AN ABSTRACT OF THE DISSERTATION OF

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Title: <u>Studies Toward the Total Synthesis of (+)-Providencin.</u>

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James D. White

Studies toward the total synthesis of (+)-providencin (1), a highly oxygenated cembranoid dipterpene with a unique bicyclo[12.2.0]hexadecane skeleton and pronounced biological activity, are described. These studies resulted in the synthesis of advanced intermediates **320** and **332** which contain all of the carbon atoms of **1**. In a first generation approach toward **1**, a zirconium-mediated deoxygenative ring contraction of furanose **177** was used to furnish enantiopure cyclobutanol **176**. Olefination of furan aldehyde **197** with phosphonate **214** completed cyclobutylfuran segment **215**.

A second generation approach toward cyclobutylfuryl subunit 221 via ringclosing metathesis of diene 237 was unproductive, but the iodolactone subunit 228 needed for 1 was prepared successfully using carbometallation-iodination of alkyne 231. Nucleophilic substitution of tosylate 230 with the dianion of phenylselenyl acetic acid (252) followed by acid-catalyzed lactone formation was employed for construction of the γ -lactone moiety of 228. A third generation route to the cyclobutylfuryl subunit of **1** involved a tin(II) chloride-mediated stereoselective allenol synthesis by reaction of aldehyde **302** with propargyllic bromide **264**. A silver-catalyzed allenone-to-furan isomerization of **309** completed the synthesis of cyclobutylfuran subunit **288**. Attempts to couple the two major fragments, **228** and **288**, using palladium-catalyzed C-H activation of the furan component were unsuccessful, but linkage of two major subunits was achieved at the C12-C13 bond via an *inter*molecular aldol reaction to give **332** and at the C6-C7 bond using *inter*molecular palladium-catalyzed cross-coupling to afford **320**.

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Studies Toward the Total Synthesis of (+)-Providencin

by

Somnath Jana

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Presented on February 16, 2012.

APPROVED:

Major Professor, representing Chemistry

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I understand that my dissertation will become part of the permanent collection of Oregon State University libraries. My signature below authorizes release of my dissertation to any reader upon request.

Somnath Jana, Author

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Dedicated with love to

my parents

CHAPTER 1

Providencin

1.1 Isolation and biological activity

In 2003, the Rodriguez group isolated a highly oxygenated cembranoid from the sea plume *Pseudopterogorgia Kallos* (Bielschowski, 1918), which was collected near Providencia (Old Providence) Island located in the southwestern Caribbean Sea.¹ *Pseudopterogorgia Kallos* and related gorgonian octocorals have proven to be an abundant source of secondary metabolites that possess a diverse range of structural features and biological activities.² Dry animal specimens of 1.07 kilograms gave 20.0 milligrams of providencin (1) (corresponding to 0.012% yield dry weight) as a colorless amorphous solid.



Figure 1.1 Molecular structure and X-ray crystal structure of (+)-providencin (1)

The structure and relative configuration of providencin (1) was secured through a combination of NMR spectroscopy and single-crystal X-ray analysis. The absolute configuration is still unknown. Providencin (1) manifests a unique bicyclo[12.2.0]hexadecane scaffold incorporating a trisubstituted furan linked directly to a tetrasubstituted cyclobutane as well as an unusual α,γ -bridged α,β -epoxy- γ lactone.¹

Providencin (1) displays modest *in vitro* cytotoxicity against MCF7 breast, NCI-H460 lung, and SF-268 CNS cancer cell lines.¹ The growth inhibition of treated cells to untreated cells was 57%, 39%, and 94% respectively. The natural scarcity of providencin has prevented further biological testing. The intriguing structure as well as the promising biological activity of providencin has stimulated several efforts towards the total synthesis of this natural product.

1.2 Diterpenes Related to Providencin

Rodriguez' group isolated several structurally intriguing diterpenes from *Pseudopterogorgia Kallos*. They proposed that the carbon skeletons represented by providencin and six other compound or classes of compounds share a common biogenic precursor.³ Cembranoid natural products, the class to which providencin belongs, arise from the cyclization of geranylgeranyl pyrophosphate (GGPP).^{3c} Different modes of cyclization or ring contraction can lead to seven novel carbon

skeletons. It has been proposed that a C_2 - C_{17} cyclization leads to the providenciane carbon skeleton.



Figure 1.2 Proposed biosynthetic pathway to cembrane and related diterpenes

Providencin (1) is closely related to a class of diterpenes with the cembranebased skeleton known as furanocembranoids. The structure of providencin (1) shows a connectivity similar to that of the bipinnatins, a family of diterpenes isolated from the gorgonian *Pseudopterogorgia bipinnata*.⁴ The common structural features between these cembranoids include a 2,3,5-substituted furan, a 14-membered macrocycle and a butenolide or epoxidized butenolide.² The bipinnatins display inhibition of the nicotinic receptor, as well as anticancer and inflammatory activities.



Figure 1.3 Bipinnatin family of natural products

Pseudopterogorgia Kallos has been known to be a source of many other intriguing diterpene natural products.^{3,5} These compounds show a wide variety of structural and biological properties and have been the target of numerous synthetic efforts. Kallolide A (**13**), the first and most abundant metabolite isolated from *Pseudopterogorgia Kallos*, shows anti-inflammatory activity, and was synthesized by Marshall and co-workers.¹⁵ A hexacyclic compound, beilschowskysin (**16**), exhibits anitimalarial activity and strong anticancer activity against two cancer cell lines.^{3c} Several synthetic studies towards the cyclobutane core of **16** have been reported using intramolecular [2+2] photocycloaddition, but to date there is no total synthesis of this natural product.⁷ Intricarene (**17**) was synthesized from its proposed biogenic precursor bipinnatin J (**11**) by both the Trauner^{8a} and Pattenden groups.^{8b,c} However, no synthetic work has been reported on kallolide E (**14**), kallosin A (**15**), and ciereszkolide (**18**).



Figure 1.4 Diterpenes isolated from Pseudopterogorgia Kallos

1.3 Previous Syntheses of Related Furanocembranoids

Furanocembrane-based diterpenes have been attractive targets for several research groups. Those studies have shown that a variety of approaches can effectively produce the structural moieties found in these molecules.

1.3.1 Paquette's Synthesis of Dihydropseudopterolide, Gorgiacerone, and Acerosolide

In 1990, Paquette and co-workers reported a total synthesis of dihydropseudopterolide 32.⁹ Their strategy was to build the furan first and then functionalize the remainder of the macrocycle. The Paquette synthesis began with an aldol reaction between glyceraldehyde acetonide (19) and α -thiomalonate 20 (Scheme 1.1). Treatment of the crude with acetic acid in hot ethanol then gave furan 21. Swern oxidation of 21 to an aldehyde, followed by addition of 2-propenyl magnesium bromide gave allylic alcohol 22. Acetylation of this alcohol was followed by palladium catalyzed allylic stannylation gave tributylstannane 23 with only modest stereoselectivity. Subsequent treatment of allylic stannane 23 with boron trifluoride etherate in the presence of aldehyde 24 gave predominately the *erythro* product, which was treated with camphorsulfonic acid to afford lactone 25.



Scheme 1.1 Paquette's synthesis of furanolactone 25

Formation of a dianion by treating 25 with two equivalents of potassium hexamethyldisilazide and trapping with phenylselenyl chloride gave an α -selenolactone-selenothioacetal that was treated with silver perchlorate to hydrolyze the selenothioacetal to an aldehyde (Scheme 1.2). The selenolactone was oxidatively eliminated with sodium periodate to give butenolide 26. Sodium borohydride reduction of 26 followed by mesylation and displacement with bromide yielded furyl bromide 27. Cross-coupling of bromide 27 with vinyl stannane 28 gave 29, which was converted to bromoaldehyde 30 by a three step sequence. An intramolecular Nozaki-Hiyama-Kishi reaction on 30 in the presence of excess of chromium(II) chloride resulted in stereoselective macrocyclization to afford dihydropseudopterolide 31, albeit in low yield. The selectivity was explained by an intramolecular π -facially selective attack at the aldehyde carbonyl by the flanking π -bond such that both large groups are equatorially placed on the oxachromium six-membered transition state.



Scheme 1.2 Completion of Paquette's synthesis of dihydropseudopterolide (32)

In 1992, Paquette and co-workers published the total synthesis of gorgiacerone (**36**) following the same strategy they used in previous furanocembranoid syntheses.¹⁰ Exposure of **30** to a large excess of chromium(II) chloride, led to the cyclized product **34** (Scheme 1.3). Finally, Swern oxidation of alcohol **34** and epimerization of the isopropenyl group in the same reaction yielded gorgiacerone (**36**).



Scheme 1.3 Paquette's synthesis of gorgiacerone (36)

A year later in 1993, Paquette and Astles completed a synthesis of the furanocembranoid acerosolide (46).¹¹ This synthesis featured similar disconnections to those employed previously yet they were able to tune the reactivity of stannane 23 in order to access the larger ring system. Reaction of (*E*)-allylic stannane 23, once again with aldehyde 24 in the presence of tin(IV) chloride at low temperature, yielded the desired regioreversed product 38 (Scheme 1.4). This was treated with camphorsulfonic acid to give lactone 39 in good yield. It is believed that primary stannane 23 initially rearranges to a secondary stannane, which then undergoes the metallo-ene bond-forming reaction. A three-step sequence was used to convert the sulfide of 39 to an aldehyde and to oxidize the lactone to a butenolide to yield 40. The aldehyde was reduced with sodium borohydride, and the resulting alcohol was treated with *N*-bromosuccinimide and dimethyl sulfide to give bromide 41. An analogous four-step sequence to that used with 27, including the cross-coupling with 28, gave allylic



bromide **43**. Finally, chromium-mediated macrocyclization of **43** followed by oxidation of the resulting alcohols **44** and **45** afforded acerosolide (**46**).

Scheme 1.4 Paquette's synthesis of acerosolide (46)

1.3.2 Marshall's Synthesis of Kallolide B and Rubifolide

In the 1990's and early 2000's, the Marshall group was one of the key contributors to synthetic efforts towards furanocembranoids and pseudopteranes. In this context, a broad range of novel methodology was investigated to construct key features of these natural products. Marshall developed a new way to isomerize an allenone to a furan as well as an allenoic acid to a butenolide moiety to fabricate the pseudopterane ring system.



Scheme 1.5 Marshall's synthesis of kallolide B (57)

Marshall and co-workers first reported an asymmetric synthesis of the enantiomer of natural kallolide B (57) in 1996.¹² They targeted the non-natural enantiomer based on the large cost difference between their starting material (S)-(-)perillyl alcohol (47) and its enantiomer. The synthesis began with a directed epoxidation of 47 using vanadyl acetylacetonate and tert-butyl hydroperoxide, followed by oxidative cleavage of the resulting epoxide with periodic acid to afford carboxylic acid 48 (Scheme 1.5). Esterification of 48 with diazomethane followed by addition of 1-bromo-2-butyne in the presence of tin(II) chloride and sodium iodide in N,N'-dimethyl-N,N'-propylene urea (DMPU) gave allenyl carbinol **49** as a mixture of diastereomers in high yield. Swern oxidation produced an allenone that was treated with a catalytic amount of silver nitrate to affect cyclization to furan 50. Reduction of the ester to an aldehyde with diisobutylaluminum hydride followed by Corey-Fuchs olefination gave a gem dibromide, which was treated with *n*-butyllithium to afford an acetylide. The latter was trapped with paraformaldehyde to yield propargyllic alcohol **51**. Lithiation of the 5-position of the furan with *sec*-butyllithium followed by addition of dimethylformamide gave a formylated furan, which underwent Stille-Gennari olefination with phosphonate 52. Subsequent conversion of the propargyllic alcohol to the chloride with mesyl chloride and lithium chloride gave (Z)-enoate 53. The ester was reduced with diisobutylaluminum hydride and the resulting alcohol upon treatment with sodium hydride underwent intramolecular etherification to give a macrocyclic ether 54. Treatment of allylic ether 54 with *n*-butyllithium effected the key [2,3]-Wittig rearrangement to furnish propargyllic alcohol 55 as the sole product

and in excellent yield. The alcohol was converted to its mesylate which was treated with palladium(0), carbon monoxide, and β -trimethylsilylethanol to give allenic ester **56**. The allenoate could be isomerized with triphenylphosphine in hot acetonitrile to yield an equilibrium mixture of separable allenoates. The desired epimer was treated with tetra-*n*-butylammonium fluoride to give an allenoic acid which was exposed to silver nitrate to furnish kallolide B (**57**).

A year later in 1997, Marshall and Sehon reported a synthesis of the enantiomer of rubifolide (64).¹³ First, the allenvl stannane 58 was prepared from (S)-(-)-perillyl alcohol (47), using standard protocols. Deprotection of silyl ether 58 with tetrabutylammonium fluoride followed by oxidation and treatment with ethylmagnesium bromide and 1-1'-(azodicarbonyl)dipiperdine (ADD) gave aldehyde 59 (Scheme 1.6). Macrocyclization of 59 using boron trifluoride etherate yielded homopropargyllic alcohol, which was oxidized to allenone 60 using Dess-Martin periodinane. Exposure of the allenone 60 with 10% silver nitrate on silica gel afforded furan 61 in good yield. Treatment of the protected tertiary alcohol of 61 with ptoluenesulfonic acid resulted in elimination to give exclusively the (Z)-olefin in macrocycle 62. Saponification of carbonate 62 with potassium carbonate in methanol gave separable epimeric propargyllic alcohols. Oxidation of the mixture with manganese dioxide and subsequent reduction with K-Selectride led exclusively to synproduct 63. Finally, a one-pot reaction of 63 with trifluoroacetic anhydride (TFAA), followed by palladium catalyzed carbonylation and lastly silver nitrate-mediated cyclization yielded rubifolide (64).



Scheme 1.6 Marshall's synthesis of rubifolide (64)

Marshall and co-workers also accomplished the synthesis of deoxypukalide in 2001^{14} and kallolide A (13) in 1998^{15} using their same strategy described earlier.

1.3.3 Pattenden's Synthesis of Lophotoxin

The Pattenden group has been working toward a total synthesis of the elusive target lophotoxin (84), and they completed an asymmetric synthesis of bisdeoxylophotoxin (83) in 2001.¹⁶ Their approach hinges upon a Stille coupling to close the macrocycle, while the butenolide is formed by unsaturation of a lactone as in Paquette's synthesis. The Pattenden route began with addition of lithiated trimethylsilylacetylene to epichlorohydrin (65), followed by alkyne deprotection and carbometallation/iodination to yield vinyl iodide 67 (Scheme 1.7). The chlorohydrin 67 was treated with sodium hydroxide to give an epoxide 68 which was reacted with lithiated ethoxyacetylene to give a secondary alcohol. Acid catalyzed cyclization then yielded lactone 69 which was deprotonated with lithium hexamethyldisilazide and selenated to give α -phenylselenolactone 70.



Scheme 1.7 Pattenden's synthesis of iodolactone 70

The furan building block of the lophotoxin structure was synthesized from Evans' auxiliary **71** (Scheme 1.8). Oxazolidinone **71** was acylated with 3-methylbuten-2-oyl chloride (**72**) to afford **73**, and allylic deprotonation of **73** with sodium hexamethyldisilazide followed by alkylation with furyl bromide **74** gave **75**. Cleavage of the auxiliary with Super Hydride yielded alcohol **76**, which was converted to its tosylate, subsequent ester reduction afforded alcohol **77**. Displacement of tosylate from **77** with tetrabutylammonium cyanide and silyl protection of the alcohol yielded nitrile **78**.



Scheme 1.8 Completion of Pattenden's synthesis of bis-deoxylophotoxin (83)

Reduction of nitrile **78** first with diisobutylaluminum hydride and then sodium borohydride gave alcohol **79**. Exposure of furan **70** with excess *n*-butyllithium and trapping with trimethyltin chloride followed by perruthenate oxidation yielded coupling partner **80**. The lithium anion of **70** was treated with aldehyde **80** and then with hydrogen peroxide to give butenolide **81** as a mixture diastereomers. Stille macrocyclization and subsequent acetylation yielded furanocembranoid **82** in low yield. Finally, deprotection of the primary alcohol and oxidation to an aldehyde gave bis-deoxylophotoxin (83). To date, there have been no reports of successful oxidation of this intermediate to yield the lophotoxin (84) itself.

1.3.4 Trauner's Synthesis of (-)-Bipinnatin J, Rubifolide, Isoepilophodione, and Intricarene

In early 2006, Trauner's group reported a racemic and protecting-group-free synthesis of bipinnatin J (11).¹⁷ Subsequently, they published an asymmetric synthesis of **11** in order to study its biosynthetic relationship with other furanocembranoids and related natural products such as rubifolide (64), isoepilophodione (93), and intricarene (17).^{8a} The Trauner synthesis began with Dess-Martin periodinane oxidation of known vinyl iodide 85 followed by addition of lithiated trimethylsilylacetylene and reoxidation (Scheme 1.9). Asymmetric reduction of the resulting ynone 86 with Midland's (S)-Alpine borane gave propargyllic alcohol 87 in 92% ee. The alkyne 87 was elaborated to the butenolide 88 in nine steps, and Stille cross-coupling of 88 with furylstannane 89 afforded linear precursor 90 which was converted to allylic bromide 91 using triphenylphosphine and *N*-bromosuccinimide. Finally, highly a diasteroselective Nozaki-Hiyama-Kishi macrocyclization gave enantiomerically enriched (-)-bipinnatin J (11) in good yield.



Scheme 1.9 Trauner's asymmetric synthesis of (-)-bipinnatin J (11)

Trauner's group also showed that bipinnatin J (11) could be converted into rubifolide (64) in essentially quantitative yield by deoxygenation with trifluoroacetic acid and triethylsilane (Scheme 1.10). Rubifolide (64) was then oxidized with *m*-chloroperbenzoic acid to isoepilophodione (93). In a separate exercise, bipinnatin J (11) was oxidized with *m*-chloroperbenzoic acid which promoted an Achmatowicz rearrangement to 94. Acetylation yielded enone 95 which upon heating with tetramethylpiperidine in dimethyl sulfoxide presumably gave an oxidopyrylium ion that underwent a transannular 1,3-dipolar cycloaddition to afford intricarene (17), albeit in low yield.



Scheme 1.10 Trauner's synthesis of rubifolide (64), isoepilophodione (93) and intricarene (17) from bipinnatin J (11)

1.3.5 Rawal's Synthesis of (±)-Bipinnatin J

Shortly after Trauner's report on the racemic synthesis of bipinnatin J (11), Rawal and Huang also reported a racemic synthesis of 11 using Negishi crosscoupling and Nozaki-Hiyama-Kishi macrocyclization as key reactions.¹⁸ They prepared the butenolide **97** starting from commercially available 5-bromo methylpent-2-ene (**96**) (Scheme 1.11).



Scheme 1.11 Rawal's synthesis of (±)-bipinnatin J (11)

Negishi cross-coupling of vinyl iodide **97** with organozinc species **98** derived from 3-methylfurfural yielded coupled product **99** in quantitative yield. Acidic deprotection of **99** and bromination of the allylic alcohol gave the same intermediate, **91** used in Trauner's synthesis. Final diasteroselective Nozaki-Hiyama-Kishi macrocyclization of **91** gave (\pm)-bipinnatin J (**11**) in good yield.
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CHAPTER 2

Previous Synthetic Studies Towards Providencin

Three research groups have published their progress towards the synthesis of providencin (1). The Pattenden group at the University of Nottingham, the Mulzer group at the University of Vienna, and the Wood group at Colorado State University have explored different approaches toward providencin but neither group has completed a synthesis.

2.1 Pattenden's Synthetic Approach

In 2006, Pattenden and co-workers proposed that the cyclobutanol moiety in providencin (1) could be obtained from bipinnatin E (6) via a photochemically mediated intramolecular C-H insertion reaction.¹



Scheme 1.1 Pattenden's proposed biosynthetic pathway to providencin (1)

To verify this hypothesis, Pattenden and co-workers prepared a model compound 104. Commercially available 2-methyl-3-furoic acid (100) was esterified and bromination of the resulting ester with N-bromosuccinimide gave bromomethyl furan 101. Deprotonation 3-methylbut-2-enoate (102)using lithium of diisopropylamide, followed by alkylation of the resulting α -organolithium species with bromide 101, gave desired β , γ -unsaturated ester 103. Allylic oxidation of 103 by selenium oxide furnished α,β -unsaturared aldehyde **104**. Irradiation of a dilute solution of 104 in benzene in a Pyrex photoreactor using light from a 400 Watt medium pressure mercury lamp led to cyclobutanol **105** in a low 19% yield. To make matters worse, NOE experiments showed that 105 possessed the undesired trans-1,2-trans-2,3-relationship around the cyclobutane ring. Pattenden proposed that the conformational bias imposed by the macrocycle of furanocembrane would promote the formation of the desired cyclobutane diastereomer in the photo product (i.e., *cis*-1,2-*trans*-2,3; cf. 1) but this idea was never pursued.



Scheme 2.1 Pattenden's synthesis of cyclobutanol 105

2.2 Mulzer's Synthetic Studies

Mulzer and co-workers have published three reports on their synthetic effort towards providencin (1).² In his first generation approach, Mulzer envisioned that 1 could be formed by late stage epoxidations of **106** with introduction of the missing hydroxyl group in the cyclobutane moiety.



Scheme 2.2 Mulzer's first generation retrosynthetic analysis of providencin (1)

Butenolide **106** would be generated by a palladium-catalyzed carbonylation of vinyl iodide **107**, which would be formed by an intramolecular Horner-Wadsworth-Emmons olefination of phosphonate **108**. The aldehyde **108** would be prepared from

seco-intermediate **109** using standard chemistry. The furan of **109** would arise from two major fragments **110** and **111** via alkylation followed by palladium-catalyzed Wipf furan synthesis.³

Mulzer's synthesis began with the preparation of cyclobutanone **112** in optically pure form. Commercially available racemic bicyclic ketone **112** was first resolved via its diastereomeric ketal **114** and **115**, and the crystalline diastereomer **115** was cleaved to enantiomerically pure (+)-**112** by Birch reduction and Swern oxidation (Scheme 2.3).



Scheme 2.3 Synthesis of (+)-112

In an alternative route, Mulzer used enzymatic resolution to obtain alcohol (+)-117. Sodium borohydride reduction and acylation of *rac*-112 gave *rac*-116 (Scheme 2.4), after which lipase-catalyzed resolution of *rac*-116 afforded cyclobutanol (+)-117 in 45% yield and >95% ee.



Scheme 2.4 Lipase-catalyzed synthesis of (+)-117

Triisopropylsilyl protection of the secondary alcohol of (+)-117 followed by ozonolysis and reduction gave diol 118 (Scheme 2.5). Monoprotection of 118 with monomethoxytrityl chloride afforded 119, which was converted to aldehyde 120 by oxidation and subsequent base-catalyzed epimerization. A Reformatsky reaction of aldehyde 120 with methyl bromoacetate (121) followed by oxidation yielded β -keto ester 111 in good yield.



Scheme 2.5 Mulzer's synthesis of cyclobutane fragment 111

Synthesis of propargyllic iodide **110** commenced with sodium borohydride reduction of (*S*)-malic ester **122** and silyl protection of the resultant diol (Scheme 2.6). Conversion of ester **123** to its Weinreb amide followed by addition of methylmagnesium bromide provided methyl ketone **124**. Addition of deprotonated alkyne **125** to this ketone furnished tertiary alcohol **126**, and after benzoylation, *p*-methoxybenzyl group removal and iodination of the resulting alcohol the coupling partner **110** was produced in 28% overall yield from **122**.



Scheme 2.6 Mulzer's synthesis of propargyllic iodide 110

The coupling of fragments **110** and **111** was accomplished via alkylation of β keto ester 111 with propargyllic iodide 110 in the presence of sodium hydride (Scheme 2.7). Wipf cyclization of benzoate 127 gave a 1:1 mixture of the corresponding E- and Z- propenylfurans, which was equilibrated to E-isomer 109 using diphenyl diselenide. Detritylation of 109 with hexafluoroisopropyl alcohol (HFIP) and oxidation of the resulting alcohol gave the corresponding aldehyde, which was converted to keto phosphonate 128 in two steps. The primary alcohol of 128 was subsequently oxidized to the deprotected and aldehyde which underwent macrocyclization via Horner-Wadsworth-Emmons olefination to furnish the providencin skeleton 129 in modest yield. No further elaboration of 129 toward providencin (1) has been reported to date.



Scheme 2.7 Mulzer's synthesis of macrocycle 129

In a second generation approach towards providencin (1), Mulzer envisioned that ring-closing metathesis (RCM) of 130 would be used to provide the macrocycle of the natural product (Scheme 2.8). Diene 130 would be formed by an aldol reaction between α -seleno lactone 131 and aldehyde 132, while the furanocyclobutane of 132 segment would be prepared from cyclobutanone 112.



Scheme 2.8 Mulzer's second generation strategy for providencin (1)

First, lactone 131 was synthesized in four steps from (R)-glycidol tosylate (133). Cuprate addition to epoxide 133 provided homoallylic alcohol 135 which was converted to epoxide 136 (Scheme 2.9). The latter was then treated with the dianion of (phenylseleno)acetic acid (137) to give hydroxy acid 138. Acid-catalyzed cyclization of 138 yielded the desired lactone 131 in good overall yield.



Scheme 2.9 Mulzer's synthesis of α -seleno lactone 131

Synthesis of cyclobutylfuran 132 began with the alkylation of β -ketoester 111 with propargyllic iodide 139 (Scheme 2.10). Wipf cyclization of 140 furnished furan 141 in excellent yield, after which removal of the monomethoxytrityl group followed by oxidation provided aldehyde 132. Deprotonation of lactone 131 with lithium hexamethyldisilazide and reaction of the anion with aldehyde 132 gave a product, which upon treatment with hydrogen peroxide yielded butenolide 130. Ring-closing metathesis of diene 130 using Grubbs' II catalyst (155) gave Z-isomer 142 as a 1:1 mixture of alcohol diastereomers. All the attempts to convert the Z-olefin of 142 to the desired *E*-olefin were unsuccessful.



Scheme 2.10 Mulzer's synthesis of macrocycle 142

In his third generation approach toward providencin (1), Mulzer attempted to build the core structure prior to Wipf cyclization. In this plan, the furan would be introduced into macrocycle **143** by palladium-catalyzed cyclization with the macrocycle again prepared by ring-closing metathesis (Scheme 2.11). The RCM precursor, diene **144**, would be prepared by an aldol reaction between α -thiolactone **145** and aldehyde **146** while cyclobutanone **112** would again be the source of the tetrasubstituted cyclobutane **146**.



Scheme 2.11 Mulzer's third generation retrosynthetic plan for providencin (1)

In the forward direction, the new route began with conversion of cyclobutanone **112** to *tert*-butyldimethylsilyl enol ether **147** (Scheme 2.12). Hydroboration, triisopropylsilyl ether formation, ozonolysis and subsequent sodium borohydride reduction yielded tetrasubstituted cyclobutane **148**. Mono-protection of diol **148**, oxidation of the remaining alcohol to an aldehyde and epimerization was followed by conversion to alkyne **149**. Deprotonation of alkyne **149** and treatment of

epoxide 133 furnished secondary alcohol 150 which was converted to protected allylic alcohol 151 in three steps. Cleavage of the trityl group from 151 and subsequent oxidation provided an aldehyde, which upon exposure to the anion of lactone 152 afforded the ring-closing metathesis precursor 144. Unfortunately, all attempts to effect ring-closing metathesis of 144 were unsuccessful; a result that led Mulzer to speculate that the alkyne moiety of this molecule was the source of the problem. However, converting 144 into its cobalt complex 153 also failed to generate macrocycle 154 by ring-closing metathesis. At this stage, Mulzer appears to have abandoned efforts to synthesize providencin and no further work has been published by the Vienna group.



Scheme 2.12 Mulzer's attempted synthesis of macrocycles 143 and 154

2.3 Wood's Synthetic Approach

In 2011, Wood's group⁴ published their progress towards providencin (1), where a route to the furanyl-cyclobutanone fragment of 1 involving a [2+2] photocycloaddition and an alkylation reaction was employed. Wood envisioned construction a preassembled furanyl-cyclobutane segment (156) that would be derived from coupling of 3-furoic acid (157) with bromide 158 (Scheme 2.13).



Scheme 2.13 Wood's strategy for synthesis of the furanyl-cyclobutane segment of providencin (1)

Wood's synthesis of bromocyclobutanone **158** commenced with [2+2] photocycloaddition of 1,1-diethoxyethylene (**159**) and diethyl fumarate (**160**) which, under the conditions developed by Brunner,⁵ furnished cycloadduct **161** in excellent yield (Scheme 2.14). Reduction of diester **161** to diol **162** and subsequent conversion to the corresponding bis-benzyl ether was followed by removal of acetal hydrolysis to provide cyclobutanone **163** in good overall yield. After extensive experimentation, the most efficient conversion of **163** to **158** was found to be initial transformation to its silyl enol ether **164** and subsequent treatment with *N*-bromosuccinimide. This gave cyclobutanone **158** as a 6:1 mixture of diastereomers, with the major diastereomer having a trans-trans relationship between the three substituents.



Scheme 2.14 Wood's synthesis of bromocyclobutanone 158

In an initial reaction of the dianion of 3-furoic acid (**157**) with bromide **158** at low temperature, the crude extract was treated with diazomethane (Scheme 2.14). The desired product **156** was isolated as a single diastereomer in 20 % yield. NMR studies, including 1D NOE and 2D ROESY experiments, confirmed that **156** possessed a trans-trans relationship between substituents.



Scheme 2.15 Wood's initial coupling between 157 and 158

During the course of efforts to improve this yield of the key coupling, Wood observed that quenching the reaction at -78 °C prior to treatment with diazomethane resulted in the formation of tertiary alcohol **165** (Scheme 2.16). This alcohol proved to be very labile and difficult to isolate, but it could be transformed to **156** in an improved yield upon exposure of a base.



Scheme 2.16 Wood's initial synthesis of 166

Wood and co-workers found that lithium hexamethyldisilazide was the base of choice for promoting the conversion of **165** to **156** and the optimized sequence, wherein addition of the dianion of **157**, esterification and rearrangement were carried out without isolation of intermediates **165** and **166**, yielded the desired cyclobutanone **156** in 60% overall yield (Scheme 2.17). The utilization of key fragment **156** towards

the synthesis of providencin (1) is presumably underway in Wood's laboratory although no further progress has been reported.



Scheme 2.17 Wood's optimized synthesis of furanyl-cyclobutanone fragment 156

2.4 References

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CHAPTER 3

First Generation Approach Towards Providencin

3.1 Retrosynthetic analysis of providencin (1)

In our initial approach, we envisioned that epoxidation of butenolide 167 from the less sterically hindered face would provide providencin (1) (Scheme 3.1). The macrocycle 167 would be closed from aldehyde 168 by an intramolecular Nozaki-Hiyami-Kishi reaction,¹ which would be followed by Mitsunobu inversion and acetate formation. Butenolide 168 would be acquired from vinyl dibromide 169 by Hoye's protocol using a palladium catalyst and carbon monoxide.² Dibromide **169** would be derived from 170 using standard operations. Asymmetric epoxidation of 171 using Onaka's protocol for *E*-homoallylic alcohols³ would yield **170**, and alcohol **171** would be prepared via Wittig olefination of aldehyde 172. Selective protection of the primary alcohol of diol 173, followed by oxidation of the remaining alcohol to a cyclobutanone and Wittig olefination would yield 161. The furan ring would be introduced into 173 by Knoevenagel condensation⁴ of ester 175 with glyceraldehyde acetonide (174) and subsequent acid-catalyzed cyclization. β -Ketoester 175 would be prepared from vinyl cyclobutanol **176** using a Wacker oxidation⁵ followed by acylation with Mander's reagent.⁶ The key transformation of **177** to **176** is predicated upon zirconium-mediated deoxygenative ring contraction of a furanoside, a process first developed by

Taguchi^{7a,b} and subsequently employed by Paquette^{7c,d,e} as an entry to enantiopure cyclobutanols. Furanose **177** would be obtained from readily available D-glucose (**178**).



Scheme 3.1 First generation retrosynthetic analysis of providencin (1)

3.2 Synthesis of an enantiopure cyclobutanol via a zirconiummediated ring contraction of a furanoside

In the forward direction, D-glucose (178) was first converted into diacetonide 179 in the presence of anhydrous zinc chloride and polyphosphoric acid in dry acetone (Scheme 3.2).^{8a} The secondary alcohol of 179 was oxidized to the corresponding ketone using Swern's conditions and the ketone was subjected to a Wittig reaction to give $\alpha_{\gamma}\beta$ -unsaturated ester 180.^{8b} Hydrogenation of alkene 180 over palladium on charcoal gave 181 as a single diasteroisomer and the latter was reduced to alcohol 182⁹ which was protected as its *p*-methoxybenzyl ether 183.¹⁰ Selective deketalization of 183 using 60% acetic acid gave diol 184 which was converted to the 2vinyltetrahydrofuran 185 using triphenylphosphine, imidazole, and iodine in hot toluene.¹¹ The remaining acetonide group in 185 was cleaved in acidic methanol to give a methyl glycoside 186 and the free secondary alcohol in 186 was then protected as its *t*-butyldimethylsilyl ether 177.



Scheme 3.2 Synthesis of furanoside 177

The pivotal oxygen atom abstraction from tetrahydrofuran **177** was carried out with dicyclopentadienylzirconium(0), generated by treatment of dicyclopentadienylzirconium(II) dichloride with 2 equivalents of *n*-butyllithium in toluene (Scheme 3.3). That all four stereocenters of **177** had been faithfully transcribed into **176** was confirmed by careful analysis of the ¹H NMR spectrum of the cyclobutane which showed NOE's (2.7% and 3.3%) consistent with a cis relationship between H₁ and H₄ and between H₂ and H₃.



Scheme 3.3 Mechanism for the formation of cyclobutanol 176 from furanoside 177

3.3 Elaboration of a cyclobutanol into a cyclobutylfuran

Initially, cyclobutanol **177** was protected as its 2-trimethylsilylethoxymethoxy ether **187** and the latter was subjected to Wacker oxidation conditions (Scheme 3.4). The desired methyl ketone **188** was obtained from this reaction along with substantial amounts of aldehyde **189** (**188:189** 2.3:1). The alternate mode of regiochemical oxidation leading to **189** is explained by coordination of the 2trimethylsilylethoxymethoxy protecting group of **187** with the palladium metal center leading to complex **187a**. In this complex, water can attack at the more accessible site, leading to the undesired byproduct **189**.¹²



Scheme 3.4 Wacker oxidation of 2-trimethylsilylethoxymethoxy ether 187

To circumvent this problem, cyclobutanol **176** was converted to its triisopropylsilyl ether **190** (Scheme 3.5) which would be expected to show little tendency to complex with palladium(II). In the event, exposure of **190** to Wacker conditions provided our desired methyl ketone **192** with only a trace amount of aldehyde **191** (**191**:**192** 1:11). Treatment of the kinetically formed lithium enolate of **192** with methyl cyanoformate (**193**) furnished β -keto ester **194** which was condensed with 2,3-*O*-isopropylidene–D-glyceraldehyde (**174**)¹³ under acid catalysis. The initial Knoevenagel product **195**, detectable by thin-layer chromatography, was converted slowly to a mixture of **196** and desilylated alcohol **173**.¹⁴ Oxidation of **196** with tetra-*n*-propyl perruthenate¹⁵ then yielded aldehyde **197**.



