),JP(J))
12    CONTINUE
13    CONTINUE
L=0
JXAM1=JXA-1
IF(JMA.GT.JXAM1) GO TO 100
DO 30 I=JMA,JXAM1
IP1=I+1
DO 36 J=IP1,JXA
L=L+1
H(J,I)=H(J,I)+2.0D0*(TA(L)+ETH(-Z(I),-Z(J)))
DO 28 K=JMC,JXC
H(J,I)=H(J,I)+MOLAL(K)*PSIA(L,JP(K))
H(I,K)=H(I,K)+MOLAL(J)*PSIA(L,JP(K))
H(J,K)=H(J,K)+MOLAL(I)*PSIA(L,JP(K))
28    CONTINUE
GT(I)=GT(I)+MOLAL(J)*ETHP(-Z(I),-Z(J))
GT(J)=GT(J)+MOLAL(I)*ETHP(-Z(I),-Z(J))
FZ2=FZ2+MOLAL(I)*MOLAL(J)*ETHP2(-Z(I),-Z(J))
36    CONTINUE
30    CONTINUE
100   JXCM1=JXC-1
IF(JMC.GT.JXCM1) GO TO 101
L=0
DO 34 I=JMC,JXCM1
IP1=I+1
DO 33 J=IP1,JXC
L=L+1
H(J,I)=H(J,I)+2.0D0*(TC(L)+ETH(Z(I),Z(J)))
DO 32 K=JMA,JXA
H(J,I)=H(J,I)+MOLAL(K)*PSIC(L,JP(K))
H(K,I)=H(K,I)+MOLAL(J)*PSIC(L,JP(K))
H(K,J)=H(K,J)+MOLAL(I)*PSIC(L,JP(K))
32    CONTINUE
GT(I)=GT(I)+MOLAL(J)*ETHP(Z(I),Z(J))
GT(J)=GT(J)+MOLAL(I)*ETHP(Z(I),Z(J))
FZ2=FZ2+MOLAL(I)*MOLAL(J)*ETHP2(Z(I),Z(J))
33    CONTINUE
34    CONTINUE
101   CONTINUE

102   CONVR=1.0D0/(WH2O*X(JM))
FZ2=0.5D0*FZ2
DO 91 I=JMC,JX
ZI=IABS(Z(I))
H(I,I)=CONVR*(H(I,I)+2.0D0*ZI*(HT(I)+ZI*GT(I))+1.0D0/MOLAL(I)
1      +ZI*ZI*ZI*ZI*FZ2)
H(I,JM)=H(I,JM)-MOLAL(I)*H(I,I)
IM1=I-1
IF(JMC.GT.IM1) GO TO 91
DO 90 J=JMC,IM1

```

```

ZJ=IABS(Z(J))
H(I,J)=CONVR*( H(I,J)+ZI*(HT(J)+ZI*GT(J))+ZJ*(HT(I)+ZJ*GT(I))
1      +ZI*ZI*ZJ*ZJ*FZ2)
H(J,JM)=H(J,JM)-MOLAL(I)*H(I,J)
H(I,JM)=H(I,JM)-MOLAL(J)*H(I,J)
90    CONTINUE
91    CONTINUE

600    T=0.0D0
      DO 92 J=JMC,JX
      H(J,JM)=WH2O*H(J,JM)
92    T=T-MOLAL(J)*H(J,JM)
      H(JM,JM)=T*WH2O
      RETURN

C IDEAL SOLUTION HESSIAN ELEMENTS
500    JM=JMIN(KPP)
      JX=JMAX(KPP)
      S=0.0D0
      DO 70 J=JM,JX
70    S=S+X(J)
      S=1.0D0/S
      DO 71 J=JM,JX
      DO 72 I=J,JX
      H(I,J)=-S
72    CONTINUE
      H(J,J)=1.0D0/X(J)-S
71    CONTINUE
      RETURN

C SAS
C DAVIES SOLUTION HESSIAN ELEMENTS
700    JM=JMIN(KPP)
      JX=JMAX(KPP)
      S=0.0D0
      DO 80 J=JM,JX
      S=S+X(J)
80    CONTINUE
      S=1.0D0/S
      IONIC=0.0D0
      CONVR=1.0D0/(WH2O*X(JM))
      DO 81 J=JMC,JX
      IONIC=IONIC+CONVR*X(J)*Z(J)*Z(J)
81    CONTINUE
      IONIC=IONIC*0.5D0
      DAVIES=1.0D0/(DSQRT(IONIC)+2.0D0*IONIC+IONIC**1.5D0)-0.6D0
      DO 82 J=JM,JX
      DO 83 I=J,JX
      H(I,J)=-0.75D0*APHI*Z(I)*Z(I)*Z(J)*Z(J)*DAVIES-S
83    CONTINUE
      H(J,J)=H(J,J)+1.0D0/X(J)
82    CONTINUE
      RETURN

C SAS
      END

```

```

C-----  

C  

C      THIS ROUTINE CALCULATES A REFINEMENT STEP FOR SOLUTION  

C      SPECIES WHOSE MOLE NUMBERS ARE AT OR BELOW THE  

C      BOUNDARY VALUE. FOR DEGENERATE SPECIES A NEWTON  

C      ITERATION PROCEDURE IS USED TO SOLVE FOR THE REFINEMENT  

C      DIRECTION.  

C  

C-----  

C      SUBROUTINE REFINE(Q2MAX,NCM)  

C  

C      SUBROUTINE ESTIMATES EQUILIBRIUM CONCENTRATIONS FOR  

C      SMALL CONCENTRATION SOLUTION PHASE VARIABLES AND  

C      COMPUTES A DESCENT DIRECTION  

C  

C      IMPLICIT DOUBLE PRECISION(A-H,O-Z)  

C      IMPLICIT INTEGER*2 (I-N)  

C      LOGICAL NEWCON,NOVAR  

C      COMMON/CONST/Q(96,96),MACT(96),JA(96),JI(96),NSC,NCC,NDC,NC0C  

C      COMMON/DUAL2/DEG(96,96)  

C      COMMON/DUALB/QC(96,96),IR(96)  

C      COMMON/SEARCH/X(96),P(96),G(96),W0(96),H(64,64),B(96)  

C      COMMON/PHASE/R11(96,16),JMIN(96),JMAX(96),KSPEC(96),KTYP(96)  

1     ,NP0C,N00C  

      COMMON/PRINT/PP,PP2,PD  

      LOGICAL PP,PP2,PD  

      DIMENSION W1(96),YS(96),V(20),DD(20),JDX(96)  

      DIMENSION KBCON(96)  

      PARAMETER(DELMP=0.5D0**55)  

      PARAMETER(DEL2=1.0D-12)  

      PARAMETER(DEL2P=DEL2*(1.0D0+DEL2),DEL2M=DEL2*(1.0D0-DEL2))  

      PARAMETER(DEL4=1.0D-13)  

      PARAMETER(DEL1=1.0D-07, DELLB=1.0D-20)  

      NC=NCC  

      NC0=NC0C  

      Q2MAX=0.0D0  

      IF(NC.EQ.NC0) RETURN  

      NS=NSC  

      ND=NDC  

      NINACT=NS-NC-ND+NC0  

      NC0P1=NC0+1  

      NOVAR=.TRUE.  

C  

C      IDENTIFY CONSTRAINED SOLUTION SPECIES (KBCON=-1)  

C  

      DO 962 J=1,96  

      DO 962 I=1,96  

962      DEG(I,J)=0.0D0  

      DO 1 I=NC0P1,NC  

      IF(PP) THEN  

        WRITE(6,8324) I,JA(I),Q(JA(I),I)  

8324      FORMAT(2I5,1PD20.4)

```

```

END IF
IF(KTYP(KSPEC(JA(I))).GT.1) THEN
  KBCON(I)--1
  NOVAR=.FALSE.
ELSE
  KBCON(I)=0
END IF
1  CONTINUE
IF(NOVAR) RETURN
NU=0
NSS=0
C
C  IDENTIFY CONSTRAINTS ON SMALL SOLUTION SPECIES DIRECTION
C  VECTOR
IF(ND.GT.0) THEN
  NUX=0
  DO 8 JJ=1,ND
    J1=JA(JJ)
    KBCON(JJ)=0
    IF(X(J1).LT.DEL2*100.0D0) THEN
      NUX=NUX+1
      IF(NSS.GT.0) THEN
        DO 2 K=1,NSS
          V(K)=0.0D0
2    CONTINUE
      END IF
      IF(KTYP(KSPEC(J1)).GT.1) THEN
        NM1=NSS
        NSS=NSS+1
        JDX(NSS)=J1
        W1(NSS)=0.0D0
        V(NSS)=-1.0D0
        DEG(NSS,NUX)=-1.0D0
        KBCON(JJ)=NSS
        IF(NM1.GT.0) THEN
          DO 3 K=1,NM1
            QC(NSS,K)=0.0D0
            QC(K,NSS)=0.0D0
3    CONTINUE
        END IF
        QC(NSS,NSS)=1.0D0
      END IF
      DO 7 II=NC0P1,NC
        I=(NC0P1+NC)-II
        TAU=Q(J1,I)
        IF(I.LT.NC) THEN
          IP1=I+1
          DO 4 I2=IP1,NC
            TAU=TAU-YS(I2)*Q(JA(I2),I)
4    CONTINUE
      END IF
      YS(I)=TAU/Q(JA(I),I)
      IF(DABS(YS(I)).GT.DEL4) THEN
        KK=KBCON(I)

```