Scheme 3.5 Synthesis of cyclobutylfuranaldehyde 197

3.4 Installation of an (E)-trisubstituted alkene at C7-C8 of providencin

Our initial plan envisioned elaboration of aldehyde **197** to an (*E*)-trisubstituted alkene **198** via Wittig olefination. To make the requisite Wittig salt **199**, diol **200** was selectively protected as the primary *t*-butyldimethylsilyl ether **201**¹⁶ (Scheme 3.6) and the remaining secondary alcohol was converted to the corresponding bromide **202** by *N*-bromosuccinimide.¹⁷ However, when bromide **202** was refluxed with triphenylphosphine in toluene or acetonitrile, no reaction occurred. Fortunately, an

alternative route to **206** was more successful. In this sequence, 3-bromobutanol (**204**) was converted to phosphonium salt **205**¹⁸ which was reacted with *n*-butyllithium and methyl iodide to give phosphonium salt **206** in good yield.¹⁹



Scheme 3.6 Synthesis of Wittig salt 206

Before attempting Wittig olefination of aldehyde **197** with phosphonium salt **206**, a model reaction of furfural (**207**) with **206** was investigated. When an excess amount of *n*-butyllithium was used to generate the ylide from **206**, it gave the addition product to furfural but when the phosphonium salt **205** was methylated and reacted in situ with furfural it afforded alkene **208** as an inseparable E/Z mixture (1.5:1) in good yield (Scheme 3.7).²⁰ Unfortunately, when the same reaction conditions were applied to aldehyde **197**, only decomposition occurred. This failure caused us to revise our initial plan for accessing (*E*)-trisubstituted alkene **171**.



Scheme 3.7 Attempted synthesis of trisubstituted alkene 209

An alternative plan was envisioned in which conversion of aldehyde **197** to an alkyne could be used to set in place the (*E*)-trisubstituted alkene at C7-C8 of providencin. To this end, aldehyde **197** was reacted with dimethyl 1-diazo-2-oxo-propylphosphonate (**210**)²¹ in the presence of base to furnish terminal alkyne **211** (Scheme 3.8). Alkyne **211** was methylated²² to give disubstituted alkyne **212**, but all attempts to functionalize this alkyne through hydrozirconation with Schwartz reagent²³ returned starting material or destroyed the furan. It is known that zirconium is oxyphilic in nature and since several oxygen atoms including a furan are present in alkyne **212**, we speculate that coordination of a zirconium species with those oxygen atoms could be responsible for destroying the Schwartz reagent.



Scheme 3.8 Attempted synthesis of vinyl bromide 213

The failure to functionalize alkyne 212 prompted a new strategy employing Horner-Wadsworth-Emmons olefination of 197 with ethyl 2-(diethoxyphosphono)propionate (214). This reaction produced (E)- α , β -unsaturated ester 215 in excellent yield (Scheme 3.9).²⁴ At this point, an attempt was made to install the exo methylene function at C15 of providencin. Thus, the tbutyldimethylsilyl ether of **215** was cleaved²⁵ to liberate alcohol **216** which was oxidized to cyclobutanone 217. However, all attempts to convert 217 to 218 using Wittig olefination.²⁶ Petasis reagent.²⁷ or Lombardo reagent²⁸ resulted only in return of the starting cyclobutanone. It seemed that the steric congestion around the carbonyl moiety of cyclobutanone 217 was responsible for the failure of the methylenation.



Scheme 3.9 Synthesis of (E)- α , β -unsaturated ester 217

Subsequently, a more direct route to cyclobutanone **217** was found through exhaustive oxidation of diol **173** with tetra-*n*-propylammonium perruthenate to furnish keto aldehyde **219** (Scheme 3.10). Horner-Wadsworth-Emmons olefination of **219** with phosphonate **214** took place exclusively at the aldehyde carbonyl but in this case the result was a 3:1 E/Z mixture of **217** and **220**, respectively, which proved difficult to separate.



Scheme 3.10 Alternative route to cyclobutylfuran 217

At this stage of our approach to providencin,²⁹ we accepted the fact that our inability to install an exo methylene moiety on cyclobutanone **217** as well as the difficulty elaborating an ethyl ester selectively in the presence of a methyl ester presented serious problems. A viable route to **1** therefore required a revised strategy.

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3.6 Experimental Section

General techniques: All reactions requiring anhydrous conditions were conducted in flame dried glass apparatus under an atmosphere of argon. THF, Et₂O, CH₂Cl₂, DMF, benzene and acetonitrile were dried by passage through an activated alumina column under argon. DMSO was distilled from CaH₂ at 15 mm Hg and stored over activated 4Å molecular sieves. Anhydrous MeOH was freshly distilled from calcium hydride. Preparative chromatographic separations were performed on silica gel (35-75 µm); reactions were followed by TLC analysis using silica plates with fluorescent indicator (254 nm) and visualized with a UV lamp or phosphomolybdic acid. All commercially available reagents were purchased from TCI or Aldrich and used as received unless stated otherwise. Optical rotations were measured with a polarimeter using a 1 mL capacity cell with 1 dm path length. Infrared spectra were recorded using a thin film supported on KBr discs or dispersed in a KBr pellet. ¹H and ¹³C NMR spectra were recorded in Fourier transform mode at the field strength specified on either a 300 or 400 or 700 MHz spectrometer. Spectra were obtained on CDCl₃ solutions in 5 mm diameter tubes, and chemical shifts in ppm are quoted relative to the residual signals of chloroform (δ H 7.26 ppm, or δ C 77.0 ppm). Multiplicities in the ¹H NMR spectra are described as: s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, br =broad; coupling constants are reported in Hz. Low (MS) and high (HRMS) resolution mass spectra are reported with ion mass/charge (m/z) ratios as values in atomic mass units.



Ethyl 2-((3a*R*,5*R*,6a*R*)-5-((*R*)-2,2-Dimethyl-1,3-dioxolan-4-yl)-2,2dimethylfuro[2,3-*d*][1,3]dioxol-6(3a*H*,5*H*,6a*H*)-ylidene) Acetate (180)

A solution of DMSO (0.75 mL, 10.04 mmol) in CH_2Cl_2 (1.5 mL) was added to a stirred solution of (COCl)₂ (0.39 mL, 4.56 mmol) in CH_2Cl_2 (20.0 mL) at – 78 °C. After 20 min, a solution of alcohol **179** (1 g, 3.8 mmol) in CH_2Cl_2 (25 mL) was added over 20 min. The mixture was stirred for 30 min at -78 °C, and Et_3N (3.5 mL, 25.84 mmol) was added and warmed to room temperature. After 1 h, the reaction mixture was poured into water (100 mL), the layers were separated, and the aqueous layer was extracted with CH_2Cl_2 (2 x 50 mL). The combined organic extract was dried (Na₂SO₄), filtered and concentrated in vacuo. The crude product was used directly for the next step.

To a stirred solution of the crude product obtained above in CH_2Cl_2 (15 mL) was added $Ph_3P=CH(CO_2Et)$ (1.6 g, 4.8 mmol) and the solution was stirred at room temperature for 18 h. The reaction mixture was poured in water (100 mL) and extracted with CH_2Cl_2 (2 x 50 mL), and the combined extract was dried (Na₂SO₄), filtered, and concentrated. The crude residue was purified by silica gel chromatography (9:1 hexanes/ethyl acetate) to afford **180** (1.08 g, 85% from **179**) as a

colorless solid: $R_f 0.69$ (1:1 hexanes/ethyl acetate); mp 66-70 °C (lit.^{8b} mp 69-71 °C); [α]²² _D +116.7 (*c* 1.0, CHCl₃) [lit.^{8b} [α]²⁹ _D +114.0 (*c* 1.15, CHCl₃)]; IR (neat) 2983, 2937, 1722, 1373, 1209, 1160, 1055, 1021 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.34 (t, *J* = 7.1 Hz, 3H), 1.37 (s, 3H), 1.43 (s, 3H), 1.46 (s, 3H), 1.54 (s, 3H), 4.00-4.06 (m, 1H), 4.10-4.18 (m, 2H), 4.26 (q, *J* = 7.2 Hz, 2H), 4.70 (dt, *J* = 7.0, 1.7 Hz, 5.76 (dt, *J* = 4.2, 1.3 Hz, 1H), 5.86 (d, *J* = 4.2 Hz, 1H), 6.65 (dt, *J* = 7.0, 1.7 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 14.1, 25.4, 26.7, 27.1, 60.1, 67.3, 76.7, 78.4, 79.9, 104.9, 110.2, 112.8, 117.9, 151.6, 165.2.



Ethyl 2-((3a*R*,5*R*,6*R*,6a*R*)-5-((*R*)-2,2-Dimethyl-1,3-dioxolan-4-yl)-2,2-dimethyl tetrahydrofuro[2,3-*d*][1,3]dioxol-6-yl) Acetate (181)

To a 95% ethyl alcohol (201 mL) solution of **180** (16.54 g, 50.42 mmol) was added 10% palladium on carbon (7.47 g, 7.05 mmol). The reaction mixture was purged with hydrogen for 5 min and stirred under a hydrogen atmosphere for 16 h. The reaction mixture was filtered through Celite which was rinsed with ethanol and the filtrate was concentrated. The crude residue was purified by silica gel flash chromatography (9:1 hexanes/ethyl acetate) to give **181** (15.02 g, 90%) as a colorless oil: $R_f 0.70$ (1:1

hexanes/ethyl acetate); $[\alpha]^{22}_{D} +63.7$ (*c* 1.15, CHCl₃); IR (neat) 2986, 2937, 1735, 1381, 1241, 1214, 1066, 1017 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.34 (t, *J* = 6.7 Hz, 3H), 1.37 (s, 3H), 1.43 (s, 3H), 1.54 (s, 3H), 2.23-2.39 (m, 1H), 2.65 (dd, *J* = 17.5, 10.2 Hz, 1H), 2.85 (dd, *J* = 17.5, 4.2 Hz, 1H), 3.70 (dd, *J* = 9.8, 7.5 Hz, 2H), 3.95-4.05 (m, 2H), 4.10-4.25 (m, 2H), 4.85 (t, *J* = 4.2 Hz, 1H), 5.80 (d, *J* = 3.6 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 14.1, 25.2, 26.6, 26.7, 29.9, 44.5, 60.4, 67.8, 77.8, 80.4, 104.9, 109.9, 111.7, 172.2; EI-HRMS calcd for C₁₇H₂₄O₅ [M] ⁺ (*m*/*z*) 308.1624, found 308.1634.



2-((3a*R*,5*S*,6*R*,6a*R*)-5-((*R*)-2,2-Dimethyl-1,3-dioxolan-4-yl)-2,2dimethyltetrahydrofuro[2,3-d][1,3]dioxol-6-yl)ethanol (182)

To a solution of LiAlH₄ (3.64 g, 182.2 mmol) in dry Et₂O (364 mL) under argon was added dropwise at 0 °C a solution of ester **181** (15.03 g, 45.5 mmol) in dry Et₂O (30 mL). After stirring for 2 h at room temperature, the reaction was quenched with satd aqueous ammonium chloride and extracted with Et₂O (2 x 50 mL). The combined organic extract was washed with brine and dried over anhydrous Na₂SO₄ and the solvent was removed under reduced pressure. The crude mixture was purified by silica gel chromatography (7:3 hexanes/ethyl acetate) to give **182** (11.28 g, 86%) as a
colorless oil: R_f 0.31 (7:3 hexanes/ethyl acetate); $[\alpha]^{23}_{D}$ +49.0 (*c* 1.00, CHCl₃); IR (neat) 3412, 2980, 2937, 2887, 2325, 1376, 1212, 1061, 1021, 844 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.34 (s, 3H), 1.36 (s, 3H), 1.45 (s, 3H), 1.55 (s, 3H), 1.70 (s, broad, OH), 1.85-2.1 (m, 4H), 3.75-3.85 (m, 3H), 3.90-4.08 (m, 2H), 4.10-4.20 (m, 2H), 4.75 (t, *J* = 4.1 Hz, 1H), 5.8 (d, *J* = 3.7 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 25.3, 26.4, 26.8, 28.0, 45.6, 61.2, 67.6, 77.9, 81.8, 81.9, 105.0, 109.6, 111.7; EI-HRMS calcd for C₁₄H₂₅O₆ [M + H] ⁺ (*m/z*) 289.1651, found 289.1648.



(3aR,5S,6R,6aR)-5-((R)-2,2-Dimethyl-1,3-dioxolan-4-yl)-6-(2-(4methoxybenzyloxy)ethyl)-2,2-dimethyltetrahydrofuro[2,3-d][1,3]dioxole (183)

A cold (0 °C) suspension of sodium hydride (4.86 g, 121.96 mmol) in dry THF (125 mL) was treated with a solution of **182** (8.78 g, 30.49 mmol) in dry THF (20 mL) and the mixture was stirred for 20 min at 0 °C and for 20 min at room temperature. After cooling the solution to 0 °C, *p*-methoxybenzyl chloride (5.37 mL, 39.64 mmol) in THF (10 mL) containing tetra-*n*-butylammonium iodide (1.12 g, 3.04 mmol) was introduced. The reaction mixture was stirred at room temperature for 10 h and the reaction was quenched with satd ammonium chloride solution (50 mL) at 0 °C. The separated aqueous layer was extracted with ethyl acetate (2 x 50 mL) and the

combined organic extract was dried (Na₂SO₄), filtered and concentrated, leaving a residue that was chromatographed on silica gel (4:1 hexanes/ethyl acetate) to furnish **183** (10.52 g, 84%) as a colorless oil: R_f 0.50 (1:1 hexanes/ethyl acetate); $[\alpha]^{22}_{D}$ +54.5 (*c* 1.25, CHCl₃); IR (neat) 2986, 2933, 2878, 1617, 1515, 1367, 1246, 1212, 1064, 1033, 875, 820, 844 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.34(s, 3H), 1.36 (s, 3H), 1.45 (s, 3H), 1.55 (s, 3H), 1.80-1.88 (m, 1H), 1.89-2.15 (m, 2H), 3.60 (ddd, *J* = 7.3, 4.9, 1.6 Hz, 2H), 3.78-3.80 (m, 1H), 3.86 (s, 3H), 3.95 (dd, *J* = 7.1, 4.7 Hz, 1H), 4.10 (m, 2H), 4.48 (d, *J* = 5.4 Hz, 2H), 4.60 (t, *J* = 3.9 Hz, 1H), 5.76 (d, *J* = 3.5 Hz, 1H), 6.88 (d, *J* = 8.6 Hz, 2H), 7.30 (d, *J* = 8.6 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 25.1, 25.3, 26.4, 26.5, 26.8, 45.4, 55.3, 67.3, 68.2, 72.7, 77.7, 81.5, 81.9, 104.9, 109.5, 111.7, 113.7 (2C), 129.3 (2C), 130.6, 159.1; EI-HRMS calcd for C₂₂H₃₂O₇ [M] ⁺ *m*/z 408.2148, found *m*/z 408.2143.



(*R*)-1-((3a*R*,5*S*,6*R*,6a*R*)-6-(2-(4-Methoxybenzyloxy)ethyl)-2,2dimethyltetrahydrofuro[2,3-d][1,3]dioxol-5-yl)ethane-1,2-diol (184)

A solution of **183** (2.16 g, 5.29 mmol) in 60% AcOH (44 mL) was stirred at room temperature for 18 h, after which toluene (90 mL) was added to the mixture and the solvent was removed under reduced pressure. The residual oil was chromatographed

on silica gel (1:1 hexanes/ethyl acetate) to give **184** (1.60 g, 86%) as a colorless oil: R_f 0.30 (1:4 hexanes/ethyl acetate); $[\alpha]^{23}_D$ +52.1 (*c* 1.15, CHCl₃); IR (neat) 3420, 2934, 2869, 1513, 1372, 1247, 1031, 874, 820 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.32 (s, 3H), 1.50 (s, 3H), 1.90 (m, 2H), 2.10 (m, 1H), 2.47 (s, broad, OH), 3.33 (s, broad, OH), 3.55 (dd, *J* = 15.8, 6.8 Hz, 1H), 3.65 (m, 1H), 3.70 (m, 3H), 3.80 (s, 3H), 3.90 (dd, *J* = 9.7, 5.1 Hz, 1H), 4.48 (d, *J* = 6.1 Hz, 2H), 4.60 (t, *J* = 4.0 Hz, 1H), 5.75 (d, *J* = 3.6 Hz, 1H), 6.90 (d, *J* = 8.6 Hz, 2H), 7.30 (d, *J* = 8.6 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 25.2, 26.3, 26.7, 44.2, 55.2, 60.4, 63.8, 68.3, 72.7, 73.2, 81.7, 82.7, 104.6, 111.7, 113.8 (2C), 129.8 (2C), 130.0, 159.3; CI-HRMS calcd for C₁₉H₂₈O₇ [M] ⁺ *m/z* 368.1835, found *m/z* 368.1838.



(3a*R*,5*S*,6*R*,6a*R*)-6-(2-(4-Methoxybenzyloxy)ethyl)-2,2-dimethyl-5-vinyltetrahydrofuro[2,3-*d*][1,3]dioxole (185)

To a stirred solution of **184** (1.50 g, 4.24 mmol) in toluene (36 mL) were added triphenylphosphine (4.46 g, 16.98 mmol) and imidazole (1.15 g, 16.98 mmol) and the mixture was heated at reflux. When reflux started, iodine (3.20 g, 12.72 mmol) was added to the reaction mixture slowly (portion wise). The solution was refluxed for 2.5 h, cooled to room temperature and was washed with satd sodium thiosulphate solution (3 x 30 mL) followed by water (50 mL) and brine (50 mL). The organic layer was

separated and dried over anhydrous Na₂SO₄, filtered and concentrated. The crude residue was purified by silica gel chromatography (4:1 hexanes/ethyl acetate) to afford **185** (1.33 g, 94%) as a colorless oil: R_f 0.56 (1:1 hexanes/ethyl acetate); $[\alpha]^{20}_{D} + 34.0$ (*c* 1.10, CHCl₃); IR (neat) 2911, 1612, 1513, 1248, 1098, 1033, 820 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.32 (s, 3H), 1.52 (s, 3H), 1.65 (m, 1H), 1.90 (m, 2H), 3.55 (t, *J* = 5.1 Hz, 2H), 3.72 (s, 3H), 4.18 (t *J* = 7.8 Hz, 1H), 4.45 (s, 2H), 4.60 (t, *J* = 3.1 Hz, 1H), 5.24 (d, *J* = 9.9 Hz, 1H), 5.35 (d, *J* = 16.9 Hz, 1H), 5.75 (m, 2H), 6.85 (m, 2H), 7.25 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 24.4, 26.3, 26.7, 46.6, 55.3, 67.9, 72.4, 80.9, 83.0, 104.9, 111.4, 113.7 (2C), 118.7, 129.2 (2C), 130.2, 135.5, 159.1; EI-HRMS calcd for C₁₉H₂₆O₅ [M]⁺ *m*/z 334.1780, found *m*/z 334.1764.



(2*R*,3*R*,4*S*,5*R*)-2-Methoxy-4-(2-(4-methoxybenzyloxy)ethyl)-5vinyltetrahydrofuran-3-ol (186)

A solution of **185** (1.33 g, 4.00 mmol) in HCl-MeOH (1% mol/V) was stirred at room temperature for 16 h. The mixture was neutralized with aq. NaOH (5N) and filtered, and the filtrate was concentrated and extracted with CH₂Cl₂ (3 x 30 mL). The combined organic extract was concentrated and the crude residue was purified by chromatography on silica gel (3:1 hexanes/ethyl acetate) to give **186** (0.76 g, 62%) as a colorless oil: R_f 0.62 (1:1 hexanes/ethyl acetate); $[\alpha]^{20}_{D}$ -69.8 (*c* 1.05, CHCl₃); IR

(neat) 3443, 2972, 2325, 1614, 1512, 1252, 1101, 1033, 934, 820 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.70 (m, 1H), 1.85 (m, 1H), 2.10 (m, 1H), 3.10 (d, *J* = 3.0 Hz, OH), 3.40 (s, 3H), 3.45 (ddd, *J* = 10.5, 9.6, 2.5 Hz, 1H), 3.68 (ddd, *J* = 8.3, 4.5, 3.3 Hz, 1H), 3.82 (s, 3H), 4.14 (t, *J* = 3.7 Hz, 1H), 4.22 (t, *J* = 8.6 Hz, 1H), 4.40 (s, 2H), 4.90 (s, 1H), 5.14 (d, *J* = 10.0 Hz, 1H), 5.22 (d, *J* = 17.0 Hz, 1H), 5.80 (ddd, *J* = 17.3, 10.1, 7.8 Hz, 1H), 6.90 (d, *J* = 8.5 Hz, 2H), 7.30 (d, *J* = 8.5 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 25.4, 47.4, 54.6, 55.3, 69.3, 73.3, 76.9, 85.8, 110.6, 113.9, 116.7, 129.4, 129.5 (3C), 139.1, 159.4; CI-HRMS calcd for C₁₇H₂₄O₅ [M] ⁺ *m*/*z* 308.1624, found *m*/*z* 308.1634.



tert-Butyl((2*R*,3*R*,4*R*,5*R*)-2-methoxy-4-(2-(4-methoxybenzyloxy)ethyl)-5-vinyltetrahydrofuran-3-yloxy)dimethylsilane (177)

To a solution of **186** (586 mg, 1.90 mmol) and 2,6-lutidine (0.3 mL, 2.47 mmol) in CH₂Cl₂ (15 mL) at 0 °C was added TBSOTf (0.57 mL, 2.47 mmol) and the mixture was stirred at room temperature for 3 h. The reaction was quenched with satd aqueous NaHCO₃ solution and the aqueous layer was extracted with EtOAc (3 x 20 mL). The combined organic extract was washed with brine, dried (Na₂SO₄), and concentrated in vacuo. The crude residue was purified by chromatography on silica gel (9:1 hexanes/ethyl acetate) to afford **177** (788 mg, 98%) as a colorless oil: $R_f 0.77$ (7:3

hexanes/ethyl acetate); $[\alpha]^{23}_{D}$ +32.0 (*c* 1.00, CHCl₃); IR (neat) 2953, 2929, 2856, 1523, 1250, 1099, 1039, 835 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.03 (s, 3H), 0.08 (s, 3H), 0.93 (s, 9H), 1.65 (m, 1H), 1.85 (m, 1H), 2.10 (m, 1H), 3.38 (s, 3H), 3.44 (t, *J* = 6.7 Hz, 2H), 3.82 (s, 3H), 4.02 (d, *J* = 3.7 Hz, 1H), 4.22 (t, *J* = 2.3 Hz, 1H), 4.44 (d, *J* = 2.3 Hz, 2H), 4.70 (s, 1H), 5.14 (d, *J* = 10.0 Hz, 1H), 5.24 (d, *J* = 17.1 Hz, 1H), 5.80 (ddd, *J* = 17.8, 10.0, 7.8 Hz, 1H), 6.90 (d, *J* = 8.6 Hz, 2H), 7.30 (d, *J* = 8.6 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ -5.0, -4.5, 18.1, 24.8, 25.7 (3C), 44.2, 54.5, 55.3, 68.2, 74.3, 76.9, 85.5, 109.1, 113.7 (2C), 117.0, 129.3 (2C), 130.5, 139.2, 159.1; CI-HRMS calcd for C₂₃H₃₉O₅Si [M + H] ⁺ *m*/z 423.2567, found *m*/z 423.2566.



(1*R*,2*R*,3*R*,4*R*)-2-(*tert*-Butyldimethylsilyloxy)-3-(2-(4-methoxybenzyloxy)ethyl)-4vinylcyclobutanol (176)

To a solution of $Cp_2Zr(n-Bu)_2$, prepared in situ by reaction of zirconocene dichloride (4.49 g, 15.37 mmol) in toluene (30 mL) with *n*-butyllithium in hexane (2 eq) at -78 °C for 1.5 h, was added a solution of **177** (2.59 g, 6.15 mmol) in toluene (30 mL) at - 78 °C. The reaction mixture was stirred for 30 min at -78 °C and allowed to warm to ambient temperature. The mixture was stirred for 3 h, a solution of BF₃.OEt₂ (3.9 mL, 30.75 mmol) in toluene (5.0 mL) was added at 0 °C, and stirring was continued for 2 h at ambient temperature. The reaction was quenched with 1N HCl (50 mL) and the

mixture was extracted with CH₂Cl₂ (3 x 25 mL). The combined organic extract was washed with brine, dried (Na₂SO₄), and concentrated under reduced pressure. The crude residue was purified by chromatography on silica gel (9:1 hexanes/ethyl acetate) to give **176** (2.43 g, 86%) as a colorless oil: R_f 0.64 (7:3 hexanes/ethyl acetate); $[\alpha]^{22}_{D}$ -4.8 (*c* 1.00, CHCl₃); IR (neat) 3416, 2985, 2930, 2859, 1509, 1249, 1150, 1095, 1036, 891, 835, 733 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.09 (s, 3H), 0.11 (s, 3H), 0.93 (s, 9H), 1.60 (td, *J* = 14.4, 6.4 Hz, 1H), 1.80 (s, broad, OH), 2.05 (td, *J* = 13.6, 6.9 Hz, 1H), 2.30 (dd, *J* = 7.2, 6.6 Hz, 1H), 2.75 (t, *J* = 8.0 Hz, 1H), 3.50 (ddd, *J* = 16.0, 9.1, 6.8 Hz, 2H), 3.82 (s, 3H), 4.14 (dd, *J* = 11.7, 5.8 Hz, 2H), 4.45 (s, 2H), 5.15 (dd, *J* = 17.3, 2.9 Hz, 1H), 5.25 (dd, *J* = 10.5, 2.5 Hz, 1H), 5.87 (ddd, *J* = 17.1, 10.2, 8.5 Hz, 1H), 6.90 (d, *J* = 8.6 Hz, 2H), 7.30 (d, *J* = 8.6 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ -5.0, -4.6, 18.1, 25.8 (3C), 29.7, 36.9, 43.7, 55.3, 68.7, 72.6, 75.0, 75.4, 113.7 (2C), 118.0, 129.2 (2C), 130.7, 136.5, 159.1; CI-HRMS calcd for C₂₂H₃₇O₄Si [M + H] ⁺ *m/z* 393.2461, found *m/z* 393.2455.



tert-Butyl((1*R*,2*R*,3*R*,4*R*)-2-(2-(4-methoxybenzyloxy)ethyl)-4-((2-(trimethylsilyl)ethoxy)methoxy)-3-vinylcyclobutoxy)dimethylsilane (187)

Alcohol **176** (80 mg, 0.20 mmol) was dissolved at room temperature in 2 mL of CH_2Cl_2 containing 62 μ L (0.81 mmol) of diisopropylethylamine and 71.8 μ L (0.40

mmol) of [2-(trimethylsilyl)]ethoxymethyl chloride (SEM chloride). The reaction mixture was heated at reflux for 10 h, at which point 10 mL of CH₂Cl₂ and 10 mL of 10% aqueous HCl solution were added. The organic layer was separated and washed successively with 10 mL of 10% aqueous HCl, 10 mL of satd aqueous sodium bicarbonate and 20 mL of brine. The organic layer was dried (Na₂SO₄), filtered, and concentrated leaving a reddish oil, which was placed on a silica gel column and eluted with 3% ethyl acetate in hexanes to give 187 (85 mg, 80%) as a colorless oil: $R_f 0.64$ (9:1 hexanes/ethyl acetate); $[\alpha]^{22}$ p -43.2 (c 1.25, CHCl₃); IR (neat) 2952, 2928, 2855, 1513, 1249, 1111, 1061, 1039, 1008, 859, 835 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.02 (m, 16H), 0.90 (m, 11H), 1.60-1.70 (m, 1H), 1.88-2.20 (m, 1H), 2.02-2.22 (m, 1H), 2.75 (t, J = 8.8 Hz, 1H), 3.40-3.60 (m, 2H), 3.60-3.70 (m, 2H), 3.82 (s, 3H), 4.16 (t, J = 7.5 Hz, 1H), 4.34 (t, J = 7.8 Hz, 1H), 4.44 (s, 2H), 4.64 (s, 2H), 5.00-5.10 (m, 100)2H), 5.85-6.00 (m, 1H), 6.85-6.90 (m, 2H), 7.25-7.28 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ -5.0, 4.7, 1.3, 18.0, 18.1, 25.8, 29.2, 37.9, 43.2, 55.3, 65.3, 65.5, 68.7, 72.2, 72.6, 79.2, 93.4, 113.7, 115.7, 129.2, 130.7, 137.8, 159.1; CI-HRMS calcd for $C_{28}H_{51}O_5Si_2 [M + H]^+ (m/z) 523.3275$, found 523.3273.



1-((1*R*,2*R*,3*R*,4*R*)-3-(*tert*-Butyldimethylsilyloxy)-2-(2-(4-methoxybenzyloxy)ethyl)-4-((2-(trimethylsilyl)ethoxy)methoxy)cyclobutyl)ethanone (188)

In a Schlenk flask fitted with an oxygen filled balloon was added a suspension of PdCl₂ (2.5 mg, 0.014 mol) and CuCl (14.7 mg, 0.149 mmol) in DMF/H₂O 7:1 (1.00 mL) and the suspension was stirred for 60 min. A solution of 187 (78 mg, 0.149 mmol) in DMF/H₂O 7:1 (0.50 ml) was added and the mixture was stirred for 48 h under an oxygen atmosphere, after which the reaction was quenched by adding icecold 3 N HCl (4 mL). The mixture was extracted with Et₂O (2 \times 10 mL), and the combined organic extract was dried (Na₂SO₄), filtered and concentrated under reduced pressure. The crude residue was purified by flash chromatography (hexanes/ethyl acetate 9:1) to yield 80.0 mg (98%) of inseparable 188 and 189 (2.3:1) as a colorless oil: $R_f 0.77$ (2:8 hexanes-ethyl acetate); $[\alpha]^{24} - 26.6$ (c 0.9, CHCl₃); IR (neat) 2955, 2927, 2853, 1710, 1614, 1515, 1463, 1113, 1061, 1036, 1005, 835 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.30 (s, 9H), 0.60 (s, 3H), 0.65 (s, 3H), 1.55 (m, 1H), 1.90 (m, 1H), 2.70 (m, 1H), 3.24 (dd, J = 9.2, 2.6 Hz, 1H), 3.40 (m, 1H), 3.50 (m, 1H), 3.62 (dd, J = 0.23 Hz, 10.23 Hz)9.1, 7.8 Hz, 2H), 3.83 (s, 3H), 4.26 (m, 1H), 4.35 (m, 1H), 4.42 (s, 2H), 4.67 (s, 2H), 6.89 (d, J = 8.5 Hz, 2H), 7.25 (d, J = 8.3 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ -4.9, -4.8, -1.3, 18.0, 25.7, 28.3, 31.5, 34.3, 51.2, 55.2, 65.7, 68.8, 72.6, 79.2, 93.9, 113.7, 129.2, 130.6, 159.1, 201.8, 207.9; EI-HRMS calcd for $C_{28}H_{49}O_6Si_2$ [M] ⁺ (*m/z*) 561.3068, found 561.3061.



tert-Butyl((1R,2R,3R,4R)-2-(2-(4-methoxybenzyloxy)ethyl)-4-

(triisopropylsilyloxy)-3-vinylcyclobutoxy)dimethylsilane (190)

To a solution of **176** (409 mg, 1.04 mmol) and DMAP (152.5 mg, 1.25 mmol) in dry pyridine (3.2 mL) was added TIPSOTf (1.1 mL, 4.16) via syringe and the mixture was stirred for 15 h at room temperature. Excess TIPSOTf was consumed by addition of dry MeOH (5 mL) and after 15 min the solution was diluted with Et₂O. The solution was washed with 5% HCl (2 mL) and then with a mixture of satd aqueous NaHCO₃ and brine (1:1, 20 mL). The aqueous layer was extracted with Et₂O (2 x 20 mL) and the combined organic extract was dried (Na_2SO_4). Filtration through a short plug of silica gel and concentration of the filtrate in *vacuo* yielded crude product which was purified by chromatography on silica gel (hexanes/ethyl acetate 4:96) to afford 190 (565 mg, 98%) as a colorless oil: $R_f 0.65$ (2:8 hexanes-ethyl acetate); $\left[\alpha\right]_{D}^{20} - 53.9$ (c 1.15, CHCl₃); IR (neat) 2952, 2859, 1642, 1611, 1583, 1515, 1463, 1364, 1302, 1252, 1163, 1101, 996, 777, 687 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.09 (s, 3H), 0.10 (s, 3H), 0.91 (s, 9H), 1.11 (m, 21H), 1.60 (m, 1H), 2.00 (m, 1H), 2.15 (m, 1H), 2.70 (t, J = 8.08 Hz, 1H), 3.53 (t, J = 6.82 Hz, 2H), 3.82 (s, 3H), 4.24 (t, J = 7.78 Hz, 1H), 4.30 (t, J = 7.78 Hz, 1H), 4.47 (d, J = 3.54 Hz, 2H), 5.05 (d, J = 17.3 Hz, 1H), 5.10 (d, J = 17.3 Hz, 100 Hz)9.7 Hz, 1H), 6.10 (ddd, J = 17.3, 10.2, 8.7 Hz, 1H), 6.90 (m, 2H), 7.25 (m, 2H); ¹³C

NMR (100 MHz, CDCl₃) δ -4.9, -4.7, 12.1, 12.2, 17.9, 18.0 (2C), 25.8, 28.9, 29.2, 37.9, 38.9, 45.0, 45.1, 55.2, 68.7, 68.9, 72.5, 75.1, 75.2, 75.6, 76.0, 113.7 (2C), 115.3, 129.1 (2C), 130.7, 138.4, 138.3, 159.0; EI-HRMS calcd for C₃₁H₅₆O₄Si₂Na [M + Na] ⁺ *m*/*z* 571.3615, found *m*/*z* 571.3598.



1-((1*R*,2*R*,3*R*,4*R*)-3-(*tert*-Butyldimethylsilyloxy)-2-(2-(4-methoxybenzyloxy)ethyl)-4-(triisopropylsilyloxy)cyclobutyl)ethanone (192)

In a Schlenk flask fitted with an O₂ filled balloon was placed a suspension of PdCl₂ (203 mg, 1.15 mol) and CuCl (190.0 mg, 1.92 mmol) in DMF/H₂O 7:1 (4.8 mL). The mixture was stirred for 60 min and a solution of **190** (527.0 mg, 0.96 mmol) in DMF – H₂O (7:1, 4.8 mL) was added. The mixture was stirred at 40 °C for 48 h under an O₂ atmosphere and the reaction was quenched by adding ice-cold 3N HCl (4 mL). The mixture was extracted with Et₂O (2 × 10 mL) and the combined organic extract was dried (MgSO₄), filtered and concentrated under reduced pressure. The crude residue was purified by chromatography on silica gel (hexanes/ethyl acetate, 1:9) to yield an inseparable mixture (1:11) of **191** and **192** (417 mg, 77%) as a colorless oil: $R_f 0.62$ (2:8 hexanes-ethyl acetate); $[\alpha]^{24}_{D} - 56.0$ (*c* 1.25, CHCl₃); IR (neat) 2940, 2869, 1707, 1617, 1509, 1459, 1357, 1246, 1175, 1107, 888, 838, 773, 681 cm⁻¹; ¹H NMR (400

MHz, CDCl₃) δ 0.06 (s, 3H), 0.08 (s, 3H), 0.91 (s, 9H), 1.08 (m, 21H), 1.55 (m, 1H), 1.90 (m, 1H), 2.17 (s, 3H), 2.60 (dd, *J* = 11.6, 8.6 Hz, 1H), 3.20 (dd, *J* = 9.5, 1.7 Hz, 1H), 3.45 (m, 2H), 3.82 (s, 3H), 4.30 (t, *J* = 7.5 Hz, 1H), 4.41 (s, 2H), 4.45 (dd, *J* = 7.1, 6.2 Hz 1H), 6.90 (d, *J* = 8.6 Hz, 2H), 7.22 (d, *J* = 8.6 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ -5.0, -4.5, 12.1 (3C), 17.9 (6C), 18.0, 25.8 (3C), 28.1, 32.0, 33.6, 52.9, 55.9, 58.8, 72.6, 74.8, 75.6, 113.7 (2C), 129.1 (2C), 130.6, 159.1, 208.2; EI-HRMS calcd for C₃₁H₅₆O₅Si₂Na [M + Na] ⁺ *m*/*z* 578.3564, found *m*/*z* 578.3571.



Methyl 3-((1*R*,2*R*,3*R*,4*R*)-3-(*tert*-Butyldimethylsilyloxy)-2-(2-(4methoxybenzyloxy)ethyl)-4-(triisopropylsilyloxy)cyclobutyl)-3-oxopropanoate (194)

To a solution of **191** and **192** (216 mg, 0.38 mmol) in THF (3.8 mL) at -78 $^{\circ}$ C was added a solution of LDA (0.95 mL, 0.95 mmol, freshly prepared as a 1M solution in THF). The pale yellow solution was stirred for 1 h then was treated with methyl cyanoformate (90.4 μ L, 1.14 mmol). The resulting solution was stirred for 1 h at -78 $^{\circ}$ C and the reaction was quenched with H₂O (10 mL). The mixture was extracted with Et₂O (3 x 10 mL) and the combined organic extract was washed with H₂O and brine. The extract was dried (Na₂SO₄) and concentrated and the residue was

chromatographed on silica gel (4:96 hexanes/ethyl acetate) to afford **194** (182 mg, 70%) as a colorless oil: $R_f 0.71$ (7:3 hexanes-ethyl acetate); $[\alpha]^{23}_D - 61.2$ (*c* 1.2, CHCl₃); IR (neat) 2949, 2859, 1747, 1716, 1611, 1459, 1246, 1163, 1104, 882, 835, 780, 681 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.06 (s, 3H), 0.08 (s, 3H), 0.91 (s, 9H), 1.08 (m, 21H), 1.50 (m, 1H), 1.90 (m, 1H), 2.65 (dd, *J* = 11.4, 8.7 Hz, 1H), 3.30 (dd, *J* = 9.5, 1.5 Hz, 1H), 3.45 (m, 2H), 3.50 (s, 2H), 3.72 (s, 3H), 3.82 (s, 3H), 4.30 (t, *J* = 7.7 Hz, 1H), 4.44 (s, 2H), 4.48 (dd, *J* = 8.9, 6.5 Hz, 1H), 6.90 (d, *J* = 8.5 Hz, 2H), 7.25 (d, *J* = 8.5 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ -5.1, -4.5, 12.1 (3C), 17.9 (6C), 18.0, 25.8 (3C), 27.9, 33.8, 50.8, 52.0, 55.2 (2C), 68.8, 72.6, 74.7, 75.8, 113.7 (2C), 129.1 (2C), 130.5, 159.1, 167.6, 202.3; EI-HRMS calcd for C₃₃H₅₈O₇Si₂ [M] ⁺ *m*/z 622.3721, found *m*/z 622.3683.



Methyl2-((1R,2R,3R,4R)-3-(tert-Butyldimethylsilyloxy)-2-(2-(4-methoxybenzyloxy)ethyl)-4-(triisopropylsilyloxy)cyclobutyl)-5-(hydroxymethyl)furan-3-carboxylate(196)andMethyl2-((1R,2R,3R,4R)-3-Hydroxy-2-(2-(4-methoxybenzyloxy)ethyl)-4-(triisopropylsilyloxy)cyclobutyl)-5-(hydroxymethyl)furan-3-carboxylate(173)

To a solution of **194** (104 mg, 0.170 mmol) in MeOH (1.7 mL) under argon at room temp was added glyceraldehyde acetonide (**174**, 63.4 μ L, 0.510 mmol) followed by PPTS (4.27 mg, 0.017 mmol). The reaction mixture was stirred at 70 °C for 20 h and quenched with 2% NaOH solution (2 mL). The aqueous layer was extracted with CH₂Cl₂ (3 x 10 mL) and the combined extract was washed with H₂O, dried (Na₂SO₄) and concentrated. The residue was purified by chromatography on silica gel (14:8 – 40.60 hexanes/ethyl acetate) to afford **196** (47 mg, 42%) along with **173** (23 mg, 25%) as colorless oils.

196: R_f 0.57 (7:3 hexanes-ethyl acetate); $[\alpha]^{22}{}_{D}$ – 38.2 (*c* 1.7, CHCl₃); IR (neat) 3425, 2930, 2865, 1719, 1614, 1574, 1515, 1466, 1249, 1107, 1070, 900, 835, 773, 684 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.12 (s, 6H), 0.90 (s, 30H), 1.60 (m, 1H), 1.90 (m 1H), 2.6 (m, 1H), 3.50 (m, 2H), 3.76 (s, 3H), 3.82 (s, 3H), 4.20 (dd, *J* = 8.9, 3.4 Hz, 1H), 4.40 (dd, *J* = 8.5, 11.4 Hz, 2H), 4.51 (dd, *J* = 8.7, 6.4 Hz, 1H), 4.60 (dd, *J* = 8.5, 6.1 Hz, 2H), 4.65 (m, 1H), 6.57 (s, 1H), 6.84 (d, *J* = 8.5 Hz, 2H), 7.20 (d, *J* = 8.5 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ -3.9, -3.8, 11.8 (3C), 17.7 (6C), 18.0, 25.9 (3C), 29.3, 37.2, 39.2, 51.1, 55.2, 57.3, 68.1, 72.5, 75.3, 76.7, 108.5, 113.6 (2C), 115.4, 129.1 (2C), 130.6, 152.5, 159.0, 161.1, 164.2; EI-HRMS calcd for C₃₆H₆₀O₈Si₂Na [M + Na] ⁺ *m*/*z* 699.3727, found *m*/*z* 699.3724.

173: $R_f 0.13$ (7:3 hexanes-ethyl acetate); $[\alpha]^{22}_D + 1.13$ (*c* 1.85, CHCl₃); IR (neat) 3416, 2941, 2925, 2855, 1719, 1618, 1516, 1466, 1244, 1244, 1065, 882, cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.95 (m, 21H), 1.78 (m, 1H), 2.10 (m, 1H), 2.82 (septet, *J* = 4.5)

Hz, 1H), 3.43 (ddd, J = 12.2, 9.1, 3.4 Hz, 1H), 3.62 (pentet, J = 4.5 Hz, 1H), 3.77 (s, 3H), 3.82 (s, 3H), 4.21 (dd, J = 4.3, 8.3 Hz, 1H), 4.47 (m, 3H), 4.53 (m, 1H), 4.53 (m, 2H), 6.55 (s, 1H), 6.87 (d, J = 8.6 Hz, 2H), 7.24 (d, J = 8.6 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 11.8 (3C), 17.6 (6C), 29.4, 37.9, 39.6, 51.2, 55.2, 57.2, 68.7, 73.0, 74.9, 76.0, 108.6, 113.8 (2C), 115.3, 129.4 (3C), 152.4, 159.3, 160.7, 164.3; EI-HRMS calcd for C₃₀H₄₆O₈SiNa [M + Na] ⁺ m/z 585.2860, found m/z 585.2844.



Methyl2-((1R,2R,3R,4R)-3-(tert-Butyldimethylsilyloxy)-2-(2-(4-methoxybenzyloxy)ethyl)-4-(triisopropylsilyloxy)cyclobutyl)-5-formylfuran-3-carboxylate (197)

To a solution of **196** (8.0 mg, 0.011 mmol) in CH₂Cl₂ (0.1 mL) was added tetra-*n*propylammonium perruthenate (TPAP, 4.63 mg, 0.008 mmol) and the reaction mixture was stirred at room temperature for 1.5 h. The suspension was filtered through a pad of silica gel which was washed with CH₂Cl₂ (10 mL). Evaporation of the filtrate and chromatography of the crude residue on silica gel (93:7 hexanes/ethyl acetate) afforded **197** (7.0 mg, 88%) as a colorless oil: R_f 0.64 (3:7 hexanes-ethyl acetate); $[\alpha]^{16}_{D} - 53.7$ (*c* 1.0, CHCl₃); IR (neat) 2920, 2858, 1732, 1693, 1456, 1254, 1114, 1075 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.14 (s, 6H), 0.92 (s, 21H), 0.95 (s, 9H), 1.70 (m, 1H), 2.10 (m, 1H), 2.70 (m, 1H), 3.50 (m, 2H), 3.81 (s, 3H), 3.82 (s, 3H), 4.32 (dd, J = 8.4, 3.0 Hz, 1H), 4.40 (dd, J = 21.5, 11.5 Hz, 2H), 4.60 (dd, J = 8.5, 6.4 Hz, 1H), 4.72 (dd, J = 8.5, 6.2 Hz, 1H), 6.82 (d, J = 8.70Hz, 2H), 7.19 (d, J = 8.7 Hz, 2H), 7.45 (s, 1H), 9.62 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -4.9, -4.8, 11.6 (3C), 17.6 (6C), 17.7, 25.8 (3C), 29.7, 37.2, 39.9, 51.6, 55.2, 68.0, 72.5, 75.2, 76.7, 113.6 (2C), 117.4, 121.2, 129.1 (2C), 130.5, 150.7, 159.0, 162.9, 166.6, 177.2; HRMS calcd for C₃₆H₅₈O₈Si₂Na [M + Na] ⁺ *m*/*z* 697.3568, found *m*/*z* 697.3545.



4-(tert-Butyldiphenylsilyloxy)butan-2-ol (201)

To an ice-cold, stirred solution of **200** (1.0 mL, 11.09 mmol), DMAP (6.77 mg, 0.05 mmol), and NEt₃ (1.68 mL, 12.08 mmol) in dry CH₂Cl₂ (30 mL) was added dropwise a solution of TBDPSCI (2.8 mL, 11.09 mmol) in CH₂Cl₂ (5 mL). The mixture was stirred at room temperature for 12 h, and then MeOH (2 mL) was added to the reaction mixture which was stirred for 15 min. The solvent was removed under reduced pressure and the residue was dissolved in Et₂O and the solution was washed with H₂O. The separated Et₂O layer was dried (Na₂SO₄), filtered, and concentrated and the residual oil was purified by silica gel chromatography (hexanes/ethyl acetate, 94:6) to afford **201** (3.25 g, 89%) as a colorless oil: R_f 0.62 (3:7 hexanes-ethyl acetate); IR (neat) 3413, 3067, 1959, 1928, 1858, 1475, 1421, 1110, 1071, 827, 733, 695 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.11 (s, 9H), 1.26 (d, *J* = 6.2 Hz, 3H), 1.69 (m, 1H), 1.80

(m, 1H), 3.32 (d, J = 2.5 Hz, 1H), 3.90 (m, 2H), 4.15 (m, 1H), 7.46 (m, 6H), 7.74 (d, J = 6.9 Hz, 4H); ¹³C NMR (100 MHz, CDCl₃) δ 19.0, 23.4, 26.8, 40.1, 63.4, 67.9, 127.8, 129.8, 133.0, 133.1, 135.6; CI-HRMS calcd for C₂₀H₂₉O₂Si [M + H] ⁺ (*m*/*z*) 329.1936, found 329.1934.



(3-Bromobutoxy)(tert-butyl)diphenylsilane (202)

To a solution of **201** (1.066 g, 3.24 mmol) in CH₂Cl₂ (30 mL) was added PPh₃ (1.27 g, 4.86 mmol). The solution was cooled to 0 °C and NBS (1.14 g, 6.49 mmol) was added. The reaction mixture was stirred at room temperature for 3 h, and water (20 mL) was added to the mixture. The layers were separated and the aqueous layer was extracted with CH₂Cl₂ (2 x 20 mL) and the combined organic extract was dried (Na₂SO₄), filtered, and concentrated. The crude residue was purified by silica gel chromatography (hexanes/ethyl acetate 3:97) to afford **202** (1.00 g, 79%) as a colorless oil: R_f 0.85 (2:8 hexanes-ethyl acetate); IR (neat) 3071, 3048, 2959, 2924, 1846, 1467, 1429, 1188, 1106, 879, 815 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.26 (s, 9H), 1.86 (d, *J* = 6.7 Hz, 3H), 2.20 (m, 2H), 4.00 (m, 2H), 4.60 (sextet, *J* = 6.7 Hz, 1H), 7.58 (m, 6H), 7.88 (m, 4H); ¹³C NMR (100 MHz, CDCl₃) δ 19.4, 26.8, 27.0, 43.8, 48.0, 61.9, 127.9, 129.8, 133.6, 133.8, 135.6, 135.7; CI-HRMS calcd for C₂₀H₂₈O₂Si Br [M + H] ⁺ (m/z) 391.1092, found 391.1080.



Phophonium salt 205: To a solution of **204** (0.65 mL, 7.2 mmol) in acetonitrile (14.4 mL) was added triphenylphosphine (3.76 g, 14.38 mmol) and the solution was refluxed for 4 h, during which a white precipitate formed. The solid was filtered off and was washed with Et₂O to give **205** as a colorless solid (2.07 g, 72%): IR (neat) 3311, 2890, 2855, 1435, 1120, 746, 695 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.80 (m, 2H), 3.80 (m, 4H), 7.70 (m, 15H); ¹³C NMR (100 MHz, CDCl₃) δ 19.9, 20.4, 25.8, 25.9, 60.2, 60.4, 130.5, 130.6, 133.4, 135.0, 135.1; EI-HRMS calcd for C₂₁H₂₂OP [M - Br] ⁺ (*m*/*z*) 321.1408, found 321.1409.



Phosphonium salt 206: To a THF (8.26 mL) solution of **205** (500 mg, 1.24 mmol) at 0 °C was added *n*-BuLi (1.55 mL, 1.6M in hexane) and the solution was stirred for 1 h at 0 °C. Methyl iodide (84.6 μ L, 1.36 mmol) was added to the reaction mixture at 0 °C which was stirred for 3 h at room temperature. A white precipitate formed which was filtered off and washed with Et₂O to provide **206** (476 mg, 92%) as a yellowish solid: IR (neat) 3355, 1429, 1110, 990, 726, 691, 648 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.12 (m, 1H), 1.44 (dd, *J* = 19.6, 6.7 Hz, 3H), 2.25 (m, 1H), 3.79 (m, 1H), 4.01 (m, 1H), 4.67 (m, 1H), 5.11 (m, 1H), 7.81 (m, 15H); ¹³C NMR (100 MHz, CDCl₃) δ 14.0,

21.8, 22.3, 34.0, 56.8, 56.8, 56.9, 117.4, 118.2, 130.5, 133.7, 133.8, 134.8; EI-HRMS calcd for C₂₂H₂₄OP [M - Br] ⁺ (*m*/*z*) 335.1565, found 335.1522.



5-(Furan-2-yl)-4-methylpent-4-en-1-ol (208)

To a stirred suspension of phosphonium salt 205 (400 mg, 1.00 mmol) in dry THF (10.0 mL) under argon at 0 °C was added dropwise n-BuLi (1.38 mL, 1.6 M, 2.2 mmol). The initial yellow color turned a dark red color after 10 min. After 15 min, methyl iodide (74.7 µL, 1.2 mmol) was added dropwise at 0 °C, producing an immediate white precipitate. After stirring for 15 min at room temperature, n-BuLi (0.70 mL, 1.6 M, 1.1 mmol) was added at 0 °C, again producing a deep red coloration. After 15 min, a solution of 207 (82.8 µL, 1.0 mmol) in dry THF (1 mL) was introduced and the mixture was stirred for 1 h at 0 °C. Water (10 mL) was added and the separated aqueous layer was extracted with Et₂O (3 x 15 mL). The combined organic extract was dried (Na₂SO₄), filtered, concentrated and the residue was purified by silica gel chromatography (hexanes/ethyl acetate, 83:17) to afford an inseparable mixture of E/Z (1.5:1) of **208** (102 mg, 73%) as a colorless oil: $R_f 0.30$ (3:7 hexanesethyl acetate); IR (neat) 3346, 2933, 2879, 1657, 1493, 1151, 1042, 1014, 734 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.94 (s, 3H), 2.04 (s, 3H), 2.08 (s, broad, 2H), 2.43 (t, J = 6.7 Hz, 2H), 2.75 (t, J = 6.7 Hz, 2H), 6.77 (t, J = 6.5 Hz, 2H), 6.83 (t, J = 6.7 Hz,

2H), 6.18 (s, 2H), 6.24 (s, 2H), 6.38 (m, 1H), 6.41 (m, 1H), 7.36 (d, J = 1.4 Hz, 1H), 7.38 (d, J = 1.4 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 18.3, 24.8, 36.8, 43.8, 60.5, 60.9, 107.9, 108.0, 111.0, 116.4, 116.5, 134.3, 134.8, 140.8, 140.9, 152.9, 153.1; EI-HRMS calcd for C₉H₁₂O₂ [M] ⁺ (*m*/*z*) 152.0837, found 152.0836.



Methyl 2-((1*R*,2*R*,3*R*,4*R*)-3-(*tert*-Butyldimethylsilyloxy)-2-(2-(4methoxybenzyloxy)ethyl)-4-(triisopropylsilyloxy)cyclobutyl)-5-ethynylfuran-3carboxylate (211)

To a solution of **197** (10.0 mg, 0.017 mmol) and K₂CO₃ (15.01 mg, 0.108 mmol) in MeOH (0.20 mL) was added **210** (7.54 mg, 0.039 mmol) and the reaction mixture was stirred for 3.5 h at room temperature. The solution was diluted with Et₂O and washed with satd aqueous NaHCO₃. The ethereal layer was dried (Na₂SO₄), filtered and concentrated, and the residue was purified by chromatography on silica gel (hexanes/ethyl acetate 93:7) to give **211** (8.0 mg, 81%) as a colorless oil: R_f 0.65 (2:8 hexanes-ethyl acetate); $[\alpha]^{19}_{D}$ – 59.8 (*c* 0.65, CHCl₃); IR (neat) 3311, 3256, 1960, 2929, 2859, 1719, 1594, 1509, 1462, 1248, 1120, 835, 777, 680 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.11 (s, 6H), 0.90 (m, 30H), 1.70 (m, 1H), 2.00 (m, 1H), 2.60 (m, 1H), 3.41 (s, 1H), 3.50 (m, 2H), 3.75 (s, 3H), 3.81 (s, 3H), 4.22 (dd, *J* = 8.6, 3.2 Hz, 1H),

4.40 (dd, J = 25.3, 11.5 Hz, 2H), 4.52 (dd, J = 8.7, 6.2 Hz, 1H), 4.64 (dd, J = 8.5, 6.0 Hz, 1H), 6.83 (d, J = 8.5 Hz, 2H), 6.88 (s, 1H), 7.19 (d, J = 8.5 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ -4.9, -4.5, 11.9 (3C), 17.6 (6C), 17.7, 25.9 (3C), 29.7, 37.1, 39.5, 51.3, 55.2, 68.1, 72.5, 73.5, 75.2, 76.7, 81.9, 113.6 (2C), 115.5, 116.8, 129.1 (2C), 130.6, 134.5, 159.0, 162.3, 163.4; CI-HRMS calcd for C₃₇H₅₈O₇Si₂Na [M + Na] ⁺ m/z 693.3619, found m/z 693.3626.



Methyl 2-((1*R*,2*R*,3*R*,4*R*)-3-(*tert*-Butyldimethylsilyloxy)-2-(2-(4methoxybenzyloxy)ethyl)-4-(triisopropylsilyloxy)cyclobutyl)-5-(prop-1ynyl)furan-3-carboxylate (212)

To a solution of **211** (14 mg, 0.02 mmol) in THF (0.20 mL) at -78 °C under argon was added dropwise NaHMDS (0.25M in THF, 0.29 mL, 0.07 mmol). The resulting solution was stirred for 1.5 h at -78 °C, MeI (10.0 µL, 0.16 mmol) was added and the reaction mixture was stirred for an additional 1.5 h at -78 °C. The reaction was quenched with H₂O (10 mL) and the separated aqueous layer was extracted with Et₂O (3 x 10 mL). The combined organic layer was washed with brine, dried (Na₂SO₄) and concentrated under reduced pressure. Chromatography of the crude residue on silica gel (ethyl acetate: hexanes 7:93) afforded **212** (12 mg, 84%) as a colorless oil: R_f 0.65 (2:8 hexanes-ethyl acetate); $[\alpha]^{20}_{D}$ – 59.7 (*c* 0.40, CHCl₃); IR (neat) 2941, 2925, 2859, 2630, 2606, 2489, 1723, 1555, 1513, 1462, 1256, 1116, 1057, 832, 777 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.11 (s, 6H), 0.91 (m, 30H), 1.70 (m, 1H), 2.00 (m, 1H), 2.11 (s, 3H), 2.60 (m, 1H), 3.52 (m, 2H), 3.74 (s, 3H), 3.80 (s, 3H), 4.20 (dd, *J* = 8.7, 3.2 Hz, 1H), 4.40 (dd, *J* = 24.9, 11.3 Hz, 2H), 4.50 (dd, *J* = 8.8, 6.1 Hz, 1H), 4.60 (dd, *J* = 8.7, 6.1 Hz, 1H), 6.68 (s, 1H), 6.81 (d, *J* = 8.5 Hz, 2H), 7.19 (d, *J* = 8.5 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ -4.9, -4.5, 4.5, 11.9 (3C), 17.6 (6C), 17.7, 25.8 (3C), 29.3, 37.0, 39.5, 51.1, 55.2, 68.1, 69.8, 72.5, 75.2, 76.5, 97.4, 113.6 (2C), 114.3, 115.4, 129.1 (2C), 130.6, 135.9, 159.0, 161.0, 163.7; CI-HRMS calcd for C₃₈H₆₀O₇Si₂Na [M + Na] ⁺ *m*/z 707.3775, found *m*/z 707.3737.



Methyl 2-((1*R*,2*R*,3*R*,4*R*)-3-(*tert*-Butyldimethylsilyloxy)-2-(2-(4methoxybenzyloxy)ethyl)-4-(triisopropylsilyloxy)cyclobutyl)-5-((*E*)-3-ethoxy-2methyl-3-oxoprop-1-enyl)furan-3-carboxylate (215)

To a solution of **214** (11.0 μ L, 0.05 mmol) in THF (0.1 mL) at 0 °C was added a solution of sodium bis(trimethylsilyl)amide (0.25 mL, 0.06 mmol, 0.25 M in THF). After 15 min, a solution of **197** (17 mg, 0.025 mmol) in THF (0.15 mL) was added dropwise. The reaction mixture was allowed to warm to room temp and was stirred for

1.5 h. The reaction was quenched with a mixture of H₂O and satd NH₄Cl (3:1, 10 mL) and was diluted with Et₂O (20 mL). The layers were separated and the aqueous layer was extracted with Et₂O (2 x 10 mL). The combined organic extract was washed with brine (20 mL), dried (Na₂SO₄), filtered, and concentrated in vacuo. The residue was purified by column chromatography on silica gel (1:9 hexanes/ethyl acetate) to afford **215** (16 mg, 84%) as a colorless oil: $R_f 0.52$ (8:2 hexanes-ethyl acetate); $[\alpha]^{24}_{D} - 111.0$ (c 0.45, CHCl₃); IR (neat) 2960, 2925, 2855, 1723 (2), 1641, 1610, 1583, cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.10 (s, 6H), 0.90 (m, 30H), 1.38 (t, J = 7.2 Hz, 3H), 1.68 (m, 1H), 2.06 (m, 1H), 2.34 (s, 3H), 2.60 (m, 1H), 3.54 (m, 2H), 3.76 (s, 3H), 3.80 (s, 3H), 4.25 (m, 1H), 4.28 (q, J = 7.2 Hz, 2H), 4.40 (dd, J = 11.4, 33.0 Hz, 2H), 4.58 (dd, J = 6.5, 8.4 Hz, 1H), 4.76 (dd, J = 6.3, 8.4 Hz, 1H), 6.81 (s, 1H), 6.85 (d, J = 8.5 Hz, 2H), 7.19 (d, J = 8.5 Hz, 2H), 7.35 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -4.9, -4.6, 11.8 (3C), 14.1, 14.3, 17.6 (6C), 18.0, 25.8 (3C), 29.2, 37.2, 39.2, 39.4, 51.3, 55.3, 60.9, 68.0, 72.5, 75.3, 113.6 (2C), 115.6, 116.8, 124.7, 125.3, 129.1 (2C), 130.5, 150.0, 159.0, 162.8, 163.7, 168.5; EI-HRMS calcd for $C_{41}H_{66}O_9SiNa [M + Na]^+ m/z$ 781.4166, found *m/z* 781.4143.



Methyl 5-((*E*)-3-Ethoxy-2-methyl-3-oxoprop-1-enyl)-2-((1*R*,2*R*,3*R*,4*R*)-3-hydroxy-2-(2-(4-methoxybenzyloxy)ethyl)-4-(triisopropylsilyloxy)cyclobutyl)furan-3carboxylate (216)

To a solution of 215 (16 mg, 0.02 mmol) in EtOH (0.4 mL) was added PPTS (8 mg, 0.03 mmol) and the mixture was heated for 48 h at 65 °C. After the solution had cooled by an ice bath, the reaction was quenched with satd aqueous NaHCO₃. The separated aqueous layer was extracted with Et₂O (2 x 10 mL) and the combined organic extract was dried (Na₂SO₄), filtered, and concentrated. The crude residue was purified by chromatography on silica gel (15:85 ethyl acetate/hexanes) to furnish alcohol **216** (11 mg, 85%) as a colorless oil: $R_f 0.45$ (7:3 hexanes-ethyl acetate); $[\alpha]^{24}_{D}$ - 74.8 (c 0.35, CHCl₃); IR (neat) 3416, 2941, 2914, 2863, 1723, 1699, 1520, 1454, 1244, 1112, 1073, 878 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.90 (m, 21H), 1.39 (t, J = 7.2 Hz, 3H), 1.80 (m, 1H), 2.14 (m, 1H), 2.27 (s, 3H), 2.85 (septet, J = 4.5 Hz, 1H), 3.44 (ddd, J = 12.4, 9.2, 3.2 Hz, 1H), 3.64 (ddd, J = 9.2, 5.3, 3.8 Hz, 1H), 3.81 (s, 3H),3.83 (s, 3H), 4.28 (m, 3H), 4.50 (m, 3H), 4.58 (dd, J = 9.2, 4.7 Hz, 1H), 6.83 (s, 1H), 6.90 (d, J = 8.6 Hz, 2H), 7.24 (d, J = 8.6 Hz, 2H), 7.38 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 11.8 (3C), 14.2, 14.3, 17.6 (6C), 29.4, 29.7, 37.9, 39.7, 51.3, 55.2, 60.9, 68.6, 73.1, 75.0, 76.4, 113.8 (2C), 115.1, 116.8, 124.9, 125.7, 129.4 (2C), 150.0,

159.3, 162.1, 163.9, 168.4; EI-HRMS calcd for $C_{35}H_{52}O_9SiNa$ [M + Na] ⁺ m/z 667.3278, found m/z 667.3276.



Methyl5-((E)-3-Ethoxy-2-methyl-3-oxoprop-1-enyl)-2-((1R,2R,4R)-2-(2-(4-methoxybenzyloxy)ethyl)-3-oxo-4-(triisopropylsilyloxy)cyclobutyl)furan-3-carboxylate (217)

To a solution of **216** (2.0 mg, 0.003 mmol) in CH₂Cl₂ (0.2 mL) was added TPAP (1.2 mg, 0.003 mmol) and the reaction mixture was stirred at room temperature for 2 h. The suspension was filtered through a short pad of silica gel which was washed with CH₂Cl₂ (2 x 10 mL). Evaporation of the filtrate and purification of the crude residue by chromatography on silica gel (14:86 hexanes/ethyl acetate) gave **217** (1.6 mg, 84%) as a colorless oil: R_f 0.62 (3:7 hexanes-ethyl acetate); $[\alpha]^{23}_D$ – 59.6 (*c* 0.32, CHCl₃); IR (neat) 2921, 2855, 1797, 1715, 1696 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.90 (m, 21H), 1.35 (t, *J* = 7.4 Hz, 3H), 2.10 (m, 2H), 2.17 (s, 3H), 3.57 (m, 2H), 3.70 (m, 1H), 3.81 (s, 3H), 3.83 (s, 3H), 4.27 (q, *J* = 7.3 Hz, 2H), 4.40 (d, *J* = 4.0 Hz, 2H), 4.52 (dd, *J* = 6.1, 9.2 Hz, 1H), 5.32 (dd, *J* = 2.9, 9.2 Hz, 1H), 6.84 (m, 3H), 7.18 (d, J = 8.8 Hz, 2H), 7.36 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 11.7 (3C), 14.0, 14.3, 17.4 (6C), 29.4, 29.7, 37.6, 51.4, 55.2, 56.4, 61.0, 67.2, 72.7, 84.0, 113.6 (2C), 114.5, 124.5,

126.9, 128.9, 129.1 (2C), 130.0, 150.7, 159.1, 160.0, 163.6, 168.3, 208.1; EI-HRMS calcd for $C_{35}H_{50}O_9SiNa [M + Na]^+ m/z$ 665.3122, found m/z 665.3112.



Methyl 5-Formyl-2-((1*R*,2*R*,4*R*)-2-(2-(4-methoxybenzyloxy)ethyl)-3-oxo-4-(triisopropylsilyloxy)cyclobutyl)furan-3-carboxylate (219)

A solution of **173** (7.0 mg, 0.017 mmol) in CH₂Cl₂ (0.3 mL) containing activated 4Å molecular sieves, TPAP (0.49 mg, 0.001 mmol) and *N*-methylmorpholine *N*-oxide (3.58 mg, 0.03 mmol) was stirred at room temperature for 1.5 h. The suspension was filtered through a short pad of silica gel which was washed with CH₂Cl₂ (10 mL). Evaporation of the filtrate and purification of the residue by chromatography on silica gel (15:85 hexanes/ethyl acetate) afforded **219** (5.0 mg, 72%) as a colorless oil: R_f 0.65 (6:4 hexanes-ethyl acetate); $[\alpha]^{24}_{D}$ + 5.2 (*c* 0.5, CHCl₃); IR (neat) 2964, 2918, 2859, 1797, 1723, 1684, 1536, 1516, 1443, 1260, 1069, 800 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.90 (m, 21H), 2.1 (m, 2H), 3.57 (t, *J* = 6.1 Hz, 2H), 3.81 (s, 3H), 3.84 (s, 3H), 3.92 (dd, *J* = 7.4, 3.3 Hz, 1H), 4.36 (s, 2H), 4.56 (dd, *J* = 6.9, 8.9 Hz, 1H), 5.30 (dd, *J* = 2.9, 9.1, 1H), 6.85 (d, *J* = 8.6 Hz, 2H), 7.17 (d, *J* = 8.6 Hz, 2H), 7.79 (s, 1H), 9.62 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 11.8 (3C), 17.4 (6C), 29.9, 38.3, 51.8, 55.2, 56.8, 67.2, 72.7, 84.3, 113.8 (2C), 117.9, 120.5, 129.1 (2C), 130.0, 151.1, 159.1,

162.8, 164.0, 177.4, 207.5; EI-HRMS calcd for $C_{30}H_{42}O_8SiNa$ [M + Na] ⁺ m/z 581.2547, found m/z 581.2573.



(*E* and *Z*) Methyl 5-(3-Ethoxy-2-methyl-3-oxoprop-1-enyl)-2-((1R,2R,4R)-2-(2-(4-methoxybenzyloxy)ethyl)-3-oxo-4-(triisopropylsilyloxy)cyclobutyl)furan-3carboxylate (217 and 220)

To a solution of **214** (2.17 µL, 0.01 mmol) in THF (0.1 mL) at 0 °C was added a solution of sodium bis(trimethylsilyl)amide (55 µL, 0.013 mmol, 0.25M in THF). After 15 min, a solution of **219** (3 mg, 0.005 mmol) in THF (0.1 mL) was added dropwise. The reaction mixture was allowed to warm to room temperature and was stirred for 1.5 h. The reaction was quenched with a mixture of H₂O and satd NH₄Cl (3:1, 10 mL) and was diluted with Et₂O (20 mL). The layers were separated and the aqueous layer was extracted with Et₂O (2 x 10 mL). The combined organic extract was washed with brine (20 mL), dried (Na₂SO₄), filtered, and concentrated in vacuo. The crude mixture was purified by chromatography on silica gel (1:9 hexanes/ethyl acetate) to yield an inseparable mixture of (*E*)-isomer **217** and (*Z*)-isomer **220** (2.5 mg, 73%, *E/Z* 3:1 by NMR) as a colorless oil: R_f 0.52 (8:2 hexanes-ethyl acetate); $[\alpha]^{24}_{D}$ –

111.0 (*c* 0.45, CHCl₃); IR (neat) 2914, 2844, 1789, 1711(2), 1633, 1505, 1466 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.90 (m, 35H), 1.38 (t, *J* = 7.1 Hz, 6H), 2.06 (m, 3H), 2.16 (s, 1H), 2.17 (s, 3H), 3.50 (m, 2H), 3.64 (m, 1H), 3.85 (m, 9H), 4.25 (q, *J* = 7.2 Hz, 3H), 4.40 (m, 3H), 4.53 (dd, *J* = 6.0, 9.1 Hz, 1H), 5.32 (dd, *J* = 2.2, 9.1 Hz, 1H), 6.85 (m, 5H), 7.20 (m, 3H), 7.30 (s, 1H), 7.35 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 11.7 (3C), 14.0, 14.3, 17.4(6C), 29.7, 30.4, 37.7, 51.5, 55.2, 56.4, 61.0, 67.2, 72.5, 72.7, 84.0, 113.7 (2C), 114.3, 114.5, 124.5, 126.7, 129.1 (2C), 130.0, 150.7, 159.1, 160.0, 163.7, 208.1; EI-HRMS calcd for C₃₅H₅₀O₉SiNa [M + Na] ⁺ *m*/*z* 665.3122, found *m*/*z* 665.3112.

CHAPTER 4

Second Generation Approach Towards Providencin

4.1 Second generation strategy for providencin (1)

In a second generation approach toward providencin (1), we envisioned that late stage epoxidations of macrocycle 221 would deliver our target molecule (Scheme 4.1). The core structure of providencin in 221 would be generated via an intramolecular aldol reaction of aldehyde 222 which would result from an intermolecular Stille cross-coupling¹ between two major fragments, vinylstannane 223 and furyl triflate 224.