```

IF(KK.GT.0) THEN
  V(KK)=YS(I)
  DEG(KK,NUX)=YS(I)
ELSE IF(KK.EQ.-1) THEN
  NM1=NSS
  NSS=NSS+1
  JDX(NSS)=JA(I)
  V(NSS)=YS(I)
  DEG(NSS,NUX)=YS(I)
  KBCON(I)=NSS
  IF(NM1.GT.0) THEN
    DO 5 K=1,NM1
    QC(NSS,K)=0.0D0
    QC(K,NSS)=0.0D0
    CONTINUE
    END IF
    QC(NSS,NSS)=1.0D0
  END IF
END IF
CONTINUE
C
7
IF(NU.LT.NSS) THEN
  NEWCON=.FALSE.
  SIG=0.0D0
  NUP1=NU+1
  DO 84 I=NUP1,NSS
  TAU=0.0D0
  DO 85 J=1,NSS
  TAU=TAU+V(J)*QC(J,I)
  CONTINUE
  IF(DABS(TAU).GT.DEL4) NEWCON=.TRUE.
  SIG=SIG+TAU*TAU
  YS(I)=TAU
  CONTINUE
  IF(NEWCON) THEN
    NU=NUP1
    IF(NU.LT.NSS) THEN
      SIG=DSIGN(DSQRT(SIG),YS(NU))
      YS(NU)=YS(NU)+SIG
      PI=1.0D0/(YS(NU)*SIG)
      DO 81 J=1,NSS
      TAU=0.0D0
      DO 82 I=NU,NSS
      TAU=TAU+QC(J,I)*YS(I)
      CONTINUE
      TAU=TAU*PI
      DO 83 I=NU,NSS
      QC(J,I)=QC(J,I)-YS(I)*TAU
    CONTINUE
    CONTINUE
    END IF
  END IF
END IF
C

```

```

        END IF
8      CONTINUE
END IF
IF(PP) THEN
    WRITE(6,8567) NU
    IF(NU.GT.0) THEN
        DO 901 JS=1,NSS
        WRITE(6,777) JDX(JS),(DEG(JS,I),I=1,NUX)
777      FORMAT(I5,7F10.6,/,5X,7F10.6,/,5X,7F10.6)
901      CONTINUE
8567      FORMAT(' CONSTRAINT EQUATIONS  NU= ',I5)
8568      FORMAT(7F10.6)
        END IF
    END IF

C
C      ESTIMATE EQUILIBRIUM MOLE NUMBER USING MULTIPLIER AND
C      ASSUMING TRACE CONCENTRATION
C
DO 12 II=NC0P1,NC
I=(NC0P1+NC)-II
J1=JA(I)
TAU=0.0D0
IF(NINACT.GT.0) THEN
    DO 9 J=1,NINACT
    TAU=TAU+G(J1(J))*Q(J1(J),I)
9      CONTINUE
    END IF
IF(ND.GT.0) THEN
    DO 10 J=1,ND
    TAU=TAU+G(JA(J))*Q(JA(J),I)
10     CONTINUE
    END IF
IF(I.LT.NC) THEN
    IP1=I+1
    DO 11 J=IP1,NC
    TAU=TAU+YS(JA(J))*Q(JA(J),I)
11     CONTINUE
    END IF
YS(J1)=--TAU/Q(J1,I)
IF(KBCON(I).EQ.-1) THEN
    PI=DMAX1(X(J1)*DEXP(YS(J1)-G(J1)), DELLB)
    P(J1)=PI-X(J1)
    IF(DABS(P(J1)).GT.Q2MAX*X(J1)) Q2MAX=DABS(P(J1))/X(J1)
    NCM=I
    TAU3=G(J1)-YS(J1)
    TAU2=X(J1)+P(J1)
    IF(PP) THEN
        WRITE(6,3232) J1,TAU3,TAU2,P(J1),X(J1)
    END IF
3232      FORMAT(I4,1P4D16.8)
ELSE IF(KBCON(I).GT.0) THEN
    W1(KBCON(I))=YS(J1)-G(J1)
    IF(PP) THEN
        WRITE(6,3232) J1,-W1(KBCON(I))

```

```

        END IF
        NCM=I
    END IF
12   CONTINUE
    IF(PP) THEN
        IF(NU.EQ.0) WRITE(6,3845) Q2MAX
3845      FORMAT(' Q2MAX =',1PD12.5)
    END IF
    IF(NU.EQ.0) RETURN
C
C -----
C       NEWTON ITERATION FOR A DEGENERATE OPTIMAL SOLUTION
C
C       INITIALIZATION (LEAST SQUARE ESTIMATE FOR V(I) I=1,NU)
C
        FUNC0=1.0D+30
        NUP1=NU+1
        NUP2=NU+2
        DO 16 I=1,NU
        IP1=I+1
        DO 14 K=1,NU
        TAU=0.0D0
        DO 13 J=1,NSS
        TAU=TAU+X(JDX(J))*QC(J,I)*QC(J,K)
13   CONTINUE
        H(K,IP1)=TAU
14   CONTINUE
        TAU=0.0D0
        PI=0.0D0
        DO 15 J=1,NSS
        SIG=X(JDX(J))*QC(J,I)
        TAU=TAU+SIG
        PI=PI+SIG*W1(J)
15   CONTINUE
        V(I)=0.0D0
        DD(I)=TAU
        H(I,NUP2)--PI
16   CONTINUE
        ITER=0
        IFUN=0
C
C       SOLVE LINEAR EQUATIONS FOR NEWTON DIRECTION (OR
C       INITIALIZATION)
1002  GAMMA=0.0D0
        ZETA=0.0D0
        THETA=0.0D0
        DO 27 I=1,NU
        IP1=I+1
        IF(DABS(H(I,IP1)).GT.GAMMA) GAMMA=DABS(H(I,IP1))
        IF(I.LT.NU) THEN
            DO 26 J=IP1,NU
            IF(DABS(H(I,J+1)).GT.ZETA) ZETA=DABS(H(I,J+1))
26   CONTINUE
        IF(I.EQ.1) THETA=ZETA

```

```

END IF
27  CONTINUE
    DELTA=(0.5D0***45)*DMAX1(GAMMA,ZETA,DELLB)
    BETA2=1.0D0/DMAX1(GAMMA,DELTA,ZETA/NU)
    TAU=DMAX1(DABS(H(1,2)), DELTA, THETA*THETA*BETA2)
    H(1,2)=1.0D0/TAU
    IF(NU.EQ.1) THEN
        TAU=H(1,3)
        H(1,3)=H(1,3)*H(1,2)
    ELSE
        DO 31 J=2,NU
        JM1=J-1
        JP1=J+1
        JP2=J+2
        PHI=H(J,JP1)
        DO 28 K=1,JM1
        PI=H(K,JP1)*H(K,K+1)
        PHI=PHI-PI*H(K,JP1)
        H(K,JP1)=PI
    28  CONTINUE
        THETA=0.0D0
        DO 30 I=JP2,NUP2
        TAU=H(J,I)
        DO 29 K=1,JM1
        TAU=TAU-H(K,JP1)*H(K,I)
    29  CONTINUE
        H(J,I)=TAU
        IF(I.LT.NUP2) THEN
            IF(DABS(TAU).GT.THETA) THETA=DABS(TAU)
        END IF
    30  CONTINUE
        SIG=DMAX1( DABS(PHI), DELTA, THETA*THETA*BETA2)
        H(J,JP1)=1.0D0/SIG
    31  CONTINUE
        H(NU,NUP2)=H(NU,NUP2)*H(NU,NUP1)
        DO 33 JMN=2,NU
        J=NUP1-JMN
        JP1=J+1
        TAU=H(J,NUP2)*H(J,JP1)
        DO 32 K=JP1,NU
        TAU=TAU-H(J,K+1)*H(K,NUP2)
    32  CONTINUE
        H(J,NUP2)=TAU
    33  CONTINUE
    END IF
C
C      COMPUTE MOLE NUMBER ESTIMATES, SHORTEN STEP LENGTH TO
C      INSURE DECREASE OF OBJECTIVE FUNCTION
C
    ALPHA=1.0D0
1001  FUNC=0.0D0
        DO 50 I=1,NSS
        J=JDX(I)
        TAU0=W1(I)

```

```

PI=0.0D0
DO 51 K=1,NU
TAU0=TAU0+QC(I,K)*V(K)
PI=PI+QC(I,K)*H(K,NUP2)
51  CONTINUE
TAU=TAU0+ALPHA*PI
IF(TAU.GT.60.0D0) THEN
    ALPHA=(59.9D0-TAU0)/PI
    GO TO 1001
END IF
YS(I)=DMAX1( X(J)*DEXP(TAU), DELLB)
FUNC=FUNC+YS(I)
P(J)=YS(I)-X(J)
50  CONTINUE
DO 52 I=1,NU
FUNC=FUNC-DD(I)*(V(I)+ALPHA*H(I,NUP2))
52  CONTINUE
IFUN=IFUN+1
IF(FUNC.GE.FUNC0) THEN
    IF(ITER.EQ.0) GO TO 1000
    PTG=0.0D0
    DO 53 J=1,NSS
    TAU=0.0D0
    DO 54 I=1,NU
    TAU=TAU+QC(J,I)*H(I,NUP2)
54  CONTINUE
    PTG=PTG+P(JDX(J))*TAU
53  CONTINUE
    IF(PTG.LE.0.0D0) GO TO 1000
    ALPHA=ALPHA*0.5D0
    GO TO 1001
END IF
C
C      TEST FOR FEASIBILITY (ZERO GRADIENT) AND COMPUTE HESSIAN
C      MATRIX FOR NEWTON ITERATION
C
1000 FUNC0=FUNC
ITER=ITER+1
SIG=0.0D0
DO 64 K=1,NU
V(K)=V(K)+ALPHA*H(K,NUP2)
DO 65 I=K,NU
TAU=0.0D0
DO 66 J=1,NSS
TAU=TAU+YS(J)*QC(J,K)*QC(J,I)
66  CONTINUE
H(K,I+1)=TAU
65  CONTINUE
TAU=0.0D0
DO 67 J=1,NSS
TAU=TAU+P(JDX(J))*QC(J,K)
67  CONTINUE
H(K,NUP2)--TAU
IF(DABS(TAU).GT.SIG) SIG=DABS(TAU)

```

```
64  CONTINUE
IF(SIG.GT.DELMP) GO TO 1002
C
C      CONVERGED WITHIN DELMP OF FEASIBILITY
C      COMPUTE REMAINING PART OF Q2MAX
C
DO 68 I=1,NSS
J=JDX(I)
IF(DABS(P(J)).GT.Q2MAX*X(J)) Q2MAX=DABS(P(J))/X(J)
IF(PP) THEN
    TAU=X(J)+P(J)
    WRITE(6,3234) J,-W1(I),TAU,P(J),X(J)
3234      FORMAT(I4,1P4D16.8)
END IF
68  CONTINUE
IF(PP) WRITE(6,3845) Q2MAX
RETURN
END
```

```

C-----  

C THIS SUBROUTINE ADDS A SPECIES CONSTRAINT (SPECIES J0) TO  

C THE CONSTRAINT MATRIX. IF THE NEW CONSTRAINT IS  

C DEGENERATE THE CONSTRAINT SET IS NOT ALTERED, BUT A NEW  

C ELEMENT TO JA IS ADDED TO NOTE THE DEGENERATE  

C CONSTRAINT.  

C-----  

C-----  

SUBROUTINE CONADD(J0)  

IMPLICIT DOUBLE PRECISION(A-H,O-Z)  

IMPLICIT INTEGER*2 (I-N)  

LOGICAL REDUN,NODEG  

COMMON/CONST/Q(96,96),MACT(96),JA(96),JI(96),NS,NC,ND,NC0  

DIMENSION V(96)  

PARAMETER(DEL4=1.0D-13)  

IF(NS.EQ.NC) GO TO 99  

NC=NC+1  

SIG=0.0D0  

NODEG=.TRUE.  

DO 1 I=NC,NS  

V(I)=Q(J0,I)  

IF(DABS(V(I)).GT.DEL4) NODEG=.FALSE.  

Q(J0,I)=0.0D0  

SIG=SIG+V(I)*V(I)  

1 CONTINUE  

IF(NODEG) GO TO 99  

SIG=DSIGN(DSQRT(SIG),V(NC))  

NINACT=(NS+NC0)-(NC+ND)  

JI(-MACT(J0))-JI(NINACT+1)  

MACT(JI(NINACT+1))=MACT(J0)  

JA(NC)=J0  

MACT(J0)=NC  

IF(NS.EQ.NC) THEN  

Q(J0,NC)=V(NC)  

IF(NINACT.GT.0) THEN  

DO 2 J=1,NINACT  

ND=ND+1  

JA(ND)=JI(J)  

MACT(JI(J))=ND  

2 CONTINUE  

END IF  

ELSE  

Q(J0,NC)--SIG  

IF(NINACT.GT.0) THEN  

NDS=ND  

V(NC)=V(NC)+SIG  

PI=1.0D0/(V(NC)*SIG)  

DO 5 MI=1,NINACT  

J=JI(MI)  

TAU=0.0D0  

DO 3 I=NC,NS  

TAU=TAU+V(I)*Q(J,I)  

CONTINUE  

3

```

```

TAU=TAU*PI
REDUN=.TRUE.
DO 4 I=NC,NS
T=Q(J,I)-V(I)*TAU
IF(DABS(T).GT.DEL4) THEN
  Q(J,I)=T
  IF(I.GT.NC) REDUN=.FALSE.
ELSE
  Q(J,I)=0.0D0
END IF
4  CONTINUE
IF(REDUN) THEN
  ND=ND+1
  JA(ND)=J
END IF
5  CONTINUE
IF(ND.GT.NDS) THEN
  NDS=NDS+1
  DO 6 JD=NDS,ND
  J=JA(JD)
  IF(NINACT.GT.0) THEN
    JI(-MACT(J))=JI(NINACT)
    MACT(JI(NINACT))=MACT(J)
    NINACT=NINACT-1
  END IF
  MACT(J)=JD
  CONTINUE
6  END IF
  END IF
  RETURN
C
C -----
C      ATTEMPT TO ADD A CONSTRAINED VARIABLE
C
99  WRITE(6,9999) JO
9999 FORMAT(' CONADD ERROR',I5)
STOP
END

```

```

C-----
C THIS SUBROUTINE DELETES A SPECIES CONSTRAINT (SPECIES JR).
C-----
C-----  

SUBROUTINE CONDEL(JR)
IMPLICIT DOUBLE PRECISION(A-H,O-Z)
IMPLICIT INTEGER*2 (I-N)
COMMON/CONST/Q(96,96),MACT(96),JA(96),JI(96),NS,NC,ND,NC0
PARAMETER(DEL4=1.0D-13)
MREM=MACT(JR)
IF(MREM.LE.NC0) GO TO 99
NC=NC-1
NINACT=(NS+NC0)-(NC+ND)
JI(NINACT)=JR
MACT(JR)=NINACT
IF(MREM.LE.NC) THEN
  DO 4 MM=MREM,NC
    MP1=MM+1
    J0=JA(MP1)
    JA(MM)=J0
    MACT(J0)=MM
    CMM=Q(J0,MM)
    CMP1=Q(J0,MP1)
    SIG=DSIGN(DSQRT(CMM*CMM+CMP1*CMP1),CMM)
    CMM=CMM+SIG
    Q(J0,MM)=-SIG
    Q(J0,MP1)=0.0D0
    V1=1.0D0/SIG
    V2=CMP1/(SIG*CMM)
    IF(NINACT.GT.0) THEN
      DO 1 I=1,NINACT
        J=JI(I)
        TAU=Q(J,MM)*V1+Q(J,MP1)*V2
        Q(J,MM)=Q(J,MM)-CMM*TAU
        Q(J,MP1)=Q(J,MP1)-CMP1*TAU
1      CONTINUE
    END IF
    IF(ND.GT.0) THEN
      DO 2 I=1,ND
        J=JA(I)
        TAU=Q(J,MM)*V1+Q(J,MP1)*V2
        Q(J,MM)=Q(J,MM)-CMM*TAU
        Q(J,MP1)=Q(J,MP1)-CMP1*TAU
2      CONTINUE
    END IF
    IF(MM.LT.NC) THEN
      DO 3 I=MP1,NC
        J=JA(I+1)
        TAU=Q(J,MM)*V1+Q(J,MP1)*V2
        Q(J,MM)=Q(J,MM)-CMM*TAU
        Q(J,MP1)=Q(J,MP1)-CMP1*TAU
3      CONTINUE
    END IF

```

```
4      CONTINUE
END IF
IF(ND.GT.0) THEN
  NDS=ND
  NCP1=NC+1
  ND=0
  DO 5 JD=1,NDS
    JX=JA(JD)
    JA(JD)=0
    T=Q(JX,NCP1)
    IF(DABS(Q(JX,NCP1)).LT.DEL4) THEN
      ND=ND+1
      JA(ND)=JX
      MACT(JX)=ND
      Q(JX,NCP1)=0.0D0
    ELSE
      NINACT=NINACT+1
      JI(NINACT)=JX
      MACT(JX)--NINACT
    END IF
5      CONTINUE
END IF
RETURN
C
C -----
C      ATTEMPT TO DELETE A NON-EXPLICITLY CONSTRAINED SPECIES
C
99  WRITE(6,9999) JR
9999 FORMAT(' CONDEL ERROR',I5)
STOP
END
```

```

C-----  

C THIS SUBROUTINE SOLVES THE SUBMINIZATION PROBLEM FOR  

C SOLUTION PHASES THAT ARE NOT PRESENT IN ORDER TO  

C CALCULATE THE PHASE MULTIPLIER. IT ALSO CALCULATES THE  

C PHASE MULTIPLIER FOR PURE PHASES. IT THEN PICKS THE  

C LOWEST NEGATIVE MULTIPLIER.  

C-----  

C-----  

C SUBROUTINE DUAL(SUBM,ADDPHS,ZOUTEN,ERROR,WMIN,NPP,NCM,KPH)  

C-----  

C-----  

C SUBROUTINE DETERMINES MINIMUM DESCENT DIRECTION WHEN  

C CONSTRAINTS ARE TO BE RELAXED  

C-----  

IMPLICIT DOUBLE PRECISION(A-H,O-Z)  

IMPLICIT INTEGER*2 (I-N)  

INTEGER*2 Z  

CHARACTER*24 A1  

LOGICAL ZDUAL(96),ZOUTEN(96),VTYP(96),ERROR,NODEG,SVAR  

LOGICAL TEST,REDUN  

LOGICAL NORMAL,ADDCON,SENS,SUBMIN, SUBM,POSDEF,ADDPHS,  

1,NOMIN,CONCAV  

COMMON/CONST/Q(96,96),MACT(96),JA(96),JI(96),NSC,NCC,NDC,NC0C  

COMMON/DUALB/QB(96,96),MACB(96),JAQ(96),JIQ(96),NSBC,NCBC,NDBC  

1,NCB0C  

COMMON/LABEL/Z(96),U0RT(96),A1(96)  

COMMON/PHASE/R11(96,16),JMIN(96),JMAX(96),KSPEC(96),KTYP(96)  

1,NP0C,N00C  

COMMON/SEARCH/X(96),P(96),G(96),PS(96),H(64,64),B(96),S(32)  

COMMON/PRINT/PP,PP2,PD  

COMMON/MMM/MUL,WWW(96)  

LOGICAL PP,PP2,PD,MUL  

DIMENSION JQX(96),JXQ(96),JBLK(96),KPH(96),KBCON(96)  

DIMENSION YS(96),V(96),BB(96),GSAV(96),XSAV(96),W1(96)  

DIMENSION SL(96),SLP(96)  

PARAMETER(DEL1=1.0D-07,DEL2=1.0D-12,DEL3=1.0D-12,DEL4=1.0D-13)  

PARAMETER(DEL5=1.0D-01,DEL6=1.0D-04)  

PARAMETER(DEL3P=DEL3*(1.0D0+DEL4),DEL3M=DEL3*(1.0D0-DEL4))  

PARAMETER(DEL2P=DEL2*(1.0D0+DEL2))  

PARAMETER(DELLB=1.0D-20)  

PARAMETER(NFUN=2000)  

C-----  

C-----  

C AT A SUBSPACE MINIMUM RESET ALL ZOUTENDIJK FLAGS  

C-----  

NS=NSC  

NC=NCC  

NC0=NC0C  

IF(SUBM) THEN  

  DO 1 J=1,NS  

    ZOUTEN(J)=.FALSE.  