Scheme 4.1 Second generation retrosynthetic analysis of providencin (1)

Furyl triflate **224** would result from butenolide **225** by reaction with a base and triflic anhydride (Scheme 4.2).² A ring-closing metathesis (RCM) of diene **226** would lead to **225** and RCM precursor **226** would be prepared from aldehyde **227**. The latter would be easily available from cyclobutane **190**.



Scheme 4.2 Retrosynthesis of furyl triflate 224

The second component for Stille cross-coupling with **224**, vinyl stannane **223**, would be prepared from the corresponding vinyl iodide **228** by standard chemistry (Scheme 4.3). Following a similar procedure reported by Pattenden and co-workers,³ α -selenolactone **228** would be obtained from hydroxy alkyne **229**, which would be prepared from tosylate **230**. Vinyl iodide **230** would be synthesized from alkyne **231** via Negishi carbometallation-iodination,⁴ and the latter would be acquired from commercially available (*R*)-glycidol (**232**).



Scheme 4.3 Retrosynthesis of vinyl stannane 223

4.2 Preparation of a ring closing metathesis precursor

Our new route started with the preparation of vinyl iodide **235** for alkylation of aldehyde **227**. Propargyl alcohol (**233**) was converted to vinyl iodide in the presence of trimethylsilyl chloride and sodium iodide, and alcohol **234** was protected as its *t*-butyldimethylsilyl ether **235** (Scheme 4.4).⁵



Scheme 4.4 Synthesis of vinyl iodide 235

The vinyl moiety of cyclobutane **190** was oxidatively cleaved by osmium tetroxide and sodium periodate to provide aldehyde **227** in moderate yield.⁶ Unfortunately, all attempts to alkylate aldehyde **227** with vinyl iodide **235** in the

presence of *n*-butyllithium or *t*-butyllithium resulted only in return of the starting aldehyde.⁷ It was surmised that, in the presence of *n*-butyllithium or *t*-butyllithium, the vinyl lithium species **237** was formed but that **237** underwent facile elimination of the *t*-butyldimethylsilyloxy group to give allene which was lost upon evaporative work up (Scheme 4.5).



Scheme 4.5 Attempted synthesis of alcohol 236

The problem with elimination of the lithio derivative of **235** was avoided by treating iodo alcohol **234** with two equivalents of *n*-butyllithium at low temperature and then reacting the dilithio species with aldehyde **227** (Scheme 4.6).⁸ Diol **238** was isolated in moderate yield and as a mixture of diastereomers (2.5:1) from this reaction. After selective protection of the primary alcohol of **238** as its pivaloyl ester **239**,⁹ the secondary alcohol of **239** was esterified with acryloyl chloride⁹ to provide RCM precursor **240**.



Scheme 4.6 Synthesis of ring closing metathesis precursor 240

4.3 Attempted Ring closing metathesis of diene 240

When diene **240** was subjected to ring closing metathesis with Grubbs' first generation catalyst,^{10,11} even in the presence of titanium(IV) isopropoxide,¹² it returned starting material. However, when Grubbs' second generation catalyst (**241**) was used, cinnamate **242** was observed (10% by ¹H NMR) in the reaction mixture (Scheme 4.7) but, no signals for our desired butenolide **243** could be seen in the NMR of the crude mixture. Cinnamate **242** was difficult to separate from starting diene **240**, but the signals due to the phenyl ring as well as the to the *trans*-alkene in the NMR spectra of **242** confirmed its presence. It was found that variation of both the reaction temperature and the solvent gave similar results. It appears that the combination of electron deficiency in the conjugated double bond and steric hindrance due to *gem*-disubstitution in the allylic alkene moiety of **240** contributed to the failure of this ring closing metathesis.



Scheme 4.7 Attempted ring closing metathesis of 240

4.4 Synthesis of vinyl iodide 225

Alongside experiments focused on synthesis of the furylcyclobutane segment of providencin, a route to vinyl iodide **228** was developed starting from (*R*)-glycidol (**232**). The epoxide of **232** was opened by reaction with the lithiated derivative of trimethylsilylacetylene (**244**) to afford diol **245** which was treated with potassium carbonate in methanol to release alkyne **231** (Scheme 4.8).¹³ Negishi carbometallationiodination¹⁴ of alkyne **231** yielded *E*-iodoalkene **246** with a trace amount of the alternate regioisomer. The primary alcohol of **246** was converted selectively to tosylate **230** which was treated with an excess of the lithium salt of ethoxyacetylene (**247**). The product was immediately hydrolyzed using *p*-toluenesulfonic acid as a catalyst to yield lactone **248**. Finally, deprotonation of **248** using lithium hexamethyldisilazide at -78 °C, followed by trapping of the enolate with trimethylsilyl chloride and then quenching of the resultant silyl ether with phenylselenyl bromide gave substituted lactone **228**. Although **228** was formed in low yield and was accompanied by bis-selenelated lactone **249**, these lactones were easily separated by chromatography.



Scheme 4.8 Synthesis of vinyl iodide 228

In view of the mixture of 228 and 249 obtained by selenylation of 248, an alternative route to our desired lactone 228 was devised. First phenylseleno acetic acid (252) was prepared from ethyl bromoacetate (250) in two steps (72% overall yield).¹⁵ Treatment carboxylic equivalents of lithium of acid 252 with two hexamethyldisilazide at -78 °C followed by addition of tosylate 230 yielded hydroxy acid 253 which underwent acid-catalyzed cyclization in hot benzene.¹⁶ This sequence afforded vinyl iodide 228 as a mixture of two diastereomers (1.4:1) in 54% overall yield based on 230 (Scheme 4.9).



Scheme 4.9 Alternative route to vinyl iodide 228

Although our retrosynthetic plan called for conversion of vinyl iodide **228** to its corresponding vinyl stannane **223**, our failure to synthesize its furyl triflate coupling partner **224** caused us to delay this transformation. At this stage of our approach, we realized that we required an improved route to our cyclobutylfuran segment, specifically one that would allow us to use our already prepared vinyl iodide fragment **228**. This new route to the cyclobutylfuran segment of providencin is discussed in Chapter 5.
4.5 References

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4.6 Experimental Section



(1*R*,2*R*,3*R*,4*R*)-3-(*tert*-Butyldimethylsilyloxy)-2-(2-(4-methoxybenzyloxy)ethyl)-4-(triisopropylsilyloxy)cyclobutanecarbaldehyde (227)

To a solution of **190** (118 mg, 0.21 mmol) in dioxane-water (3:1, 2.1 mL) were added 2,6-lutidine (48.6 μ L, 0.42 mmol), OsO₄ (86 mg, 0.008 mmol), and NaIO₄ (179 mg, 0.84 mmol, 2.5% in 2-methyl-2-propanol) an the reaction mixture was stirred at room temperature for 15 h. Water (10 mL) and CH₂Cl₂ (10 mL) were added to the reaction mixture and the organic layer was separated. The aqueous layer was extracted with CH₂Cl₂ (3 x 20 mL) and the combined organic extract was washed with brine, dried over anhydrous Na₂SO₄, and filtered. The solvent was removed under vacuum and the crude residue was purified by silica gel chromatography (1:9 ethyl acetate/hexanes) to afford **227** (65 mg, 60%) as a colorless oil: R_f 0.52 (9:1 hexanes-ethyl acetate); $[\alpha]^{22}$ D – 56.7 (*c* 1.35, CHCl₃); IR (neat) 2937, 2859, 1715, 1606, 1509, 1462, 1240, 1104 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.08 (s, 6H), 0.90 (s, 9H), 1.20 (m, 21H), 1.51 (m, 1H), 2.00 (m, 1H), 2.65 (m, 1H), 3.02 (m, 1H), 3.45 (m, 2H), 3.82 (m, 3H), 4.33 (m, 1H), 4.41 (s, 2H), 4.52 (m, 1H), 6.89 (d, *J* = 8.6 Hz, 2H), 7.24 (d, *J* = 8.6 Hz, 2H), 9.85 (d, *J* = 3.57 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.0, -4.7, 11.8, 12.4 (3C),

17.8(6), 25.8 (3C), 18.0, 32.4, 52.7, 55.7, 55.2, 68.4, 72.6, 75.2, 76.8, 113.7 (2C), 129.1 (2C), 130.4, 159.1; EI-HRMS calcd for $C_{30}H_{54}O_5SiNa$ [M + Na] ⁺ (*m*/*z*) 573.3408, found 573.3401.



1-((1*S*,2*R*,3*R*,4*R*)-3-(*tert*-Butyldimethylsilyloxy)-2-(2-(4-methoxybenzyloxy)ethyl)-4-(triisopropylsilyloxy)cyclobutyl)-2-methylenepropane-1,3-diol (238)

To a solution of **234** (103.2 mg, 0.56 mmol) in 3.35 mL of Et₂O at -78 °C was added slowly 0.2 mL of 2.6 M *n*-butyllithium in hexanes. The solution was quickly warmed to 0 °C and stirred for 2 h at 0 °C. A solution of **227** (103 mg, 0.187 mmol) in 1 mL of Et₂O was added to the reaction mixture which was stirred for 1 h at 0 °C. The mixture was hydrolyzed with MeOH (4 mL) and water (1 mL), the layers were separated and the aqueous phase was extracted with Et₂O (2 x 10 mL). The combined organic extract was dried (anhydrous Na₂SO₄), filtered, and concentrated. Purification of the crude residue by silica gel chromatography (6:4 hexanes-ethyl acetate) afforded **238** (68 mg, 60%, *dr* 2.5:1) as a brown oil: R_f 0.24 (8:2 hexanes-ethyl acetate); $[\alpha]^{19}_{D} - 49.6$ (*c* 0.50, CHCl₃); IR (neat) 3443, 2925, 1516, 1462, 1248, 835 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.05 (s, 3H), 0.07 (s, 3H), 0.91 (s, 9H), 1.10 (m, 21H), 1.50 (m, 1H), 1.93 (m, 1H), 2.29 (m, 1H), 2.36 (m, 1H), 3.40 (m, 2H), 3.82 (s, 3H), 4.04 (s, 2H), 4.30 (dd, J = 6.3, 9.2 Hz, 1H), 4.39 (s, 2H), 4.49 (m, 1H), 4.65 (s, 1H), 5.15 (s, 2H), 6.88 (d, J = 8.4 Hz, 2H), 7.25 (d, J = 8.4 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ -5.0, -4.5, 11.9 (3C), 17.9 (6C), 18.0, 25.8 (3C), 28.7, 32.5, 44.9, 55.3, 63.7, 65.0, 68.9, 70.0, 72.4, 76.3, 111.7, 113.7 (2C), 129.3 (2C), 130.2, 147.9, 159.2; EI-HRMS calcd for C₃₃H₆₀O₆Si₂Na [M + Na] ⁺ (*m*/*z*) 631.3826, found 631.3817.



2-(((1*S*,2*R*,3*R*,4*R*)-3-(*tert*-Butyldimethylsilyloxy)-2-(2-(4methoxybenzyloxy)ethyl)-4(triisopropylsilyloxy)cyclobutyl)(hydroxy)methyl)allyl pivalate (239)

To a solution of **238** (10.0 mg, 0.016 mmol) in dry CH₂Cl₂ (0.2 mL) at 0°C was added pyridine (5.3 µL, 0.065 mmol), trimethylacetyl chloride (8.0 µL, 0.065 mmol) and *N,N*-dimethylaminopyridine (0.4 mg, 0.0032 mmol). The resultant mixture was stirred for 2 h at 0 °C and then quenched with satd aqueous NH₄Cl (10 mL). The organic layer was separated and the aqueous layer was extracted with CH₂Cl₂ (3 x 10 mL). The combined organic extract was washed with brine, dried over anhydrous Na₂SO₄, filtered, and concentrated under reduced pressure. The crude residue was purified by flash column chromatography (silica gel, 15:85 ethyl acetate/hexanes) to provide **239** (10 mg, 90%) as a colorless oil: R_f 0.75 (8:2 hexanes-ethyl acetate); $[\alpha]^{20} _{D} - 42.3$ (*c* 0.80, CHCl₃); IR (neat) 3517, 2960, 2937, 1867, 1731, 1614, 1520, 1466, 1252, 1147, 905, 843 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.05 (s, 3H), 0.07 (s, 3H), 0.90 (s, 9H), 1.09 (m, 21H), 1.21 (s, 9H), 1.50 (m, 1H), 1.92 (m, 1H), 2.30 (m, 2H), 3.34 (m, 2H), 3.82 (s, 3H), 4.30 (dd, *J* = 6.4, 9.2 Hz, 1H), 4.39 (s, 2H), 4.48 (m, 1H), 4.51 (s, 2H), 4.59 (s, 1H), 5.21 (s, 1H), 5.33 (s, 1H), 6.87 (d, *J* = 8.6 Hz, 2H), 7.23 (d, *J* = 8.3 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ -5.0, -4.5, 11.9 (3C), 17.9 (6C), 18.0, 25.8 (3C), 27.1 (3C), 29.1, 29.7, 32.1, 38.8, 43.8, 55.2, 65.3, 68.4, 68.7, 72.5, 76.0, 76.3, 113.5, 113.7 (2C), 129.1 (2C), 130.6, 142.9, 159.0, 178.0; EI-HRMS calcd for C₃₈H₆₉O₇Si₂ [M + H] ⁺ (*m*/*z*) 693.4582, found 693.4604.



1-((1*S*,2*R*,3*R*,4*R*)-3-(*tert*-Butyldimethylsilyloxy)-2-(2-(4-methoxybenzyloxy)ethyl)-4-(triisopropylsilyloxy)cyclobutyl)-2-(pivaloyloxymethyl)allyl acrylate (240)

To a CH₂Cl₂ (0.3 mL) solution of **239** (17 mg, 0.024 mmol) at 0 °C were added triethylamine (10.24 μ L, 0.073 mmol), acryloyl chloride (4.0 μ L, 0.048 mmol), and *N*,*N*-dimethylaminopyridine (0.58 mg, 0.2 mmol). The mixture was stirred at room temperature for 15 h. The reaction was quenched with satd aqueous NH₄Cl solution. The separated aqueous layer was extracted with CH₂Cl₂ (2 x 10 mL) and the combined organic extract was dried (Na₂SO₄), filtered, and concentrated. Purification of the

crude residue by silica gel chromatography (hexanes-ethyl acetate, 94:6) furnished **240** (12.8 mg, 70%) as a colorless oil: $R_f 0.55$ (8.5:1.5 hexanes-ethyl acetate); $[\alpha]^{20}_{D} - 38.0$ (*c* 0.15, CHCl₃); IR (neat) 2935, 2868, 1732, 1512, 1460, 1146, 1105, 840 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.05 (s, 3H), 0.06 (s, 3H), 0.91 (s, 9H), 1.05 (m, 21H), 1.21 (s, 9H), 1.50 (m, 1H), 1.95 (m, 1H), 2.37 (m, 2H), 3.37 (m, 2H), 3.82 (s, 3H), 4.37 (m, 4H), 4.55 (d, J = 13.8 Hz, 1H), 4.60 (d, J = 13.8 Hz, 1H), 5.16 (s, 2H), 5.54 (s, 1H), 5.85 (dd, J = 1.2, 10.3 Hz, 1H), 6.15 (dd, J = 10.4, 17.3 Hz, 1H), 6.40 (dd, J = 1.2, 17.2 Hz, 1H), 6.87 (d, J = 8.6 Hz, 2H), 7.25 (d, J = 8.6 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ -5.0, -4.6, 12.1 (3C), 17.9 (6C), 18.0, 25.8 (3C), 27.2 (3C), 29.2, 29.7, 32.9, 38.8, 42.9, 55.2, 64.7, 68.6, 70.0, 72.5, 74.9, 75.6, 113.4, 113.7 (2C), 128.8, 129.1 (2C), 130.1, 130.4, 142.5, 159.0, 164.1, 178.0; EI-HRMS calcd for C₄₁H₇₀O₈Si₂Na[M + Na] ⁺ (*m*/z) 769.4507, found 769.4487.



(S)-Pent-4-yne-1,2-diol (231)

To a solution of (trimethylsilyl)acetylene (**244**, 2.80 mL, 20.23 mmol) in THF (10 mL) at -78 °C was added dropwise *n*-butyllithium in hexane (2.5M, 8.10 mL, 10.23 mmol). After 30 min, the mixture was added to a stirred solution of (*R*)-glycidol (**232**, 0.89 mL, 13.49 mmol) and *n*-butyllithium in hexane (5.90 mL, 14.83 mmol) in THF (20 mL) at -78 °C. The reaction mixture was allowed to warm to ambient temperature and was stirred for 18 h. The reaction was quenched by addition of satd aqueous NH₄Cl

solution (50 mL), the layers were separated and the aqueous layer was extracted with Et_2O (2 x 20 mL). The combined organic layer was washed with brine, dried (Na₂SO₄) and concentrated. The crude residue (1.72 g) was used directly for the next step.

The crude residue (1.72 g) obtained above was dissolved in MeOH (25 mL) and to the solution was added K₂CO₃ (4.12 g, 29.9 mmol). The reaction mixture was stirred at room temperature for 15 h, then filtered and concentrated. The crude residue was purified by silica gel chromatography (6:4, ethyl acetate/hexanes) to give **231** (0.85 g, 85%) as a colorless sticky oil: R_f 0.25 (3:7 hexanes-ethyl acetate); $[\alpha]^{21}_{D}$ + 6.5 (*c* 1.0, CHCl₃) [lit.^{13b} $[\alpha]^{23}_{D}$ + 7.1 (*c* 1.01, CH₂Cl₂)]; ¹H NMR (300 MHz, CDCl₃) δ 2.08 (t, *J* = 2.6 Hz, 1H), 2.45 (ddd, *J* = 1.2, 2.6, 6.3 Hz, 2H), 3.61 (dd, *J* = 6.5, 11.3 Hz, 1H), 3.77 (dd, *J* = 3.4, 11.3 Hz, 1H), 3.91 (m, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 23.3, 65.3, 70.1, 70.8, 80.1.



(*S*,*E*)-5-Iodo-4-methylpent-4-ene-1,2-diol (246)

To a suspension of Cp_2ZrCl_2 (334.4 mg, 1.14 mmol) in DCE (1.0 mL) at room temperature was added slowly AlMe₃ (0.3 mL, 3.12 mmol) and the mixture was stirred for 15 min. The solution was cooled to 0 °C and a solution of **231** (105.0 mg, 1.04 mmol) in DCE (1.1 mL) was added via cannula. The mixture was stirred for 41 h at room temperature, then was cooled to – 30 °C and a solution of I₂ in Et₂O (3.0 mL) was added. The mixture was stirred for 45 min – 30 °C, then was allowed to warm to 0 °C and the reaction was quenched with satd sodium potassium tertarate solution (30 mL) and pentane (20 mL). The layers were separated and the aqueous layer was extracted with EtOAc (6 x 30 mL). The combined organic extract was dried (Na₂SO₄), filtered and concentrated, and the crude residue was purified by silica gel chromatography (55:45 ethyl acetate/hexanes) to give **246** (154 mg, 60%) as a colorless oil: R_f 0.25 (3:7 hexanes-ethyl acetate); $[\alpha]^{21}_{D}$ – 10.4 (*c* 0.65, CHCl₃); IR (neat) 3361, 2921, 2871, 1271, 1151 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.91 (d, *J* = 0.8 Hz, 3H), 2.39 (dd, *J* = 8.3, 11.1 Hz, 2H), 3.46 (dd, *J* = 10.6, 6.9 Hz, 1H), 3.88 (m, 1H), 6.06 (q, *J* = 0.8 Hz, 1H) ¹³C NMR (100 MHz, CDCl₃) δ 24.2, 43.2, 66.2, 69.6, 77.6, 144.6; EI-HRMS calcd for C₆H₁₁O₈IO₂[M]⁺ (*m*/*z*) 241.9804, found 241.9910.



(*S*,*E*)-2-Hydroxy-5-iodo-4-methylpent-4-enyl 4-methylbenzenesulfonate (230)

To a solution of **246** (2.53 g, 10.45 mmol) in pyridine (25.0 mL) at 0 °C was added *p*tolylsulfonyl chloride (2.19 g, 11.50 mmol) and the mixture was stirred at 4 °C for 20 h. The reaction was quenched with water (100 mL), the layers were separated and the aqueous layer was extracted with EtOAc (4 x 50 mL). The combined organic extract was dried over Na₂SO₄, filtered, and concentrated. The crude mixture was purified by silica gel chromatography (22:88 ethyl acetate/hexanes) to afford **230** (3.18 g, 76%) as a colorless oil: R_f 0.66 (7:3 hexanes/ethyl acetate); $[\alpha]^{18}_{D}$ + 5.8 (*c* 0.85, CHCl₃); IR (neat) 3517, 2910, 2848, 1361, 1170, 1096, 976 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.86 (d, J = 0.8 Hz, 3H), 2.38 (d, J = 6.1 Hz, 2H), 2.48 (s, 3H), 3.92 (dd, J = 5.6, 9.4 Hz, 1H), 4.03 (m, 2H), 6.00 (s, 1H), 7.39 (d, J = 8.06 Hz, 2H), 7.82 (d, J = 7.9 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 21.7, 24.0, 42,7, 67.1, 72.7, 78.4, 128.0 (2C), 130.0 (2C), 132.4, 143.1, 145.2; CI-HRMS calcd for C₁₃H₁₈O₄SI [M + H]⁺ m/z396.9971, found m/z 396.9956.



(*R*,*E*)-5-(3-Iodo-2-methylallyl)dihydrofuran-2(3H)-one (248)

To a stirred solution of ethoxyacetylene (247, 40% solution w/w in hexanes, 78.9 μ L, 0.33 mmol) in THF (2.0 mL) at – 78 °C was added a solution of *n*-BuLi (2.5 M) in hexanes (0.13 mL, 0.32 mmol). The mixture was stirred at – 78 °C for 1 h and a solution of 230 (22.0 mg, 0.05 mmol) in THF (1.0 mL) was added dropwise. The reaction mixture was stirred at – 78 °C for 2 h, then was warmed to 0 °C and BF₃.OEt₂ (27.9 μ L, 0.22 mmol) was added. The mixture was stirred at 0 °C for 1 h, after which the reaction was quenched by addition of a satd aqueous solution of NaHCO₃. The resulting mixture was diluted with Et₂O (10 mL) and water (10 mL), and the separated aqueous phase was extracted with Et₂O (2 x 20 mL). The combined organic extract was dried (Na₂SO₄), filtered, and concentrated to leave a crude oil.

To an EtOH (1.00 mL) solution of the crude oil obtained above was added *p*toluenesulfonic acid (10.0 mg) and the mixture was stirred at room temperature for 18 h. The reaction was quenched with satd aqueous NaHCO₃ and the mixture was diluted with EtOAc (5 mL). The separated aqueous phase was extracted with EtOAc (2 x 10 mL) and the combined organic extract was dried (Na₂SO₄), filtered, and concentrated. Chromatographic purification (silica gel, 25:75 ethyl acetate/hexanes) of the residue gave **248** (8.0 mg, 57%) as a colorless oil: R_f 0.4 (6:4 hexanes-ethyl acetate); $[\alpha]^{20}$ _D -82.8 (*c* 0.85, CHCl₃); IR (neat) 2923, 2851, 1768, 1456, 1174 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.86-1.90 (m, 1H), 1.92 (d, *J* = 1.0 Hz, 3H), 2.28-2.38 (m, 1H), 2.51 (dd, *J* = 8.7, 14.1 Hz, 1H), 2.56 (dd, *J* = 6.9, 9.5 Hz, 2H), 2.66 (dd, *J* = 7.2, 14.3 Hz, 1H), 4.63 (m, 1H), 6.11 (q, *J* = 1.0 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 24.4, 27.6, 28.5, 44.9, 78.3, 78.6, 142.7, 176.6; EI-HRMS calcd for C₈H₁₁O₂I [M] ⁺ (*m*/*z*) 265.9804, found 265.9799.



(S,E)-5-(3-Iodo-2-methylallyl)-3-(phenylselenyl)dihydrofuran-2(3H)-one (228)

To a solution of α -selenoacetic acid (**252**, 4.48 g, 20.74 mmol) in anhydrous THF (20 mL) was added slowly LiHMDS (45.6 mL, 45.65 mmol, 1M solution in THF) at -78 °C. The mixture was stirred at – 78 °C for 1 h and a solution of **230** (3.18 g, 8.01

mmol) in THF (20 mL) was added. The reaction mixture was stirred at -78 $^{\circ}$ C for 30 min and then at 0 $^{\circ}$ C for 1 h. The mixture was warmed to room temperature and was stirred at room temperature for 2 h. To the mixture was added 1N NaOH (85 mL) and the layers were separated. The organic layer was discarded and the aqueous layer was acidified with 1N HCl (125.0 mL) and extracted with Et₂O (3 x 50 mL). The combined organic extract was dried (Na₂SO₄), filtered and concentrated to give a crude oil.

The crude oil obtained above was immediately dissolved in dry benzene (20 mL) and TsOH.H₂O (180 mg, 0.94 mmol, 10 mol%) was added to the solution. The mixture was refluxed for 16 h and the reaction was quenched with satd aqueous NaHCO₃ solution. The aqueous layer was extracted with Et₂O (2 x 30 mL) and the organic extract was dried (Na₂SO₄), filtered and concentrated. Purification of the crude residual oil by silica gel chromatography (15:85 ethyl acetate: hexanes) gave an inseparable mixture (1.4:1) of stereoisomers 228 (1.82 g, 54% based on 230) as a viscous yellow oil: Rf 0.31 (8:2 hexanes/ethyl acetate); IR (neat) 2918, 2847, 1767, 1434 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ major: 1.85 (s, 3H), 2.35 (dd, J = 5.4, 7.5Hz, 2H), 2.45 (dd, J = 5.8, 14.4 Hz, 1H), 2.61 (dd, J = 7.1, 14.2 Hz, 1H), 3.96 (t, J =5.4 Hz, 1H), 4.40 (m, 1H), 6.06 (s, 1H), 7.40 - 7.50 (m, 3H), 7.65 - 7.75 (2H); minor: 1.83 (d, J = 1.0 Hz, 3H), 1.96 (ddd, J = 7.9, 9.1, 13.7 Hz, 1H), 2.29 – 2.36 (m, 1H), 2.48 (dd, J = 7.1, 14.4 Hz, 1H), 2.75 (ddd, J = 6.7, 9.4, 13.7 Hz, 1H), 4.02 (t, J = 9.3Hz, 1H), 4.55 - 4.65 (m, 1H), 6.01 (q, J = 1.0 Hz, 1H), 7.40 - 7.50 (m, 3H), 7.65 - 1007.75 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ major: 24.2, 36.4, 36.7, 44.5, 76.9,

78.7, 126.5, 129.3, 129.5 (2C), 135.9 (2C), 142.3, 175.2; minor: 24.5, 35.2, 36.9,
44.8, 76.9, 78.8, 126.8, 129.1, 129.4 (2C), 135.9 (2C), 142.3, 175.2; CI-HRMS calcd
for C₁₄H₁₆O₂SeI [M + H] ⁺ m/z 422.9361, found m/z 422.9350.

CHAPTER 5

Third Generation Approach Towards Providencin

5.1 Third generation retrosynthetic analysis of providencin

In a variant of our previous strategy, we foresaw that the core structure **221** of providencin could be assembled in principle from two major fragments, vinyl iodide **228** and cyclobutylfuran **254**. We envisioned that the C₆-C₇ bond of **221** could be forged either via palladium-catalyzed C-H activation¹ at C6 of furan **254** or by Stille cross-coupling² between vinyl iodide **228** and a 6-stannyl derivative of **254**. In this scenario, the C₁₂-C₁₃ bond would be acquired by an aldol coupling (Scheme 5.1).



Scheme 5.1 Third generation strategy for providencin (1)

Since synthesis of the α -seleno lactone fragment **228** had already been accomplished in our second generation synthesis, we now focused on synthesis of the

cyclobutylfuran subunit **254**. We envisioned that aldehyde **254** would be available from **255** and that the furan ring of this moiety would be installed from **256** using Marshall's allenone-to-furan isomerization³ protocol (Scheme 5.2). Allenic ketone **256** would be prepared from allenic alcohol **257** which would be obtained from alkynyl bromide **258** and aldehyde **227** using Harada's procedure.⁴ The latter was already in hand from our second generation approach to providencin.



Scheme 5.2 Retrosynthetic analysis of cyclobutylfuran fragment 254

5.2 Synthesis of the cyclobutylfuran fragment of providencin

This third generation approach commenced with the preparation of bromide **258** from propargyl alcohol (**259**). Formation of the tetrahydropyranyl ether of **260** and subsequent deprotonation of alkyne **260** with *n*-butyllithium was followed by reaction with methyl chloroformate to afford ester **261**. Acid-catalyzed cleavage of the

tetrahydropyranyl ether of **261** gave alcohol **262** which was converted to the corresponding bromide **258**.⁵



Scheme 5.3 Synthesis of alkynyl bromide 258

When the organostannane derived from alkyne **258** in the presence of tin(II) chloride and sodium iodide was treated with aldehyde **227**, we were pleased to find that allenol **257** was formed in 62% yield and as a single diastereomer (Scheme 5.4). The configuration at the secondary alcohol center of **257** was not determined due to the fact that it would be oxidized at the next step to a ketone. However, all attempts to oxidize this alcohol to allenone **256** using Dess Martin reagent, tetra-*n*-propylammonium perruthenate or Swern conditions led to a complex mixture of unknown compounds. We speculate that since allenone **256** possesses an allenic moiety conjugated with two different carbonyl groups, it would be a highly reactive Michael acceptor and could lead to a complex mixture of products under these oxidation conditions.



Scheme 5.4 Synthesis of allenol 257

This oxidation problem was solved by using a modified alkynyl bromide 264^6 as the alkylating partner for aldehyde 227. First, propargyl alcohol (259) was protected as its *t*-butyldimethylsilyl ether, after which deprotonation of the alkyne with *n*-butyllithium followed by addition of paraformaldehyde gave alcohol 263. This alcohol was converted to bromide 264 with carbon tetrabromide and triphenylphosphine in good overall yield (Scheme 5.5).



Scheme 5.5 Synthesis of alkynyl bromide 264



Scheme 5.6 Synthesis of cyclobutylfuran 267

Treatment of the organostannane species prepared from **264** with aldehyde **227** afforded a single diastereomer of allenol **265** in 60% yield (Scheme 5.6). Oxidation of **265** with Dess-Martin periodinane smoothly converted the alcohol to allenone **266**, and exposure of this ketone to 10% silver nitrate on silica gel in a mixed solvent system furnished our desired furan **267**. The overall yield of **267** from **227** was 35%.



Scheme 5.7 Attempted synthesis of furylstannane 268

Our initial intention was to convert furan **267** to the stannane derivative **268** so that we could link **268** with vinyl iodide **228** via a Stille cross-coupling experiment. Unfortunately, all efforts to functionalize the C-5 position of furan **267** with either *n*-butyllithium or *t*-butyllithium and tri-*n*-butyltin chloride led to a complex mixture of unknown compounds (Scheme 5.7). We speculated that the *p*-methoxybenzyl protecting group in **267** was interfering with deprotonation of the furan in these experiments and to prove this hypothesis we used the same reaction conditions on model furan **270**.



Scheme 5.8 Rearrangement of furan 270 with *t*-butyllithium

When *p*-methoxybenzyl ether **270** was treated with *t*-butyllithium and tetramethylethylenediamine (TMEDA) at low temperature and the mixture was quenched with deuterated methanol, rearranged product **271** was isolated in 54% yield. The structure of **271** was established by NMR analysis and mass spectrometry (Scheme 5.8). A [1,2]-Wittig rearrangement⁷ of dianion **272** would generate **273** which was trapped by deuterated methanol.



Scheme 5.9 Synthesis of furyl stannane 275 and a model Stille coupling with 276

Since the presence of a *p*-methoxybenzyl group in **270** turned out to be the problematic actor in the deprotonation of this furan, we explored analogous chemistry with *t*-butyldimethylsilyl ether **274**. When furan **274** was allowed to react with *t*-butyllithium and tetramethylethylenediamine (TMEDA) at -78 $^{\circ}$ C and then with trimethyltin chloride, we were pleased to obtain furyl stannane **275** after purification on neutral alumina, albeit in low yield (Scheme 5.9). With furyl stannane **275** in hand, we attempted a model Stille cross-coupling with vinyl iodide **276**. Vinyl iodide **276** was prepared from diol **246** in two steps, first by selective formation of the pivaloyl

ester of the primary alcohol (yield 70%) and then trimethylsilyl protection of the remaining secondary alcohol (yield 80%). Coupling of stannane 275 with 276 in the presence of tris(dibenzylideneacetone)dipalladium(0) and triphenylarsine in hot dimethylformamide produced (E)-alkene 277 in an unoptimized 41% yield, a result that gave confidence to our plan for the final stage assembly of the complete providencin skeleton.



Scheme 5.10 Attempted cleavage of *p*-methoxybenzyl ether 267

However, when **267** was treated with 4,5-dichloro-3,6-dioxocyclohexa-1,4diene-1,2-dicarbonitrile (DDQ) to cleave the *p*-methoxybenzyl ether, a complex mixture of compounds was obtained. Proton NMR of the crude reaction mixture showed the presence of *p*-methoxybenzyl aldehyde, the expected byproduct from oxidation of the *p*-methoxybenzyl group of **267**, along with many unknown compounds. We suspected that the electron-rich disubstituted furan ring of **267**, acting as a potential diene, might undergo Diels-Alder cycloaddition with 4,5-dichloro-3,6dioxocyclohexa-1,4-diene-1,2-dicarbonitrile (DDQ) along with other side reactions, generating a complex mixture. In light of this disappointing result, we adopted a modified route to the cyclobutylfuran portion of providencin. The important modification made in this revision was removal of the *p*-methoxybenzyl ether before furan formation and its replacement by a pivaloyl group.



Scheme 5.11 Synthesis of allenol 282 and alkynol 283

The new route began with removal of the *p*-methoxybenzyl group from **190** with 4,5-dichloro-3,6-dioxocyclohexa-1,4-diene-1,2-dicarbonitrile (DDQ) and subsequent esterification of alcohol **279** with pivaloyl chloride to afford **280** (Scheme 5.11). Oxidative cleavage of the vinyl moiety of **280** then gave aldehyde **281** in moderate yield. Addition of the organostannane, prepared in situ from bromide **264**, to aldehyde **281** at 0 $^{\circ}$ C gave **282** and **283** with the desired allenic alcohol **282** as the major product.