1   CONTINUE  

END IF

```

```

IF(NC.EQ.NC0) RETURN
NSP1=NS+1
ND=NDC
NC0P1=NC0+1
NINACT=NS-NC-ND+NC0
C
C IDENTIFY BLOCKS OF ACTIVE CONSTRAINTS
C
C KBCON(I) = 0, SINGLE SPECIES BLOCK
C           K0, SPECIES IN NON-DEG. ZERO SOLUTION PHASE K0
C           -JJ, DEG. BLOCK ASSOCIATED WITH JA(JJ)
C           NSP1, CANNOT BE DELETED (TOO LARGE TO BE
C           CONSIDERED DEGENERATE OR ZOUTEN=TRUE)
C
DO 2 I=1,NC
IF(I.GT.NC0) THEN
  J=JA(I)
  IF(ZOUTEN(J)) THEN
    KBCON(I)=NSP1
  ELSE IF(KTYP(KSPEC(J)).LT.-1) THEN
    KBCON(I)=KSPEC(J)
  ELSE
    KBCON(I)=0
  END IF
ELSE
  KBCON(I)=NSP1
END IF
2 CONTINUE
IF(ND.GT.0) THEN
  DO 7 JJ=1,ND
  J1=JA(JJ)
  IF(X(J1).LT.DEL2P) THEN
    W1(J1)=G(J1)
    KBCON(JJ)=-JJ
    NOMIN=.FALSE.
    IF(ZOUTEN(JJ)) NOMIN=.TRUE.
    DO 5 II=NC0P1,NC
    I=(NC0P1+NC)-II
    TAU=Q(J1,I)
    IF(I.LT.NC) THEN
      IP1=I+1
      DO 3 I2=IP1,NC
      TAU=TAU-YS(I2)*Q(JA(I2),I)
    3 CONTINUE
    END IF
    YS(I)=TAU/Q(JA(I),I)
    IF(DABS(YS(I)).GT.DEL4) THEN
      NBLK=KBCON(I)
      IF(NBLK.EQ.NSP1) NOMIN=.TRUE.
      IF(NBLK.EQ.0) THEN
        KBCON(I)=-JJ
      ELSE IF(NBLK.NE.-JJ) THEN
        DO 4 K=1,NC
        IF(KBCON(K).EQ.NBLK) KBCON(K)=-JJ
      4 CONTINUE
    END IF
  END IF
END IF

```

```

4      CONTINUE
      END IF
      END IF
5      CONTINUE
      IF(NOMIN) THEN
          DO 6 K=1,NC
              IF(KBCON(K).EQ.-JJ) KBCON(K)=NSP1
6      CONTINUE
      END IF
      END IF
7      CONTINUE
END IF
C
C      LEAST SQUARE MULTIPLIER CALCULATION (GRADIENT ASSUMED
C      FOR DEGENERATE SPECIES)
C
NOMIN=.TRUE.
DO 11 II=NC0P1,NC
I=(NC0P1+NC)-II
J1=JA(I)
TAU=0.0D0
IF(NINACT.GT.0) THEN
    DO 8 J=1,NINACT
        TAU=TAU+G(JI(J))*Q(JI(J),I)
8      CONTINUE
END IF
IF(ND.GT.0) THEN
    DO 9 J=1,ND
        TAU=TAU+G(JA(J))*Q(JA(J),I)
9      CONTINUE
END IF
IF(ILT.NC) THEN
    IP1=I+1
    DO 10 J=IP1,NC
        TAU=TAU+W1(JA(J))*Q(JA(J),I)
10     CONTINUE
END IF
W1(J1)--TAU/Q(J1,I)
K0=KBCON(I)
IF(K0.EQ.0) THEN
    KBCON(I)=NSP1
    GFE0=G(J1)-W1(J1)
    WWW(KSPEC(J1))=GFE0
    IF(PP.OR.PD) THEN
        WRITE(6,5656) GFE0,J1
5656    FORMAT(' DUAL SIMPLE ',1PD20.12,I5)
    END IF
    IF(GFE0.LT.WMIN) THEN
        NBMIN=1
        JBMIN(1)--J1
        WMIN=GFE0
        ADDPHS=.TRUE.
    END IF
    ELSE

```

```

      IF(K0.NE.NSP1) NOMIN=.FALSE.
      END IF
11   CONTINUE
C      IF NOMIN=TRUE, THERE ARE NO MULTIPLE SPECIES BLOCKS
C      FOR WHICH THE MULTIPLIER NEED BE EVALUATED. IF
C      ADDPHS IS SET INITIALIZE FOR ADDPHS MODIFIED DIRECTION.
C      RETURN
C
C      IF(NOMIN) GO TO 3000
C
C -----
C      EVALUATION OF MULTIPLIERS FOR MULTIPLE SPECIES BLOCKS
C
      DO 100 ICON=NC0P1,NC
      NBLK=KBCON(ICON)
      IF(NBLK.EQ.NSP1) GO TO 100
      NSB=0
      NBLOCK=0
      NOMIN=.FALSE.
      CONCAV=.FALSE.
      NCON=0
      DO 13 I=1,NC
      IF(I.EQ.NC0P1) NCON=NSB+1
      IF(KBCON(I).EQ.NBLK) THEN
          KBCON(I)=NSP1
          J=JA(I)
          NSB=NSB+1
          JQX(NSB)=J
          JXQ(J)=NSB
          JIQ(NSB)=NSB
          MACB(NSB)--NSB
          ZDUAL(NSB)=.FALSE.
          KK=KTYP(KSPEC(J))
          IF(KK.LT.-1) THEN
              IF(-KK.EQ.3.OR.-KK.EQ.5.OR.-KK.EQ.7) CONCAV=.TRUE.
              NB=NBLOCK
              NBLOCK=NBLOCK+1
              K0=KSPEC(J)
              KBLK(NBLOCK)=K0
              IF(NB.GT.0) THEN
                  DO 12 K=1,NB
                  IF(KBLK(K).EQ.KSPEC(J)) NBLOCK=NBLOCK-1
12       CONTINUE
              END IF
              VTYP(NSB)=.TRUE.
              SL(NSB)=DEL3
              SLP(NSB)=DEL3P
              X(J)=PS(J)
              IF(X(J).LT.DEL3P) X(J)=DEL3
              GSAV(NSB)=U0RT(J)
              UORT(J)=U0RT(J)-W1(J)
          ELSE
              VTYP(NSB)=.FALSE.
              XSAV(NSB)=X(J)
      END IF
  END IF
END IF
IF(K0.NE.NSP1) NOMIN=.FALSE.
END IF

```

```

X(J)=.01D0
GSAV(NSB)=G(J)
G(J)=G(J)-W1(J)
NBLOCK=NBLOCK+1
KBLK(NBLOCK)--J
END IF
END IF
13 CONTINUE
GFE0=0.0D0
ALPHA=0.0D0
PTGA=0.0D0
DO 14 KP=1,NBLOCK
K=KBLK(KP)
IF(K.GT.0) THEN
  WRITE(*,*) 'CALLING GRAD FROM DUAL @ LN 231'
  CALL GRAD(K,ALPHA,PTGA,GFE0)
ELSE
  GFE0=GFE0+X(-K)*G(-K)
END IF
14 CONTINUE
SVAR=.FALSE.
KK=KBLK(1)
IF(NBLOCK.EQ.1.AND.KK.GT.0) THEN
  IF(CONCAV) THEN
    JM=JMIN(KK)
    JX=JMAX(KK)
    DO 123 J=JM,JX
    IF(DABS(S(J)).GT.DEL3P) THEN
      SL(JXQ(J))=S(J)
      SLP(JXQ(J))=S(J)*(1.0D0+DEL4)
      IF(X(J).LT.SLP(JXQ(J))) X(J)=SL(JXQ(J))
      SVAR=.TRUE.
    END IF
  123   CONTINUE
  END IF
ELSE
END IF
GLOBLB=-1.0D+20
ITER=0
IFUN=1
ISENS=0
C
C   INITIALIZE ORTHOGONAL Q MATRIX FOR BLOCK MINIMIZATION
C
NSBP1=NSB+1
AN=NSB
SQN=DSQRT(AN)
AN=-1.0D0/(AN+SQN)
BN=AN*(1.0D0+SQN)
QB(1,1)=1.0D0+BN*(1.0D0+SQN)
QB(1,NSBP1)=0.0D0
DO 16 I=2,NSB
DO 15 J=2,NSB
QB(J,I)=AN

```

```

15 CONTINUE
    QB(I,1)=BN
    QB(1,I)=BN
    QB(I,I)=1.0D0+AN
    QB(I,NSBP1)=0.0D0
16 CONTINUE
    BB(1)=-1.0D0/SQN
    NCB=1
C
C      ADD ADDITIONAL CONSTRAINTS DUE TO DEGENERACY, IF ANY
C
    NINACT=NSB
    NDB=0
    IF(NCON.GT.1) THEN
        NCONM1=NCON-1
        DO 24 JQD=1,NCONM1
            JXD=JQX(JQD)
            SIG=0.0D0
            PI=0.0D0
            NCBP1=NCB+1
            DO 18 II=NCON,NSB
                JQ=(NCON+NSB)-II
                JX=JQX(JQ)
                I=MACT(JX)
                TAU=Q(JXD,I)
                IF(JQ.LT.NSB) THEN
                    JQP1=JQ+1
                    DO 17 JQ2=JQP1,NSB
                        TAU=TAU-YS(JQ2)*Q(JQX(JQ2),I)
17        CONTINUE
                    END IF
                    YS(JQ)=TAU/Q(JX,I)
18        CONTINUE
                    NODEG=.FALSE.
                    DO 20 I=1,NSB
                        TAU-=QB(JQD,I)
                    DO 19 J=NCON,NSB
                        TAU=TAU+YS(J)*QB(J,I)
19        CONTINUE
                    IF(I.LT.NCBP1) THEN
                        PI=PI-BB(I)*TAU
                    ELSE
                        IF(DABS(TAU).GT.DEL4) NODEG=.TRUE.
                        V(I)=TAU
                        SIG=SIG+TAU*TAU
                    END IF
20        CONTINUE
                    IF(NODEG) THEN
                        NCB=NCBP1
                        SIG=DSIGN(DSQRT(SIG),V(NCB))
                        IF(NCB.EQ.NSB) THEN
                            BB(NCB)=PI/V(NCB)
                        ELSE
                            V(NCB)=V(NCB)+SIG

```

```

      BB(NCB)=-PI/SIG
      PI=1.0D0/(V(NCB)*SIG)
      DO 23 J=1,NSB
      TAU=0.0D0
      DO 21 I=NCB,NSB
      TAU=TAU+QB(J,I)*V(I)
21      CONTINUE
      TAU=TAU*PI
      DO 22 I=NCB,NSB
      QB(J,I)=QB(J,I)-TAU*V(I)
22      CONTINUE
23      CONTINUE
      END IF
      END IF
24      CONTINUE
      DO 26 J=1,NSB
      REDUN=.TRUE.
      DO 25 I=NCBP1,NSB
      IF(DABS(QB(J,I)).GT.DEL4) REDUN=.FALSE.
25      CONTINUE
      IF(NSB.EQ.NCB) REDUN=.TRUE.
      IF(REDUN) THEN
          NDB=NDB+1
          MACB(JIQ(NINACT))=MACB(J)
          JIQ(-MACB(J))=JIQ(NINACT)
          NINACT=NINACT-1
          JAQ(NDB)=J
          MACB(J)=NDB
      END IF
26      CONTINUE
      END IF
      NCB0=NCB
      NDBC=NDB
      NCB0P1=NCB0+1
      NCBC=NCB
      NSBC=NSB
      NCB0C=NCB0
C
C      ADD CONSTRAINTS FOR SMALL SPECIES, IF ANY
C
      DO 27 JQ=1,NSB
      IF(X(JQX(JQ)).LT.SLP(JQ)) THEN
          IF(MACB(JQ).LT.0) THEN
              WRITE(*,*) 'CALLING DADD @ LN 368'
              CALL DADD(JQ)
          END IF
          END IF
27      CONTINUE
      NORMAL=.FALSE.
      ADDCON=.FALSE.
C
C      TEST FOR FEASIBILITY
C
1000    SENS=.FALSE.

```

```

DO 81 I=1,NCB0
TAU=BB(I)
DO 80 JQ=1,NSB
TAU=TAU-X(JQX(JQ))*QB(JQ,I)
80  CONTINUE
IF(DABS(TAU).GT.DEL4) SENS=.TRUE.
YS(I)=TAU
81  CONTINUE
C
C   IF FEASIBLE COMPUTE NORMAL DESCENT DIRECTION, OTHERWISE
C   COMPUTE A SENSITIVITY DIRECTION TO FEASIBLE DOMAIN
IF(SENS) THEN
  ISENS=ISENS+1
  IF(ISENS.GT.100) THEN
    WRITE(6,9973) ISENS
9973    FORMAT(' SUBROUTINE DUAL #9973, ISENS=',I5)
    ERROR=.TRUE.
    RETURN
  END IF
  NCMIN=NCB0P1
  NJMIN=1
  AMAX=1.0D0
ELSE
  NORMAL=.TRUE.
  GO TO 1002
END IF
C
C   MODIFIED STEP DIRECTION CALCULATION
C
1001  NCB=NCBC
      NDB=NDBC
      NU=NSB-NCB
      NUP1=NU+1
      NUT=NUP1
      NUP2=NU+2
      NINACT=NU-NDB+NCB0
      PTG=0.0D0
      J0Q=0
      AMAX=1.0D0
      IF(NCB.GE.NCMIN) THEN
        DO 28 I=NCMIN,NCB
          JQ=JAQ(I)
          JX=JQX(JQ)
          IF(DABS(P(JX)).GT.DEL4) THEN
            TAU=P(JX)
            PTG=PTG+TAU*G(JX)
            QB(JQ,NSBP1)=TAU
          ELSE
            TAU=0.0D0
          END IF
          IF(I.GT.NJMIN) THEN
            IM1=I-1
            DO 82 J=NJMIN,IM1
              TAU=TAU-QB(JQ,J)*YS(J)
            82  CONTINUE
          END IF
        28  CONTINUE
      END IF
    END IF
  END IF
END IF

```

```

82      CONTINUE
        END IF
        YS(I)=TAU/QB(JQ,I)
28      CONTINUE
        END IF
        IF(NDB.GT.0) THEN
          DO 30 J=1,NDB
            JQ=JAQ(J)
            JX=JQX(JQ)
            TAU=0.0D0
            DO 29 I=NJMIN,NCB
              TAU=TAU+QB(JQ,I)*YS(I)
29      CONTINUE
            P(JX)=TAU
            PTG=PTG+TAU*G(JX)
            QB(JQ,NSBP1)=TAU
            IF(VTYP(JQ)) THEN
              IF(X(JX)+AMAX*TAU.LT.SL(JQ)*(1.0D0-DEL4)) THEN
                AMAX=DMAX1( (SL(JQ)-X(JX))/TAU, 0.0D0)
                J0Q=JQ
              END IF
            ELSE
              IF(X(JX)+AMAX*TAU.LT.0.0D0) THEN
                AMAX=DMAX1( -X(JX)/TAU, 0.0D0)
                J0Q=JQ
              END IF
            END IF
30      CONTINUE
            IF(SENS.AND.J0Q.GT.0) THEN
              J1Q=0
              DO 32 II=NCB0P1,NCB
                I=(NCB0P1+NCB)-II
                J2Q=JAQ(I)
                TAU=QB(J0Q,I)
                IF(I.LT.NCB) THEN
                  IP1=I+1
                  DO 31 I2=IP1,NCB
                    TAU=TAU-YS(I2)*QB(JAQ(I2),I)
31      CONTINUE
                END IF
                YS(I)=TAU/QB(J2Q,I)
                IF(YS(I).GT.DEL4.AND.J2Q.GT.J1Q) THEN
                  J1Q=J2Q
                END IF
32      CONTINUE
                IF(J1Q.EQ.0) THEN
                  WRITE(6,9999) J0Q
                  FORMAT(15,' SUBROUTINE DUAL NO FEASIBLE SOLUTION')
                  ERROR=.TRUE.
                  RETURN
                ELSE
                  WRITE(*,*) 'CALLING DDEL @ LN 485'
                  CALL DDEL(J1Q)
                  WRITE(*,*) 'CALLING DADD @ LN 487'
9999

```

```

        CALL DADD(J0Q)
        P(JQX(J0Q))=0.0D0
        QB(J0Q,NSBP1)=0.0D0
    END IF
    GO TO 1001
ELSE IF(NINACT.EQ.0) THEN
    GO TO 1006
END IF
END IF
IF (NINACT.EQ.0) GO TO 1006
DO 34 J=1,NINACT
JQ=JIQ(J)
TAU=0.0D0
DO 33 I=NJMIN,NCB
TAU=TAU+QB(JQ,I)*YS(I)
33  CONTINUE
QB(JQ,NSBP1)=TAU
34  CONTINUE
DO 36 I1=1,NUP2
DO 35 I2=1,I1
H(I2,I1)=0.0D0
35  CONTINUE
36  CONTINUE
H(NUP1,NUP2)=-1.0D0
GO TO 1005
C
C      NEWTON STEP DIRECTION CALCULATION
C
1002  NCB=NCBC
NDB=NDBC
NU=NSB-NCB
NUT=NU
NUP1=NU+1
NUP2=NU+2
NINACT=NU-NDB+NCB0
IF(NU.EQ.0) THEN
    SUBMIN=.TRUE.
    Q3DMAX=0.0D0
    POSDEF=.TRUE.
    GO TO 1004
END IF
POSDEF=.FALSE.
C
C      REDUCED GRADIENT VECTOR CALCULATION
C
1003  Q3DMAX=0.0D0
J0Q=0
PTG=0.0D0
AMAX=0.999D0
DO 39 L=1,NU
LP1=L+1
DO 37 L2=1,L
H(L2,LP1)=0.0D0
37  CONTINUE
37  CONTINUE

```

```

LQ=L+NCB
TAU=0.0D0
DO 38 JJ=1,NINACT
TAU=TAU+G(JQX(JIQ(JJ)))*QB(JIQ(JJ),LQ)
38 CONTINUE
H(L,NUP2)=TAU
IF(DABS(TAU).GT.Q3DMAX) Q3DMAX=DABS(TAU)
39 CONTINUE
IF(Q3DMAX.LT.DEL1) THEN
  SUBMIN=.TRUE.
ELSE
  SUBMIN=.FALSE.
END IF
C
C      BLOCK SPECIES MULTIPLIER EVALUATION, IF ANY
C
1004 IF(POSDEF) THEN
  GFELB=GFE0-NSB*Q3DMAX
  IF(SUBMIN) THEN
    DO 85 JQ=1,NSB
    ZDUAL(JQ)=.FALSE.
  85 CONTINUE
  END IF
  IF(NCB.GT.NCB0) THEN
    JADD=0
    WLB=1.0D+20
    WM=-DMAX1(Q3DMAX,1.0D-03)
    DO 43 II=NCB0P1,NCB
    I=(NCB0P1+NCB)-II
    JQ1=JAQ(I)
    JX1=JQX(JQ1)
    TAU=0.0D0
    IF(NINACT.GT.0) THEN
      DO 40 J=1,NINACT
      TAU=TAU+G(JQX(JIQ(J)))*QB(JIQ(J),I)
    40 CONTINUE
    END IF
    IF(NDB.GT.0) THEN
      DO 41 J=1,NDB
      TAU=TAU+G(JQX(JAQ(J)))*QB(JAQ(J),I)
    41 CONTINUE
    END IF
    IF(I.LT.NCB) THEN
      IP1=I+1
      DO 42 J=IP1,NCB
      TAU=TAU+YS(J)*QB(JAQ(J),I)
    42 CONTINUE
    END IF
    YS(I)--TAU/QB(JQ1,I)
    SIG=G(JX1)-YS(I)
    IF(SIG.GT.0) THEN
      GFELB=GFELB-DEL3*SIG
    ELSE
      GFELB=GFELB+SIG
  END IF
END IF

```

```

        END IF
        IF(SIG.LT.WM.AND..NOT.ZDUAL(JQ1)) THEN
            WM=SIG
            JADD=JQ1
        END IF
        CONTINUE
        IF(JADD.GT.0) THEN
            NCMIN=MACB(JADD)
            ZDUAL(JADD)=.TRUE.
            NJMIN=NCMIN
            ADDCON=.TRUE.
            NORMAL=.FALSE.
            JXADD=JQX(JADD)
            IF(VTYP(JADD)) THEN
                K0=KSPEC(JXADD)
                JM=JMIN(K0)
                JX=JMAX(K0)
                TAU=0.0D0
                DO 44 J=JM,JX
                TAU=TAU+X(J)
            CONTINUE
            PI=DEXP(DMIN1(-WM,17.0D0))
            P(JXADD)=DMIN1(X(JXADD)*PI, 0.1D0*TAU)
        ELSE
            P(JXADD)=0.01D0
        END IF
        GO TO 1001
    END IF
    END IF
    IF(GFE0.LT.WMIN) THEN
        IF(SUBMIN) THEN
            WMIN=GFE0
            NBMIN=0
            DO 65 KP=1,NBLOCK
            NBMIN=NBMIN+1
            JBLK(NBMIN)=KBLK(KP)
        65      CONTINUE
        ADDPHS=.TRUE.
        GO TO 2001
    END IF
    ELSE
        IF(GFELB.GT.WMIN.OR.SUBMIN) GO TO 2001
    END IF
END IF
C
C      PROJECTED HESSIAN MATRIX CALCULATION
C
1005  DO 50 KP=1,NBLOCK
      K=KBLK(KP)
      IF(K.GT.0) THEN
          JM=JMIN(K)
          JX=JMAX(K)
          WRITE(*,*) 'CALLING HESS FROM DUAL @ LN 649'
          CALL HESS(K)