Scheme 5.12 Synthesis of "northern sector" of providencin

After chromatographic purification, allenol **282** was oxidized to allenic ketone **284** with Dess-Martin reagent, and the ketone was treated with silver nitrate in the dark to furnish furan **285** (Scheme 5.12). At this stage, we attempted to install the exo methylene moiety at C15 of providencin and thus **285** was treated with catalytic amount of *p*-toluenesulfonic acid in ethanol to cleave both *t*-butyldimethylsilyl ethers. Monitoring this reaction by thin layer chromatography revealed that the cleavage of the primary *t*-butyldimethylsilyl group of **285** was a relatively fast process and that cleavage of the more hindered secondary silyl ether was very slow at room temperature. Heating the reaction mixture at 60 °C to accelerate silyl ether cleavage led to decomposition. The best result was obtained when **285** was allowed to react with 20 mol% of *p*-toluenesulfonic acid for 48 hours at room temperature to give a 45% yield of diol **286** after which the primary alcohol of **286** was reprotected selectively as its *t*-butyldimethylsilyl ether **287**. Oxidation of secondary alcohol **287**

by tetra-*n*-propylammonium perruthenate to a ketone and Wittig olefination⁸ of this cyclobutanone yielded **288**, thus completing a route to the "northern sector" of providencin. However, the low efficiency (yield 45%) of the deprotection step (**285** \rightarrow **286**) caused us to change the protecting group at C15 in an earlier stage of the synthesis.



Scheme 5.13 A modified route to the cyclobutylfuran segment of providencin

In this revised plan, we intended to install the exo-methylene moiety of the cyclobutane at an early stage of our synthesis after removal of both the *p*-methoxybenzyl and *t*-butyldimethylsilyl groups from **190**. This would deliver diol **289** (Scheme 5.13), the primary alcohol of which would be selectively protected as **290**. Oxidation of the secondary alcohol to a ketone followed by Wittig olefination would then afford diene **292**. Selective oxidative cleavage of the more exposed alkene functionality of **292** would provide aldehyde **293** which would be transformed into the desired cyclobutylfuran **294** following the protocol used earlier.



Scheme 5.14 Preparation of diene 298

In the forward direction, cleavage of the *p*-methoxybenzyl ether⁹ of **190** by 4,5dichloro-3,6-dioxocyclohexa-1,4-diene-1,2-dicarbonitrile (DDQ) to alcohol 295 was followed by acid-catalyzed removal of the *t*-butyldimethylsilyl group to afford diol 289 (Scheme 5.14). The primary alcohol of 289 was selectively protected as its triethylsilyl ether 296, and subsequent oxidation of the remaining secondary alcohol cyclobutanone 297. olefination gave Wittig of 297 with methylenetriphenylphosphorane smoothly installed the exo methylene function in 298 to yield diene 298. However attempts, to react the more exposed vinyl moiety of 298 with either osmium tetraoxide-sodium periodate¹⁰ or with potassium osmate, or with AD-mix- β^{11} returned only starting material. When catalytic osmium tetraoxide and Nmethylmorpholine N-oxide (NMO) were used with 298, tetraol 299 was isolated in 84% yield. The lack of selectivity in osmylation of **298** and consequently our inability to transform diene 298 into aldehyde 293 caused us to abandon this strategy. Instead of installing the exo methylene group at this early stage of our synthesis, it was decided to defer this step. Also, the labile nature of the triethylsilyl protection in **296** made it advisable to replace this entity with a more robust protecting group.



Scheme 5.15 Synthesis of alkynol 303 and allenol 304

In this revised plan, the primary hydroxyl group of diol **289** was selectively converted¹² to its pivalate ester **300** and the secondary alcohol was protected as an acetate¹³ to afford diester **301** (Scheme 5.15). Osmium tetraoxide-sodium periodate¹⁴ mediated cleavage of alkene **301** yielded aldehyde **302** and then the latter was treated with the organostannane species prepared in situ from alkynyl bromide **264**. The desired allenol **304** was obtained as a single diastereoisomer along with alkynol **303** which was separated from **304** by silica gel chromatography.

5.3 Mosher ester analysis of allenol 304

Although the hydroxyl configuration of **304** is inconsequential from the viewpoint of access to cyclobutylfuran **310**, we nevertheless decided to determine the

stereochemistry at the secondary alcohol center of this allenol. In order to accomplish that, the method of Mosher was used.¹⁵ The Mosher method employs esterification of a chiral secondary alcohol with (R) and (S)-2-methoxy-2-trifluoromethylphenylphenylacetyl chloride, analysis of chemical shift differences in NMR spectra of the resultant diastereoisomer provides a reliable means for assigning their absolute configuration.



Figure 5.1 Mosher model for determination of the absolute configuration of a secondary alcohol

Figure 5.1 shows the concept underlying the Mosher method (A), including its extended application (B) using high-field NMR measurements of proton chemical shifts. An idealized conformation of a Mosher ester is depicted in Figure 5.1B (for convenience, the plane and the conformation of the α -methoxy- α -

trifluoromethylphenylacetyl (MTPA) group will be called the MTPA plane and the ideal conformation, respectively). Due to the diamagnetic shielding effect of the phenyl ring, the H_{A,B,C,...} NMR signals of the (*R*)-MTPA ester should appear upfield relative to those of the (*S*)-MTPA ester. The reverse should hold true for H_{X,Y,Z}. Therefore, if $\Delta \delta = \delta S - \delta R$, protons on the right side of the MTPA plane (Figure 5.1B) should have positive values ($\Delta \delta > 0$) and protons on the left side of the plane should have negative values ($\Delta \delta < 0$). This is illustrated in Figure 5.1C. Now, Mosher's method can be extended using the following protocol:

(1) Assign as many proton chemical shifts as possible with respect to each of the (R)-and (S)-MTPA esters.

(2) Obtain $\Delta\delta$ values for those protons.

(3) Put the protons with positive $\Delta\delta$ on the right side and those with negative $\Delta\delta$ on the left side of model A (Figure 5.1C).

(4) Construct a molecular model of the compound in question and confirm that all the assigned protons with positive and negative $\Delta\delta$ values are actually found on the right side and left side of the MTPA plane, respectively. The absolute values of $\Delta\delta$ must be proportional to the distance from the MTPA moiety.



Scheme 5.16 Preparation of (S)- and (R)-MTPA esters of 304



Figure 5.2 Conformations of MTPA esters of 304 with (5R)- and (5S)-configuration

Thus, (*R*)- and (*S*)-MTPA esters were prepared from allenol **304** by reaction with (*R*)- and (*S*)- α -methoxy- α -trifluoromethylphenylacetyl chloride (**305** and **307** respectively) and 4-dimethylaminopyridine (DMAP) (Scheme 5.16). In Figure 5.2, the conformations of MTPA esters with possible (*R*)- and (*S*)-C5 configuration have been depicted. In the case where C5 has (*R*) configuration, the $\Delta\delta = \delta S - \delta R$ values for the cyclobutane substituent would be positive and those for the allene substituent would be negative. On the other hand, if C5 has (*S*) configuration, the $\Delta\delta = \delta S - \delta R$ values for the cyclobutane substituent would be negative and those for the allene substituent would be positive. Comparison of proton NMR spectra of (*S*)-MTPA ester **306** and (*R*)-MTPA ester **308** proved that allenol **304** has (*S*) configuration at the alcohol center.



Chemical	H ₁	H ₂	H ₃	H_4	H ₁₀	H ₁₁	H ₅	H ₈	H ₉
Shift									
(ppm)									
δS	4.50	5.14	2.62	2.67	1.71	3.91	5.52	4.95	4.36
					1.54			4.86	
δR	4.55	5.34	2.65	2.65	1.76	3.93	5.53	4.72	4.33
					1.57			4.38	
$\Delta\delta(\delta S-$	-0.05	-0.20	-0.02	+0.02	-0.05	-0.02	-0.01	+0.23	+0.03
δR)					-0.03			+0.48	

Table 5.1 Measured δ values of protons in the NMR spectra of (*R*)- and (*S*)- MTPA

esters of 304

This conclusion mandates that attack of the stannane nucleophile at the aldehyde carbonyl of **302** takes place from the *si*-face. A molecular model shows that attack at the *re*-face of aldehyde **302** would be hindered by the bulky triisopropylsilyl group whereas approach of the stannane from the *si*-face is opposed only by a hydrogen atom. The much more favorable attack at the aldehyde of **302** from its *si*-face thus leads to (*S*)-allenol **304**, exclusively (Scheme 5.17).



Scheme 5.17 Stereoelectronic explanation for the formation of (S)-allenol 304

5.4 Completion of the synthesis of the "northern sector" of providencin

With pure allenol **304** in hand, oxidation to allenone **309** followed by isomerization to furan **310** in the presence of silver nitrate proceeded in excellent yield (70% over two steps, Scheme 5.18). Exposure of **310** to potassium carbonate in methanol¹⁶ smoothly cleaved the acetate to give free alcohol **287** which was oxidized to cyclobutanone **311** with tetra-*n*-propylammonium perruthenate. Wittig olefination of ketone **311** using standard conditions yielded **288**, the "northern sector" of providencin, along with a minor amount **312** in which the pivalate was cleaved.



Scheme 5.18 Synthesis of the cyclobutylfuran portion of providencin

At this stage, two major fragments required for providencin, iodolactone **228** and cyclobutylfuran **288** had been synthesized. Specifically, lactone **228** had been prepared from (*R*)-glycidol (**232**) in six steps and 15% overall yield while furan **288** was obtained from (D)-glucose (**178**) in 23 steps and 0.43% overall yield (Scheme 5.19). We now turned into our attention to strategies for coupling these two fragments. These are discussed in the following section.



Scheme 5.19 Summary of the preparation of major fragments 228 and 288

5.5 Coupling of vinyl iodide and cyclobutylfuran segments 228 and 288 or 312

5.5.1 Palladium-catalyzed C-H activation

Transition-metal-catalyzed C-H bond functionalization of arenes and hetero arenes is a very powerful tool for forming new C-C bonds.¹⁷ In particular, palladium-catalyzed arylation via C-H bond activation using aryl halides to prepare biaryls has undergone intensive development over the past decade. The Fagnou research group developed a variety of reaction conditions to couple hetero aromatics with aryl halides.¹⁸ For example, when 2,3-disubstituted furan **313** and aryl bromide **314** were treated with a catalytic amount of palladium acetate and tricyclohexylphosphonium tetrafluoroborate in *N*,*N*-dimethyl acetamide (DMA) at elevated temperature (100 °C), C-H activation at the C-5 position of furan **313** took place to give trisubstituted furan **315** in 61% yield (Scheme 5.20).



Scheme 5.20 Fagnou's protocol for arylation of heteroarenes via C-H activation

The Fagnou group also reported that 2-chlorothiophene **316** could be coupled with aryl iodide **317** in the presence of palladium acetate under biphasic conditions at 60 $^{\circ}$ C to yield 2,5-disubstituted thiophene **318** in good yield.



Scheme 5.21 Initial strategy for segment coupling using palladium-catalyzed C-H activation

Inspired by Fagnou's work on C-H activation of hetero aromatics, our initial coupling strategy envisioned linkage of either furan **288** or **312** with vinyl iodide **228** via palladium-catalyzed C-H activation at the C-5 position of the furan. In principle, this would lead to (*E*)-trisubstituted alkene **319** or **320** (Scheme 5.21). Either pivaloyl ester **319** or alcohol **320** would be transformed into aldehyde **321** which would be our candidate for an *intra*molecular aldol coupling to create the macrocycle **322**.

Thus, vinyl iodide 228 and either furan 288 or 312 were treated with a substoichiometric amount of palladium acetate in either a biphasic medium at 60 $^{\circ}$ C or

in *N*,*N*-dimethyl acetamide at 100 °C (Scheme 5.22). Only starting materials remained after 16 hours at 60 °C, and when the reaction mixture was heated for a longer period of time (30 hours) at 60 °C decomposition of the starting materials was observed with no evidence for the formation of either **319** or **320**. The same result was observed when the reaction was run at 100 °C for 16 hours.



Scheme 5.22 Attempted coupling of 228 and 288 or 312 using palladium-catalyzed C-

H activation

Our failure to couple **228** with **288** using Fagnou's protocol prompted us to investigate a related procedure published by the Doucet group.¹⁹ Doucet reported that coupling 2,3-disubstituted furan **323** with aryl bromide **324** in the presence of a substoichiometric amount of palladium acetate and potassium acetate under phosphine-free conditions yielded trisubstituted furan **325** in good yield (Scheme 5.23). He also showed that furan **326** which contains a free hydroxyl group coupled with **324** under the same reaction conditions to produce **327** in 67% yield.


Scheme 5.23 Doucet's procedure for C-H activation and coupling of hetero arenes

When furan **288** was treated with vinyl iodide **228** in the presence of a catalytic amount of palladium acetate and a stoichiometric quantity of potassium acetate in hot dimethylformamide, only cleavage of the *t*-butyldimethylsilyl ether of **288** occurred to give alcohol **328** along with unreacted vinyl iodide **228** (Scheme 5.24). However, when furan **312** containing a free hydroxyl group and **288** were subjected to the Doucet reaction conditions, a Heck coupling²⁰ took place to produce **329**. The structure of **329** was established by extensive 1D, 2D NMR experiments and mass spectroscopy. We surmise that the free hydroxyl group in **312** coordinated with the palladium complex formed from **228** and directed coupling with the exo methylene function of **312** to give conjugated diene **329**.



Scheme 5.24 Synthesis of trisubstituted alkene 329

5.5.2 Aldol coupling of subunits 228 and 330

Alongside experiments directed toward *inter*molecular coupling of **228** with **288** or **312** via palladium-catalyzed C-H activation at the C-5 position of the furan, we also foresaw the possibility of an *intra*molecular coupling of seleno lactone **228** and a cyclobutylfuran unit. In principle, this could be effected if **228** were first linked to aldehyde **330** via an *inter*molecular aldol reaction. This would lead to **331** which could then undergo oxidative elimination of the phenylselenyl subunit to give **332**. *Intra*molecular palladium-catalyzed C-H activation of the furan in **332** at the C-5 position would then furnish the macrocycle **333** (Scheme 5.25).



Scheme 5.25 Alternative strategy for coupling subunits 228 and 330

In direction, reduction of pivaloyl ester the forward 288 using diisobutylaluminum hydride yielded alcohol 312 which was oxidized with tetra-npropylammonium perruthenate to aldehyde 330 (Scheme 5.26). Exposure of the lithium enolate of α -seleno lactone 228 to aldehyde 330 at low temperature gave an aldol product which was immediately treated with hydrogen peroxide. We were pleased to observe that this sequence afforded hydroxy butenolide 332 in acceptable overall yield as a 4:3 diastereomeric mixture at the C-13 alcohol center. However, all attempts to close the macrocycle 333 from 332 using an intramolecular palladiumcatalyzed C-H activation of the furan under either Fagnou's or Doucet's conditions led only to decomposition of 332.



Scheme 5.26 Synthesis of hydroxy butenolide 332

At this stage, we realized that a coupling strategy for our two principal subunits involving either *inter*molecular or *intra*molecular palladium-catalyzed C-H activation at the C-5 position of the furan component using either Fagnou's or Doucet's protocol were destined for failure. On the other hand, we had proved that aldol coupling of the two subunits, lactone **228** and aldehyde **328**, was feasible. It was clear that we needed to modify our coupling strategy by first connecting vinyl iodide **228** to C-5 of furan **288**, for which a transition metal-mediated cross-coupling seemed the best option. An *intra*molecular aldol reaction would then be employed to close the macrocycle **333** along lines demonstrated with the formation of **332**. This strategy is discussed in the following section.

5.5.3 Stille cross-coupling of subunits 228 and 334

In light of our failed coupling of **228** with furans **288** or **312** via C-H activation at the C-5 position of the furan, we decided to explore linkage of the 5-stannyl derivative of furan **312** with vinyl iodide **228** using a Stille cross-coupling.²¹ Trauner's group, in their synthesis of bipinnatin J (**11**), successfully employed Stille crosscoupling of 5-furyl stannane **89** with vinyl iodide **88** to give **90** in high yield using a catalytic amount of tetrakis(triphenylphosphine)palladium, cuprous iodide and cesium fluoride (5.27).²²



Scheme 5.27 Trauner's Stille cross-coupling of vinyl iodide 88 with furyl stannane 89 to give 90

Inspired by Trauner's result, we hypothesized that Stille cross-coupling of vinyl iodide **228** with furyl stannane **334** would afford (*E*)-trisubstituted alkene **320** which could then be oxidized to aldehyde **321**. An *intra*molecular aldol reaction of **321** would then lead to the macrocycle **322** (Scheme 5.28).



Scheme 5.28 Strategy for coupling using a Stille cross-coupling

Initial efforts to prepare furyl stannane **334** from furan **312** were unproductive and all attempts to functionalize the C-5 position of furan **312** with *n*-butyllithium or with *t*-butyllithium and trimethyltin chloride led only to the return of starting material. This suggested that the H-5 proton of furan **312** was so weakly acidic that it could not be abstracted even by a strong base. Conceivably, the acidity of H-5 could be enhanced by an electron-withdrawing group at C-3 of furan **312** and this idea was pursued with furan **288**. Acid-catalyzed hydrolysis of the *t*-butyldimethylsilyl ether of **288** gave alcohol **328** which was oxidized to aldehyde **335** with tetra-*n*propylammonium perruthenate (Scheme 5.29). Pinnick oxidation of **335** to carboxylic acid **336** followed by esterification of **336** provided methyl ester **337**.



Scheme 5.29 Synthesis of ester 337

Although neither lithium diisopropylamide nor lithium diethylamide were able to deprotonate the C-5 position of furan 337, the reaction of s-butyllithium with alcohol 312 was found to be more promising. When furan 312 was treated with an excess of s-butyllithium and freshly distilled tetramethylethylenediamine (TMEDA) in tetrahydrofuran and then with trimethyltin chloride, furyl stannane 334 was obtained in good yield as determined by proton NMR (Scheme 5.30). Attempts at purification of 334 using deactivated silica gel or neutral alumina chromatography led to substantial destannylation, hence the crude furyl stannane was reacted directly with vinyl iodide 228. This Stille cross-coupling, carried out in the presence of tetrakis(triphenylphosphine)palladium and cuprous iodide in degassed dimethylformamide, furnished (E)-trisubstituted alkene 320 in 54% yield based on furan 312.



Scheme 5.30 Synthesis of furyl stannane 334 and Stille coupling with 228

With **320** in hand, our goal was to oxidize this alcohol to aldehyde **321**. An initial attempt to oxidize **320** was made with Dess-Martin reagent under buffered conditions which gave aldehyde **321** in low yield. This oxidation was capricious, however and when the same conditions were repeated on **320**, we invariably obtained the butenolide **338** as the major product with only a trace amount of aldehyde **321**. It was found that oxidation and subsequent elimination of the phenylselenyl oxide from **320** was competing with oxidation of the primary alcohol under Dess-Martin oxidation conditions. In the hope of avoiding overoxidation of **320**, other oxidants were investigated but Swern oxidation produced a complex mixture of compounds, whereas tetra-*n*-propylammonium perruthenate led to a less polar compound that underwent decomposition upon attempted purification. At this stage, it seemed that aldehyde **321** retaining the phenylselenyl substituent would not be available by oxidation of **320**, and

having exhausted the supply of precursors to **320** no further attempt was made to complete the synthesis of providencin from **320**. An attempt to generate the enolate of enone **338** via a Morita-Baylis-Hillman reaction using magnesium iodide, tetramethylethylenediamine (TMEDA) and 4-dimethylaminopyridine to close the macrocycle was unsuccessful.



Scheme 5.31 Oxidation of alcohol 320

A possible solution to difficulties associated with oxidation of **320** to aldehyde **321** is depicted in Scheme 5.32. Since furyl stannane **334** is already in hand, oxidation of this alcohol to aldehyde **339** would open the door to its coupling with α -seleno lactone **228** via an *inter*molecular aldol reaction. This would lead to aldol adduct **340** which would be subjected to an intramolecular Stille coupling to yield macrocycle **322**. Oxidation and elimination of the phenylselenyl substituent of **322** would provide **333**, after which chemoselective and diastereoslective epoxidations, functional group interconversions and final removal of triisopropylsilyl ether would lead to (+)-providencin (1).



Scheme 5.32 A plan for completion of the total synthesis of providencin (1)

In an alternative plan, we envisioned that Stille cross-coupling of a modified furyl stannane **341** with vinyl iodide **342** could provide **343** which could be transformed into hydroxy bromide **344** using standard conditions (Scheme 5.33). Butenolide **345** would be obtained from dibromide **344** via Hoye's protocol using a palladium catalyst and carbon monoxide. The protected primary alcohol of **345** would be converted to aldehyde **346** which would be subjected to an intramolecular NozakiHiyami-Kishi reaction to close the macrocycle **333**. The later would lead to providencin (**1**).



Scheme 5.33 An alternative strategy for the completion of total synthesis of providencin (1)

In summary, a route to advanced intermediates **320** and **332**, which possess all the required carbons for providencin (1), has been developed. The enantiopure cyclobutane moiety of 1 was synthesized using a zirconium-mediated ring contraction of a furanose, while the furan unit was obtained by isomerization of an allenone. Assembly of the two major subunits was accomplished via an *inter*molecular aldol reaction and by means of Stille cross-coupling but all attempts to close the macrocycle nucleus of providencin form these advanced systems failed.

5.6 References

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5.7 Experimental Section



2-(Prop-2-ynyloxy)tetrahydro-2H-pyran (260)

To a CH₂Cl₂ (356 mL) solution of propargyl alcohol (**259**, 10.0 g, 178.3 mmol) at 0 °C were added dihydropyran (19.4 mL, 214.0 mmol) and *p*-toluenesulfonic acid (1.69 g, 8.91 mmol) and the mixture was stirred for 1.5 h at 0 °C. A satd solution of NaHCO₃ was added slowly to the mixture which was stirred until gas evolution ceased. The separated aqueous layer was extracted with CH₂Cl₂ (2 x 50 mL) and the combined organic extract was dried over Na₂SO₄. After filtering, the organic phase was concentrated to leave a crude oil that was purified by distillation (head temperature 73 °C at 10 mmHg) to afford **260** as a colorless oil (21.2 g, 88%): R_f 0.50 (8:2 hexanes/ethyl acetate); ¹H NMR (400 MHz, CDCl₃) δ 1.50-1.90 (m, 6H), 2.43 (t, *J* = 2.4 Hz, 1H), 3.53-3.57 (m, 1H), 3.83-3.88 (m, 1H), 4.28 (dd, *J* = 11.2, 2.3 Hz, 2H), 4.84 (t, *J* = 3.3 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 18.9, 20.3, 30.2, 53.9, 61.9, 73.9, 79.7, 96.8.



Methyl 4-(Tetrahydro-2H-pyran-2-yloxy)but-2-ynoate (261)

To a solution of **260** (2.052 g, 14.64 mmol) in THF (55.0 mL) was added *n*-BuLi (6.77 mL, 2.6 M solution in hexane) at -78 °C and the reaction mixture was allowed to warm to -20 °C slowly (15 min) and then was cooled to -78 °C. A solution of methyl chloroformate (2.26 mL, 29.28 mmol) in THF (45 mL) was added slowly to the reaction mixture and the solution was stirred for 15 min at -78 °C and then warmed to 0 °C. After stirring the solution for 1.5 h at 0 °C, the reaction was quenched with a satd solution of NH₄Cl and the separated aqueous layer was extracted with Et₂O (2 x 25 mL). The combined organic extract was dried over Na₂SO₄ (anhydrous) and concentrated, and the crude residue was distilled under reduced pressure (head temperature 140 °C, 12 mmHg) to give **261** as a colorless oil (2.29 g, 79%): R_f 0.50 (8:2 hexanes/ethyl acetate); IR (neat) 2239, 1713 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) 1.52-1.93 (m, 6H), 3.54-3.59 (m, 1H), 3.80 (s, 3H), 3.81-3.86 (m, 1H), 4.40 (s, 2H), 4.82 (t, *J* = 3.8 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 18.7, 25.8, 30.0, 52.7, 53.6, 61.9, 77.3, 83.9, 97.1, 153.6.



Methyl 4-Hydroxybut-2-ynoate (262)

To a solution **261** (303 mg, 1.52 mmol) in 95% ethanol (16 mL) was added pyridinium *p*-toluenesulfonate (28.0 mg, 0.15 mmol) and the solution was refluxed at 80 °C for 2 h. The mixture was poured into a satd solution of NaHCO₃ (20 mL) and the separated aqueous layer was extracted with Et₂O (3 x 20 mL). The combined organic extract was dried over anhydrous Na₂SO₄, filtered, and concentrated and the crude residue was purified by chromatography (silica gel, 2:8 ethyl acetate/hexanes) to afford **262** (88 mg, 50%) as a colorless oil: $R_f 0.37$ (7:3 hexanes/ethyl acetate); IR (neat) 3408, 2233, 1715 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 3.81 (s, 3H), 4.42 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 50.6, 52.9, 77.3, 85.6, 153.7.



Methyl 4-Bromobut-2-ynoate (258)

To a CH_2Cl_2 (30.0 mL) solution of **262** (1.658 g, 14.5 mmol) At 0 °C were added triethylamine (5.0 mL, 36.3 mL) and mesyl chloride (2.8 mL, 36.5 mmol). The reaction mixture was stirred for 3 h at 0 °C and the reaction was quenched with aqueous satd NH₄Cl solution. The separated aqueous layer was extracted with CH_2Cl_2 and the combined organic layer was dried (anhydrous Na₂SO₄), filtered, and concentrated. The crude residue was used for next reaction without further purification.

To an ice-cold solution of the crude oil obtained above (1.002 g, 5.21 mmol) in THF (14.0 mL) was added LiBr (0.906 g, 10.43 mmol) and the mixture was allowed to stir for 10 h at room temperature. Water (20 mL) was added to the reaction mixture and the separated aqueous layer was extracted with Et_2O (3 x 15 mL). The combined organic extract was dried over anhydrous Na_2SO_4 , filtered, and concentrated under

reduced pressure. The crude residue was chromatographed (silica gel, 7:93 ethyl acetate/hexanes) to give **258**^{5b} (609 mg, 67%) as a colorless oil: R_f 0.67 (7:3 hexanes/ethyl acetate); ¹H NMR (400 MHz, CDCl₃) δ 3.82 (s, 3H), 4.97 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 11.5, 52.9, 77.3, 81.7, 153.4. CI-HRMS calcd for C₅H₆O₂Br [M + H] ⁺ *m/z* 176.9551, found *m/z* 176.9543.



Methyl 2-(((1*S*,2*R*,3*R*,4*R*)-3-(*tert*-Butyldimethylsilyloxy)-2-(2-(4methoxybenzyloxy)ethyl)-

4(triisopropylsilyloxy)cyclobutyl)(hydroxy)methyl)buta-2,3-dienoate (257)

To a solution of **258** (40.8 mg, 0.23 mmol) in 1.1 mL of DMPU at room temperature under argon were added simultaneously SnCl₂ (53.0 mg, 0.27 mmol) and NaI (40.4 mg, 0.27 mmol) and the resulting mixture was stirred for 2.5 h while protected from light. The bright yellow solution was cooled to 0 °C and a solution of **227** (32 mg, 0.28 mmol) in DMPU (1.0 mL) was added over a 10 min period. After 20 h at room temperature in the absence of light, the reaction mixture was diluted with Et₂O (10.0 mL) and a satd aqueous solution of NH₄Cl (10 mL) with efficient stirring. The separated aqueous layer was extracted with Et₂O (3 x 10 mL) and the combined organic extract was washed satd aqueous NaHCO₃ and brine, dried over anhydrous Na₂SO₄, filtered, and concentrated. Flash chromatographic purification of the crude residue on silica gel (9:91 ethyl acetate:hexanes) afforded **257** (23 mg, 62%) as a light yellow oil: R_f 0.42 (8:2 hexanes/ethyl acetate); $[\alpha]^{29}_{D} - 13.6$ (*c* 1.1, CHCl₃); IR (neat) 3527, 2941, 2857, 1960, 1711, 1608, 1517, 1253, 1085 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.05 (s, 6H), 0.90 (s, 9H), 1.11 (s, 21H), 1.41-1.50 (m, 1H), 1.89-1.96 (m, 1H), 2.35-2.40 (m, 1H), 2.49-2.51 (m, 1H), 3.33-3.37 (m, 2H), 3.38 (d, *J* = 2.4 Hz, OH), 3.74 (s, 3H), 3.82 (s, 3H), 4.29 (dd, *J* = 5.8, 9.0 Hz, 1H), 4.36 (d, *J* = 1.4 Hz, 2H), 4.47 (dd, *J* = 6.1, 9.1 Hz, 1H), 4.91 (t, *J* = 2.5 Hz, 1H), 5.24 (dd, *J* = 2.5, 14.1 Hz, 1H), 5.38 (dd, *J* = 3.0, 14.1 Hz, 1H), 6.88 (d, *J* = 8.6 Hz, 2H), 7.26 (d, *J* = 8.6 Hz, 2H) ; ¹³C NMR (100 MHz, CDCl₃) δ -5.1, -4.5, 11.9, 17.8, 25.8, 29.0, 29.7, 31.7, 44.4, 52.1, 55.2, 65.9, 68.4, 72.4, 76.1, 76.3, 81.2, 101.6, 113.7, 129.1, 130.7, 159.0, 166.2, 214.1; CI-HRMS calcd for C₃₅H₆₀O₇Si₂Na [M + Na] ⁺ *m*/z 671.3775, found *m*/z 671.3766.



4-(*tert*-Butyldimethylsilyloxy)but-2-yn-1-ol (263)

To a DMF (45 mL) solution of propargyl alcohol (**259**, 2.1 mL, 35.6 mmol) were added imidazole (3.64 g, 53.5 mmol) and TBSCl (6.45 g, 42.8 mmol) at room temperature. The reaction mixture was stirred at room temperature for 16 h and was diluted with water (100 mL) and Et₂O (50 mL). The layers were separated and the aqueous layer was extracted with Et₂O (3 x 25 mL). The combined organic extract was

dried over anhydrous Na_2SO_4 , filtered, and concentrated to leave a crude oil (6.36 g) which was used directly for the next reaction.

To a THF (64.2 mL) solution of the crude oil (6.56 g, 38.5 mmol) obtained above at -78 °C was added slowly *n*-BuLi (22.3 mL, 2.6 M in hexane) and the reaction mixture was stirred for 20 min at – 78 °C. A solution of paraformaldehyde (2.31 g, 77.0 mmol) in THF (10.0 mL) was added to the mixture at – 78 °C and the solution was warmed to room temperature and stirred for 3 h. Aqueous NH₄Cl solution (25 mL) was added to the reaction mixture and the layers were separated. The aqueous layer was extracted with Et₂O (2 x 25 mL) and the combined organic extract was dried (Na₂SO₄), filtered, and concentrated. The crude residue was purified by silica gel chromatography (1:9 ethyl acetate/hexane) to afford **263** (6.32 g, 80%, 2 steps) as a colorless oil: R_f 0.22 (9:1 hexanes/ethyl acetate); IR (neat), 3369, 2956, 2929, 2885, 1466, 1256, 1139 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.14 (s, 6H), 0.93 (s, 9H), 1.62 (t, *J* = 5.6 Hz, OH), 4.32 (td, *J* = 1.5, 5.9 Hz, 2H), 4.37 (t, *J* = 1.7 Hz, 2H); δ ¹³C NMR (100 MHz, CDCl₃) δ (ppm) -5.1, 18.3, 25.8, 51.2, 51.7, 82.9, 84.5.



(4-Bromobut-2-ynyloxy)(tert-butyl)dimethylsilane (264)

To a solution of **263** (1.00 g, 4.99 mmol) in 9.98 mL of CH_2Cl_2 was added triphenylphosphine (1.74 g, 6.63 mmol) and the mixture was cooled in an ice bath. To this solution at 0 °C was added portionwise CBr_4 (2.19 g, 6.63 mmol) over 10 min and

the brown reaction mixture was stirred for 16 h at room temperature. The solvent was removed in vacuo and the crude residue was purified by silica gel chromatography (5:95 ethyl acetate/hexanes) to provide **264** (819 mg, 63%) as a colorless oil: $R_f 0.8$ (85:15 hexanes/ethyl acetate); IR (neat), 2960, 2929, 2863 1474, 1365, 1260, 1151 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.14 (s, 6H), 0.93 (s, 9H), 3.96 (t, *J* = 1.9 Hz, 2H), 4.39 (t, *J* = 2.0 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ -5.1, 14.4, 18.3, 25.8, 51.8, 79.7, 85.5.



1-((1*S*,2*R*,3*R*,4*R*)-3-(*tert*-Butyldimethylsilyloxy)-2-(2-(4-methoxybenzyloxy)ethyl)-4-(triisopropylsilyloxy)cyclobutyl)-2-((*tert*-butyldimethylsilyloxy)methyl)buta-2,3dien-1-ol (265)

To a solution of **264** (28.5 mg, 0.10 mmol) in DMPU (1 mL) at room temperature under argon were added $SnCl_2$ (24.6 mg, 0.12 mmol) and NaI (19.3 mg, 0.12 mmol), and the resulting mixture was stirred for 2 h in a flask wrapped with aluminum foil to protect the contents from light. The foil was removed, the bright yellow solution was cooled to 0 °C, and a solution of **227** (15 mg, 0.027 mmol) in DMPU (0.5 mL) was added over a 10 min period. After 20 h at room temperature in the absence of light, the reaction mixture was diluted with Et_2O and a satd aqueous solution of NH₄Cl was

added with efficient stirring. The separated aqueous layer was extracted with Et₂O (3 x 10 mL) and the combined organic extract was washed with satd aqueous NaHCO₃ and brine, dried over Na₂SO₄, filtered, and concentrated. Flash chromatography of the crude residue on silica gel (9:91 ethyl acetate:hexanes) afforded 265 (12.0 mg, 60%) as a colorless oil: $R_f 0.50$ (8:2 hexanes/ethyl acetate); $[\alpha]^{26} - 25.1$ (c 0.35, CHCl₃); IR (neat) 3536, 2937, 2859, 1964, 1727, 1610, 1509, 1458, 1248, 1096, 1069 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.04 (s, 3H), 0.05 (s, 3H), 0.07 (s, 6H), 0.90 (s, 18H), 1.07-1.10 (m, 21H), 1.54-1.57 (m, 1H), 1.93-2.02 (m, 1H), 2.32-2.44 (m, 2H), 3.18 (d, J = 2.6 Hz, OH), 3.37-3.48 (m, 2H), 3.82 (s, 3H), 4.13 (td, J = 2.3, 11.6 Hz, 1H), 4.22 (td, J = 2.1, 11.8 Hz, 1H), 4.32 (dd, J = 5.9, 8.5 Hz, 1H), 4.42 (d, J = 1.4 Hz, 2H), 4.45 (m, 1H), 4.65 (t, J = 2.6 Hz, 1H), 4.84 (qd, J = 2.3, 10.3 Hz, 4.93 (qd, J = 2.3, 10.5 Hz, 1H), 6.88 (d, J = 8.7 Hz, 2H), 7.26 (d, J = 8.5 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ -5.3, -5.2, -5.0, -4.5, 11.9, 17.9, 18.2, 25.8, 29.4, 29.7, 32.1, 44.4, 55.2, 63.0, 66.4, 72.5, 76.0, 77.5, 104.4, 113.7, 129.1, 130.7, 159.0, 205.8; CI-HRMS calcd for $C_{40}H_{74}O_6Si_3Na [M + Na]^+ m/z$, 757.4691, found m/z, 757.4681.