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```

DO 49 L1=1,NU
LQ1=L1+NCB
DO 46 J1=JM,JX
TAU=0.0D0
DO 45 J2=JM,JX
JP=MAX0(J1,J2)
TAU=TAU+H(JP,J1+J2-JP)*QB(JXQ(J2),LQ1)
45  CONTINUE
YS(JXQ(J1))=TAU
46  CONTINUE
DO 48 L2=L1,NUT
LQ2=L2+NCB
TAU=H(L1,L2+1)
DO 47 J1=JM,JX
JQ=JXQ(J1)
TAU=TAU+YS(JQ)*QB(JQ,LQ2)
47  CONTINUE
H(L1,L2+1)=TAU
48  CONTINUE
49  CONTINUE
END IF
50  CONTINUE
C
C      REDUCED DIRECTION VECTOR CALCULATION
C
POSDEF=.TRUE.
GAMMA=0.0D0
ZETA=0.0D0
THETA=0.0D0
DO 52 I=1,NU
IP1=I+1
IF(DABS(H(I,IP1)).GT.GAMMA) GAMMA=DABS(H(I,IP1))
IF(I.LT.NU) THEN
    DO 51 J=IP1,NU
        IF(DABS(H(I,J+1)).GT.ZETA) ZETA=DABS(H(I,J+1))
51    CONTINUE
        IF(I.EQ.1) THETA=ZETA
    END IF
52  CONTINUE
DELTA=DEL4*DMAX1(1.0D0,GAMMA,ZETA)
BETA2=1.0D0/DMAX1(GAMMA,DELTA,ZETA/NU)
TAU=DMAX1(DABS(H(1,2)),DELTA,THETA*THETA*BETA2)
IF(TAU.GT.H(1,2)) POSDEF=.FALSE.
H(1,2)=1.0D0/TAU
IF(NU.EQ.1) THEN
    H(1,3)=H(1,3)*H(1,2)
ELSE
    DO 56 J=2,NU
        JM1=J-1
        JP1=J+1
        JP2=J+2
        PHI=H(J,JP1)
        DO 53 K=1,JM1
            PI=H(K,JP1)*H(K,K+1)

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      PHI=PHI-PI*H(K,JP1)
      H(K,JP1)=PI
53    CONTINUE
      THETA=0.0D0
      DO 55 I=JP2,NUP2
      TAU=H(J,I)
      DO 54 K=1,JM1
      TAU=TAU-H(K,JP1)*H(K,I)
54    CONTINUE
      H(J,I)=TAU
      IF(I.LT.NUP2) THEN
         IF(DABS(TAU).GT.THETA) THETA=DABS(TAU)
      END IF
55    CONTINUE
      SIG=DMAX1(DABS(PHI),DELTA,THETA*THETA*BETA2)
      IF(SIG.GT.PHI) POSDEF=.FALSE.
      H(J,JP1)=1.0D0/SIG
56    CONTINUE
      H(NU,NUP2)=H(NU,NUP2)*H(NU,NUP1)
      DO 58 JMN=2,NU
      J=NUP1-JMN
      JP1=J+1
      TAU=H(J,NUP2)*H(J,JP1)
      DO 57 K=JP1,NU
      TAU=TAU-H(J,K+1)*H(K,NUP2)
57    CONTINUE
      H(J,NUP2)=TAU
58    CONTINUE
      END IF
C
C     TEST FOR ACCIDENTAL MAXIMUM
C
      IF(SUBMIN.AND.NORMAL) THEN
         IF(POSDEF) GO TO 1004
         WRITE(6,9991)
9991      FORMAT(' SUBROUTINE DUAL 9991: LOCAL MAXIMUM')
         ERROR=.TRUE.
         RETURN
      END IF
C
C     FEASIBLE DIRECTION VECTOR CALCULATION AND MAXIMUM STEP
C     LENGTH
C
      DO 60 JJ=1,NINACT
      JQ=JIQ(JJ)
      JX=JQX(JQ)
      TAU=0.0D0
      DO 59 I=1,NUT
      TAU=TAU-QB(JQ,I+NCB)*H(I,NUP2)
59    CONTINUE
      P(JX)=TAU
      PTG=PTG+TAU*G(JX)
      IF(VTYP(JQ)) THEN
         IF(X(JX)+AMAX*TAU.LT.SL(JQ)*(1.0D0-DEL4)) THEN

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```

        AMAX=DMAX1((SL(JQ)-X(JX))/TAU,0.0D0)
        J0Q=JQ
    END IF
ELSE
    IF(X(JX)+AMAX*TAU.LT.0.0D0) THEN
        AMAX=DMAX1( -X(JX)/TAU,0.0D0)
        J0Q=JQ
    END IF
END IF
60 CONTINUE
C
C     TEST FOR NEGATIVE DERIVATIVE ON DESCENT DIRECTIONS (IERR=1)
C
1006 IF(SENS) GO TO 1007
IF(PTG.GT.0.0D0) THEN
    IERR=1
    IF(ADDCON.AND..NOT.SUBMIN) THEN
        P(JQX(JADD))=0.0D0
        QB(JADD,NSBP1)=0.0D0
        IF(NDB.GT.0) THEN
            DO 76 JQ=1,NDB
            P(JQX(JAQ(JQ)))=0.0D0
            QB(JAQ(JQ),NSBP1)=0.0D0
        76 CONTINUE
    END IF
    ADDCON=.FALSE.
    NORMAL=.TRUE.
    GO TO 1002
ELSE
    WRITE(6,9992) JADD,PTG
9992 FORMAT(' SUBROUTINE DUAL 9992: PTG= ',I2,1PD12.4)
    ERROR=.TRUE.
    RETURN
END IF
END IF
C
C     LIMIT MAXIMUM CHANGE IN SOLUTION PHASE VARIABLE
C
IF(J0Q.GT.0) THEN
    IF(VTYP(J0Q)) THEN
        JX=JQX(J0Q)
        IF((DEL6-1.0D0)*X(JX).GT.AMAX*P(JX)) THEN
            AMAX=(DEL6-1.0D0)*X(JX)/P(JX)
            IF(X(JX)+AMAX*P(JX).GT.SL(J0Q)) J0Q=0
        END IF
    END IF
END IF
C
C     GRADIENT AT MAXIMUM STEP LENGTH
C
1007 IFUN=IFUN+1
IF(IFUN.GT.NFUN) THEN
    WRITE(6,9993) IFUN
9993 FORMAT(' SUBROUTINE DUAL 9993: IFUN= ',I6)

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        ERROR=.TRUE.
        RETURN
    END IF
    ALPHA=AMAX
    PTGA=0.0D0
    GFEA=0.0D0
    DO 61 KP=1,NBLOCK
    K=KBLK(KP)
    IF(K.GT.0) THEN
        WRITE(*,*) 'CALLING GRAD FROM DUAL @ LN 821'
        CALL GRAD(K,ALPHA,PTGA,GFEA)
    ELSE
        GFEA=GFEA+(X(-K)+ALPHA*P(-K))*G(-K)
        PTGA=PTGA+P(-K)*G(-K)
    END IF
61   CONTINUE
    IF(SENS) GO TO 1010
C
C      TEST FOR MAXIMUM STEP LENGTH ON DESCENT DIRECTIONS
C          IF FREE ENERGY DECREASE TAKE FULL STEP
C
C      IF(GFEA.LT.GFE0) GO TO 1010
C
C          IF FUNCTION DECREASING AND MAXIMUM IMPROVEMENT
C          OF FREE ENERGY IS SMALL COMPARED TO GFE0 TAKE
C          FULL STEP
C
C      IF(PTGA.LE.0.0D0) THEN
        IF(DABS(ALPHA*PTG).LT.DABS(GFE0)*DEL4) GO TO 1010
        WRITE(6,8811) J0Q,ADDCON,AMAX,PTG,PTGA,GFE0,GFEA
8811      FORMAT(' DUAL 8811 ',2I5,1P3D12.3,1P2D24.16)
        GO TO 1010
    END IF
C
C      FUNCTION INCREASING AT MAXIMUM STEP  LINESEARCH
C
C      AMIN=0.0D0
C      ETA=DEL5*DABS(PTG)
1008      AMAX=ALPHA
C      PTG2=PTGA
        IF(DABS(PTGA).LT.ETA) THEN
            IF(GFEA.LT.GFE0) GO TO 1009
            IF(PTGA.LE.0.0D0) GO TO 1009
        END IF
        ALPHA=AMIN-(AMAX-AMIN)*PTG/(PTG2-PTG)
        IFUN=IFUN+1
        IF(IFUN.GT.NFUN) THEN
            WRITE(6,9993) IFUN
            ERROR=.TRUE.
            RETURN
        END IF
        PTGA=0.0D0
        GFEA=0.0D0
        DO 62 KP=1,NBLOCK

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K=KBLK(KP)
IF(K.GT.0) THEN
  WRITE(*,*) 'CALLING GRAD FROM DUAL @ LN 868'
  CALL GRAD(K,ALPHA,PTGA,GFEA)
ELSE
  GFEA=GFEA+(X(-K)+ALPHA*P(-K))*G(-K)
  PTGA=PTGA+P(-K)*G(-K)
END IF
62  CONTINUE
IF(PTGA*PTG2.LT.0.0D0) THEN
  PTG=PTG2
  AMIN=AMAX
ELSE
  PTG=PTG*0.5D0
END IF
GO TO 1008
1009 IF(J0Q.GT.0) THEN
  TAU=X(JQX(J0Q))+ALPHA*P(JQX(J0Q))
  IF(VTYP(J0Q)) THEN
    IF(TAU.GT.SLP(J0Q)) J0Q=0
  ELSE
    IF(TAU.GT.DEL4) J0Q=0
  END IF
END IF
C
C   UPDATE X = X + ALPHA*P
C
1010 ITER=ITER+1
DELGX=GFE0-GFEA
GFE0=GFEA
IF(NORMAL) THEN
  DO 63 J=1,NINACT
  JX=JQX(JIQ(J))
  X(JX)=X(JX)+ALPHA*P(JX)
63  CONTINUE
IF(PD) THEN
  IF(J0Q.GT.0) THEN
    J0=JQX(J0Q)
  ELSE
    J0=0
  END IF
  WRITE(6,5001) ITER,IFUN,J0,GFE0,ALPHA
  FORMAT(' DUAL NORMAL',3I5,1PD25.17,1P2D20.8)
  DO 6000 JJ=1,NSB
  J=JQX(JJ)
  WRITE(6,5002) JJ,J,VTYP(JJ),MACB(JJ),X(J),G(J),P(J)
6000  CONTINUE
5002  FORMAT(4I5,F25.21,1P3D13.4)
  END IF
  IF(J0Q.EQ.0) THEN
    GO TO 1003
  ELSE
    WRITE(*,*) 'CALLING DADD @ LN 918'
    CALL DADD(J0Q)
  END IF

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J0X=JQX(J0Q)
P(J0X)=0.0D0
QB(J0Q,NSBP1)=0.0D0
IF(NDB.LT.NDBC) THEN
    NDBP1=NDB+1
    DO 71 JD=NDBP1,NDBC
        P(JQX(JAQ(JD)))=0.0D0
        QB(JAQ(JD),NSBP1)=0.0D0
    CONTINUE
END IF
IF(VTYP(J0Q)) THEN
    X(J0X)=SL(J0Q)
ELSE
    X(J0X)=0.0D0
END IF
GO TO 1002
END IF
ELSE
    DO 64 JQ=1,NSB
        JX=JQX(JQ)
        X(JX)=X(JX)+ALPHA*P(JX)
        YS(JX)=P(JX)
        P(JX)=0.0D0
        QB(JQ,NSBP1)=0.0D0
    CONTINUE
64 IF(ADDCON) THEN
    WRITE(*,*) 'CALLING DDEL @ LN 946'
    CALL DDEL(JADD)
END IF
IF(J0Q.GT.0) THEN
    IF(MACB(J0Q).LT.0) THEN
        WRITE(*,*)'CALLING DADD @ LN 951'
        CALL DADD(J0Q)
    END IF
END IF
IF(PD) THEN
    IF(J0Q.GT.0) THEN
        J0=JQX(J0Q)
    ELSE
        J0=0
    END IF
    IF(ADDCON) THEN
        WRITE(6,5003) ITER,IFUN,J0,GFE0,ALPHA
    ELSE
        WRITE(6,5004) ITER,IFUN,J0,GFE0,ALPHA
    END IF
5003 FORMAT(' DUAL ADDCON',3I5,1PD25.17,1PD20.8)
5004 FORMAT(' DUAL SENS',3I5,1PD25.17,1PD20.8)
    DO 6001 JJ=1,NSB
        J=JQX(JJ)
        WRITE(6,5002) JJ,J,VTYP(JJ),MACB(JJ),X(J),G(J),YS(J)
    CONTINUE
6001 END IF
IF(SENS) THEN

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        GO TO 1000
    ELSE
        ADDCON=.FALSE.
        NORMAL=.TRUE.
        GO TO 1002
    END IF
END IF
C
C      REPLACE X AND G VECTORS FOR PRIMAL
2001  DO 67 JQ=1,NSB
      JX=JQX(JQ)
      IF(MACB(JQ).GT.0.AND.SUBM.AND..NOT.ADDPHS) THEN
          WRITE(6,3100) JX,MACB(JQ),X(JX),S(JX)
3100      FORMAT('WARNING: DUAL SPECIES ON BOUNDARY AT GLOBAL
1 MINIMUM',2I5,1P2D12.3)
      END IF
      IF(.NOT.SVAR.OR.NBLOCK.EQ.1) PS(JX)=X(JX)
      P(JX)=0.0D0
      IF(VTYP(JQ)) THEN
          U0RT(JX)=GSAV(JQ)
          G(JX)=G(JX)+W1(JX)
          X(JX)=0.0D0
      ELSE
          X(JX)=XSAV(JQ)
          G(JX)=GSAV(JQ)
      END IF
67      CONTINUE
      IF(PP.OR.PD) THEN
          WRITE(6,5657) GFE0,(JQX(JQ),JQ=1,NSB)
5657      FORMAT(' DUAL BLOCK ',1PD20.12,96I3)
      END IF
C
100  CONTINUE
C -----
3000  IF(ADDPHS) THEN
        IF(NBMIN.EQ.1.AND.JBMIN(1).LT.0) THEN
            J1=-JBMIN(1)
            NCM=MACT(J1)
            K0=KSPEC(J1)
            IF(KTYP(K0).GT.1) THEN
                JM=JMIN(K0)
                JX=JMAX(K0)
                TAU=0.0D0
                DO 97 J=JM,JX
                    TAU=TAU+X(J)
97            CONTINUE
                PI=DEXP( DMIN1(-WMIN,20.0D0) )
                SIG=X(J1)*PI
                IF(TAU.GE.DEL2) THEN
                    TAU=DMIN1( SIG ,0.1D0*TAU,.01D0)
                    IF((TAU-X(J1)).LE.0.0D0) TAU=SIG
                    TAU=TAU-X(J1)
                ELSE
                    TAU=DMAX1(SIG-X(J1))
                END IF
            END IF
        END IF
    END IF
END IF

```

```

1           ,DMIN1(0.1D0*X(J1),DEL2*0.01D0))
      END IF
      P(J1)=DMAX1(TAU,0.0D0)
   ELSE
      PS(J1)=DMIN1(-WMIN,1.0D0)
      P(J1)=PS(J1)
      IF(KTYP(K0).LT.0) THEN
         NPP=NPP+1
         KPH(NPP)=K0
      END IF
   END IF
ELSE
   NCM=NC
   TEST=.FALSE.
   ALPHA=DMIN1(-WMIN,1.0D0)
   SIG=0.0D0
   DO 68 KP=1,NBMIN
   K0=JBMIN(KP)
   IF(K0.GT.0) THEN
      JM=JMIN(K0)
      JX=JMAX(K0)
      TEST=.TRUE.
   ELSE
      JM=-K0
      JX=-K0
      K0=KSPEC(JM)
   END IF
   IF(KTYP(K0).LT.0) THEN
      NPP=NPP+1
      KPH(NPP)=K0
   END IF
   DO 66 J=JM,JX
   SIG=SIG+X(J)
   IF(PS(J).GT.DELLB) THEN
      P(J)=ALPHA*PS(J)
      II=MACT(J)
      IF(II.GT.NC0.AND.II.LT.NCM) NCM=II
   END IF
66   CONTINUE
68   CONTINUE
   IF(.NOT.TEST) THEN
      SIG=DMAX1(SIG,DEL2)
      DO 160 KP=1,NBMIN
      J=-JBMIN(KP)
      IF(J.LT.0) STOP
      IF(KTYP(KSPEC(J)).LT.2) GO TO 170
      P(J)=P(J)*SIG/ALPHA
160   CONTINUE
170   CONTINUE
      END IF
   END IF
END IF
RETURN
END

```

```

C-----
C THIS SUBROUTINE ADDS A SPECIES CONSTRAINT FOR SPECIES J0 IN
C THE SUB-MINIZATION PROBLEM. DEGENERATE CONSTRAINTS ARE
C HANDLED EXACTLY AS IN CONADD.F (CONSTRAINT ADDITION FOR THE
C MAIN PROBLEM).
C-----
C-----  

SUBROUTINE DADD(J0)
IMPLICIT DOUBLE PRECISION(A-H,O-Z)
IMPLICIT INTEGER*2 (I-N)
LOGICAL REDUN,NODEG
COMMON/DUALB/Q(96,96),MACT(96),JA(96),JI(96),NS,NC,ND,NC0
DIMENSION V(96)
PARAMETER(DEL4=1.0D-13)
IF(NS.EQ.NC) GO TO 99
NC=NC+1
SIG=0.0D0
NODEG=.TRUE.
DO 1 I=NC,NS
V(I)=Q(J0,I)
IF(DABS(V(I)).GT.DEL4) NODEG=.FALSE.
Q(J0,I)=0.0D0
SIG=SIG+V(I)*V(I)
1 CONTINUE
IF(NODEG) GO TO 99
SIG=DSIGN(DSQRT(SIG),V(NC))
NINACT=(NS+NC0)-(NC+ND)
JI(-MACT(J0))=JI(NINACT+1)
MACT(JI(NINACT+1))=MACT(J0)
JA(NC)=J0
MACT(J0)=NC
IF(NS.EQ.NC) THEN
  Q(J0,NC)=V(NC)
  IF(NINACT.GT.0) THEN
    DO 2 J=1,NINACT
    ND=ND+1
    JA(ND)=JI(J)
    MACT(JI(J))=ND
 2 CONTINUE
  END IF
ELSE
  Q(J0,NC)=-SIG
  IF(NINACT.GT.0) THEN
    NDS=ND
    V(NC)=V(NC)+SIG
    PI=1.0D0/(V(NC)*SIG)
    DO 5 MI=1,NINACT
    J=JI(MI)
    TAU=0.0D0
    DO 3 I=NC,NS
    TAU=TAU+V(I)*Q(J,I)
    CONTINUE
    TAU=TAU*PI
 5 CONTINUE
  END IF
END IF

```

```

REDUN=.TRUE.
DO 4 I=NC,NS
T=Q(J,I)-V(I)*TAU
IF(DABS(T).GT.DEL4) THEN
  Q(J,I)=T
  IF(I.GT.NC) REDUN=.FALSE.
ELSE
  Q(J,I)=0.0D0
END IF
4  CONTINUE
IF(REDUN) THEN
  ND=ND+1
  JA(ND)=J
END IF
5  CONTINUE
IF(ND.GT.NDS) THEN
  NDS=NDS+1
  DO 6 JD=NDS,ND
  J=JA(JD)
  IF(NINACT.GT.0) THEN
    JI(-MACT(J))=JI(NINACT)
    MACT(JI(NINACT))=MACT(J)
    NINACT=NINACT-1
  END IF
  MACT(J)=JD
  6  CONTINUE
  END IF
  END IF
  RETURN
C
C -----
C   ATTEMPT TO ADD A CONSTRAINED VARIABLE
C
99  WRITE(6,9999) J0
9999 FORMAT(' DADD ERROR',15)
STOP
END

```