1-((1*R*,2*R*,3*R*,4*R*)-3-(*tert*-Butyldimethylsilyloxy)-2-(2-(4-methoxybenzyloxy)ethyl)-4-(triisopropylsilyloxy)cyclobutyl)-2-((*tert*-butyldimethylsilyloxy)methyl)buta-2,3dien-1-one (266) To a solution of 265 (7.0 mg, 0.009 mmol) in 0.2 mL of CH₂Cl₂ at 0 °C was added NaHCO₃ (3.2 mg, 0.038 mmol) and Dess-Martin periodinane (8.0 mg, 0.019 mmol). The slurry was stirred for 2.5 h at 0 °C and the reaction mixture was diluted with CH_2Cl_2 (1.0 mL) and satd $Na_2S_2O_3$ solution (10 mL). After stirring the biphasic mixture vigorously for 20 min, the layers were separated, and the aqueous phase was extracted with CH₂Cl₂ (2 x 10 mL). The combined organic extract was washed with satd NaHCO₃, water, and brine and dried over anhydrous Na₂SO₄. The drying agent was removed by filtration and the filtrate was concentrated in vacuo. The crude mixture was purified by silica gel chromatography (9:1 pentane/ethyl acetate) to yield **266** (6 mg, 87%) as a colorless oil: $R_f 0.50$ (9:1 hexanes/ethyl acetate); $[\alpha]^{26} - 39.4$ (c 1.5, CHCl₃); IR (neat), 2925, 2859, 1933, 1731, 1664, 1474, 1244, 1108, 843 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.06 (s, 12H), 0.90 (s, 18H), 1.03 (s, 21H), 1.47-1.55 (m, 1H), 1.91-1.96 (m, 1H), 2.43 (m, 1H), 3.43 (m, 2H), 3.82 (s, 3H), 3.85 (dd, J =1.8, 8.9 Hz, 1H), 4.29 (td, J = 3.6, 13.5 Hz, 1H), 4.36-4.42 (m, 3H), 4.46 (td, J = 3.4, 13.5 Hz, 1H), 4.52 (dd, J = 6.8, 8.2 Hz, 1H), 5.16 (ddd, J = 3.2, 4.2, 7.2 Hz, 2H), 6.88 (d, J = 8.7 Hz, 2H), 7.23 (d, J = 8.7 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ -5.3 (2), -5.0, -4.5, 12.1, 17.9, 18.3, 25.8 (2), 28.5, 19.7, 35.2, 48.7, 55.2, 58.4, 68.8, 72.6, 74.9, 76.2, 81.9, 112.2, 113.7, 129.1, 130.6, 159.1, 199.5, 216.2; CI-HRMS calcd for $C_{40}H_{72}O_6Si_3$ [M] + m/z 733.4715, found m/z 733.4733.



tert-Butyl((2-((1*R*,2*R*,3*R*,4*R*)-3-(*tert*-butyldimethylsilyloxy)-2-(2-(4methoxybenzyloxy)ethyl)-4-(triisopropylsilyloxy)cyclobutyl)furan-3vl)methoxy)dimethylsilane (267)

To a solution of **266** (30.0 mg, 0.04 mmol) in 0.5 mL of hexane and 0.1 mL of CH₂Cl₂ at room temperature was added 34.0 g of AgNO₃ on SiO₂ (10 wt%, 0.02 mmol) in a single portion. The suspension was stirred at room temperature in the absence of light for 2 h and the mixture was diluted with Et₂O (5 mL) and filtered through a short pad of Celite which was washed with Et₂O (10 mL). The filtrate was concentrated in vacuo and the crude oil was purified by silica gel chromatography (96/4, hexanes/ethyl acetate) to give **267** (20 mg, 66%) as a pale yellow oil: R_f 0.76 (9:1 hexanes/ethyl acetate); $[\alpha]^{26}$ D – 25.8 (*c* 0.35, CHCl₃); IR (neat) 2912, 2851, 1738, 1618, 1505, 1458, 1248 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.07 (s, 3H), 0.08 (s, 3H), 0.10 (s, 3H), 0.11 (s, 3H), 0.9-1.0 (m, 39H), 1.60-1.72 (m, 1H), 2.02-2.06 (m, 1H), 2.41-2.48 (m, 1H), 3.29 (dd, *J* = 2.6, 8.7 Hz, 1H), 3.46-3.50 (m, 2H), 3.81 (s, 3H), 4.41 (s, 2H), 4.43-4.47 (m, 3H), 4.63 (dd, *J* = 6.6, 8.7 Hz, 1H), 6.32 (d, *J* = 1.7 Hz, 1H), 6.86 (d, *J* = 8.5 Hz, 2H), 7.23 (d, *J* = 8.5 Hz, 2H), 7.31 (d, *J* = 1.7 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.2 (2C), -4.9, -4.6, 11.9, 14.1, 17.7 (2C), 18.0, 18.4, 22.7, 25.9 (2C), 29.7, 30.3, 37.4,

38.7, 55.2, 57.3, 68.9, 72.5, 75.3, 110.6, 113.7, 121.0, 129.1, 130.6, 140.5, 19.4, 159.0; CI-HRMS calcd for C₄₀H₇₂O₆Si₃ [M] ⁺ *m/z* 732.4637, found *m/z* 732.4606.



2-((4-Methoxybenzyloxy)methyl)furan (270)

To a cold (0 °C) suspension of sodium hydride (0.61 g, 15.3 mmol) in dry THF (10.2 mL) was added a solution of **269** (0.44 mL, 30.49 mmol) in dry THF (3 mL) and the mixture was stirred for 30 min at 0 °C and for 20 min at room temperature. To this mixture at 0 °C was added a solution of *p*-methoxybenzyl chloride (1.0 mL, 7.63 mmol) in THF (5 mL) and the reaction mixture was stirred at room temperature for 10 h. The reaction was quenched with satd ammonium chloride solution (30 mL) and the separated aqueous layer was extracted with Et₂O (2 x 30 mL). The combined organic extract was dried (Na₂SO₄), filtered, and concentrated and the crude residue was chromatographed on silica gel (4:1 hexanes/ethyl acetate) to furnish **270** (0.87 g, 80%) as a colorless oil: R_f 0.46 (1:1 hexanes/ethyl acetate); ¹H NMR (400 MHz, CDCl₃) δ 3.83 (s, 3H), 4.48 (s, 2H), 4.51 (s, 2H), 6.34-6.38 (m, 2H), 6.90 (d, *J* = 8.5 Hz, 2H), 7.30 (d, *J* = 8.5 Hz, 2H), 7.44 (d, *J* = 0.8 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 55.2, 63.5, 71.5, 109.2, 110.2, 113.8, 129.5, 130.0, 142.7, 151.9, 159.3.



2-((5'-Deutero)furan-2-yl)-1-phenylethanol (271)

To a stirred THF (1.6 mL) solution of 270 (35.0 mg, 0.164 mmol) at -78 °C were added slowly t-BuLi (0.2 mL, 0.33 mmol, 1.73 M solution in hexane) and freshly distilled TMEDA (0.1 mL, 0.65 mmol). The solution turned yellow and was allowed to warm to room temperature and was stirred for 4 h during which the color changed yellow to brown. The mixture was cooled to -78 °C and the reaction was quenched with CD₃OD (1.0 mL). The mixture was poured into water (5 mL), the layers were separated and the aqueous layer was extracted with Et₂O (2 x 5 mL). The combined organic extract was washed with brine, dried (Na₂SO₄), filtered, and concentrated. The crude residue was purified by silica gel chromatography (95:5 hexanes/ethyl acetate) to afford 271 (19 mg, 54%) as a colorless oil: Rf 0.25 (9:1 hexanes-ethyl acetate); IR (neat) 3424, 2921, 2836, 1618, 1516 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 3.06 (dd, J = 4.4, 7.8 Hz, 1H), 3.82 (s, 3H), 4.97 (dd, J = 5.8, 7.6 Hz, 1H), 6.08 (d, J = 3.1 Hz, 1H), 6.31 (d, J = 3.1, 1H), 6.80 (d, J = 8.7 Hz, 2H), 7.30 (d, J = 8.7 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 38.2, 55.2, 72.5, 107.2, 110.1, 113.8, 126.9, 135.5, 152.4, 159.1; EI-HRMS calcd for $C_{13}H_{13}O_{3}D [M]^{+} m/z$ 219.1006, found m/z 219.1010.



tert-Butyl(furan-2-ylmethoxy)dimethylsilane (274)

To a stirred solution of **269** (0.44 mL, 5.09 mmol) in dry DMF (10.0 mL) at room temperature were added imidazole (0.46 g, 7.12 mmol) and TBSCl (0.84 g, 5.59 mmol). The reaction mixture was stirred at room temperature for 3 h and was poured

into water (25 mL). The separated aqueous layer was extracted with Et₂O (2 x 50 mL) and the combined organic extract was dried over Na₂SO₄, filtered, and concentrated. Purification of the crude oil by silica gel chromatography gave **274** (0.7 g, 65%) as a colorless oil: R_f 0.82 (9:1 hexanes-ethyl acetate); ¹H NMR (400 MHz, CDCl₃) δ 0.10 (s, 6H), 0.92 (s, 9H), 4.66 (s, 2H), 6.24 (d, *J* = 3.2 Hz, 1H), 6.33 (dd, *J* = 1.8, 3.1 Hz, 1H), 7.39 (dd, *J* = 0.8, 1.6 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.2 (2C), 18.4, 25.8(3C), 58.1, 107.2, 110.1, 142.0, 154.3.



(S,E)-5-Iodo-4-methyl-2-(triethylsilyloxy)pent-4-enyl Pivalate (276)

To a CH_2Cl_2 solution (0.3 mL) of **246** (18 mg, 0.074 mmol) at 0 °C was added slowly pyridine (59.7 µL, 0.74 mmol) and trimethylacetyl chloride (10.0 µL, 0.081 mmol). The reaction mixture was allowed to warm to room temperature and was stirred for 18 h. The reaction was quenched by addition of a saturated aqueous NaHCO₃ solution, the separated aqueous layer was extracted with CH_2Cl_2 (2 x 10 mL) and the combined organic extract was dried (Na₂SO₄), filtered, and concentrated. The crude mixture was used for the next reaction without further purification.

To a solution of the crude oil (19 mg, 0.058 mmol) obtained above in DMF (0.5 mL) at room temperature were added imidazole (11.9 mg, 0.17 mmol) and TESC1 (28.46 μ L, 0.17 mmol). The reaction mixture was stirred at room temperature for 15 h,

when MeOH (3 mL) was added. The solution was stirred for 10 min, then was poured into a mixture of H₂O (3 mL) and Et₂O (10 mL). The separated aqueous layer was extracted with Et₂O (2 x 20 mL) and the combined organic extract was dried (Na₂SO₄), filtered and concentrated. The crude mixture was purified by silica gel chromatography (2:98 ethyl acetate/hexanes) to give **276** (20 mg, 56% 2 steps) as a colorless oil: R_f 0.60 (95:5 hexanes/ethyl acetate); $[\alpha]^{20}_{D}$ - 5.5 (*c* 0.80, CHCl₃); IR (neat) 2960, 2877, 1729, 1475, 1453, 1281, 1150 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.61 (q, *J* = 7.8 Hz, 6H), 0.97 (t, *J* = 7.9 Hz, 9H), 1.24 (s, 9H), 1.88 (s, 3H), 2.40 (dd, *J* = 4.7, 7.7 Hz, 2H), 3.95 (m, 3H), 6.00 (s, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 4.8 (3C), 6.8 (3C), 24.5, 27.2 (3C), 38.7, 44.7, 67.4, 68.7, 77.9, 143.9, 178.3. EI-HRMS calcd for C₁₇H₃₄O₃SiI [M + H] ⁺ (*m/z*) 441.1322, found 441.1329.



tert-Butyldimethyl((5-(trimethyllstannyl)furan-2-yl)methoxy)silane (275)

To an Et₂O (1.5 mL) solution of **274** (56.0 mg, 0.26 mmol) and TMEDA (0.15 mL, 1.04 mmol) at 0 $^{\circ}$ C was added dropwise *n*-BuLi (0.3 mL, 0.52 mmol, 1.73M in pentane). The solution turned brown and was warmed to room temperature. After stirring at room temperature for 1 h, the reaction mixture was cooled to 0 $^{\circ}$ C and Me₃SnCl (1.28 mL, 1.3 mmol, 1M in THF) was added. The mixture was warmed to room temperature and was stirred for overnight. Water (10.0 mL) was added to the reaction mixture, the layers were separated and the aqueous layer was extracted with

Et₂O (3 x 10 mL). The combined organic extract was dried over anhydrous Na₂SO₄, filtered, and concentrated. The crude residue was purified by neutral alumina chromatography (pentane) to provide **275** (35 mg, 35%) as a colorless oil: R_f 0.65 (5:95 Et₂O-hexanes on neutral alumina); IR (neat) 2956, 2925, 2855, 1478,1252 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.09 (s, 6H), 0.33 (s, 9H), 0.92 (s, 9H), 4.70 (s, 2H), 6.26 (d, J = 3.0 Hz, 1H), 6.52 (d, J = 3.1 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -9.2 (3C), -5.1 (2C), 18.4, 25.9 (3C), 58.3, 107.1, 121.5, 158.8, 160.2.



(*S*,*E*)-5-(5-((*tert*-Butyldimethylsilyloxy)methyl)furan-2-yl)-4-methyl-2-(triethylsilyloxy)pent-4-enyl Pivalate (277)

Furyl stannane **275** (9.0 mg, 0.023 mmol) and vinyl iodide **276** (12.3 mg, 0.028 mmol) were dissolved in 0.3 mL of DMF and the mixture was degassed three times. Then Pd₂(dba)₃ (6.3 mg, 0.006 mmol) and AsPh₃ (7.0 mg, 0.023 mmol) were added to the reaction mixture under argon and the mixture was heated at 95 °C for 5 h. The mixture was diluted with Et₂O (10 mL) and water (10.0 mL), and the aqueous layer was extracted with Et₂O (3 x 10 mL). The combined organic layer was dried (anhydrous Na₂SO₄), filtered and concentrated, and the crude residue was purified by silica gel chromatography (3% Et₂O in pentane) to afford **277** (5.0 mg, 41%) as a pale yellow oil: R_f 0.29 (95:5 hexanes-Et₂O); $[\alpha]^{28}$ _D – 1.2 (*c* 0.4, CHCl₃); IR (neat) 2959, 2924,

2880, 2851,1732 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.11 (s, 6H), 0.63 (q, *J* = 7.6 Hz, 6H), 0.93 (s, 9H), 0.97 (t, *J* = 7.6 Hz, 9H), 1.24 (s, 9H), 2.03 (d, J = 0.9 Hz, 3H), 2.37 (dd, *J* = 5.7, 9.9 Hz, 2H), 4.01-4.09 (m, 3H), 4.65 (s, 2H), 6.11 (s, 1H), 6.14 (d, *J* = 3.3 Hz, 1H), 6.24 (d, *J* = 3.1 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.2 (2C), 4.9, 6.7, 18.4, 19.3, 25.8, 27.2, 29.6, 30.3, 38.7, 46.3, 58.3, 67.7, 69.0, 108.6, 108.7, 117.2, 133.6, 152.6, 152.8, 178.3; EI-HRMS calcd for C₂₈H₅₂O₅Si₂Na [M + Na] ⁺ (*m*/*z*) 547.3251, found 547.3231.



2-((1*R*,2*R*,3*R*,4*R*)-2-(*tert*-Butyldimethylsilyloxy)-3-(triisopropylsilyloxy)-4vinylcyclobutyl)ethanol (279)

To a stirred solution of **190** (44.0 mg, 0.08 mmol) in CH₂Cl₂ (1.0 mL) and pH 7 buffer (0.05 mL) at 0 °C was added DDQ (22.0 mg, 0.096 mmol). The solution was stirred at 0 °C for 1 h, after which the mixture was diluted with pH 7 buffer (50.0 μ L). The aqueous layer was separated and extracted with CH₂Cl₂ (2 x 10 mL). The combined organic extract was dried (Na₂SO₄), filtered, and concentrated in vacuo. The crude residue was purified by silica gel chromatography (6:94 ethyl acetate:hexanes) to yield **279** (22.0 mg, 64%,) as a pale yellow oil: R_f 0.43 (9:1 hexanes-ethyl acetate); $[\alpha]^{22}$ D – 39.5 (*c* 1.0, CHCl₃); IR (neat) 3330, 2943, 2863, 1478 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.08 (s, 3H), 0.09 (s, 3H), 0.90 (s, 9H), 1.06 (m, 21H), 1.59-1.68 (m, 1H),

1.86-1.95 (m, 1H), 2.10-2.19 (m, 1H), 2.65 (t, J = 8.3 Hz, 1H), 3.63-3.74 (m, 2H), 4.22 (dd, J = 6.9, 8.3 Hz, 1H), 4.31 (dd, J = 6.3, 7.5 Hz, 1H), 5.05 (dd, J = 6.3, 7.5 Hz, 1H), 5.05 (dd, J = 9.5, 10.2 Hz, 2H), 6.01 (ddd, J = 8.8, 10.2, 17.2 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -4.7, -4.6, 12.1 (3C), 18.0 (6C), 25.9 (3C), 29.7, 32.4, 38.4, 45.1, 61.3, 75.1, 75.6, 115.6, 138.1; CI-HRMS calcd for C₂₃H₄₉O₃Si₂ [M + H] ⁺ m/z428.3164, found m/z 428.3145.



2-((1*R*,2*R*,3*R*,4*R*)-2-(*tert*-Butyldimethylsilyloxy)-3-(triisopropylsilyloxy)-4vinylcyclobutyl)ethyl Pivalate (280)

To a CH₂Cl₂ solution (1.5 mL) of **279** (67 mg, 0.156 mmol) at 0 °C was added slowly pyridine (25.2 μ L, 0.312 mmol) followed by trimethylacetyl chloride (38.4 μ L, 0.312 mmol) and DMAP (1.9 mg, 0.015 mmol). The mixture was allowed to warm to room temperature and was stirred for 12 h. The reaction was quenched by addition of satd aqueous NaHCO₃ solution and the separated aqueous layer was extracted with CH₂Cl₂ (2 x 20 mL). The combined organic extract was washed with satd CuSO₄ solution, dried (Na₂SO₄), filtered, and concentrated. The crude mixture was purified by silica gel chromatography (4:96 ethyl acetate/hexanes) to afford **280** (64 mg, 80%) as a colorless oil: R_f 0.69 (9:1 hexanes/ethyl acetate); [α]²² _D – 55.8 (*c* 1.9, CHCl₃); IR (neat) 2964, 2867, 1734, 1470 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.06 (s, 3H), 0.07

(s, 3H), 0.90 (s, 9H), 1.06 (m, 21H), 1.20 (s, 9H), 1.59 (m, 1H), 1.97 (m, 1H), 2.10 (m, 1H), 2.68 (t, J = 8.7 Hz, 1H), 4.10 (q, J = 6.4 Hz, 2H), 4.20 (dd, J = 6.7, 8.5 Hz, 1H), 4.28 (dd, J = 6.2, 7.4 Hz, 1H), 5.04 (m, 2H), 6.00 (ddd, J = 8.5, 10.5, 17.4 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.0, -4.8, 11.9 (3C), 17.9 (6C), 25.8 (3C), 27.2 (3C), 28.1, 31.9, 38.7, 44.8, 63.0, 75.0, 75.6, 115.5, 137.9, 178.5; CI-HRMS calcd for C₂₈H₆₄O₄Si₂ [M] ⁺ *m*/*z* 512.3757, found *m*/*z* 512.3769.



2-((1R,2R,3R,4R)-2-(tert-Butyldimethylsilyloxy)-4-formyl-3-

(triisopropylsilyloxy)cyclobutyl)ethyl Pivalate (281)

To a solution of **280** (64 mg, 0.124 mmol) in dioxane-water (3:1, 1.24 mL) were added 2,6-lutidine (28.8 μ L, 0.248 mmol), OsO₄ (0.12 mL, 0.006 mmol, 0.05M in 2-methyl-2-propanol), and NaIO₄ (106.0 mg, 0.50 mmol). The reaction mixture was stirred at room temperature for 15 h and was diluted with water (10 mL) and CH₂Cl₂ (10 mL). The organic layer was separated and the aqueous layer was extracted with CH₂Cl₂ (3 x 20 mL). The combined organic extract was washed with aqueous CuSO₄ solution and brine, dried over Na₂SO₄ and filtered. The solvent was removed under reduced pressure and the crude mixture was purified by silica gel chromatography (5:95 ethyl acetate/hexanes) to afford **281** (38.5 mg, 60%) as a colorless oil: R_f 0.5 (9:1 hexanes-

ethyl acetate); $[\alpha]^{23}_{D} - 68.6 \ (c \ 1.15, CHCl_3)$; IR (neat) 2953, 2863, 1731(2), 1147 cm⁻¹; ¹H NMR (400 MHz, CDCl_3) δ 0.08 (s, 6H), 0.91 (s, 9H), 1.06 (s, 21H), 1.19 (s, 9H), 1.55 (m, 1H), 1.99 (m, 1H), 2.65 (m, 1H), 3.04 (d, *J* = 8.9 Hz, 1H), 4.07 (m, 2H), 4.31 (dd, *J* = 6.2, 7.7 Hz, 1H), 4.52 (dd, *J* = 6.6, 9.2 Hz, 1H), 9.87 (d, *J* = 2.9 Hz, 1H); ¹³C NMR (100 MHz, CDCl_3) δ -5.0, -4.7, 11.8 (3C), 17.9 (6C), 25.7 (3C), 27.1 (4C), 30.3, 31.9, 38.7, 52.4, 62.7, 75.1, 76.8, 178.4, 202.2; EI-HRMS calcd for C₂₇H₅₅O₅Si₂ [M + H] ⁺ (*m/z*) 515.3588, found 515.3581.



2-((1*R*,2*R*,3*R*,4*S*)-2-(*tert*-Butyldimethylsilyloxy)-4-(5-(*tert*-butyldimethylsilyloxy)-1-hydroxypent-3-ynyl)-3-(triisopropylsilyloxy)cyclobutyl)ethyl Pivalate (282) and 2-((1*R*,2*R*,3*R*,4*S*)-2-(*tert*-Butyldimethylsilyloxy)-4-(2-((*tert*butyldimethylsilyloxy)methyl)-1-hydroxybuta-2,3-dienyl)-3-(triisopropylsilyloxy)cyclobutyl)ethyl Pivalate (283)

To a solution of **264** (51.0 mg, 0.19 mmol) in 1.0 mL of DMF at room temperature under nitrogen were added $SnCl_2$ (44.3 mg, 0.23 mmol) and NaI (34.4 mg, 0.23 mmol), and the resulting mixture was stirred for 3 h while protected from light in a flask wrapped with aluminum foil. The foil was removed, the bright yellow solution

was cooled to 0 °C, and a solution of **281** (40 mg, 0.077 mmol) in 1.0 mL of DMF was added to the mixture over a 10 min period. After 30 h at 0 °C in the absence of light, the reaction mixture was diluted with Et₂O and a satd aqueous solution of NH₄Cl while being stirred. The separated aqueous layer was extracted with Et₂O (3 x 10 mL) and the combined organic extract was washed with saturated aqueous NaHCO₃ and brine, dried over Na₂SO₄ and filtered. The solvent was removed under reduced pressure after which flash chromatography of the crude residue on silica gel (3:97 ethyl acetate: hexanes) afforded **282** (32.0 mg, 60%) and **283** (10.0 mg, 18%) as colorless oils.

282: $R_f 0.44$ (9:1 hexanes-ethyl acetate); $[\alpha]^{29}_{D} - 61.0$ (*c* 1.0, CHCl₃); IR (neat) 3539, 2957, 2931, 2872, 1729, 1470, 1460, 1247, 1147 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.05 (s, 6H), 0.10 (s, 6H), 0.91 (s, 18H), 1.09 (m, 21H), 1.20 (s, 9H), 1.49 (m, 1H), 1.98 (m, 1H), 2.36 (m, 4H), 3.02 (s, 1H), 3.98 (m, 1H), 4.15 (m, 1H), 4.22-4.29 (m, 2H), 4.31 (t, J = 1.7 Hz, 2H), 4.46 (dd, J = 6.0, 8.4 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.3, -5.2, -5.0, -4.6, 11.9 (3C), 17.9 (6C), 25.8 (6C), 27.2 (3C), 28.7, 30.3 (2C), 31.7, 38.6, 44.2, 63.0 (2C), 66.4, 75.9, 76.5, 77.9, 104.4, 178.4, 205.6; CI-HRMS calcd for C₃₇H₇₄O₆Si₃ [M] ⁺ (*m/z*) 699.4872, found 699.4899.

283: $R_f 0.58$ (9:1 hexanes-ethyl acetate); $[\alpha]^{24}_{D} - 38.2$ (*c* 1.15, CHCl₃); IR (neat) 3539, 2954, 2928, 1726, 1475, 1460, 1262, 1153 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.05 (s, 6H), 0.07 (s, 6H), 0.90 (s, 18H), 1.08 (21H), 1.19 (s, 9H), 1.57 (m, 1H), 1.97 (m, 1H), 2.39 (m, 2H), 3.18 (d, *J* = 2.6 Hz, 1H). 3.97 (m, 1H), 4.12 (m, 1H), 4.16 (td, *J*

= 1.9, 13.6 Hz, 1H), 4.27 (td, J = 1.9, 13.6 Hz, 1H), 4.34 (dd, J = 5.2, 7.9 Hz, 1H), 4.48 (dd, J = 6.0, 8.9 Hz, 1H), 4.69 (m, 1H), 4.88 (qd, J = 2.4, 10.5 Hz, 1H), 4.96 (qd, J = 2.4, 10.5 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.1 (2C), -4.6 (2C), 11.9 (3C), 17.9 (6C), 23.3, 25.8 (6C), 27.2 (3C), 28.3, 30.3 (2C), 31.2, 38.7, 43.9, 51.9, 62.2, 68.1, 75.8, 76.3, 80.5, 81.4, 178.3; CI-HRMS calcd for C₃₇H₇₄O₆Si₃ [M] ⁺ (*m*/*z*) 698.4793, found 698.4820.



2-((1*R*,2*R*,3*R*,4*R*)-2-(*tert*-Butyldimethylsilyloxy)-4-(2-((*tert*butyldimethylsilyloxy)methyl)buta-2,3-dienoyl)-3-(triisopropylsilyloxy)cyclobutyl)ethyl Pivalate (284)

To a solution of **282** (24.0 mg, 0.034 mmol) in 0.4 mL of CH_2Cl_2 at 0 °C was added NaHCO₃ (11.0 mg, 0.131 mmol) and Dess-Martin periodinane (28.0 mg, 0.065 mmol). The slurry was stirred at 0 °C for 2.5 h and the mixture was diluted with CH_2Cl_2 (1.0 mL) and 10 mL of satd $Na_2S_2O_3$ solution. After stirring the biphasic mixture vigorously for 20 min, the layers were separated and the aqueous phase was extracted with CH_2Cl_2 (2 x 10 mL). The combined organic extract was washed with satd $NaHCO_3$ and brine, and dried over Na_2SO_4 . The drying agent was removed by filtration, and the solvent was removed in vacuo. Purification of the crude oil by silica

gel chromatography (98:2 hexane/ethyl acetate) provided **284** (23 mg, 96%) as a yellow oil: $R_f 0.65$ (9:1 hexanes/ethyl acetate); $[\alpha]^{25}_{D} - 24.0$ (*c* 1.2, CHCl₃); IR (neat) 2958, 2928, 2864, 1932, 1736, 1667, 1474, 1455, 1255, 1153, 1111 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.05 (s, 6H), 0.07 (s, 6H), 0.9 (s, 18H), 1.34 (s, 21H), 1.18 (s, 9H), 1.53 (m, 1H), 1.97 (m, 1H), 2.45 (m, 1H), 3.83 (dd, *J* = 1.8, 9.0 Hz, 1H), 3.95-4.11 (m, 2H), 4.28 (td, *J* = 3.7, 13.5 Hz, 1H), 4.41 (dd, *J* = 6.6, 9.7 Hz, 1H), 4.44 (td, *J* = 3.7, 13.5 Hz, 1H), 4.51 (dd, *J* = 7.2, 8.3 Hz, 1H), 5.26 (td, *J* = 3.5, 14.1 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.4, -5.3, -5.0, -4.6, 12.1 (3C), 17.9 (6C), 25.8 (6C), 27.1 (3C), 27.5, 30.3 (2C), 34.3, 38.6, 48.4, 58.4, 62.6, 74.7, 76.2, 82.1, 112.1, 178.4, 198.9, 216.2; CI-HRMS calcd for C₃₇H₇₂O₆Si₃ [M] ⁺ *m/z* 696.4637, found *m/z* 696.4653.



2-((1R,2R,3R,4R)-2-(tert-Butyldimethylsilyloxy)-4-(3-((tert-

butyldimethylsilyloxy)methyl)furan-2-yl)-3-(triisopropylsilyloxy)cyclobutyl)ethyl Pivalate (285)

To a solution of **284** (23.0 mg, 0.04 mmol) in 2.74 mL of hexane and 0.54 mL of CH_2Cl_2 at room temperature was added 28.0 mg of AgNO₃ on SiO₂ (10 wt%, 0.016 mmol) in a single portion. The suspension was stirred at room temperature in the
absence of light for 2.5 h, then was diluted with Et₂O and filtered through a short pad of Celite which was washed with Et₂O (20 mL). The solvent was removed in vacuo and the crude mixture was purified by silica gel chromatography (98/2, hexanes/ethyl acetate) to give **285** (17.0 mg, 74%) as a pale yellow oil: R_f 0.62 (9:1 hexanes/ethyl acetate); $[\alpha]^{25}_{D} - 36.3$ (*c* 0.65, CHCl₃); IR (neat) 2954, 2924, 2856, 1729, 1460, 1251 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.07 (s, 3H), 0.08 (s, 3H), 0.10 (s, 6H), 0.90 (m, 39 H), 1.17 (s, 9H), 1.67 (m, 1H), 2.04 (m, 1H), 2.47 (m, 1H), 3.30 (dd, *J* = 2.7, 9.0 Hz, 1H), 4.02 (m, 1H), 4.14 (m, 1H), 4.45 (dd, *J* = 6.7, 8.6 Hz, 1H), 4.49 (d, *J* = 2.3 Hz, 2H), 4.64 (dd, *J* = 6.5, 8.5 Hz, 1H), 6.32 (d, *J* = 1.7 Hz, 1H), 7.31 (d, *J* = 1.7 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.2, -5.1, -4.9, -4.7, 11.9 (3C), 17.7 (6C) 25.8 (3C), 25.9 (3C), 27.1 (4C), 29.7, 30.3 (2C), 36.9, 38.6 (2C), 57.3, 62.8, 75.2, 76.6, 110.6, 121.1, 140.5, 149.0, 178.3; CI-HRMS calcd for C₃₇H₇₂O₆Si₃ [M] ⁺ *m/z* 697.4715, found *m/z* 697.4744.



2-((1*R*,2*R*,3*R*,4*R*)-2-(3-((*tert*-Butyldimethylsilyloxy)methyl)furan-2-yl)-4-hydroxy-3-(triisopropylsilyloxy)cyclobutyl)ethyl Pivalate (287)

To a solution of **285** (11 mg, 0.015 mmol) in EtOH (0.2 mL) was added PPTS (1.0 mg, 0.005 mmol) and the mixture was stirred for 48 h at room temperature. The

reaction was quenched with satd aq NaHCO₃ solution and the separated aqueous layer was extracted with Et_2O (2 x 10 mL). The combined organic extract was dried (Na₂SO₄), filtered, and concentrated. The crude oil was used for the next step without further purification.

The crude residue (4.0 mg) obtained above was dissolved in 0.2 mL of THF, and imidazole (1.74 mg, 0.025 mmol) and TBSCl (3.20 mg, 0.021 mmol) were added to the solution. The mixture was stirred at room temperature for 16 h and the reaction was quenched with satd aqueous NaHCO₃ solution. The separated aqueous layer was extracted with Et₂O (2 x 10 mL) and the combined organic extract was dried over anhydrous Na₂SO₄, filtered, and concentrated. The crude residue was purified by silica gel chromatography (7:93 ethyl acetate:hexanes) to afford **283** (4.0 mg, 36%, 2 steps) as a colorless oil: $R_f 0.68$ (8:2 hexanes/ethyl acetate); $\left[\alpha\right]^{20} = -6.0$ (c 1.45, CHCl₃); IR (neat) 3466, 2933, 2871, 1734, 1711, 1454 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.08 (s, 3H), 0.09 (s, 3H), 0.95 (s, 30H), 1.20 (s, 9H), 1.72-1.81 (m, 1H), 2.06-2.14 (m, 1H), 2.16 (d, J = 4.7 Hz, 1H), 2.64 (dd, J = 3.6, 8.0 Hz, 1H), 3.34 (dd, J = 6.2, 8.4 Hz, 1H), 4.06-4.22 (m, 2H), 4.44 (dd, J = 6.2, 8.4 Hz, 1H), 4.49 (d, J = 2.0 Hz, 2H), 4.66 $(dd, J = 5.7, 9.4 Hz, 1H), 6.32 (d, J = 1.5 Hz, 1H), 7.29 (d, J = 1.4 Hz, 1H); {}^{13}C NMR$ (100 MHz, CDCl₃) δ -5.2 (2C), 11.9 (3C), 17.6 (6C), 25.9 (3C), 27.1 (3C), 28.8, 29.7, 38.6, 57.2, 63.1, 75.3, 76.0, 110.7, 121.3, 121.3, 140.6, 148.7, 178.6; CI-HRMS calcd for $C_{33}H_{57}O_6NaSi_2 [M + Na]^+ m/z$ 605.3670, found m/z 605.3654.



2-((1*R*,2*R*,3*R*)-2-(3-((*tert*-Butyldimethylsilyloxy)methyl)furan-2-yl)-4-methylene-3-(triisopropylsilyloxy)cyclobutyl)ethyl Pivalate (288)

To a solution of **287** (4.0 mg, 0.0068 mmol) in CH_2Cl_2 (0.1 mL) was added TPAP (2.9 mg, 0.0082 mmol) and the reaction mixture was stirred at room temperature for 1 h. The suspension was filtered through a short pad of silica gel which was washed with CH_2Cl_2 (2 x 5 mL). Evaporation of the filtrate gave a crude residue which was used immediately for the next reaction.

To a solution of the crude residue (3.2 mg, 0.0065 mmol) obtained above in dry THF (0.1 mL) at 0 °C was added Ph₃P=CH₂ (0.3 mL, 0.022 mmol, 0.076 M in THF, freshly prepared from Ph₃PCH₃Br (100.0 mg, 0.28 mmol) and *n*-BuLi (95.6 μ L, 0.23 mmol, 2.5 M solution in hexane) in THF (3.0 mL) at 0 °C) and the mixture was stirred at 0 °C for 10 min. The solution was warmed to room temperature and was stirred for 30 min. The reaction was quenched with satd NH₄Cl (10 mL) solution, the layers were separated, and the aqueous layer was extracted by Et₂O (2 x 5 mL). The combined organic extract was dried, filtered and concentrated, and the crude residue was purified by silica gel chromatography (3:97 Et₂O:pentane) to provide **288** (2.0 mg, 50%, 2 steps) as a colorless oil: R_f 0.8 (9:1 hexanes/ethyl acetate); $[\alpha]^{19}_{D} - 4.1$ (*c*

0.6, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 0.08 (s, 3H), 0.09 (s, 3H), 0.95 (m, 30H), 1.18 (s, 9H), 1.94 (dd, *J* = 6.9, 7.5 Hz, 2H), 3.53 (m, 1H), 3.46 (dd, *J* = 5.9, 7.8 Hz, 1H), 4.10 (t, *J* = 6.7 Hz, 2H), 4.54 (d, *J* = 2.6 Hz, 2H), 5.03 (s, 1H), 5.08 (m, 1H), 5.22 (s, 1H), 6.33 (d, *J* = 1.5 Hz, 1H), 7.30 (d, *J* = 1.4 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.2 (2C), 12.0 (3C), 17.7 (6C), 25.9 (3C), 27.1 (3C), 29.6, 33.4, 38.6, 41.6, 43.1, 57.3, 62.3, 72.7, 77.1, 107.3, 110.8, 121.4, 148.8, 154.9, 178.4; CI-HRMS calcd for C₃₂H₅₈O₅Si₂ [M] ⁺ *m*/*z* 578.3823, found *m*/*z* 578.3815.



(1*R*,2*R*,3*R*,4*R*)-2-(2-Hydroxyethyl)-4-(triisopropylsilyloxy)-3-vinylcyclobutanol (285)

To an EtOH (absolute) solution (0.3 mL) of **279** (24.0 mg, 0.056 mmol) was added *p*toluenesulfonic acid (1.0 mg, 0.0056 mmol) at room temperature and the mixture was stirred for 4 h. When TLC showed that all the starting material had been consumed, the reaction was quenched with satd aqueous NaHCO₃ solution. The aqueous layer was extracted with EtOAc (3 x 5 mL) and the combined organic extract was dried over anhydrous Na₂SO₄, filtered, and concentrated. The crude oil was purified by silica gel chromatography (6:4 hexanes:ethyl acetate) to afford **289** (9.0 mg, 53%) as a colorless oil: R_f 0.28 (6:4 hexanes-ethyl acetate); $[\alpha]^{22}_{D} - 29.0$ (*c* 0.40, CHCl₃); IR (neat) 3302, 2920, 2868, 1455, 1154 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.07 (m, 21H), 1.68-1.75 (m, 1H), 1.93-2.03 (m, 1H), 2.25-2.32 (m, 1H), 2.65 (dt, J = 2.8, 7.9 Hz, 1H), 3.14 (bs, 2OH), 3.63 (ddd, J = 7.8, 11.7, 17.5 Hz, 1H), 3.76-3.83 (m, 1H), 4.26 (t, J = 7.7 Hz, 2H), 5.00-5.08 (m, 2H), 6.00 (ddd, J = 8.6, 10.3, 17.0 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 11.9 (3C), 17.9 (6C), 32.3, 39.6, 45.7, 62.1, 74.6, 75.3, 115.3, 137.8; EI-HRMS calcd for C₁₇H₃₅O₃Si [M] ⁺ (*m*/*z*) 315.2356, found 315.2328.



(1*R*,2*R*,3*R*,4*R*)-2-(2-(Triethylsilyloxy)ethyl)-4-(triisopropylsilyloxy)-3vinylcyclobutanol (296)

To a THF (0.5 mL) solution of **289** (23.0 mg, 0.073 mmol) at 0 °C was added imidazole (10.9 mg, 0.161 mmol) and TESCI (14.6 μ L, 0.087 mmol). The reaction mixture was allowed to warm to room temperature and was stirred for 40 min, when satd aqueous NaHCO₃ solution was added. The separated aqueous layer was extracted with Et₂O (2 x 10 mL) and the combined organic extract was dried over Na₂SO₄, filtered, and concentrated in vacuo. The crude residue was purified by silica gel chromatography (12:88 hexanes/ethyl acetate) to afford **296** (16 mg, 51%) as a colorless oil: R_f 0.56 (7:3 hexanes-ethyl acetate); $[\alpha]^{20}$ _D – 23.7 (*c* 0.75, CHCl₃); IR (neat) 3431, 2871, 1458 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.65 (q, *J* = 7.9 Hz, 6H), 0.98 (t, *J* = 8.0 Hz, 9H), 1.08 (m, 21H), 1.64-1.71 (m, 1H), 1.95-2.04 (m, 1H), 2.24-2.30 (m, 1H), 2.68 (dt, *J* = 3.7, 8.4 Hz, 1H), 3.59 (dt, *J* = 3.4, 9.7 Hz, 1H), 3.75 (t, *J* = 5.2 Hz, 1H), 3.78 (d, J = 4.4 Hz, 1H), 4.18 (ddd, J = 4.0, 5.0, 7.8 Hz, 1H), 4.24 (dd, J = 5.3, 8.1 Hz, 1H), 5.02 (dd, J = 2.1, 12.3 Hz, 2H), 6.02 (ddd, J = 8.8, 10.3, 17.8 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 4.1 (3C), 6.6 (3C), 12.0 (3C), 17.9 (6C), 32.0, 40.0, 45.7, 62.2, 74.7, 75.5, 114.9, 138.3; CI-HRMS calcd for C₂₃H₄₉O₃Si₂ [M + H] ⁺ (*m*/*z*) 429.3220, found 429.3203.



(2*R*,3*R*,4*R*)-2-(2-(Triethylsilyloxy)ethyl)-4-(triisopropylsilyloxy)-3vinylcyclobutanone (297)

To a solution of **296** (16.0 mg, 0.037 mmol) in CH₂Cl₂ (0.4 mL) under argon atmosphere were added activated 4Å molecular sieves, TPAP (1.0 mg, 0.003 mmol) and NMO (8.66 mg, 0.074 mmol). The reaction mixture was stirred at room temperature for 20 min. The suspension was filtered through a short pad of silica gel which was washed with CH₂Cl₂ (5.0 mL). Evaporation of the filtrate and purification of the crude residue by chromatography on silica gel (1:9 ethyl acetate:hexanes) afforded **297** (12.0 mg, 75%) as a colorless oil: R_f 0.56 (7:3 hexanes-ethyl acetate); $[\alpha]^{20}$ D + 38.7 (*c* 0.40, CHCl₃); IR (neat) 2955, 2863, 1790, 1465 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.61 (q, *J* = 7.6 Hz, 6H), 0.97 (t, *J* = 7.8 Hz, 9H), 1.08 (m, 21H), 1.84-1.95 (m, 2H), 2.94-3.03 (m, 2H), 3.69 (t, *J* = 6.2 Hz, 2H), 5.07-5.20 (m, 3H), 5.92 (ddd, *J* = 8.6, 10.5, 17.5 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 4.3 (3C), 6.7 (3C), 11.9 (3C), 17.6 (6C), 33.8, 43.1, 57.4, 60.4, 82.6, 116.7, 135.6, 209.8; CI-HRMS calcd for C₂₃H₄₇O₃Si₂ [M + H] ⁺ (*m/z*) 427.3064, found 427.3050.



Triethyl-(2-((1R,3R,4R)-2-methylene-3-(triisopropylsilyloxy)-4-

vinylcyclobutyl)ethoxy)silane (298)

To a solution of **297** (21.0 mg, 0.049 mmol) in dry THF (0.2 mL) at 0 °C was added Ph₃P=CH₂ (2.72 mL, 0.197 mmol, 0.076 M in THF, freshly prepared from Ph₃PCH₃Br (100.0 mg, 0.28 mmol) in and *n*-BuLi (95.6 μ L, 0.23 mmol, 2.5 M solution in hexane) THF (3.0 mL) at 0 °C) and the mixture was stirred at 0 °C for 1 h. The reaction was quenched with satd NH₄Cl solution and the separated aqueous layer was extracted with Et₂O (2 x 5 mL). The combined organic extract was dried (Na₂SO₄), filtered and concentrated, and the crude oil was purified by silica gel chromatography (3:97 ethyl acetate:pentane) to afford **298** (12.0 mg, 60%) as a colorless oil: R_f 0.74 (95:5 hexanes/Et₂O); [α]²⁰ _D – 13.0 (*c* 0.6, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 0.61 (q, *J* = 7.4 Hz, 6H), 0.97 (t, *J* = 7.8 Hz, 9H), 1.07 (m, 2H), 1.78 (dd, *J* = 5.8, 7.0 Hz, 2H), 2.62 (m, 1H), 2.75 (dd, *J* = 3.9, 4.0 Hz, 1H), 3.64 (t, *J* = 6.9 Hz, 2H), 4.87-5.13 (m, 5H), 6.01 (m, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 4.4, 6.7, 12.0, 17.8, 37.5, 42.3, 49.5, 61.1, 72.2, 105.9, 115.2, 137.5, 155.6; CI-HRMS calcd for C₂₄H₄₉O₂Si₂ [M + H] ⁺ *m*/z 425.3271, found *m*/z 425.3282.



1-((1*S*,2*R*,4*R*)-3-Hydroxy-3-(hydroxymethyl)-2-(2-(triethylsilyloxy)ethyl)-4-(triisopropylsilyloxy)cyclobutyl)ethane-1,2-diol (299)

To a solution of **298** (4.0 mg, 0.0094 mmol) in a mixture of THF-H₂O (0.13 mL, 1:1, v/v) were added NMO (3.3 mg, 0.028 mmol) and OsO₄ (4.71 µL, 0.1M in 2-methyl-2propanol, 0.0004 mmol). The mixture was stirred at room temperature for 24 h and water (5 mL) was added. The separated aqueous layer was extracted with EtOAc (3 x 10 mL) and the combined organic extract was washed with satd NaHCO₃ solution. The organic phase was dried over Na₂SO₄, filtered, and concentrated in vacuo and the crude residue was purified by silica gel chromatography (35:65 ethyl acetate:hexanes) to afford **299** (4.0 mg, 87%) as a colorless oil: $R_f 0.35$ (6:4 hexanes/ethyl acetate); ¹H NMR (400 MHz, CDCl₃) δ 0.65 (q, J = 7.8 Hz, 6H), 0.98 (t, J = 7.9 Hz, 10H), 1.09-1.11 (m, 21H), 1.60-1.63 (m, 1H), 1.94-2.05 (m, 1H), 2.08-2.14 (m, 1H), 2.24 (ddd, J = 2.9, 7.5, 9.4 Hz, 1H), 2.60-2.65 (m, 1H), 3.50-3.54 (m, 2H), 3.60 (dt, J = 2.6, 10.0Hz, 1H), 3.74-3.81 (m, 3H), 4.13-4.17 (m, 1H), 4.18 (bs, OH), 4.55 (dd, J = 0.7, 9.2Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 4.0, 6.6, 12.0, 17.8, 17.9, 29.7, 30.3, 36.6, 41.7, 62.5, 64.8, 65.4, 70.4, 74.8, 76.6; CI-HRMS calcd for $C_{24}H_{53}O_6Si_2$ [M + H] $^+ m/z_1$ 493.3381, found *m/z* 493.3362.



2-((1*R*,2*R*,3*R*,4*R*)-2-Hydroxy-3-(triisopropylsilyloxy)-4-vinylcyclobutyl)ethyl Pivalate (300)

To a CH₂Cl₂ solution (0.8 mL) of **289** (27 mg, 0.085 mmol) at 0 °C were added pyridine (13.8 µL, 0.171 mmol), trimethylacetyl chloride (13.6 µL, 0.11 mmol) and DMAP (1.04 mg, 0.085 mmol). The mixture was allowed to warm to room temperature and was stirred for 18 h, after which the reaction was guenched by addition of a satd aqueous NaHCO₃ solution (10 mL). The separated aqueous layer was extracted with CH₂Cl₂ (2 x 20 mL) and the combined organic extract was washed with saturated CuSO₄ solution, dried (Na₂SO₄), filtered, and concentrated. The crude residue was purified by silica gel chromatography (7:93 ethyl acetate/hexanes) to give **300** (22 mg, 64%) as a colorless oil: $R_f 0.66$ (8:2 hexanes/ethyl acetate); $[\alpha]^{25} - 86.3$ (c 1.0, CHCl₃); IR (neat) 3470, 2941, 2859, 1731 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.07 (s, 21H), 1.21 (s, 9H), 1.63-1.73 (m, 1H), 1.96-2.06 (m, 1H), 2.08 (d, J = 2.0 Hz, OH), 2.16-2.23 (m, 1H), 2.67-2.72 (m, 1H), 4.06-4.20 (m, 2H), 4.23-4.31 (m, 2H), 5.00-5.08 (m, 2H), 5.98 (ddd, J = 8.3, 10.1, 17.2 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) § 11.9 (3C), 17.8 (6C), 27.2 (3C), 28.1, 36.8, 38.7, 44.9, 63.2, 75.1, 75.3, 115.5, 137.4, 178.7; CI-HRMS calcd for $C_{22}H_{42}O_4Si [M + H]^+ m/z$ 399.2930, found *m/z* 399.2943.



2-((1*R*,2*R*,3*R*,4*R*)-2-Acetoxy-3-(triisopropylsilyloxy)-4-vinylcyclobutyl)ethyl Pivalate (301)

To a CH₂Cl₂ solution (0.5 mL) of **300** (22 mg, 0.055 mmol) at 0 °C was slowly added pyridine (13.4 µL, 0.165 mmol), acetic anhydride (15.6 µL, 0.165 mmol) and DMAP (0.65 mg, 0.085 mmol). The mixture was allowed to warm to room temperature and stirred for 1 h. The reaction was quenched by addition of satd aqueous $NaHCO_3$ solution (5 mL), the separated aqueous layer was extracted with CH₂Cl₂ (2 x 20 mL) and the combined organic extract was washed with satd CuSO₄ solution, dried (Na₂SO₄), filtered, and concentrated. The crude mixture was purified by silica gel chromatography (5:95 ethyl acetate/hexanes) to afford 301 (21 mg, 87%) as a colorless oil: $R_f 0.51$ (9:1 hexanes/ethyl acetate); $\left[\alpha\right]^{25} = -77.5$ (c 1.0, CHCl₃), IR (neat) 2945, 2870, 1751, 1729, 1462 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 1.04 (s, 21H), 1.19 (s, 9H), 1.58-1.62 (m, 1H), 1.73-1.83 (m, 1H), 2.08 (s, 3H), 2.38 (m, 1H), 2.81 (t, J = 8.5 Hz, 1H), 4.05 (t, J = 6.6 Hz, 2H), 4.48 (dd, J = 6.8, 8.5 Hz, 1H), 5.04-5.14 (m, 3H), 6.01 (ddd, J = 8.4, 10.6, 17.1 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 11.8 (3C), 17.6 (6C), 20.6, 27.1 (3C), 28.2, 35.7, 38.7, 45.3, 62.4, 71.7, 76.1, 116.2, 136.5, 170.2, 178.5; CI-HRMS calcd for $C_{24}H_{44}O_5NaSi [M + Na]^+ m/z$ 463.2856, found *m/z* 463.2842.



2-((1*R*,2*R*,3*R*,4*R*)-2-Acetoxy-4-formyl-3-(triisopropylsilyloxy)cyclobutyl)ethyl Pivalate (302)

To a solution of **301** (22 mg, 0.049 mmol) in dioxane-water (3:1, 0.5 mL) were added 2,6-lutidine (11.35 μL, 0.248 mmol), OsO₄ (0.05M in 2-methyl-2-propanol, 50.0 μL, 0.0024 mmol), and NaIO₄ (42.0 mg, 0.196 mmol). The reaction mixture was stirred at room temperature for 2 h and was diluted with water (10 mL) and CH₂Cl₂ (10 mL). The organic layer was separated and the aqueous layer was extracted with CH₂Cl₂ (2 x 20 mL). The combined organic extract was washed with aqueous CuSO₄ solution, brine and dried over anhydrous Na₂SO₄. The solvent was removed under reduced pressure and the crude residue was purified by silica gel chromatography (1:9 ethyl acetate: hexanes) to afford 302 (15.0 mg, 68%) as a pale brown oil: $R_f 0.41$ (15:85 hexanes-ethyl acetate); $[\alpha]^{26}$ _D - 53.8 (c 1.0, CHCl₃); IR (neat) 2944, 2861, 1754, $1724(2) \text{ cm}^{-1}$; ¹H NMR (400 MHz, CDCl₃) δ 1.04 (s, 21H), 1.19 (s, 9H), 1.54-1.61 (m, 1H), 1.78-1.86 (m, 1H), 2.09 (s, 3H), 2.90-2.97 (m, 1H), 3.13 (d, J = 9.3 Hz, 1H), 3.97-4.07 (m, 2H), 4.72 (dd, J = 6.8, 9.2 Hz, 1H), 5.24 (dd, J = 6.4, 7.9 Hz, 1H), 9.88 (d, J = 2.7 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 11.7 (3C), 17.5 (6C), 20.5, 27.1 (3C), 27.2, 30.1, 38.7, 52.5, 62.1, 72.7, 75.7, 169.8, 178.4, 200.7; EI-HRMS calcd for $C_{23}H_{42}O_6NaSi [M + Na]^+ (m/z) 465.2648$, found 465.2659.





To a solution of **264** (149.0 mg, 0.568 mmol) in 3.0 mL of DMF at room temperature under argon were added $SnCl_2$ (129.0 mg, 0.68 mmol) and NaI (102.0 mg, 0.68 mmol) and the resulting mixture was stirred for 3 h while protected from light in a flask wrapped with aluminum foil. After the foil was removed, the bright yellow solution was cooled to 0 °C and a solution of **302** (84.0 mg, 0.189 mmol) in 2.0 mL of DMF was added over a 10 min period. After stirring for 30 h at 0 °C in the absence of light, the reaction mixture was diluted with Et_2O (20 mL) and satd aqueous solution of NH₄Cl (20 mL) while being vigorously stirred. The separated aqueous layer was extracted with Et_2O (3 x 20 mL) and the combined organic extract was washed with satd aqueous NaHCO₃ and brine, dried over anhydrous Na₂SO₄, and filtered. The solvent was removed under reduced pressure and flash chromatographic purification of the residual oil on silica gel (6:94 ethyl acetate:hexanes) afforded **303** (21.0 mg, 17%) and **304** (61.0 mg, 53%) as colorless oils. **303:** $R_f 0.42$ (85:15 hexanes-ethyl acetate); $[\alpha]^{20}_{D} - 49.0$ (*c* 1.60, CHCl₃); IR (neat) 3544, 2964, 2929, 2867,1746, 1731, 1458, 1376, 1232, 1143 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.11 (s, 6H), 0.91 (s, 9H), 1.07 (s, 21H), 1.20 (s, 9H), 1.47-1.57 (m, 1H), 1.75-1.84 (m, 1H), 2.08 (s, 3H), 2.33 (tdd, *J* = 1.8, 10.3, 15.9 Hz, 1H), 2.46-2.54 (m, 2H), 2.63-2.70 (m, 1H), 2.87 (d, *J* = 2.3 Hz, OH), 3.94-4.00 (m, 1H), 4.14-4.19 (m, 2H), 4.30 (t, *J* = 1.8 Hz, 2H), 4.63 (dd, *J* = 6.3, 9.7 Hz, 1H), 5.18 (dd, *J* = 6.3, 9.5 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.1 (2C), 11.7 (3C), 17.6 (6C), 20.6, 23.5, 25.8 (3C), 27.2 (3C), 28.2, 29.5, 38.7, 44.4, 51.9, 61.8, 67.8, 71.9, 76.6, 17.8, 80.7, 81.2, 169.8, 178.3 ; CI-HRMS calcd for C₃₃H₆₃O₇Si₂ [M] ⁺ (*m*/*z*) 627.4112, found 627.4120.

304: $R_f 0.50 (85:15 \text{ hexanes-ethyl acetate}); [\alpha]^{20} _{D} - 53.1 ($ *c* $1.20, CHCl₃); IR (neat) 3540, 2953, 2867, 1960, 1742, 1731, 1458, 1369, 1236, 1151 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) <math>\delta$ 0.07 (s, 3H), 0.07 (s, 3H), 0.89 (s, 9H), 1.06 (s, 21H), 1.19 (s, 9H), 1.58-1.67 (m, 1H), 1.76-1.85 (m, 1H), 2.07 (s, 3H), 2.48 (td, *J* = 2.7, 9.2 Hz, 1H), 2.68 (dd, *J* = 3.0, 8.5 Hz, 1H), 3.01 (d, *J* = 3.1 Hz, OH), 3.96 (t, *J* = 7.3 Hz, 2H), 4.18 (td, J = 2.0, 11.6 Hz, 1H), 4.31 (td, *J* = 2.0, 11.6 Hz, 1H), 4.62 (dd, *J* = 6.2, 9.4 Hz, 1H), 4.69 (pentet, *J* = 2.9 Hz, 1H), 4.91 (qq, *J* = 2.2, 10.5 Hz, 2H), 5.27 (dd, *J* = 5.9, 9.1 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.4, -5.3, 11.8 (3C), 17.6 (3C), 17.8 (3C), 20.7, 25.8 (3C), 27.1 (3C), 28.9, 29.7, 30.2, 38.6, 44.9, 62.6, 63.1, 66.2, 72.1, 76.5, 78.1, 104.2, 169.8, 178.4, 205.5 ; CI-HRMS calcd for C₃₃H₆₃O₇Si₂ [M] ⁺ (*m*/z) 627.4112, found 627.4088.



306 (S-Mosher ester)

2-((1*R*,2*R*,3*R*,4*S*)-2-Acetoxy-4-((*R*)-2-((*tert*-butyldimethylsilyloxy)methyl)-1-((*S*)-1'-methoxy-1'-phenyl-1'-triflouromethyl-acetoxy)buta-2,3-dienyl)-3-(triisopropylsilyloxy)cyclobutyl)ethyl Pivalate (304)

To a solution of **304** (7.0 mg, 0.011 mmol) in CH_2Cl_2 (0.2 mL) were sequentially added DMAP (*R*)-(-)- α -methoxy- α -(13.4)mg, 0.11 mmol) and trifluoromethylphenylacetyl chloride (14.1 mg, 10.4 µL, 0.055 mmol). After stirring for 2 h at room temperature, the solvent was evaporated and the residual oil was loaded directly on to a silica gel column and was purified by chromatography (5:95 ethyl acetate /hexanes) to provide (S)-MTPA ester **306** (9.0 mg, 95%) as a colorless oil: $R_f 0.66$ (85:15 hexanes/ethyl acetate): $[\alpha]^{21} - 87.8$ (c 0.5, CHCl₃); IR (neat) 2952, 2870, 1752, 1729, 1471 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.05 (s, 3H), 0.06 (s, 3H), 0.89 (s, 9H), 1.02 (s, 21H), 1.19 (s, 9H), 1.50–1.57 (m, 1H), 1.68-1.76 (m, 1H), 2.05 (s, 3H), 2.59-2.69 (m, 2H), 3.62 (s, 3H), 3.84-3.97 (m, 2H), 4.24 (td, J = 2.1, 11.9 Hz, 1H), 4.33 (td, J = 2.9, 12.3 Hz, 1H), 4.50 (dd, J = 6.9, 8.9 Hz, 1H), 4.86 (td, J = 3.0, 10.9 Hz, 1H), 4.95 (td, J = 2.6, 11.0 Hz, 1H), 5.14 (dd, J = 7.2, 8.4 Hz, 1H), 5.52 (s, 1H), 7.37-7.39 (m, 3H), 7.62-7.64 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ -5.4 (2C), 11.9 (3C), 17.6 (3C), 17.8 (3C), 20.6, 25.8 (3C), 27.1 (3C), 28.7, 31.0, 38.6, 44.8, 55.3, 62.5, 63.5, 70.6, 71.3, 80.1, 104.5, 127.7 (2C), 128.2 (3C), 129.3 (2C), 132.1, 165.5, 169.7, 206.3; CI-HRMS calcd for $C_{43}H_{69}O_9F_3NaSi_2$ [M + Na] ⁺ m/z 865.4330, found m/z 865.4362.



308 (R-Mosher ester)

2-((1*R*,2*R*,3*R*,4*S*)-2-Acetoxy-4-((*R*)-2-((*tert*-butyldimethylsilyloxy)methyl)-1-((*R*)-1'-methoxy-1'-phenyl-1'-triflouromethyl-acetoxy)buta-2,3-dienyl)-3-(triisopropylsilyloxy)cyclobutyl)ethyl Pivalate (304)

To a solution of **304** (20.0 mg, 0.031 mmol) in CH₂Cl₂ (0.6 mL) were sequentially added DMAP (38.9)mg, 0.11 mmol) and (S)-(+)- α -methoxy- α trifluoromethylphenylacetyl chloride (40.1 mg, 30.0 µL, 0.159 mmol). After stirring for 1 h at room temperature, the solvent was evaporated and the crude residue was loaded directly on to silica gel and was purified by chromatography (5:95 ethyl acetate/hexanes) to give (R)-MTPA ester 308 (23.0 mg, 86%) as a colorless oil: R_f 0.66 (85:15 hexanes/ethyl acetate); $\left[\alpha\right]^{21}$ D - 83.8 (c 0.9, CHCl₃); IR (neat) 2952, 2870, 1752, 1729, 1471 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.07 (s, 3H), 0.08 (s, 3H), 0.90 (s, 9H), 1.06 (s, 21H), 1.17 (s, 9H), 1.52-1.61 (m, 1H), 1.73-1.81 (m, 1H), 2.06 (s, 3H), 2.60-2.68 (m, 2H), 3.66 (s, 3H), 3.85-3.99 (m, 2H), 4.21 (dt, J = 1.6, 11.5 Hz,

1H), 4.33 (dt, J = 2.5, 11.7 Hz, 1H), 4.88 (d, J = 11.2 Hz, 1H), 4.55 (dd, J = 6.7, 8.2 Hz, 1H), 4.72 (d, J = 11.2 Hz, 1H), 5.34 (t, J = 7.5 Hz, 1H), 5.53 (s, 1H), 7.39-7.40 (m, 3H), 7.61-7.63 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ -5.3 (2C), 11.9 (3C), 17.6 (3C), 17.8 (3C), 20.6, 25.9 (3C), 27.1 (3C), 28.8, 30.4, 38.6, 45.0, 55.8, 62.5, 63.9, 70.3, 71.1, 75.4, 79.7, 103.9, 127.6 (2C), 128.3 (3C), 129.3 (2C), 132.0, 165.6, 169.8, 178.3, 205.9; CI-HRMS calcd for C₄₃H₆₉O₉F₃NaSi₂ [M + Na] ⁺ *m*/*z* 865.4330, found *m*/*z* 865.4362.



2-((1*R*,2*R*,3*R*,4*R*)-2-Acetoxy-4-(2-((*tert*-butyldimethylsilyloxy)methyl)buta-2,3dienoyl)-3-(triisopropylsilyloxy)cyclobutyl)ethyl Pivalate (309)

To a solution of **304** (61.0 mg, 0.097 mmol) in 1.0 mL of CH_2Cl_2 at 0 °C was added NaHCO₃ (32.5 mg, 0.38 mmol) followed by Dess-Martin periodinane (82.5 mg, 0.194 mmol). The slurry was stirred at 0 °C for 1 h and was diluted with CH_2Cl_2 (5.0 mL). The reaction was quenched with satd Na₂S₂O₃ solution (15 mL) by stirring the biphasic mixture vigorously for 20 min. The layers were separated, the aqueous phase was extracted with CH_2Cl_2 (2 x 10 mL), and the combined organic extract was washed with satd NaHCO₃ solution, water and brine and dried over Na₂SO₄. The drying agent was removed by filtration and the filtrate was concentrated in vacuo. Purification of the crude residue by silica gel chromatography (94:6 hexanes/ethyl acetate) provided

309 (59 mg, 97%) as a yellow oil: $R_f 0.50$ (85:15 hexanes/ethyl acetate); $[\alpha]^{18} _{D} - 14.6$ (*c* 0.98, CHCl₃); IR (neat) 2953, 2863, 1933, 1750, 1731, 1664, 1466, 1369, 1228 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.05 (s, 6H), 0.08 (s, 9H), 1.01 (s, 21H), 1.17 (s, 9H), 1.50-1.60 (m, 1H), 1.72-1.81 (m, 1H), 2.06 (s, 3H), 2.78-2.85 (m, 1H), 3.90 (dd, *J* = 3.2, 9.8 Hz, 1H), 3.97 (t, *J* = 6.5 Hz, 2H), 4.28 (td, *J* = 3.6, 13.3 Hz, 1H), 4.44 (td, *J* = 3.4, 13.3 Hz, 1H), 4.60 (dd, *J* = 6.5, 9.5 Hz, 1H), 5.24-5.29 (m, 2H), 5.36 (td, *J* = 3.4, 13.9 Hz, 1H) ; ¹³C NMR (100 MHz, CDCl₃) δ -5.4, -5.3, 12.1 (3C), 17.6 (3C), 17.7 (3C), 18.2, 20.5, 25.8 (3C), 27.1 (3C), 27.5, 31.9, 38.6, 48.8, 58.4, 62.2, 72.2, 75.9, 82.1, 112.0, 169.6, 178.3, 197.3, 216.3; CI-HRMS calcd for C₃₃H₆₁O₇Si₂ [M] ⁺ *m*/z 625.3956, found *m*/z 625.3955.



2-((1*R*,2*R*,3*R*,4*R*)-2-Acetoxy-4-(3-((*tert*-butyldimethylsilyloxy)methyl)furan-2-yl)-3-(triisopropylsilyloxy)cyclobutyl)ethyl Pivalate (310)

To a solution of **309** (54.0 mg, 0.086 mmol) in 3.58 mL of hexane and 0.70 mL of CH_2Cl_2 at room temperature was added 73.3 mg of AgNO₃ on SiO₂ (10 wt.%, 0.043 mmol) in a single portion. The suspension was stirred at room temperature for 2 h in the absence of light, then it was diluted with Et₂O and filtered through a short pad of Celite which was washed with Et₂O (20 mL). The filtrate was concentrated in vacuo,

and the crude mixture was purified by silica gel chromatography (96/4, hexanes/ethyl acetate) to furnish **310** (43.0 mg, 79%) as a pale yellow oil: $R_f 0.57$ (85:15 hexanes/ethyl acetate); $[\alpha]^{18}_{D} - 27.3$ (*c* 1.00, CHCl₃); IR (neat) 2953, 2863, 1746, 1731, 1462, 1365 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.08 (s, 6H), 0.92 (s, 30H), 1.17 (s, 9H), 1.67-1.73 (m, 1H), 1.84-1.92 (m, 1H), 2.10 (s, 3H), 2.78-2.85 (m, 1H), 3.46 (dd, *J* = 3.2, 9.1 Hz, 1H), 3.97-4.08 (m, 2H), 4.50 (d, *J* = 2.9 Hz, 2H), 4.63 (dd, *J* = 6.6, 8.3 Hz, 1H), 5.52 (dd, *J* = 6.6, 8.8 Hz, 1H), 6.31 (d, *J* = 1.7 Hz, 1H), 7.31 (d, *J* = 1.6 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.2 (2C), 11.8 (3C), 17.5 (3C), 17.6 (3C), 18.3, 20.6, 25.9 (3C), 27.1 (3C), 28.7, 35.0, 38.6, 39.0, 57.2, 62.2, 72.5, 76.2, 110.6, 121.6, 140.9, 147.8, 170.0, 178.3; CI-HRMS calcd for C₃₃H₆₁O₇Si₂ [M] ⁺ *m*/z 625.3955, found *m*/z 625.3938.



2-((1*R*,2*R*,3*R*,4*R*)-2-(3-((*tert*-Butyldimethylsilyloxy)methyl)furan-2-yl)-4-hydroxy-3-(triisopropylsilyloxy)cyclobutyl)ethyl Pivalate (287)

To a stirred solution of **310** (38.0 mg, 0.06 mmol) in dry MeOH (0.6 mL) at 0 $^{\circ}$ C was added K₂CO₃ (21.0 mg, 0.152 mmol). The mixture was stirred at 0 $^{\circ}$ C for 3 h and water (10 mL) was added to the reaction mixture. The separated aqueous layer was extracted with EtOAc (3 x 20 mL) and the combined organic extract was dried over

anhydrous Na₂SO₄, filtered, and concentrated. The crude residue was purified by silica gel chromatography (92:8 hexane: ethyl acetate) to afford **287** (29.0 mg, 82%) as a colorless oil: R_f 0.68 (8:2 hexanes/ethyl acetate); $[\alpha]^{20}_{D} - 6.0$ (*c* 1.45, CHCl₃); IR (neat) 3466, 2933, 2871, 1734, 1711, 1454 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.08 (s, 3H), 0.09 (s, 3H), 0.95 (s, 30H), 1.20 (s, 9H), 1.72-1.81 (m, 1H), 2.06-2.14 (m, 1H), 2.16 (d, *J* = 4.7 Hz, 1H), 2.64 (dd, *J* = 3.6, 8.0 Hz, 1H), 3.34 (dd, *J* = 6.2, 8.4 Hz, 1H), 4.06-4.22 (m, 2H), 4.44 (dd, *J* = 6.2, 8.4 Hz, 1H), 4.49 (d, *J* = 2.0 Hz, 2H), 4.66 (dd, *J* = 5.7, 9.4 Hz, 1H), 6.32 (d, *J* = 1.5 Hz, 1H), 7.29 (d, *J* = 1.4 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.2 (2C), 11.9 (3C), 17.6 (6C), 25.9 (3C), 27.1 (3C), 28.8, 29.7, 38.6, 57.2, 63.1, 75.3, 76.0, 110.7, 121.3, 121.3, 140.6, 148.7, 178.6; CI-HRMS calcd for C₃₃H₅₇O₆NaSi₂ [M + Na] ⁺ *m*/z 605.3670, found *m*/z 605.3654.



2-((1*R*,2*R*,3*R*)-2-(3-((*tert*-Butyldimethylsilyloxy)methyl)furan-2-yl)-4-oxo-3-(triisopropylsilyloxy)cyclobutyl)ethyl Pivalate (311)

To a CH_2Cl_2 (0.5 mL) solution of **287** (31.0 mg, 0.053 mmol) were added 4Å MS, TPAP (1.86 mg, 0.0052 mmol) and NMO (18.69 mg, 0.159 mmol). The reaction mixture was stirred at room temperature for 30 min after which the suspension was loaded directly on a silica gel column. The crude mixture was purified by chromatography (93/7 hexanes/ethyl acetate) to yield **311** (29.0 mg, 94%) as a colorless oil: $R_f 0.75$ (2:8 hexanes-ethyl acetate); $[\alpha]^{21}_{D} + 38.7$ (*c* 1.45, CHCl₃); IR (neat) 2957, 2868, 1792, 1733, 1470, 1158 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.10 (s, 6H), 0.93-0.96 (m, 30H), 1.17 (s, 9H), 1.95-2.14 (m, 2H), 3.53-3.59 (m, 1H), 3.69 (dd, *J* = 5.9, 9.1 Hz, 1H), 4.13 (t, *J* = 6.1 Hz, 2H), 4.56 (d, *J* = 7.0 Hz, 2H), 5.18 (dd, *J* = 2.9, 9.3 Hz, 1H), 6.32 (d, *J* = 1.3 Hz, 1H), 7.29 (d, *J* = 1.3 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.2 (2C), 11.8 (3C), 17.4 (3C), 17.5 (3C), 18.3, 25.9 (3C), 27.1 (3C), 29.9, 36.8, 38.7, 56.2, 57.2, 61.8, 83.2, 110.7, 122.1, 141.3, 147.2, 178.3, 208.4; CI-HRMS calcd for C₁₃H₅₆O₆Si₂ [M] ⁺ *m*/*z* 580.3615, found *m*/*z* 580.3616.