```

C-----
C THIS SUBROUTINE DELETES A SPECIES CONSTRAINT IN THE SUB-
C MINIMIZATION PROBLEM. IT IS VIRTUALLY IDENTICAL TO CONDEL.F
C FOR THE MAIN PROBLEM.
C-----
C-----  

SUBROUTINE DDEL(JR)
IMPLICIT DOUBLE PRECISION(A-H,O-Z)
IMPLICIT INTEGER*2 (I-N)
COMMON/DUALB/Q(96,96),MACT(96),JA(96),JI(96),NS,NC,ND,NC0
PARAMETER(DEL4=1.0D-13)
MREM=MACT(JR)
IF(MREM.LE.NC0) GO TO 99
NC=NC-1
NINACT=(NS+NC0)-(NC+ND)
JI(NINACT)=JR
MACT(JR)--NINACT
IF(MREM.LE.NC) THEN
  DO 4 MM=MREM,NC
    MP1=MM+1
    J0=JA(MP1)
    JA(MM)=J0
    MACT(J0)=MM
    CMM=Q(J0,MM)
    CMP1=Q(J0,MP1)
    SIG=DSIGN(DSQRT(CMM*CMM+CMP1*CMP1),CMM)
    CMM=CMM+SIG
    Q(J0,MM)=-SIG
    Q(J0,MP1)=0.0D0
    V1=1.0D0/SIG
    V2=CMP1/(SIG*CMM)
    IF(NINACT.GT.0) THEN
      DO 1 I=1,NINACT
        J=JI(I)
        TAU=Q(J,MM)*V1+Q(J,MP1)*V2
        Q(J,MM)=Q(J,MM)-CMM*TAU
        Q(J,MP1)=Q(J,MP1)-CMP1*TAU
1      CONTINUE
    END IF
    IF(ND.GT.0) THEN
      DO 2 I=1,ND
        J=JA(I)
        TAU=Q(J,MM)*V1+Q(J,MP1)*V2
        Q(J,MM)=Q(J,MM)-CMM*TAU
        Q(J,MP1)=Q(J,MP1)-CMP1*TAU
2      CONTINUE
    END IF
    IF(MM.LT.NC) THEN
      DO 3 I=MP1,NC
        J=JA(I+1)
        TAU=Q(J,MM)*V1+Q(J,MP1)*V2
        Q(J,MM)=Q(J,MM)-CMM*TAU
        Q(J,MP1)=Q(J,MP1)-CMP1*TAU
3    CONTINUE
  END IF
END SUBROUTINE DDEL

```

```
3      CONTINUE
      END IF
4      CONTINUE
      END IF
      IF(ND.GT.0) THEN
          NDS=ND
          NCP1=NC+1
          ND=0
          DO 5 JD=1,NDS
              JX=JA(JD)
              JA(JD)=0
              T=Q(JX,NCP1)
              IF(DABS(Q(JX,NCP1)).LT.DEL4) THEN
                  ND=ND+1
                  JA(ND)=JX
                  MACT(JX)=ND
                  Q(JX,NCP1)=0.0D0
              ELSE
                  NINACT=NINACT+1
                  JI(NINACT)=JX
                  MACT(JX)=-NINACT
              END IF
5      CONTINUE
      END IF
      RETURN
C
C -----
C      ATTEMPT TO DELETE A NON-EXPLICITLY CONSTRAINED SPECIES
C
99  WRITE(6,9999) JR
9999 FORMAT(' DDEL ERROR',I5)
STOP
END
```

```

C-----
C      THIS SUBROUTINE PRINTS THE SOLUTION TO EACH PROBLEM.
C-----
C-----  

SUBROUTINE PHSPRT(SCALE,GFEA,ITER,IFUN,IREFIN,Q2MAX,Q3MAX)  

IMPLICIT DOUBLE PRECISION(A-H,O-Z)  

IMPLICIT INTEGER*2 (I-N)  

INTEGER*2 Z  

CHARACTER*24 A1  

COMMON/LABEL/Z(96),U0RT(96),A1(96)  

COMMON/PHASE/R11(96,16),JMIN(96),JMAX(96),KSPEC(96),KTYP(96)  

1,NP0C,N00C  

COMMON/CONST/Q(96,96),MACT(96),JA(96),JI(96),NSC,NCC,NDC,NC0C  

COMMON/SEARCH/X(96),P(96),G(96),W0(96),H(64,64),B(96)  

COMMON/AQSOL/JP(46),JMC,JXC,JMA,JXA,JMB,JXB  

COMMON/MMM/MUL,WWW(96)  

COMMON/PH/LACT,NFILE,P0  

COMMON/DBASE/NCAT(18),NANI(18),ICAT,IANI,NCD,NAD,NND,TEMP  

LOGICAL MUL  

PARAMETER(WH2O=1.80153D-02)  

NPHAS0=NP0C  

NC0=NC0C  

CONC2=0.0D0  

SCALEI=1.0D0/SCALE  

IF(MUL) THEN  

    WRITE(6,2020)  

2020    FORMAT(/, 'PURE PHASE MULTIPLIERS')  

        DO 11 K=1,NPHAS0  

        IF(KTYP(K).EQ.-1) THEN  

            J=JMIN(K)  

            WRITE(6,2021) J,A1(J),WWW(K)  

2021    FORMAT(I5,2X,A24,1PD20.12)  

        END IF  

11    CONTINUE  

        END IF  

        GFE=0.0D0  

        GFEA=GFEA*SCALEI  

        WRITE(6,3002)  

        DO 10 K=1,NPHAS0  

        IF(KTYP(K).GT.0) THEN  

            JM=JMIN(K)  

            JX=JMAX(K)  

            X(JM)=X(JM)*SCALEI  

            GFE=GFE+X(JM)*G(JM)  

            IF(KTYP(K).EQ.1) THEN  

                CONC=1.0D0/X(JM)  

            ELSE IF(K.EQ.1) THEN  

                CONC=1.0D0/(X(JM)*WH2O)  

            DO 20 J=JM+1,JX  

            X(J)=X(J)*SCALEI  

            GFE=GFE+X(J)*G(J)  

20        CONTINUE  

        ELSE IF(KTYP(K).EQ.2) THEN

```

```

        CONC=X(JM)
        DO 21 J=JM+1,JX
        X(J)=X(J)*SCALEI
        CONC=CONC+X(J)
        GFE=GFE+X(J)*G(J)
21      CONTINUE
        CONC=1.0D0/CONC
        END IF

        XI=0.0D0
        DO 30 J=JM,JX
        TAU=G(J)-U0RT(J)

        WRITE(6,3000) J,A1(J),X(J),TAU,X(J)*CONC,X(J)*CONC

        XI=XI+Z(J)*Z(J)*X(J)*CONC
30      CONTINUE
        XI=0.5D0*XI

        IF(KTYP(K).GE.2) THEN
          AH2O=DEXP(G(1)-U0RT(1))
          WRITE(6,3009) AH2O, XI
        END IF

3009    FORMAT(' AH2O= ',F12.8,' IONIC STRENGTH= ',F15.8,' mol/kg')
        WRITE(6,3003)
        END IF

10      CONTINUE

        SIG=0.0D0
        DO 40 I=1,NC0
        TAU=B(I)
        DO 50 K=1,NPHAS0
        IF(KTYP(K).GT.0) THEN
          JM=JMIN(K)
          JX=JMAX(K)
          DO 60 J=JM,JX
          TAU=TAU-X(J)*Q(J,I)
60      CONTINUE
        END IF
50      CONTINUE
        IF(DABS(TAU).GT.SIG) SIG=DABS(TAU)
40      CONTINUE
        SIG=SIG*SCALE
        WRITE(6,5010) SIG
5010    FORMAT(//, ' CONVERGENCE CRITERIA ',1P1D12.3)
        WRITE(6,5011) ITER,IFUN,IREFIN
5011    FORMAT('ITER= ',I5,' IFUN= ',I5,' IREFIN= ',I5)
        RETURN
3000    FORMAT(I2,1X,A24,1P3D14.5,0PF10.5)
3002    FORMAT('SPECIES',26X,'MOLES',10X,'LN(A)',8X,'CONC',8X,'CONC')
3003    FORMAT(/)
        END

```

## APPENDIX G

## DBLIST source code

PROGRAM DBLIST  
 CHARACTER\*24 A1,A2  
 CHARACTER\*8 R1,R2

```

      WRITE(*,999)
999   FORMAT(1X,'THE FOLLOWING IS A LIST OF THE CHEMICAL
1    SPECIES FOUND IN THE DATABASE'/1X,'"THERMO.DAT" ALONG
2    WITH THEIR RESPECTIVE I.D. NUMBERS AND THE
3    TEMPERATURE',/1X,'RANGES OVER WHICH THE FREE ENERGY
4    POLYNOMIAL IS VALID. AN ASTERISK INDICATES'/1X,'THAT THE
5    FREE ENERGY DATA WERE CALCULATED USING STANDARD
6    STATE DATA FOUND',/1X, 'IN THE LITERATURE. THE ABSENCE
7    OF AN ASTERISK INDICATES THAT THE FREE' /1X,'ENERGY
8    DATA WERE CALCULATED FROM PUBLISHED SOLUBILITY
9    DATA. CONSULT THE'/1X,'USERS MANUAL FOR MORE
0     INFORMATION.'//)

      OPEN(UNIT=5,FILE='THERMO.DAT',STATUS='OLD')
      REWIND(5)
      READ(5,1000)
1000  FORMAT(/////////)

      1 READ(5,1001) I1,A1,R1,I2,A2,R2
         IF(I1.EQ.900) STOP
         IF(I2.EQ.900) THEN
            WRITE(*,1002) I1,A1,R1
            STOP
            END IF

            WRITE(*,1002) I1,A1,R1,I2,A2,R2
            GO TO 1

1001  FORMAT(I4,A24 / 4X,A8)
1002  FORMAT(1X,I3,1X,A24,A8,5X,I3,1X,A24,A8)

            STOP
            END
  
```