To a solution of **311** (26.0 mg, 0.044 mmol) in dry THF (1.0 mL) at 0 °C was added $Ph_3P=CH_2$ (1.50 mL, 0.179 mmol, 0.119 M in THF, freshly prepared from Ph_3PCH_3Br (200.0 mg, 0.56 mmol) and *n*-BuLi (0.17 mL, 0.47 mmol, 2.7 M solution in hexane in THF (4.0 mL) at 0 °C) and the mixture was stirred at 0 °C for 10 min and at room temperature for 45 min. The reaction was quenched with satd aqueous NH_4Cl solution

(10 mL) and the separated aqueous layer was extracted with Et_2O (2 x 10 mL). The combined organic extract was dried (Na₂SO₄), filtered and concentrated, and the crude residue was purified by silica gel chromatography (3:97 ethyl acetate:hexanes) to give **288** (12.0 mg, 48%) and **312** (3.0 mg, 15%) as colorless oils.

288: $R_f 0.8$ (9:1 hexanes/ethyl acetate); $[\alpha]^{19}_{D} - 4.1$ (*c* 0.6, CHCl₃); IR (neat) 2953, 2959, 1725, 1466 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.08 (s, 3H), 0.09 (s, 3H), 0.95 (m, 30H), 1.18 (s, 9H), 1.94 (dd, *J* = 6.9, 7.5 Hz, 2H), 3.53 (m, 1H), 3.46 (dd, *J* = 5.9, 7.8 Hz, 1H), 4.10 (t, *J* = 6.7 Hz, 2H), 4.54 (d, *J* = 2.6 Hz, 2H), 5.03 (s, 1H), 5.08 (m, 1H), 5.22 (s, 1H), 6.33 (d, *J* = 1.5 Hz, 1H), 7.30 (d, *J* = 1.4 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.2 (2C), 12.0 (3C), 17.7 (6C), 25.9 (3C), 27.1 (3C), 29.6, 33.4, 38.6, 41.6, 43.1, 57.3, 62.3, 72.7, 77.1, 107.3, 110.8, 121.4, 148.8, 154.9, 178.4; CI-HRMS calcd for C₃₂H₅₈O₅Si₂ [M] ⁺ *m/z* 578.3822, found *m/z* 578.3814.

312: $R_f 0.46$ (8:2 hexanes/ethyl acetate); $[\alpha]^{23} {}_D - 18.0$ (*c* 0.4, CHCl₃); IR (neat) 3400, 2941, 2863, 1462, 1246 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.10 (s, 3H), 0.11 (s, 3H), 0.95 (s, 30H), 1.72 (t, *J* = 4.3 Hz, OH), 1.76-1.85 (m, 1H), 1.90-1.98 (m, 1H), 3.39-3.49 (m, 2H), 3.67 (d, *J* = 4.3 Hz, 2H), 4.55 (q, *J* = 1.4, 7.1 Hz, 2H), 5.03 (s, 1H), 5.07 (dd, *J* = 1.4, 7.1 Hz, 1H), 5.20 (s, 1H), 6.32 (d, *J* = 1.4 Hz, 1H), 7.30 (d, *J* = 1.5 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.1 (2C), 11.7 (3C), 17.7 (6C), 25.9 (3C), 37.3, 42.6, 43.2, 57.2, 58.8, 61.0, 72.9, 106.8, 111.1, 121.2, 140.8, 149.4, 155.4; EI-HRMS calcd for C₂₇H₅₀O₄Si₂ [M] ⁺ *m/z* 494.3247, found *m/z* 494.3229.



(S)-5-((2E,4E)-4-((2R,3R,4R)-3-(3-((*tert*-Butyldimethylsilyloxy)methyl)furan-2yl)-2-(2-hydroxyethyl)-4-(triisopropylsilyloxy)cyclobutylidene)-2-methylbut-2enyl)-3-(phenylselanyl)dihydrofuran-2(3H)-one (329)

To a mixture of vinyl iodide **228** (6.14 mg, 0.014 mmol) and furan **312** (6.0 mg, 0.012 mmol) in 0.2 mL of DMF were added KOAc (2.35 mg, 0.024 mmol) and Pd(OAc)₂ (0.26 mg, 0.00012 mmol). The reaction mixture was heated in a sealed vessel at 100 ^oC for 6 h and, after cooling, was diluted with water (10 mL). The mixture was extracted with Et₂O (2 x 10 mL) and the combined organic extract was dried (Na₂SO₄), filtered, and concentrated. Purification of the crude residue by silica gel chromatography (3/7 ethyl acetate/hexanes) afforded **329** (2.0 mg, 34%) as a colorless oil: R_f 0.16 (3:7 hexanes-ethyl acetate); IR (neat) 2925, 2855, 1769 cm⁻¹; ¹H NMR (700 MHz, CDCl₃) δ 0.09–0.10 (m, 14H), 0.93–0.94 (m, 20H), 0.96-0.97 (m, 37H), 1.75 (s, 3H), 1.76 (s, 3H), 1.92-1.96 (m, 4H), 1.9–2.03 (m, 1H), 2.15–2.17 (m, 1H), 2.26–2.29 (m, 1H), 2.38 (dd, *J* = 5.5, 7.5 Hz, 4H), 2.47–2.50 (m, 1H), 2.76–2.81 (m, 1H), 3.34–3.37 (m, 1H), 3.57–3.60 (m, 1H), 3.66–3.70 (m, 2H), 3.73–3.77 (m, 1H), 3.98 (t, *J* = 5.9 Hz, 1H), 4.04 (t, *J* = 8.9 Hz, 1H), 4.42–4.46 (m, 1H), 4.48–4.51 (m,

1H), 4.53 (dd, J = 2.2, 4.0 Hz, 2H), 5.21 (d, J = 7.2 Hz, 1H), 5.88 (d, J = 11.3 Hz, 1H), 5.96 (d, J = 11.5 Hz, 1H), 6.33 (d, J = 1.8 Hz, 1H), 6.34 (d, J = 1.8 Hz, 1H), 6.36 (t, J = 1.6 Hz, 1H), 6.38 (t, J = 2.2 Hz, 1H), 7.31 (d, J = 1.8 Hz, 1H), 7.32 (d, J = 1.8 Hz, 1H), 7.34–7.42 (m, 8H), 7.68–7.71 (m, 5H); ¹³C NMR (700 MHz, CDCl₃) δ -5.1, 11.9, 12.0, 14.1, 16.8, 17.5, 17.7, 17.8, 18.4, 25.9, 29.7, 30.3, 31.9, 35.5, 36.4, 36.9, 37.2, 37.5, 40.3, 43.6, 45.6, 45.9, 57.3, 61.1, 71.9, 77.3, 110.8, 117.8, 121.3, 124.4, 126.7, 129.2, 129.4, 131.2, 135.8, 135.9, 140.9, 149.3, 149.5, 176.0, 176.1; EI-HRMS calcd for C₄₀H₆₅O₆SeSi₂Na [M + Na] ⁺ (*m*/*z*) 811.3476, found 811.3398.



2-((1*R*,2*R*,3*R*)-2-(3-((*tert*-Butyldimethylsilyloxy)methyl)furan-2-yl)-4-methylene-3-(triisopropylsilyloxy)cyclobutyl)acetaldehyde (330)

To a CH₂Cl₂ solution (0.5 mL) of alcohol **312** (15.0 mg, 0.03 mmol) were added 4Å MS, TPAP (1.06 mg, 0.003 mmol) and NMO (10.54 mg, 0.09 mmol). The reaction mixture was stirred at room temperature for 1 h, after which the suspension was loaded on a silica gel column and eluted (95/5 hexanes/ethyl acetate) to give aldehyde **330** (9.0 mg, 60%) as a colorless oil: R_f 0.66 (15:85 hexanes-ethyl acetate); $[\alpha]^{20}$ _D – 10.2 (*c* 0.4, CHCl₃); IR (neat) 2947, 2866, 1721, 1462 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.08 (s, 3H), 0.09 (s, 3H), 0.92 (s, 9H), 0.95 (s, 21H), 2.72 (dd, *J* = 1.8, 7.3

Hz, 2H), 3.50 (t, J = 7.0 Hz, 1H), 3.78–3.85 (m, 1H), 4.53 (d, J = 8.2 Hz, 2H), 5.06 (t, J = 2.0 Hz, 1H), 5.09 (dd, J = 2.4, 7.5 Hz, 1H), 5.23 (t, J = 2.0 Hz, 1H), 6.33 (d, J = 1.7 Hz, 1H), 7.32 (d, J = 1.7 Hz, 1H), 9.75 (t, J = 1.9 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.2, -5.1, 11.9 (3C), 17.7 (6C), 18.3, 25.9 (3C), 39.1, 42.7, 48.1, 57.2, 72.9, 108.3, 110.9, 121.7, 141.1, 148.0, 153.9, 200.7; CI-HRMS calcd for C₂₇H₄₈O₄NaSi₂ [M + Na] ⁺ m/z 515.2989, found m/z 515.2938.



(S)-3-(2-((1R,2R,3R)-2-(3-((*tert*-Butyldimethylsilyloxy)methyl)furan-2-yl)-4methylene-3-(triisopropylsilyloxy)cyclobutyl)-1-hydroxyethyl)-5-((*E*)-3-iodo-2methylallyl)furan-2(5H)-one (332)

To a solution of **228** (13.0 mg, 0.03 mmol) in dry THF (0.2 mL) at -78 °C was added dropwise LiHMDS (46.1 μ L, 0.046 mmol, 1.0 M solution in THF). The mixture was stirred at -78 °C for 40 min and a solution of **330** (9.0 mg, 0.018 mmol) in dry THF (0.2 mL) was added dropwise over 10 min. The mixture was stirred at -78 °C for 1.5 h and then at 0 °C for 15 min. The reaction was quenched with satd aqueous NH₄Cl solution (10 mL) and the layers were separated. The aqueous layer was extracted with Et₂O (3 x 10 mL) and the combined organic extract was dried over Na₂SO₄, filtered,

and concentrated to leave a crude oil (20.0 mg) which was immediately used for the next reaction.

To a solution of the crude residue (20.0 mg) obtained above in THF (0.3 mL) at 0 $^{\circ}$ C was added dropwise H₂O₂ (0.10 mL, 30% w/w in water) and the mixture was warmed to room temperature. After stirring at room temperature for 20 min, the reaction was quenched with satd aqueous NaHCO₃ solution (5.0 mL). The separated aqueous layer was extracted with Et₂O (2 x 5 mL) and the combined organic extract was dried (Na₂SO₄), filtered, and concentrated. The crude mixture was purified by silica gel chromatography (14%-16% ethyl acetate in hexanes) to afford separable diastereomers of **332** (less polar 4.0 mg, more polar 3.0 mg, total 51% based on **330**) as colorless oils:

Less polar: $R_f 0.46$ (75:25 hexanes-ethyl acetate); $[\alpha]^{23}_{D}$ +6.9 (*c* 0.26, CHCl₃); IR (neat) 3464, 2922, 2855, 1752, 1468, 1258 cm⁻¹; ¹H NMR (700 MHz, CDCl₃) δ 0.13 (s, 3H), 0.14 (s, 3H), 0.93-0.94 (m, 30H), 1.92 (d, *J* = 1.0 Hz, 3H), 1.94 (ddd, *J* = 7.5, 10.1, 14.0 Hz, 1H), 2.27 (td, *J* = 5.0, 14.0 Hz, 1H), 2.50 (dd, *J* = 7.4, 14.9 Hz, 2H), 3.07 (d, *J* = 5.8 Hz, OH), 3.46-3.50 (m, 1H), 3.53 (dd, *J* = 7.1, 7.5 Hz, 1H), 4.50 (d, *J* = 12.2 Hz, 1H), 4.52 (dd, *J* = 4.8, 11.9 Hz, 1H), 4.60 (d, *J* = 11.9 Hz, 1H), 4.94 (ddd, *J* = 1.3, 6.1, 7.6 Hz, 1H), 5.06 (dd, *J* = 2.0, 4.0 Hz, 1H), 5.08 (t, *J* = 1.9 Hz, 1H), 5.22 (t, *J* = 1.9 Hz, 1H), 6.11 (q, *J* = 1.0 Hz, 1H), 6.33 (d, *J* = 1.8 Hz, 1H), 6.95 (t, *J* = 1.4 Hz, 1H), 7.32 (d, *J* = 1.8 Hz, 1H); ¹³C NMR (700 MHz, CDCl₃) δ -5.1, 11.9, 17.7, 18.5, 24.5, 26.0, 29.7, 30.3, 39.4, 42.7, 42.8, 43.1, 57.1, 66.4, 73.2, 79.3, 79.4, 107.8, 111.4,

121.1, 125.5, 131.7, 136.4, 140.8, 141.8, 149.9, 154.3, 171.7; EI-HRMS calcd for $C_{35}H_{57}O_6IO_6Si_2Na [M + Na]^+ (m/z)$ 779.2636, found 779.2601.

More polar: $R_f 0.42$ (75:25 hexanes-ethyl acetate); $[\alpha]^{23}_{D}$ -2.0 (*c* 0.3, CHCl₃); IR (neat) 3457, 2930, 2862, 1752, 1456 cm⁻¹; ¹H NMR (700 MHz, CDCl₃) δ 0.12 (s, 3H), 0.14 (s, 3H), 0.94-0.93 (m, 30H), 1.93 (d, *J* = 1.0 Hz, 3H), 1.99-2.03 (m, 1H), 2.10 (ddd, *J* = 3.8, 10.4, 14.0 Hz, 1H), 2.54 (dd, *J* = 5.9, 10.4 Hz, 2H), 3.52 (dd, *J* = 7.0, 7.4 Hz, 1H), 3.59-3.63 (m, 1H), 4.47-4.48 (m, 1H), 4.50 (d, *J* = 12.1 Hz, 1H), 4.67 (d, *J* = 12.1 Hz, 1H), 4.99 (tt, *J* = 1.8, 5.7 Hz, 1H), 5.04-5.05 (m, 2H), 5.20 (s, 1H), 6.12 (q, *J* = 1.0 Hz, 1H), 6.33 (d, *J* = 1.8 Hz, 1H), 7.09 (t, *J* = 1.5 Hz, 1H), 7.32 (d, *J* = 1.8 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ -5.1, 12.0, 17.7, 18.5, 24.6, 25.9, 29.7, 30.3, 39.8, 42.6, 42.7, 43.0, 57.2, 65.6, 73.3, 79.3, 79.4, 107.5, 111.5, 121.1, 125.5, 135.7, 136.8, 140.8, 141.8, 147.9, 149.6, 154.2, 171.6; EI-HRMS calcd for C₃₅H₅₇O₆IO₆Si₂Na [M + Na] ⁺ (*m*/*z*) 779.2636, found 779.2601.



2-((1R,2R,3R)-2-(3-(Hydroxymethyl)furan-2-yl)-4-methylene-3-

(triisopropylsilyloxy)cyclobutyl)ethyl Pivalate (328)

To a solution of **288** (18.0 mg, 0.031 mmol) in EtOH (0.3 mL) was added TsOH.H₂O (0.59 mg, 0.0031 mmol) and the mixture was stirred for 1.5 h at room temperature.

The reaction was quenched with satd aqueous NaHCO₃ solution, the separated aqueous layer was extracted with Et₂O (2 x 10 mL) and the combined organic extract was dried (Na₂SO₄), filtered, and concentrated. The crude residue was purified by silica gel chromatography (1:9 ethyl acetate/hexanes) to afford **328** (13.0 mg, 90%) as a pale yellow oil: R_f 0.41 (15:85 hexanes-ethyl acetate); $[\alpha]^{20} _{D} - 2.5$ (*c* 0.2, CHCl₃); IR (neat) 3459, 2921, 2871, 1723, 1458 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 0.91– 1.02 (m, 21H), 1.14 (s, 9H), 1.90–1.95 (s, 1H), 2.00–2.05 (m, 1H), 2.64 (s, broad, OH), 3.23–3.27 (m, 1H), 3.58 (dd, *J* = 5.4, 8.4 Hz, 1H), 4.03–4.06 (m, 1H), 4.25–4.28 (m, 1H), 4.39 (dd, *J* = 7.7, 12.2 Hz, 1H), 4.60–4.62 (m, 1H), 5.10 (t, *J* = 2.0 Hz, 1H), 5.13–5.15 (m, 1H), 5.31 (t, *J* = 2.0 Hz, 1H), 6.40 (d, *J* = 1.8 Hz, 1H), 7.30 (d, *J* = 1.7 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 11.9 (3C), 17.5 (3C), 17.6 (3C), 27.1 (3C), 34.2, 38.7, 41.7, 43.4, 56.2, 61.9, 71.9, 108.3, 112.0, 121.5, 140.7, 150.2, 154.5, 178.6; EI-HRMS calcd for C₂₆H₄₄O₅NaSi [M + Na] ⁺ (*m/z*) 487.2856, found 487.2844.



2-((1*R*,2*R*,3*R*)-2-(3-Formylfuran-2-yl)-4-methylene-3-

(triisopropylsilyloxy)cyclobutyl)ethyl Pivalate (335)

To a solution of **328** (18.0 mg, 0.038 mmol) in CH_2Cl_2 (0.5 mL) were added 4Å MS, TPAP (1.36 mg, 0.0038 mmol) and NMO (13.35 mg, 0.114 mmol). The reaction mixture was stirred at room temperature for 1 h and the suspension was loaded on a silica gel column and eluted (92/8 hexanes/ethyl acetate) to give **335** (12.0 mg, 67%) as a colorless oil: $R_f 0.56$ (15:85 hexanes-ethyl acetate); $[\alpha]^{23}_{D} + 36.7$ (*c* 0.7, CHCl₃); IR (neat) 2938, 2864, 1736, 1678, 1465 cm⁻¹; ¹H NMR (700 MHz, CDCl₃) δ 0.93–0.99 (m, 21H), 1.17 (s, 9H), 1.96–2.05 (m, 2H), 3.40–3.44 (m, 1H), 3.94 (dd, *J* = 5.7, 7.9 Hz, 1H), 4.12 (dt, *J* = 2.6, 5.6 Hz, 2H), 5.11 (t, *J* = 1.9 Hz, 1H), 5.23 (qd, *J* = 2.1, 7.8 Hz, 1H), 5.29 (t, *J* = 1.9 Hz, 1H), 6.74 (d, *J* = 1.9 Hz, 1H), 7.38 (d, *J* = 1.9 Hz, 1H), 10.0 (s, 1H); ¹³C NMR (700 MHz, CDCl₃) δ 11.9 (3C), 17.6 (3C), 17.7 (3C), 27.1 (3C), 33.2, 38.7, 41.9, 43.7, 62.1, 72.6, 108.3, 108.6, 124.2, 142.5, 153.6, 162.7, 178.4, 185.3; EI-HRMS calcd for C₂₆H₄₃O₅Si [M + H] ⁺ *m*/*z* 463.2880, found *m*/*z* 463.2888.



2-((1R,2R,4R)-3-Methylene-2-(2-(pivaloyloxy)ethyl)-4-

(triisopropylsilyloxy)cyclobutyl)furan-3-carboxylic Acid (336)

To a cold (0 °C) solution of **335** (14.0 mg, 0.03 mmol) in a mixture of *t*BuOH (0.3 mL) and 2-methyl-2-butene (0.3 mL) was added dropwise a solution of NaClO₂ (80%, 10.1 mg, 0.09 mmol) and NaH₂PO₄ (16.5 mg, 0.12 mmol) in water (0.2 mL). The reaction mixture was stirred for 16 h at room temperature and was diluted with H₂O (10 mL) and Et₂O (10 mL). The layers were separated and the aqueous layer was

extracted with Et₂O (2 x 10 mL). The combined organic extract was washed with brine, dried (Na₂SO₄), filtered, and concentrated under reduced pressure. The crude residue was purified by silica gel chromatography (hexanes:ethyl acetate 22:78) to give **336** (13.0 mg, 91%) as a pale yellow oil: R_f 0.31 (25:75 hexanes-ethyl acetate); $[\alpha]^{24}_{D}$ +52.7 (*c* 0.65, CHCl₃); IR (neat) 2945, 2863, 1727, 1684, 1587 cm⁻¹; ¹H NMR (700 MHz, CDCl₃) δ 0.94 - 0.98 (m, 21H), 1.17 (s, 9H), 1.89–1.95 (m, 1H), 2.01–2.06 (m, 1H), 3.55–3.59 (m, 1H), 4.11 (t, *J* = 6.5 Hz, 2H), 4.26 (dd, *J* = 6.8, 7.4 Hz, 1H), 5.07 (t, *J* = 2.1 Hz, 1H), 5.19 (ddd, *J* = 1.7, 4.3, 6.2 Hz, 1H), 5.22 (t, *J* = 1.7 Hz, 1H), 6.72 (d, *J* = 1.9 Hz, 1H), 7.37 (d, *J* = 1.9 Hz, 1H); ¹³C NMR (700 MHz, CDCl₃) δ 12.0 (3C), 17.7 (6C), 27.1 (3C), 32.8, 38.7, 41.0, 43.4, 62.4, 73.3, 107.7, 110.8, 114.2, 141.4, 154.0, 161.4, 169.4, 178.5; EI-HRMS calcd for C₂₆H₄₃O₆Si [M + H] ⁺ *m*/z 479.2829, found *m*/z 479.2845.



Methyl 2-((1*R*,2*R*,4*R*)-3-Methylene-2-(2-(pivaloyloxy)ethyl)-4-(triisopropylsilyloxy)cyclobutyl)furan-3-carboxylate (337)

To a solution of **336** (5.0 mg, 0.01 mmol) in a mixture of toluene (0.1 mL) and MeOH (0.1 mL) at room temperature was added dropwise trimethylsilyldiazomethane (16.0 μ L, 0.03 mmol, 2.0 M in hexane). The mixture was stirred for 30 min at room

temperature and the solvents were removed under reduced pressure. The residual oil was purified by silica gel chromatography (7/93 ethyl acetate/hexane) to yield **337** (5.0 mg, 98%) as a colorless oil: $R_f 0.69$ (25:75 hexanes-ethyl acetate); $[\alpha]^{25}_D$ +59.2 (*c* 0.25, CHCl₃); IR (neat) 2941, 2871, 1727 (2), 1594, 1462 cm⁻¹; ¹H NMR (700 MHz, CDCl₃) δ 0.92-0.96 (m, 21H), 1.16 (s, 9H), 1.88–1.94 (m, 1H), 2.00–2.05 (m, 1H), 3.51–3.56 (m, 1H), 3.82 (s, 3H), 4.08 (t, *J* = 6.7 Hz, 2H), 4.23 (dd, *J* = 6.6, 7.5 Hz, 1H), 5.06 (t, *J* = 1.9 Hz, 1H), 5.17 (ddd, *J* = 1.9 Hz, 1H), 5.21 (dd, *J* = 1.9 Hz, 1H), 6.66 (d, *J* = 1.9 Hz, 1H), 7.33 (d, *J* = 1.9 Hz, 1H); ¹³C NMR (700 MHz, CDCl₃) δ 12.0 (3C), 17.6 (3C), 17.7 (3C), 27.1 (3C), 32.9, 38.6, 41.0, 43.2, 51.3, 62.3, 73.2, 107.6, 110.5, 114.7, 141.2, 154.2, 159.9, 164.4, 178.5; EI-HRMS calcd for C₂₇H₄₅O₆Si [M + H] ⁺ *m/z* 493.2985, found *m/z* 493.2995.



(*S*)-5-((*E*)-3-(4-((*tert*-Butyldimethylsilyloxy)methyl)-5-((1*R*,2*R*,4*R*)-2-(2hydroxyethyl)-3-methylene-4-(triisopropylsilyloxy)cyclobutyl)furan-2-yl)-2methylallyl)-3-(phenylselanyl)dihydrofuran-2(3H)-one (320)

To a solution of **312** (9.0 mg, 0.018 mmol) in dry THF (0.2 mL) at -78 °C were added dropwise freshly distilled TMEDA (16.0 μ L, 0.108 mmol) and *s*-BuLi (50.0 μ L, 0.054 mmol, 1.08 M solution in cyclohexane). The mixture was stirred at -78 °C for 15 min

and then at 0 °C for 30 min. The solution was cooled to -78 °C and Me₃SnCl (90.0 μ L, 0.09 mmol, 1.0 M solution in hexane) was added. The reaction mixture was stirred at -78 °C for 1 h and then warmed to room temperature. After stirring at room temperature for 1 h, the reaction was quenched with water (10 mL) and the layers were separated. The aqueous layer was extracted with Et₂O (3 x 10 mL) and the combined organic extract was dried over Na₂SO₄, filtered, and concentrated to leave a crude oil which was immediately used for the next reaction.

To a solution of the crude oil (13.0 mg, 0.019 mmol) obtained above and 228 (9.0 mg, 0.021 mmol) in degassed DMF (0.3 mL) under an argon atmosphere were added Pd(PPh₃)₄ (1.09 mg, 0.00095 mmol) and CuI (0.36 mg, 0.0019 mg). After stirring at room temperature for 4 h, the mixture was diluted with water (10 mL) and Et₂O (10 mL). The separated aqueous layer was extracted with Et₂O (3 x 10 mL) and the combined organic extract was dried over Na₂SO₄, filtered, and concentrated. The crude residue was purified by silica gel chromatography (3:22:75 NEt₃:ethyl acetate:hexanes) under argon to afford 320 (8.0 mg, 54% based on 312) as a colorless oil: $R_f 0.28$ (7:3 hexanes-ethyl acetate); $[\alpha]^{22} - 10.2$ (c 0.46, CHCl₃); IR (neat) 3478, 2929, 2855, 1769, 1462, 1170 cm⁻¹; ¹H NMR (700 MHz, CDCl₃) δ 0.10-0.12 (m, 19H), 0.92-0.95 (m, 76H), 1.80-1.85 (m, 3H), 1.92-1.94 (m, 2H), 1.96 (s, 6H), 2.00-2.04 (m, 2H), 2.14-2.18 (dd, J = 7.1, 13.6 Hz, 1H), 2.31-2.35 (m, 2H), 2.38-2.42 (m, 1H), 2.54 (dd, J = 5.9, 13.9 Hz, 1H), 2.60 (dd, J = 6.4, 14.4 Hz, 1H), 2.72 (ddd, J =6.6, 9.3, 13.5 Hz, 1H), 3.37-3.41 (m, 2H), 3.45 (dd, J = 5.9, 7.1 Hz, 2H), 3.66-3.72 (m, 4H), 3.96 (dd, J = 2.5, 8.3 Hz, 1H), 4.04 (t, J = 9.5 Hz, 1H), 4.42-4.46 (m, 1H), 4.49

(d, J = 12.0 Hz, 2H), 4.59 (d, J = 12.0 Hz, 2H), 4.60 (dd, J = 6.5, 7.3 Hz, 1H), 5.00 (s, 2H), 5.08 (d, J = 7.4 Hz, 2H), 5.17 (s, 2H), 5.92 (s, 1H), 5.98 (s, 1H), 6.13-6.15 (m, 2H), 7.34-7.37 (m, 4H), 7.38-7.42 (m, 2H), 7.62-7.64 (m, 2H), 7.68-7.71 (m, 4H) ; ¹³C NMR (700 MHz, CDCl₃) δ -5.1, 11.9, 17.7, 18.4, 26.0, 26.4, 29.7, 30.3, 35.2, 36.2, 36.9, 37.3, 42.5, 43.2, 45.9, 46.4, 57.2, 61.1, 73.1, 77.6, 77.9, 106.6, 110.4, 117.4, 117.5, 122.6, 126.7, 127.0, 127.7, 128.9, 129.2, 129.4, 130.1, 131.4, 135.7, 135.9, 148.5, 151.0, 155.2, 175.7, 175.8; EI-HRMS calcd for C₄₁H₆₄O₆SeSi₂Na [M + Na] ⁺ (*m*/*z*) 811.3304, found 811.3210.



2-((1*R*,2*R*,3*R*)-2-(3-((*tert*-butyldimethylsilyloxy)methyl)-5-((*E*)-2-methyl-3-((*S*)-5oxo-4-(phenylselanyl)tetrahydrofuran-2-yl)prop-1-enyl)furan-2-yl)-4-methylene-3-(triisopropylsilyloxy)cyclobutyl)acetaldehyde (321)

To a solution of **320** (7.0 mg, 0.0076 mmol) in 0.15 mL of CH_2Cl_2 at 0 °C was added NaHCO₃ (2.0 mg, 0.024 mmol) followed by Dess-Martin periodinane (4.1 mg, 0.0097 mmol). The slurry was stirred at 0 °C for 1.5 h and was diluted with CH_2Cl_2 (5.0 mL). The reaction was quenched with satd Na₂S₂O₃ solution (5.0 mL) by stirring the biphasic mixture vigorously for 20 min. The layers were separated, the aqueous phase was extracted with CH_2Cl_2 (2 x 5.0 mL), and the combined organic extract was

washed with satd NaHCO₃ solution, water and brine and was dried over Na₂SO₄. The drying agent was removed by filtration and the filtrate was concentrated in vacuo. Purification of the crude residue by silica gel chromatography (15:85 hexanes/ethyl acetate) provided **321** (2.5 mg, 42%) as a colorless oil: R_f 0.58 (75:25 hexanes/ethvl acetate); ¹H NMR (700 MHz, CDCl₃) & 0.93 (s, 3H), 0.95 (s, 3H), 0.10 (s, 6H), 1.90-1.95 (m, 54H), 1.97 (s, 4H), 2.02-2.05 (m, 2H), 2.16-2.20 (m, 0.5H), 2.30-2.35 (m, 2H), 2.37-2.43 (m, 1.5H), 2.54 (dd, J = 6.1, 14.0 Hz, 0.5H), 2.72 (dd, J = 6.4, 13.8Hz, 1H), 2.75 (dd, *J* = 1.8, 7.3 Hz, 2H), 3.48 (dd, *J* = 6.2, 7.3 Hz, 1H), 3.78-3.81 (m, 1H), 3.96 (dd, J = 2.5, 8.3 Hz, 1H), 4.04 (t, J = 9.4Hz, 0.5H), 4.44 (dd, J = 6.5, 8.7 Hz, 1H),4.48-4.53 (m, 3H), 4.58-4.62 (m, 0.5H), 5.04 (s, 1H), 5.09-5.10 (m, 1H), 5.18-5.19 (m, 0.5H), 5.21 (s, 1H), 5.90 (s, 0.3H), 5.99 (s, 0.7H), 6.15 (s, 1H), 7.34-7.43 (m, 5H), 7.69-7.71 (m, 3H), 9.77 (t, J = 1.6, 3.2 Hz, 1H); ¹³C NMR (700 MHz, CDCl₃) δ -5.1, 11.9, 17.7, 18.4, 25.9, 29.7, 30.3, 31.9, 36.2, 36.9, 37.3, 42.6, 45.9, 46.3, 48.1, 57.2, 73.1, 77.6, 77.9, 108.1, 110.2, 117.4, 117.5, 123.2, 125.5, 126.6, 127.0, 128.9, 129.2, 129.4, 130.3, 135.7, 135.9, 147.0, 151.7, 175.7, 200.8.



2-((1*R*,2*R*,3*R*)-2-(3-((*tert*-Butyldimethylsilyloxy)methyl)-5-((*E*)-2-methyl-3-((S)-5oxo-2,5-dihydrofuran-2-yl)prop-1-enyl)furan-2-yl)-4-methylene-3-(triisopropylsilyloxy)cyclobutyl)acetaldehyde (338)

To a solution of **320** (6.0 mg, 0.0076 mmol) in 0.15 mL of CH₂Cl₂ at 0 °C was added NaHCO₃ (2.0 mg, 0.0152 mmol) followed by Dess-Martin periodinane (4.6 mg, 0.01 mmol). The slurry was stirred at 0 $^{\circ}$ C for 1.5 h and was diluted with CH₂Cl₂ (5.0 mL). The reaction was quenched with satd $Na_2S_2O_3$ solution (5.0 mL) by stirring the biphasic mixture vigorously for 20 min. The layers were separated, the aqueous phase was extracted with CH₂Cl₂ (2 x 5.0 mL), and the combined organic extract was washed with satd NaHCO₃ solution, water and brine and was dried over Na₂SO₄. The drying agent was removed by filtration and the filtrate was concentrated in vacuo. Purification of the crude residue by silica gel chromatography (72:25:3 hexanes/ethyl acetate/NEt₃) provided **338** (2.0 mg, 43%) as a colorless oil: R_f 0.27 (7:3 hexanes/ethyl acetate); $\left[\alpha\right]^{25}$ D – 4.6 (c 0.35, CHCl₃); IR (neat) 3478, 2929, 2855, 1746, 1721, 1271 cm^{-1} ; ¹H NMR (700 MHz, CDCl₃) δ 0.90 (s, 3H), 0.10 (s, 3H), 0.93-0.95 (m, 30H), 2.09 (d, J = 0.7 Hz, 3H), 2.45 (dd, J = 7.3, 13.7 Hz, 1H), 2.70 (dd, J = 6.6, 13.8 Hz, 1H), 2.75 (dd, J = 1.5, 7.0 Hz, 2H), 3.49 (dd, J = 3.4, 7.6 Hz, 1H), 3.78-3.82 (m, 1H), 4.51 (d, J = 9.6 Hz, 2H), 5.04 (t, J = 2.0 Hz, 1H), 5.10 (qd, J = 2.0, 7.6 Hz, 1H), 5.19-5.22 (m, 2H), 6.07 (q, J = 0.7 Hz, 1H), 6.16 (dd, J = 1.9, 5.7 Hz, 1H), 6.18 (s, 1H), 7.18 (s, 1H), 7.50 (dd, J = 1.4, 5.7 Hz, 1H), 9.77 (t, J = 1.8 Hz, 1H); ¹³C NMR (700) MHz, CDCl₃) δ -5.1, 11.9, 14.1, 17.7, 18.4, 19.1, 22.7, 25.9, 29.4, 39.0, 42.6, 44.5, 48.1, 57.2, 73.1, 82.2, 108.1, 110.5, 117.9, 121.7, 123.2, 129.5, 147.3, 151.1, 156.1, 172.9, 200.8; CI-HRMS calcd for $C_{35}H_{56}O_6Si_2Na [M + Na]^+ m/z$ 651.3513, found m/z651.3529.

CHAPTER 6

Conclusion

The study described in this dissertation outlines a conceptually novel approach to (+)-providencin. The key features the approach are (1) extension of Taguchi's zirconium-mediated ring contraction of a furanose to yield an enantiopure cyclobutane, (2) stereoselective allenol formation by reaction of an aldehyde with a propargyllic bromide using an extension of Harada's tin(II) chloride-mediated allenol construction, and (3) application of Marshall's allenone-to-furan isomerization to complete the cyclobutylfuran subunit **288** of providencin. Synthesis of the iodolactone subunit **228** employed Negishi carbometallation-iodination of an alkyne and nucleophilic substitution of a tosylate with the dianion of phenylselenyl acetic acid followed by acid-catalyzed lactone formation as key steps. Although attempts to couple our two advanced fragments using palladium-catalyzed C-H activation of a furan were unsuccessful, we were able to link two major subunits of **1** at C12-C13 using an aldol reaction to give **332** and at C6-C7 using a Stille cross-coupling to afford **320**.

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APPENDICES



















































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